University of Rajshahi	Rajshahi-6205	Bangladesh.
RUCL Institutional Repository		http://rulrepository.ru.ac.bd
Department of Biochemistry and Molecular	Biology	PhD Thesis

2008

Studies on the Physicochemical Properties and Shelf Life of Postharvest Mangifera indica during Storage Environments

Islam, Md.Khairul

University of Rajshahi

http://rulrepository.ru.ac.bd/handle/123456789/724 Copyright to the University of Rajshahi. All rights reserved. Downloaded from RUCL Institutional Repository. Studies on the Physicochemical Properties and Shelf Life of Postharvest *Mangifera indica* during Storage Environments



THESIS SUBMITTED TO THE UNIVERSITY OF RAJSHAHI IN FULFILMENT OF THE REQURIRMENTS FOR THE DEGREE OF DOCTOR OF PHILOSOPHY

IN

BIOCHEMISTRY AND MOLECULAR BIOLOGY

BY

Md. Khairul Islam B. Sc. Ag. (Hons). M. S. Hort. M.Phil.

May 28, 2008

1

Protein and Enzyme Laboratory Dept. of Biochemistry and Molecular Biology Unlversity of Rajshahi Rajshahi, Bangladesh Studies on the Physicochemical Properties and Shelf Life of Postharvest Mangifera indica during Storage Environments



THESIS SUBMITTED TO THE UNIVERSITY OF RAJSHAHI IN FULFILMENT OF THE REQURIRMENTS FOR THE DEGREE OF

DOCTOR OF PHILOSOPHY

IN

BIOCHEMISTRY AND MOLECULAR BIOLOGY

BY

Md. Khairul Islam B. Sc. Ag. (Hons). M. S. Hort. M.Phil.

May 28, 2008

Protein and Enzyme Laboratory Dept. of Biochemistry and Molecular Biology University of Rajshahi Rajshahi, Bangladesh

DEDICATED TO THE DEPARTED SOUL OF MY BELOVED PARENTS





I do hereby declare that the contents of this thesis entitled"Studies on the Physicochemical Properties and Shelf Life of Postharvest Mangifera indica during Storage Environments" are comprised of my bonafide research works and have at no time, been submitted for any other degree, diploma or other similar title of any university.

Date: May 28, 2008 Rajshahi, Bangladesh

(Md. Khatrul Islam)

Assistant Professor Dept. of Crop Science and Technology University of Rajshahi Rajshahi 6205



This is to certify that Md. Khairul Islam, B. Sc. Ag. (Hons), M. S. Kort., M. Phil., Assistant Professor, Department Crop Science and Technology. University of Rajshahi is the sole author of the dissertation entitled "Studies on the Physicochemical Properties and Shelf Life of Postharvest Mangifra indica during Storage Environments". We have gone through this thesis, and that, in our opinion, it is satisfactory in scope and quality as a thesis for the degree of Doctor of Philosophy in the Department of Biochemistry and Molecular Biology of the University of Rajshahi, Bangladesh.

Supervisor

Date: May 28, 2008 University of Rajshahi Rajshahi, Bangladesh (Dr. K. K. M. Rafiul Islam) Professor Department of Botany University of Rajshahi Rajshahi 6205 Co-supervisor

Nural Alsai

(Dr. Nurul Absar) Professor Dept. of Biochemistry and Molecular Biology University of Rajshahi Rajshahi 6205

ACKNOWLEDGEMENTS

All the praises, gratitude and thanks are due to Almighty Allah who created us to explore the hidden facts of the nature for the benefit of mankind and for enable me to finish the work in time.

The author takes the opportunity to express his heart-felt respect, deep sense of gratitude, immense indebtedness and profound appreciation to his reverend teacher and research supervisors Professor Dr. A.K.M. Rafiul Islam, Department of Botany and co-supervisor Professor Dr. Nurul Absar, Department Biochemistry and Molecular Biology, University of Rajshahi for their scholastic guidance, affectionate feelings, constructive criticism, encouragement and valuable suggestions during the tenure period of investigation and preparing the thesis.

The author extends his grateful thanks to the authorities of Rajshahi University and Department of Biochemistry and Molecular Biology to provide him with good academic environment and sound staying for research work. He feels proud to express his respect and cordial thanks to the honorable teachers of the said department for their sincere co-operation and valuable suggestions in conducting different experiments and preparing the thesis as well.

The author is very grateful to Chairman & Professor Dr. Narayan Roy, exchairmann Professor Dr. Md. Resaul Karim-2, Acting Chairman & Professor Dr. Md. Shahidul Haque, Department of Biochemistry and Molecular Biology and M. Phil supervisor Professor Dr. M. Firoz Alam, Dept. of Botany University of Rajshahi for their constant co-operation, inspiration and supports for conducting the research work during the tenure period of investigation.

The author would like to express his special thanks to Professor Dr. Md. Habibur Rahman, Professor Dr. Md. Matiar Rahman, Professor Dr. Tanzima Yeasmin, and Associate Professor Dr. Md. Rezaul Karim-3, Associate Professor Dr. Khaled Hossain, Assistant Professor Md. Masudul Hasan Khan, Department of Biochemistry and Molecular Biology, University of Rajshahi for their direct help in conducting research, constant co-operation, inspiration and important suggestions for conducting research and preparing the thesis.

Acknowledgements are also due to the authorities viz., Bangladesh Agricultural University (BAU), Mymensingh; Soil Resource Development Institute (SRDI), Shampur, Rajshahi; Bangladesh Council for Scientific and Industrial Laboratory (BSCIR), Shampur, Rajshahi, Bangladesh Rice Research Institute (BRRI), Shampur, Rajshahi; Bangladesh Agricultural Research Institute (BARI), Shampur, Rajshahi and Food and Agricultural Organization (FAO), Dhaka, Bangladesh for their support in providing various information's for completion of this thesis.

The author is also grateful to Professor Dr. M. A. Rahim, Department of Horticulture, BAU, Mymensingh, Associate Professor & Chairman Dr. Md. Kawser Ali, Associate Professor Dr. Md. Nurul Alam, Associate Professor Dr. Md. Ali Asgar, Department of Crop Science and Technology, University of Rajshahi, Associate Professor Dr. Abdul Awal, Department of Crop Botany, BAU, Mymensingh Dr. Md. Ibrahim, Senior Scientific Officer, BCSIR), Rajshahi for their direct help and good suggestion in conducting research of the tenure period of the investigation.

Deepest and most sincere gratitude is due to his beloved late parents who all times inspired him for advanced degree. Special thanks are due to his wife, Shabiha Yeasmin, Ph.D. fellow, Department of Geology and Mining, University of Rajshahi, for her cordial feelings, constant co-operation, all time inspiration and composing the thesis paper for the successful accomplishment of the research work.

The author would like to express his sincere thanks to his father in law Mr. Mokhtar Hossain Khan, Engineer (Rtd.), Public Works Department, Bangladesh for his good suggestion and all times inspiration for completing the investigation. Finally, the author is deeply indebted to his kid: Samin Yasar Arefin for his blessing and sacrifice during the tenure period of investigation.

The Author

Abstract

An investigation was conducted for obtaining information on the reduction of postharvest losses, pattern of physical, biochemical and mineral content changes, storability as well as shelf life of postharvest mango. The trials included four different experiments involving two factors where variety was a factor comprising of Langra and Khirshapat. These varieties were treated with different postharvest treatments *viz.*, control, paraffin coating, perforated polyethylene cover, unperforated polyethylene cover, hot water $(55\pm1)^{0}$ C, low temperature in refrigerator $(4\pm1)^{0}$ C (experiment 1); different doses of Maleic hydrazide solution namely, control, 200, 400 and 600 ppm (experiment 2); different doses of GA3 solution such as, control, 100, 200 and 400 ppm (experiment 3), and different doses of Bavistin DF solution *viz.*, control, 250, 500 and 750 ppm (experiment 4).These experiments were laid out in Randomized Complete Block Design with three replicates. Data obtained from various biochemical analyses in terms of physicochemical properties and shelf life of postharvest mango, were recorded and statistically analyzed for comparison among the mean values using DMRT and LSD.

The results of the experiments exhibited that only the single effect of varieties was found to be significant in most of the parameters studied. Variety Langra performed better in accumulating higher quantity of edible portion, pulp to peel ratio, dry matter, ash, vitamin c, tltratable acidity, TSS (after 3rd day in case of experiment 2 and 4 as well as 6th day in case of experiment 3), crude fiber, lipid, water soluble protein, phosphorus, potassium, magnesium, iron and manganese content in all four experiments over Khirshapat. On the other hand, Khirshapat showed better performance in achieving higher quantity of moisture, progressively lost physiological weight, increased pulp pH, TSS (in terms of experiment 1, initial to 3rd day for experiment 2 and 4 as well as initial to 6th day for experiment 3), produced more quantity of sugar (total, reducing and non reducing), calcium, copper and zinc content as well as extended shelf life and delayed skin color changes than Langra at all the storage duration.

Different postharvest treatments subjected to the investigation demonstrated significant variation in most of the physicochemical properties and shelf life of mango at different days after storage. The results explored that some physicochemical properties *viz.*, edible portion, pulp to peel ratio, physiological weight loss, moisture content, pulp pH, TSS, sugar (total, reducing and non reducing), lipid, water soluble protein, phosphorus, potassium, calcium, magnesium, iron and manganese content were rapidly increased as well as skin color, dry matter, ash, vitamin C, titratable acidity, crude fiber, copper and zinc content along with shelf life drastically decreased from untreated mangoes, but, low temperature in refrigerator caused delaying of these changes except physiological weight loss. The unperforated polyethylene cover was found to be the best method for reduction of physiological weight loss. Low temperature in refrigerator was the best inhibitor of ethylene synthesis that delayed ripening and prolongation of shelf life. In case of other experiments, Mallc hydrazlde, GA3, and Bavlstln DF at the doses of 600, 400 and 750 ppm, respectively showed better results in delaying the changes in physicochemical properties and extended shelf life.

The results of the interaction effect of varieties and different post harvest treatments in different experiments were found to be significant in terms of most of the physicochemical properties and shelf life. The combination of Langra and control treatment progressively augmented in edible portion, pulp to peel ratio, lipid, water soluble protein, phosphorus, potassium, magnesium and iron content up to a point of ongoing metabolic cycle and thereafter, these compositions decreased. On the other hand, Khirshapat along with low temperature in refrigerator $(4\pm1)^{0}$ C, Malic hydrazide at 600 ppm, GA3 at 400 ppm and Bavistin DF at 750 ppm acted better in retardation of rapid augmentation of these compositions resulting in prolongation of shelf life in different experiments. Khirshapat using unperforated polyethylene cover also drastically reduced physiological weight loss.

Langra using low temperature in refrigerator $(4\pm1)^{0}$ C, 600 ppm of Maleic hydrazide, 400 ppm of GA3 and 750 ppm of Bavistin DF were found to be excellent in lower diminishing tendency in terms of dry matter, ash, vitamin C, titratable acidity as well as crude fiber followed by other treatment combination in different experiments studied. Khirshapat using no treatment extensively lost physiological weight, absorbed more moisture, increased TSS (up to 6th day then it decreased), accumulated higher sugar. (total, reducing and non reducing) and calcium content up to on going metabolic activities but, using low temperature in refrigerator $(4\pm1)^{0}$ C, 600 ppm of Maleic hydrazide, 400 ppm of GA3 and 750 ppm of Bavistin DF solution strongly interrupted these activities. Khirshapat along with low temperature in refrigerator, 600 ppm of Maleic hydrazide, 400 ppm of GA3 and 750 ppm of Bavistin DF solution strongly interrupted these activities. Khirshapat along with low temperature in refrigerator, 600 ppm of S3.00, 17.00, 18.00 and 17.33 days after storage, respectively. Therefore, low temperature in refrigerator $(4\pm1)^{0}$ C was the best method for preservation and delay ripening of postharvest mango and the second suitable method in another experiment was 400 ppm of GA3 which might be easily adopted by common farmers for mango preservation.

2

36

CONTENTS

3

R.

		Page no.
Ack	nowledgements	i-ii
Abs	tract	iii-iv
Con	tents	v-xi
List	of figures	xii-xix
List	of tables	xx-xxv
List	of plates	xxvi
List	of appendices	xxvii-xxx
СНА	PTER 1: INTRODUCTION	1-6
1.1	Orientation and origin	1
1.2	Production status	1
1.3	Uses and importance	1
1.4	Nutritional defiency	2
1.5	Quantity of postharvest losses	3
1.6	Causes of postharvest losses	4
1.7	Postharvest technology	5
1.8	Objectives of the study	6
СНА	PTER 2: REVIEW OF LITERATURES	7 - 33
2.1	Effect of paraffin coating	7
2.2	Effect of polyethylene cover	10
2.3	Effect of hot water treatment	12
2.4	Effect of low temperature in refrigerator	17
2.5	Effect of Maleic hydrazide (MH)	19
2.6	Effect of Gibberellic acid (GA3)	21
2.7	Effect of Bavistin DF (BDF)	25
2.8	Changes in skin color	26
2.9	Physical changes during storage	27
	2.9.1 Edible portion	27
	2.9.2 Pulp to peel ratio	27
	2.9.3 Physiological weight loss	27
	2.9.4 Moisture content	27
	2.9.5 Dry matter content	27
	2.9.6 Ash content	28
2.10	Biochemical changes during storage	28
	2.10.1 Vitamin C content	28

	2.10.2	Titratable acidity	28
	2.10.3	Pulp pH	29
	2.10.4	Total soluble solids (Brix %)	29
	2.10.5	Total sugar content	30
	2.10.6	Reducing sugar content	31
	2.10.7	Non-reducing sugar content	31
	2.10.8	Crude fibre content	32
	2.10.9	Total lipid content	32
	2.10.10	Water soluble protein content	33
2.11	Mineral	changes during storage	33
2.12	Shelf life	e	33
CHAF	TER 3: M	IATERIALS AND METHODS	34 - 62
3.1	Location	and time	34
3.2	Weather	data	34
3.3	Experime	ental materials	34
4.0	Experime	ents and experimental design	36
4.1	Experime	ent 1: Influence of different storage environments or	ı
	physicoc	hemical changes and shelf life of postharvest mango	36
	4.1.1	Experiment design	36
	4.1.2	Treatments	36
	4.1.3	Experimental layout	37
	4.1.4	Data collection	38
	4.1.5	Methodology	38
	4.1.6	Application of storage treatments	38
	4.1.7	Preparation of postharvest storage treatments	39
		4.1.7.1 Control (T ₀)	39
		4.1.7.2 Paraffin coating (T ₁)	39
		4.1.7.3 Perforated polyethylene covers (T_2)	39
		4.1.7.4 Unperforated polyethylene covers (T ₃)	39
		4.1.7.5 Hot water treatments (T_4)	39
		4.1.7.6 Low temperature in refrigerator (T_5)	39
4.2	Experime	ent 2: Influence of different doses of Maleic hydrazide on	
	physicoch	nemical changes and shelf life of mango during storage	39
	4.2.1	Experiment design	41
	4.2.2	Treatments	41
	4.2.3	Experimental layout	41
	4.2.4	Data collection	41
	4.2.5	Methodology	41
	4.2.6	Application of storage treatments	41
	4.2.7	Preparation of Maleic hydrazide (MH) solution	41

ð

X

4	1.3	Experime	ent 3:	Effect	of	different	doses	of	Gibberelli	c acid	on	
		physicoc	hemical b	ehavior	and	shelf life o	f postha	rvest	: mango			42
		4.3.1	Experim	ent desi	gn							42
		4.3.2	Treatme	nts								42
		4.3.3	Experim	ental lay	out	:						42
		4.3.4	Data col	lection								43
		4.3.5	Methodo	logy								43
		4.3.6	Applicat	ion of st	orag	ge treatme	nts					43
		4.3.7	Preparat	tion of G	ibbe	erellic acid	(GA3) so	olutio	n			43
4	4.4	Experim	ent 4: In	fluence	of	different de	oses Bav	vistin	DF on ph	ysiochem	ical	
		propertie	es and she	elf life of	f ma	ingo variet	ies durin	g sto	rage			43
		4.4.1	Experim	ental de	esigr	n						43
		4.4.2	Treatme	ents								43
		4.4.3	Experim	ental la	yout	t						44
		4.4.4	Data col	lection								44
		4.4.5	Methodo	ology								44
		4.4.6	Applicat	ion of st	ora	ge treatme	nts					44
		4.4.7	Prepara	tion of E	Bavis	stin DF (BC	F) soluti	on				44
	4.5	Organol	eptic eval	uation								44
;	5.0	Paramet	ers studie	ed.								45
		5.1	External	fruit fe	atur	es						45
		5.2	Physico	chemica	l pa	rameters						45 - 54
			5.2.1	Edible	por	tion of ma	ngo					45
			5.2.2	Deter	mina	ation of pul	p to pee	l rati	0			45
			5.2.3	Physic	ologi	cal weight	loss of	mang	go fruit			46
			5.2.4	Percer	nt m	oisture cor	ntent of	pulp				46
			5.2.5	Percer	nt di	ry matter c	ontent o	fpul	р			46
			5.2.6	Percer	nt as	sh content	of pulp					46
			5.2.7	Estima	atior	n of vitamir	n C conte	ent of	f mango pu	lp		47
			5.2.8	Titrab	le ad	cidity in ma	ango pulp	р				48
			5.2.9	Pulp p	H of	^f mango						49
			5.2.10	Total s	solu	ble solids (% Brix)	of ma	ango pulp			49
			5.2.11	Deterr	mina	ation of tota	al sugar	conte	ent of mang	jo pulp		49
			5.2.12	Deterr	nina	tion of red	ucing su	gar d	of mango p	qlu		51
			5.2.13	Estima	atior	n of non-re	ducing s	ugar				52
			5.2.14	Estima	tior	n of crude f	ibre of m	nang	o pulp			53
			5.2.15	Deterr	nina	tion of tota	al lipid co	onten	t of mango	pulp		54
			5.2.16	Estima	tior	of water	soluble	prot	ein conten	t of mar	ngo	
				pulp								54
			5.2.17	Detern	nina	tion of diff	erent mi	neral	S			56-62

	5.2.17.1 Determination of phosphorus content	56
	5.2.17.2 Determination of potassium content	58
	5.2.17.3 Determination of calcium content	58
	5.2.17.4 Determination of magnesium content	59
	5.2.17.5 Determination of copper content	60
	5.2.17.6 Determination of iron content	61
	5.2.17.7 Determination of manganese content	61
	5.2.17.8 Determination of zinc content	62
	5.3 Shelf life of mango	62
6.0	Statistical analysis	62
СНА	APTER 4: RESULTS AND DISCUSSION	63 - 250
7.0	Experiment 1: Influence of different storage environments on	
	physicochemical changes and shelf life of postharvest mango	63 - 118
	7.1 Changes in skin color	63
	7.2 Changes in physical characters during storage environments	64
	7.2.1 Edible portion of mango	64
	7.2.2 Pulp to peel ratio	65
	7.2.3 Physiological weight loss	66
	7.2.4 Moisture content	70
	7.2.5 Dry matter content	74
	7.2.6 Ash content	75
	7.3 Changes in biochemical properties of mango during storage	
	environments	75
	7.3.1 Vitamin C content	75
	7.3.2 Titratable acidity	79
	7.3.3 Pulp pH	82
	7.3.4 Total soluble solid (Brix%) content	85
	7.3.5 Total sugar content	86
	7.3.6 Reducing sugar Content	87
	7.3.7 Non reducing sugar content	90
	7.3.8 Crude fibre content	94
	7.3.9 Total lipid content	95
	7.3.10 Water soluble protein content	98
	7.4 Changes in minerals of mango during storage environments	99
	7.4.1 Phosphorus content	99
	7.4.2 Potassium content	100
	7.4.3 Calcium content	103
	7.4.4 Magnesium content	104
	7.4.5 Copper content	107
	7.4.6 Iron content	108

ST.

		7.4.7 Manganese content	111
		7.4.8 Zinc content	112
	7.5	Shelf life	115
8.0	Exper	riment 2: Influence of different doses of Maleic hydrazide on	
	physi	icochemical changes and shelf life of mango during storage	119-162
	8.1	Changes in skin color	119
	8.2	Changes in physical characters during storage environments	119
		8.2.1 Physiological weight loss	120
		8.2.2 Moisture content	121
		8.2.3 Dry matter content	124
		8.2.4 Ash content	125
	8.3	Changes in biochemical properties of mango during storage	
		environments	125
		8.3.1 Vitamin C content	125
		8.3.2 Titratable acidity	128
		8.3.3 Pulp pH	130
		8.3.4 Total soluble solid (Brix %) content	132
		8.3.5 Total sugar content	134
		8.3.6 Reducing sugar Content	137
		8.3.7 Non reducing sugar content	138
		8.3.8 Crude fibre content	139
		8.3.9 Total lipid content	141
		8.3.10 Water soluble protein content	143
	8.4	Changes in minerals of mango during storage environments	145
		8.4.1 Phosphorus content	145
		8.4.2 Potassium content	147
		8.4.3 Calcium content	148
		8.4.4 Magnesium content	152
		8.4.5 Copper content	153
		8.4.6 Iron content	154
		8.4.7 Manganese content	155
		8.4.8 Zinc content	159
	8.5	Shelf life	159
9.0	Exper	riment 3: Effect of different doses of Gibberellic acid on	
	physic	icochemical behavior and shelf life of postharvest mango	163-205
	9.1	Changes in skin color	163
	9.2	Changes in physical characters during storage environments	165
		9.2.1 Physiological weight loss	165
		9.2.2 Moisture content	166
		9.2.3 Dry matter content	167

8

F

		9.2.4	Ash content	168
	9.3	Changes	in blochemical Droperties of mango during storage	100
		environm	ents	168
		9.3.1	Vitamin C content	172
		9.3.2	Titratable acidity	173
		9.3.3	Pulp pH	174
		9.3.4	Total soluble solid (Brix %) content	175
		9.3.5	Total sugar content	179
		9.3.6	Reducing sugar content	180
		9.3.7	Non reducing sugar content	181
		9.3.8	Crude fibre content	182
		9.3.9	Total lipid content	183
		9.3.10	Water soluble protein content	184
	9.4	Changes	in mineral contents of mango during storage environments	185
		9.4.1	Phosphorus content	185
		9.4.2	Potassium content	191
		9.4.3	Calcium content	192
		9.4.4	Magnesium content	193
		9.4.5	Copper content	197
		9.4.6	Iron content	198
		9.4.7	Manganese content	199
		9.4.8	Zinc content	199
	9.5	Shelf life		200
10.0	Experi	iment 4: I	nfluence of different doses of Bavistin DF on physiochemical	
	prope	rties and s	helf life of mango varieties during storage	206-250
	10.1	Changes	s in skin color	206
	10.2	Changes	s in physical characters during storage environments	208
		10.2.1	Physiological weight loss	208
		10.2.2	Moisture content	209
		10.2.3	Dry matter content	210
		10.2.4	Ash content	211
	10.3	Changes	in biochemical properties of mango during storage	
		environn	nents	211
		10.3.1	Vitamin C content	215
		10.3.2	Titratable acidity	216
		10.3.3	Pulp pH	217
		10.3.4	Total soluble solid (Brix %) content	218
		10.3.5	Total sugar content	222
		10.3.6	Reducing sugar content	223
		10.3.7	Non reducing sugar content	224

5

-4

x

	10.3.8	Crude fibre content	225
	10.3.9	Total lipid content	226
	10.3.10	Water soluble protein content	227
10.4	Changes in	n mineral contents of mango during storage environments	233
	10.4.1	Phosphorus content	233
	10.4.2	Potassium content	234
	10.4.3	Calcium content	236
	10.4.4	Magnesium content	237
	10.4.5	Copper content	241
	10.4.6	Iron content	242
	10.4.7	Manganese content	243
	10.4.8	Zinc content	244
10.5	Shelf life		248
CHAPTER 5	SUMMARY	AND CONCLUSION	251 - 263
11.1	Summary		251
11.2	Conclusion		262
11.3	Recommen	dation	263
REFERENCE	S		264 - 279
APPENDICE	S		280 - 305

mt:

2

A

LIST OF FIGURES

1

Ŷ

1

Figures no	Title of figures	Page no
Fig. 1	Standard curve of glucose for estimation of total sugar	51
Fig. 2	Standard curve of glucose for estimation of reducing sugar	52
Fig. 3	Standard curve of protein for estimation of water soluble protein	55
Fig. 1.1	Edible portion of mango pulp in varieties at different days after	
	storage	67
Fig. 1.2	Pulp to peel ratio of mango in varieties at different days after	
	storage	67
Fig. 1.3	Edible portion of mango pulp as influenced by storage treatments at	
5	different days after storage. Vertical bars represent LSD at 0.05 level	67
Fig. 1.4	Pulp to peel ratio of mango as influenced by storage treatments at	
-	different days after storage. Vertical bars represent LSD at 0.05 level	67
Fig. 1.5	Physiological weight loss of mango between varieties at different	
	days after storage	72
Fig. 1.6	Moisture content of mango pulp between varieties at different days	
	after storage	72
Fig. 1.7	Physiological weight loss of mango as influenced by storage	
	treatments at different days after storage. Vertical bars represent	
	LSD at 0.05 level	72
Fig.1.8	Moisture content of mango pulp as influenced by storage treatments	
	at different days after storage. Vertical bars represent LSD at 0.05	
	level	72
Fig. 1.9	Effect of varieties on dry matter content of mango pulp at different	
	days after storage	76
Fig. 1.10	Effect of varieties on ash content of mango pulp at different days	
	after storage	76
Fig. 1.11	Dry matter content of mango pulp as influenced by storage	
	treatments at different days after storage. Vertical bars represent	
	LSD at 0.05 level	76
Fig. 1.12	Ash content of mango pulp as influenced by storage treatments at	
	different days after storage. Vertical bars represent LSD at 0.05 level	76
Fig. 1.13	Effect of varieties on vitamin C content of mango pulp at different	
	days after storage	79
Fig. 1.14	Vitamin C content of mango pulp as influenced by storage treatments	
	at different days after storage. Vertical bars represent LSD at 0.05	70
Fig. 1.15	level	/9
FIG. 1.15	check of varieties on titratable acidity of mango pulp at different	83
	uays aren sturage	00

Fig. 1.16	Effect of varieties on pH of mango pulp at different days after storage	83
Fig. 1.17	Titratable acidity of mango pulp as influenced by storage treatments	
	level	83
Fig. 1.18	Pulp pH of mango as influenced by storage treatments at different days after storage. Vertical bars represent LSD at 0.05 level	83
Fig. 1.19	Effect of varieties on TSS of mango pulp at different days after storage	88
Fig. 1.20	Effect of varieties on total sugar content of mango pulp at different days after storage	88
Fig. 1.21	TSS of mango pulp as influenced by storage treatments at different days after storage. Vertical bars represent LSD at 0.05 level	88
Fig. 1.22	Total sugar content of mango pulp as influenced by storage treatments at different days after storage. Vertical bars represent	00
	LSD at 0.05 level	88
Fig. 1.23	storage	96
Fig. 1.24	Effect of varieties on total lipid content of mango pulp at different davs after storage	96
Fig. 1.25	Crude fibre of mango pulp as influenced by storage treatments at	
F 1 1 1	different days after storage. Vertical bars represent LSD at 0.05 level	96
Fig. 1.26	at different days after storage. Vertical bars represent LSD at 0.05	
	level	96
Fig. 1.27	Effect of varieties on water soluble protein content of mango pulp at different days after storage	101
Fig. 1.28	Effect of varieties on phosphorus content of mango pulp at different	
	days after storage	101
Fig. 1.29	Water soluble protein content of mango pulp as influenced by storage	
	treatments at different days after storage. Vertical bars represent	101
Fig. 1.30	LSD at 0.05 level	101
rig. 1.50	treatments at different days after storage. Vertical bars represent	
	LSD at 0.05 level	101
Fig. 1.31	Effect of varieties on potassium content of mango pulp at different	
	days after storage	105
Fig. 1.32	Effect of varieties on calcium content of mango pulp at different days	
	after storage	105

xiii

Fig. 1.33	Potassium content of mango pulp as influenced by storage treatments at different days after storage. Vertical bars represent	
	ISD at 0.05 level	105
Fig. 1.24	Calcium content of mango nulp as influenced by storage treatments	
FIG. 1.34	at different days after storage. Vertical hars represent LSD at 0.05	
		105
	level	105
Fig. 1.35	Effect of varieties on magnesium content of mango pulp at different	100
	days after storage	109
Fig. 1.36	Effect of varieties on copper content of mango pulp at different days	
	after storage	109
Fig. 1.37	Magnesium content of mango pulp as influenced by storage	
	treatments at different days after storage. Vertical bars represent	
	LSD at 0.05 level	109
Fig. 1.38	Copper content of mango pulp as influenced by storage treatments at	
	different days after storage. Vertical bars represent LSD at 0.05 level	109
Fig. 1.39	Effect of varieties on iron content of mango pulp at different days	
	after storage	113
Fig. 1.40	Effect of varieties on manganese content of mango pulp at different	
	days after storage	113
Fig. 1.41	Iron content of mango pulp as influenced by storage treatments at	
5	different days after storage. Vertical bars represent LSD at 0.05 level	113
Fig. 1.42	Manganese content of mango pulp as influenced by storage	
	treatments at different days after storage. Vertical bars represent	
	I SD at 0.05 level	113
Fig 1 43	Effect of varieties on zinc content of mango pulp at different days	115
11g.1.45	after storage	116
Fig. 1.44	Effect of varieties on shelf life of mange	116
Fig. 1.45	Zinc content of manage pulp as influenced by storage treatments at	110
Fig. 1.45	different days after storage. Vertical hars represent LCD at 0.05 level	116
	Chelf life of manage as influenced by stamps treatments. Noticel bars	110
FIG. 1.46	Shelf life of mango as influenced by storage treatments. Vertical bars	
	represent LSD at 0.05 level	116
Fig. 2.1	Effect of varieties on physiological weight loss of mango at different	122
	days after storage	122
Fig. 2.2	Effect of varieties on moisture content of mango pulp at different	177
	days after storage	122
riy. 2.3	Malele bydraalde (MH) solution at different days after stores	
	Vertical bars represent ISD at 0.05 level	122
Fig 2.4	Moisture content of mange null as influenced by different doces of	
19.2.7	Holstore content of mango pup as initiaticed by different doses of	122

M

Y

3	Maleic hydrazide (MH) solution at different days after storage. Vertical bars represent LSD at 0.05 level	
Fig. 2.5	Effect of varieties on dry matter content of mango pulp at different	
	days after storage	126
Fig. 2.6	Effect of varieties on ash content of mango pulp at different days	
	after storage	126
Fig. 2.7	Dry matter content of mango pulp as influenced by different doses of	
	MH solution at different days after storage. Vertical bars represent	
	LSD at 0.05 level	126
Fig. 2.8	Ash content of mango pulp as influenced by different doses of MH	
	solution at different days after storage. Vertical bars represent LSD at	
	0.05 level	126
Fig. 2.9	Effect of varieties on vitamin C content of mango pulp at different	
	days after storage	128
Fig. 2.10	Vitamin C content of mango pulp as influenced by different doses of	
	MH solution at different days after storage. Vertical bars represent	
	LSD at 0.05 level	128
Fig. 2.11	Effect of varieties on titratable acidity of mango pulp at different days	
	after storage	131
Fig. 2.12	Effect of varieties on pulp pH of mango at different days after	
	storage	131
Fig. 2.13	Titratable acidity of mango pulp as influenced by different doses of	
	MH solution at different days after storage. Vertical bars represent	
	LSD at 0.05 level	132
Fig. 2.14	Pulp pH of mango as influenced by different doses of MH solution at	
	different days after storage. Vertical bars represent LSD at 0.05 level	132
Fig. 2.15	Effect of varieties on total soluble solid of mango pulp at different	
	days after storage	135
Fig. 2.16	Effect of varieties on total sugar content of mango pulp at different	
	days after storage	135
Fig. 2.17	Total soluble solid of mango pulp as influenced by different doses of	
	MH solution at different days after storage. Vertical bars represent	105
E : 0.40	LSD at 0.05 level	135
Fig. 2.18	I otal sugar content of mango pulp as influenced by different doses of	
	ISD at 0.05 level	135
Fia. 2.19	Effect of varieties on crude fibre of mango pulp at different days after	100
	storage	142
Fig. 2.20	Effect of varieties on lipid content of mango pulp at different days	
	after storage	142
Fig. 2.21	Crude fibre of mango pulp as influenced by different doses of MH	143

R

A

	solution at different days after storage. Vertical bars represent LSD at	
Fig 2.22	Lipid content of mango pulp as influenced by different doses of MH	
FIY. 2.22	solution at different days after storage. Vertical bars represent LSD at	
		143
Fig 2.22	Effect of different doses of MH solution on water soluble protein	
119. 2.25	content of mango pulp at different days after storage. Vertical bars	
	represent LSD at 0.05 level	149
Fig. 2.24	Effect of different doses of MH solution on phosphorus content of	
	mango pulp at different days after storage. Vertical bars represent	
	LSD at 0.05 level	149
Fig. 2.25	Potassium content of mango pulp as influenced by different doses of	
	MH solution at different days after storage. Vertical bars represent	
	LSD at 0.05 level	149
Fig. 2.26	Calcium content of mango pulp as influenced by different doses of MH	
	solution at different days after storage. Vertical bars represent LSD at	
	0.05 level	149
Fig. 2.27	Effect of different doses of MH solution on magnesium content of	
	mango pulp at different days after storage. Vertical bars represent	
	LSD at 0.05 level	156
Fig. 2.28	Effect of different doses of MH solution on copper content of mango	
	pulp at different days after storage. Vertical bars represent LSD at	
	0.05 level	156
Fig. 2.29	Iron content of mango pulp as influenced by different doses of MH	
	solution at different days after storage. Vertical bars represent LSD at	
5. 2.20	0.05 level	156
Fig. 2.30	Manganese content of mango pulp as influenced by different doses of	
	MH solution at different days after storage. Vertical bars represent	156
Fig. 2.31	Effect of different doses of MH solution on zinc content of mango nuln	150
19.2.51	at different days after storage. Vertical bars represent LSD at 0.05	
		160
Fig 2 32	Effect of varieties, on shelf life of mango	100
Fig. 2.32	Effect of different decor of MH colution on shelf life of manage. Vertical	160
riy. 2.33	have represent LCD at 0.05 level	
Fig. 0.1		161
FIG. 3.1	Effect of different doses of Giberellic acid (GA3) on physiological	
	weight of mango at different days after storage. Vertical bars	
F1. 0.5	represent LSD at 0.05 level	169
Fig. 3.2	Effect of different doses of Gibberellic acid (GA3) on moisture content	169

×

*

ix.

Y

x

xvi

	of mango pulp at different days after storage. Vertical bars represent LSD at 0.05 level	
Fig 3 3	Dry matter content of mango pulp as influenced by different doses of	
119. 5.5	Gibberellic acid (GA3) at different days after storage. Vertical bars	
	represent LSD at 0.05 level	169
Fig 3.4	Ash content of mango pulp as influenced by different doses of	
119. 5.1	Gibberellic acid (GA3) at different days after storage. Vertical bars	
	represent LSD at 0.05 level	169
Fig. 3.5	Effect of different doses of GA3 on vitamin C content of mango pulp	
	at different days after storage. Vertical bars represent LSD at 0.05	
	level	176
Fig. 3.6	Titratable acidity of mango pulp as influenced by different doses of	
	GA3 at different days after storage. Vertical bars represent LSD at	
	0.05 level	176
Fig. 3.7	Pulp pH of mango as influenced by different doses of GA3 at different	
	days after storage. Vertical bars represent LSD at 0.05 level	176
Fig. 3.8	Effect of different doses of GA3 on total soluble solid content of	
	mango pulp at different days after storage. Vertical bars represent	. – 6
	LSD at 0.05 level	176
Fig. 3.9	Effect of different doses of GA3 on total sugar content of mango pulp	
	at different days after storage. Vertical bars represent LSD at 0.05	100
5. 0.10		186
FIG. 3.10	Crude fibre content of mango pulp as influenced by different doses of	
	GAS at different days after storage. Vertical bars represent LSD at	196
Fig. 3.11	Linid content of manon nuln as influenced by different doses of GA3	100
119. 0.11	at different days after storage. Vertical bars represent LSD at 0.05	
	level	186
Fig. 3.12	Effect of GA3 on water soluble protein content of mango pulp at	
	different days after storage. Vertical bars represent LSD at 0.05 level	186
Fig. 3.13	Effect of different doses of GA3 on phosphorus content of mango pulp	
	at different days after storage. Vertical bars represent LSD at 0.05	
	level	194
Fig. 3.14	Potassium content of mango pulp as influenced by different doses of	
	GA3 at different days after storage. Vertical bars represent LSD at	
	0.05 level	194
Fig. 3.15	Calcium content of mango pulp as influenced by different doses of	174
	GA3 at different days after storage. Vertical bars represent LSD at	
	0.05 level	194
		± 2 Ŧ

X

×

1

x

x

xvii

	Effect of different decay of CA2 on meansion content of manage suit	
Fig. 3.16	Effect of different doses of GA3 on magnesium content of mango pup	
	at different days after storage. Vertical bars represent LSD at 0.05	
	level	194
Fig. 3.17	Effect of different doses of GA3 on copper content of mango pulp at	
	different days after storage. Vertical bars represent LSD at 0.05 level	201
Fig. 3.18	Iron content of mango pulp as influenced by different doses of GA3 at	
	different days after storage. Vertical bars represent LSD at 0.05 level	201
Fig. 3.19	Manganese content of mango pulp as influenced by different doses of	
	GA3 at different days after storage. Vertical bars represent LSD at	
	0.05 level	201
Fig. 3.20	Effect of different doses of GA3 on zinc content of mango pulp at	
	different days after storage. Vertical bars represent LSD at 0.05 level	201
Fig. 3.21	Effect of varieties on shelf life of mango	204
Fig. 3.22	Effect of different doses of GA3 on shelf life of mango. Vertical bars	
	represent LSD at 0.05 level	204
Fig. 4.1	Effect of different doses of Bavistin DF (BDF) on physiological weight	
	of mango at different days after storage. Vertical bars represent LSD	
	at 0.05 level	212
Fig. 4.2	Effect of different doses of Bavistin DF (BDF) on moisture content of	
	mango pulp at different days after storage. Vertical bars represent	
	LSD at 0.05 level	212
Fig. 4.3	Dry matter content of mango pulp as influenced by different doses of	
	Bavistin DF (BDF) at different days after storage. Vertical bars	
	represent LSD at 0.05 level	212
Fig. 4.4	Ash content of mango pulp as influenced by different doses of	
	bavistin DF (BDF) at different days after storage. Vertical bars	
	represent LSD at 0.05 level	212
Fig. 4.5	Effect of different doses of BDF on vitamin C content of mango pulp	
±.	at different days after storage. Vertical bars represent LSD at 0.05	
	level	219
Fig. 4.6	Titratable acidity of mango pulp as influenced by different doses of	
	BDF at different days after storage. Vertical bars represent LSD at	
		219
Fig. 4.7	Pulp pH of mango as influenced by different doses of BDF at different	
Fig. 4.9	days after storage. Vertical bars represent LSD at 0.05 level	219
rig. 4.0	mange puls at different days after storage. Vertical bars represent	
	LSD at 0.05 level	210
Fig. 4.9	Effect of different doses of BDF on total sugar content of mango pulp	219
	at different days after storage. Vertical bars represent LSD at 0.05	
	level	229

X

X

A

X

×

A

xviii

Fig. 4.10	Crude fibre content of mango pulp as influenced by different doses of BDE at different days after storage. Vertical bars represent LSD at	
	0.05 level	229
Fig. 4.11	Lipid content of mango pulp as influenced by different doses of BDF	
	at different days after storage. Vertical bars represent LSD at 0.05	
	level	229
Fig. 4.12	Effect of different doses of BDF on water soluble protein content of	
	mango pulp at different days after storage. Vertical bars represent	
	LSD at 0.05 level	229
Fig. 4.13	Effect of different doses of BDF on phosphorus content of mango pulp	
	at different days after storage. Vertical bars represent LSD at 0.05	
	level	238
Fig. 4.14	Potassium content of mango pulp as influenced by different doses of	
	BDF at different days after storage. Vertical bars represent LSD at	
	0.05 level	238
Fig. 4.15	Calcium content of mango pulp as influenced by different doses of	
	BDF at different days after storage. Vertical bars represent LSD at	
	0.05 level	238
Fig. 4.16	Effect of different doses of BDF on magnesium content of mango pulp	
	at different days after storage. Vertical bars represent LSD at 0.05	
	level	238
Fig. 4.17	Effect of different doses of BDF on copper content of mango pulp at	
	different days after storage. Vertical bars represent LSD at 0.05 level	245
Fig. 4.18	Iron content of mango pulp as influenced by different doses of BDF at	
	different days after storage. Vertical bars represent LSD at 0.05 level	245
Fig. 4.19	Manganese content of mango pulp as influenced by different doses of	
	BDF at different days after storage. Vertical bars represent LSD at	
	0.05 level	245
Fig. 4.20	Effect of different doses of BDF on zinc content of mango pulp at	
	different days after storage. Vertical bars represent LSD at 0.05 level	245
Fig. 4.21	Effect of varieties on shelf life of mango.	249
Fig. 4.22	Effect of different doses of BDF on shelf life of mango. Vertical bars	
	represent LSD at 0.05 level	249

x

*

X

3

*

Ą.

xix

LIST OF TABLES

Î

X

S

A

x

A

Tables no.	Title of tables	Page no.
Table 1.1	Changes in skin color of two mango varieties (viz.,Langra and	
	Khirshapat) as influenced by different postharvest treatments during	
	storage at ambient condition	68
Table 1.2	Combined effects of varieties and different storage treatments on	
	edible portion and pulp to peel ratio of postharvest mango	69
Table1.3	Combined effects of variety and different storage treatments on	
	physiological weight loss and moisture content of postharvest mango	73
Table 1.4	Combined effects of varieties and different storage treatments on dry	
	matter and ash content of postharvest mango	77
Table 1.5	Combined effects of varieties and different storage treatments on	
	vitamin C content of postharvest mango	80
Table 1.6	Combined effects of varieties and different storage treatments on	
	titratable acidity and pulp pH of postharvest mango	84
Table 1.7	Combined effects of varieties and different storage treatments on	
	total soluble solid and total sugar content of postharvest mango	89
Table 1.8	Changes of reducing and non reducing sugar content between mango	
	varieties and influenced by different storage treatments	92
Table 1.9	Combined effects of varieties and different storage treatments on	
	reducing and non reducing sugar content of postharvest mango	93
Table 1.10	Combined effects of varieties and different storage treatments on	
	crude fibre and lipid content of postharvest mango	97
Table 1.11	Combined effects of varieties and different storage treatments on	
	water soluble protein and phosphorus content of postharvest mango	102
Table 1.12	Combined effects of varieties and different storage treatments on	
	potassium and calcium content of postharvest mango	106
Table 1.13	Combined effects of varieties and different storage treatments on	110
Table 1 14	magnesium and copper content of postnarvest mango	110
14016 1.14	iron and manganese content of nostbaryest mango	114
Table 1.15	Combined effects of varieties and different storage treatments on	
	zinc content and shelf life of postharvest mango	118
Table 2.1	Changes in skin color of two mango varieties as influenced by	
	different doses of Maleic hydrazide solution during storage at	
	ambient condition	123
Table 2.2	Combined effects of varieties and different doses of Malelc hydrazlde	
	solution on physiological weight loss of mango and moisture content	
	of mango pulp during storage at ambient condition	123

Table 2.3	Combined effects of varieties and different doses of Maleic hydrazide	
	starses at embient condition	129
Table 2.4	Combined effects of varieties and different doses of Maleic hydrazide	
Table 2.4	solution on vitamin C content of nostbaryest mange during storage at	
	ambient condition	129
Table 2.5	Combined effects of varieties and different doses of Maleic hydrazide	
	solution on titratable acidity and pulp pH of postbaryest manage	
	during storage at ambient condition	136
Table 2.6	Combined effects of varieties and different doses of Maleic hydrazide	100
Table 2.0	solution on total soluble solid and total sugar content of postharvest	
	mange during storage at ambient condition	136
Table 2.7	Changes of reducing and non reducing sugar content in mango	
	varieties and influenced by different doses of Maleic hydrazide	
	solution at ambient condition during storage	140
Table 2.8	Combined effects of varieties and different doses of Maleic hydrazide	
	solution on reducing and non reducing sugar content of postharvest	
	mango during storage at ambient condition	140
Table 2.9	Combined effects of varieties and different doses of Maleic hydrazide	
	solution on crude fibre and lipid content of postharvest mango during	
	storage at ambient condition	144
Table 2.10	Changes of water soluble protein and phosphorus content in	
	postharvest mango varieties during storage at ambient condition	150
Table 2.11	Combined effects of varieties and different doses of Maleic hydrazide	
	solution on water soluble protein and phosphorus content of	
	postharvest mango during storage at ambient condition	150
Table 2.12	Changes in potassium and calcium content of postharvest mango	
	during storage environments as influenced by varieties and different	
	doses of Maleic hydrazide solution at ambient condition	151
Table 2.13	Combined effects of varieties and different doses of Maleic hydrazide	
	solution on potassium and calcium content of postharvest mango	
	during storage at ambient condition	151
Table 2.14	Changes of magnesium and copper content in postharvest mango	
	varieties during storage at ambient condition	157
Table 2.15	Combined effects of varieties and different doses of Maleic hydrazide	
	solution on magnesium and copper content of postharvest mango	
	during storage at ambient condition	157
Table 2.16	Changes of iron and manganese content in postharvest mango	107
2.	varieties during storage at ambient condition	150
		120

x

X

F

A

A

Table 2.17	Combined effects of varieties and different doses of Maleic hydrazide	
	solution on iron and manganese content of postharvest mango during	
	storage at ambient condition	158
Table 2.18	Changes of zinc content in varieties and shelf life of postharvest	
	mango during storage at ambient condition	162
Table 2.19	Combined effects of varieties and different doses of Maleic hydrazide	
	solution on zinc content and shelf life of postharvest mango during	
	storage at ambient condition	162
Table 3.1	Changes in skin color of two mango varieties as influenced by	
	different doses of Gibberellic acid during storage at ambient condition	164
Table 3.2	Changes of physiological weight loss and moisture content in mango	
	varieties during storage at ambient condition	170
Table 3.3	Combined effects of varieties and different doses of Gibberellic acid	
	solution on physiological weight loss and moisture content of	
	postharvest mango during storage at ambient condition	170
Table 3.4	Dry matter and ash content of mango pulp changes in varieties	
	during storage at ambient condition	171
Table 3.5	Combined effects of varieties and different doses of Gibberellic acid	
	solution on dry matter and ash content of postharvest mango during	
	storage at ambient condition	171
Table 3.6	Changes of vitamin C content of mango pulp in varieties during	
	storage at ambient condition	177
Table 3.7	Combined effects of varieties and different doses of Gibberellic acid	
1. A	solution on vitamin C content of postharvest mango pulp during	
	storage at ambient condition	177
Table 3.8	Changes of titratable acidity and pulp pH of postharvest mango pulp	
	between varieties during at ambient condition	178
Table 3.9	Combined effects of varieties and different doses of Gibberellic acid	
	solution on titratable acidity and pulp pH of postharvest mango	470
T-11-240	during storage at ambient condition	178
1 able 3.10	Changes of total soluble solid and total sugar content of postnarvest	188
Table 3 11	Combined effects of varieties and different doses of Gibberellic acid	100
	solution on total soluble solid and total sugar content of postharvest	
	mango during storage at ambient condition	188
Table 3.12	Changes of reducing and non reducing sugar content of postharvest	
	mango pulp between varieties and influenced by different doses of	
	Gibberellic acid solution during storage at ambient condition	189
Table 3.13	Combined effects of varieties and different doses of Gibberellic acid	
	solution on reducing and non reducing sugar content of postharvest	
	mango pulp during storage at ambient condition	189

S,

2

-

X

Y

4

Table 3.14	Changes of crude fibre and lipid content of mango pulp between	
	varieties during storage at ambient condition	190
Table 3.15	Combined effects of varieties and different doses of Gibberellic acid	
	solution on crude fibre and lipid content of postharvest mango pulp	
	during storage at ambient condition	190
Table 3.16	Changes of water soluble protein and phosphorus content of mango	
	pulp in varieties during storage at ambient condition	195
Table 3.17	Combined effects of varieties and different doses of Gibberellic acid	
	solution on water soluble protein and phosphorus content of	
	postharvest mango pulp during storage at ambient condition	195
Table 3.18	Changes of potassium and calcium content of mango pulp between	
	varieties during storage at ambient condition	196
Table 3.19	Combined effects of varieties and different doses of Gibberellic acid	
	solution on potassium and calcium content of postharvest mango	
	pulp during storage at ambient condition	196
Table 3.20	Changes of magnesium and copper content of mango pulp in	
	varieties during storage at ambient condition	202
Table 3.21	Combined effects of varieties and different doses of Gibberellic acid	
	solution on magnesium and copper content of postharvest mango	
	pulp during storage at ambient condition	202
Table 3.22	Changes of iron and manganese content of postharvest mango pulp	
	between varieties during storage at ambient condition	203
Table 3.23	Combined effects of varieties and different doses of Gibberellic acid	
	solution on iron and manganese content of postharvest mango pulp	
	during storage at ambient condition	203
Table 3.24	Changes of zinc content of mango pulp and shelf life of postharvest	
	mango in varieties during storage at ambient condition	205
Table 3.25	Combined effects of varieties and different doses of Gibberellic acid	
	solution on zinc content of mango pulp and shelf life of postharvest	
	mango during storage at ambient condition	205
Table 4.1	Changes in skin color of two mango varieties as influenced by	
	different doses of Bavistin DF during storage at ambient condition	207
Table 4.2	Changes of physiological weight loss of mango and moisture content	
	of mango pulp between varieties during storage condition	213
Table 4.3	Combined effects of varieties and different doses of Bavistin DF	
	solution on physiological weight loss and moisture content of	
	postharvest mango at ambient condition.	313
Table 4.4	Changes of dry matter and ash content of mango pulp between	
	varieties during storage at ambient condition	214

F.

T

X-

-

×

Table 4.5	Combined effects of varieties and different doses of Bavistin DF	
	solution on dry matter and ash content of postharvest mango pulp at	
	room temperature	214
Table 4.6	Changes of vitamin C content of mango pulp in varieties during	
	storage at ambient condition	220
Table 4.7	Combined effects of varieties and different doses of Bavistin DF	
	solution on vitamin C content of postharvest mango at ambient	
	condition	220
Table 4.8	Changes of titratable acidity and pH of postharvest mango pulp in	
	varieties during storage at room temperature	221
Table 4.9	Combined effects of varieties and different doses of Bavistin DF	
	solution on titratable acidity and pulp pH of postharvest mango pulp	
	during storage at ambient condition	221
Table 4.10	Pattern of total soluble solid and total sugar content of postharvest	
	mango pulp between varieties during storage at ambient condition	230
Table 4.11	Combined effects of varieties and different doses of Bavistin DF	
	solution on total soluble solid and total sugar content of postharvest	
	mango pulp at ambient condition	230
Table 4.12	Behavior of reducing and non reducing sugar content of postharvest	
	mango pulp in varieties and influenced by different doses of Bavistin	
	DF solution during storage condition	231
Table 4.13	Combined effects of varieties and different doses of Bavistin DF	
	solution on reducing and non reducing sugar content of postharvest	
	mango pulp at ambient condition	231
Table 4.14	Behavior of crude fibre and lipid content of postharvest mango pulp	
	in varieties during storage environments at ambient condition	232
Table 4.15	Combined effects of varieties and different doses of BDF solution on	
	crude fibre and lipid content of postharvest mango pulp during	
	storage at ambient condition	232
Table 4.16	Changes of water soluble protein and phosphorus content of	
	postharvest mango pulp in varieties during storage environments at	
	ambient condition	239
Table 4.17	Combined effects of varieties and different doses of BDF solution on	
	water soluble protein and phosphorus content of postharvest mango	
	pulp at ambient condition	239
Table 4.18	Changes of potassium and calcium content of postharvest mango	
	pulp between varieties during storage environments at ambient	
	condition	240

r

-

 $\overline{\mathbf{x}}$

+

÷

±.

Combined effects of varieties and different doses of BDF solution on	
potassium and calcium content of postharvest mango pulp during	
storage at ambient condition	240
Pattern of magnesium and copper content of postharvest mango pulp	
during storage environments at ambient condition	246
Combined effects of varieties and different doses of BDF solution on	
magnesium and copper content of postharvest mango at ambient	
condition	246
Behavior of iron and manganese content of postharvest mango pulp	
in varieties during storage environments at ambient condition	247
Combined effects of varieties and different doses of BDF solution on	
iron and manganese content of postharvest mango pulp at ambient	
condition	247
Pattern of zinc content of mango pulp and shelf life of postharvest	
mango in varieties during storage environments at ambient condition	250
Combined effects of varieties and different doses of BDF solution on	
zinc content and shelf life of postharvest mango pulp at ambient	
condition	250
	Combined effects of varieties and different doses of BDF solution on potassium and calcium content of postharvest mango pulp during storage at ambient condition Pattern of magnesium and copper content of postharvest mango pulp during storage environments at ambient condition Combined effects of varieties and different doses of BDF solution on magnesium and copper content of postharvest mango at ambient condition Behavior of iron and manganese content of postharvest mango pulp in varieties during storage environments at ambient condition Combined effects of varieties and different doses of BDF solution on iron and manganese content of postharvest mango pulp at ambient condition Combined effects of varieties and different doses of BDF solution on iron and manganese content of postharvest mango pulp at ambient condition Pattern of zinc content of mango pulp and shelf life of postharvest mango in varieties during storage environments at ambient condition Combined effects of varieties and different doses of BDF solution on zinc content and shelf life of postharvest mango pulp at ambient condition

30-

T

t.

LIST OF PLATES

F

2

x

A

F

T)

X

Plates no.	Title of plates	Page no.
Plate 1	Photograph of Langra before using postharvest treatments	35
Plate 2	Photograph of Khirshapat before using postharvest treatments	36
Plate 3	Photograph of Langra and Khirshapat in basket after collection from	
	farmer's orchard	40
Plate 4	photograph of the application of different postharvest treatments	
	especially, paraffin coating and hot water treatments on mangoes	40

LIST OF APPENDICES

X

×

1

-

ž

	Title of appendices	Page no.
Appendices no.	Title of appendices in inimum temperature,	
Appendix 1.1	Monthly records of maximum and minimum and minimum and the	
	relative humidity and rainfall during the study period	
	Department of Biochemistry and Molecular Biology, Oniversity	280
	of Rajshahi, Bangladesh	200
Appendix 1.2	Analysis of variance of data on edible portion and pulp to peer	
	ratio of mango as influenced by varieties and different storage	
	treatments	281
Appendix 1.3	Analysis of variance of data on physiological weight loss of	
	mango and moisture content of mango pulp as influenced by	
	varieties and different storage treatments	281
Appendix 1.4	Analysis of variance of data on dry matter and ash content of	
	mango pulp as influenced by varieties and different storage	
	treatments	282
Appendix 1.5	Analysis of variance of data on vitamin C content of mango pulp	
	as influenced by varieties and different storage treatments	282
Appendix 1.6	Analysis of variance of data on titratable acidity and pH of	
	mango pulp as influenced by varieties and different storage	
	treatments	283
Appendix 1.7	Analysis of variance of data on total soluble solid and total	
	sugar content of mango pulp as influenced by varieties and	
	different storage treatments	283
Appendix 1.8	Analysis of variance of data on reducing and non-reducing	
	sugar content of mango pulp as influenced by varieties and	
	different storage treatments	284
Appendix 1.9	Analysis of variance of data on crude fibre and lipid content of	
	mango pulp as influenced by varieties and different storage	
	treatments	284
Appendix 1.10	Analysis of variance of data on water soluble protein and	
	phosphorus content of mango pulp as influenced by varieties	
	and different storage treatments	285
Appendix 1 11	Analysis of variance of data on potassium and calcium content	
	of mango nulp as influenced by varieties and different storage	
	treatments	285
Appendix 1 12	Analysis of variance of data on magnesium and conner content	205
	of mango pulp as influenced by varieties and different storage	
	treatments	286

Appendix 3.1	Analysis of variance of data on physiological weight loss of	
	mango and moisture content of mango pulp as influenced by	
	varieties and different doses of Gibberellic acid solution	294
Appendix 3.2	Analysis of variance of data on dry matter and ash content of	
	mango pulp as influenced by varieties and different doses of	
	Gibberellic acid solution	294
Appendix 3.3	Analysis of variance of data on vitamin C content of mango pulp	
	as influenced by varieties and different doses of Gibberellic acid	
	solution	295
Appendix 3.4	Analysis of variance of data on titratable acidity and pH of	
	mango pulp as influenced by varieties and different doses of	
	Gibberellic acid solution	295
Appendix 3.5	Analysis of variance of data on total soluble solid and total	
	sugar content of mango pulp as influenced by varieties and	
	different doses of Gibberellic acid solution	296
Appendix 3.6	Analysis of variance of data on reducing and non reducing	
	sugar content of mango pulp as influenced by varieties and	
	different doses of Gibberellic acid solution	296
Appendix 3.7	Analysis of variance of data on crude fibre and lipid content of	
	mango pulp as influenced by varieties and different doses of	
	Gibberellic acid	297
Appendix 3.8	Analysis of variance of data on water soluble protein and	
	phosphorus content of mango pulp as influenced by varieties	
	and different doses of Gibberellic acid solution	297
Appendix 3.9	Analysis of variance of data on potassium and calcium content	
	of mango pulp as influenced by varieties and different doses of	
	Gibberellic acid solution	298
Appendix 3.10	Analysis of variance of data in magnesium and copper content	
	of mango pulp as influenced by varieties and different doses of	
	Gibberellic acid solution	298
Appendix 3.11	Analysis of variance of data on iron and manganese content of	
	mango pulp as influenced by varieties and different doses of	
	Gibberellic acid solution	299
Appendix 3.12	Analysis of variance of data on zinc content of mango pulp and	
	shelf life of mango as influenced by varieties and different	
	doses of Gibberellic acid solution	200
Appendix 4.1	Analysis of variance of data on physiological weight loss of	299
	manon and moisture content of manon pulp as influenced by	
	variaties and different doses of Bayistin DE solution	200
	varieties and unterent duses of pavistin DF solution	300

ji.

à

*

F

Appendix 4.2	Analysis of variance of data on dry matter and ash content of	
	mango pulp as influenced by varieties and different doses of	
	Bavistin DF solution	300
Appendix 4.3	Analysis of variance of data on vitamin C content of mango pulp	
	as influenced by varieties and different doses of Bavistin DF	
	solution	301
Appendix 4.4	Analysis of variance of data on titratable acidity and pH of	
	mango pulp as influenced by varieties and different doses of	
	Bavistin DF solution	301
Appendix 4.5	Analysis of variance of data on total soluble solid and total	
	sugar content of mango pulp as influenced by varieties and	
	different doses of Bavistin DF solution	302
Appendix 4.6	Analysis of variance of data on reducing and non reducing	
	sugar content of mango pulp as influenced by varieties and	
	different doses of Bavistin DF solution	302
Appendix 4.7	Analysis of variance of data on crude fibre and lipid content of	
1	mango pulp as influenced by varieties and different doses of	
	Bavistin DF solution	303
Appendix 4.8	Analysis of variance of data on water soluble protein and	
	phosphorus content of mango pulp as influenced by varieties	
	and different doses of Bavistin DF solution	303
Appendix 4.9	Analysis of variance of data on potassium and calcium content	
	of mango pulp as influenced by varieties and different doses of	
	Bavistin DF solution	304
Appendix 4.10	Analysis of variance of data on magnesium and copper content	
	of mango pulp as influenced by varieties and different doses of	
	Bavistin DF solution	304
Appendix 4.11	Analysis of variance of data on iron and manganese content of	
	mango pulp as influenced by varieties and different doses of	
	Bavistin DF solution	305
Appendix 4.12	Analysis of variance of data on zinc content of mango pulp and	
	shelf life of mango as influenced by varieties and different	
	doses of Bavistin DF solution	305

J.

X

X

-

F



P

X

X

7

- The

F

5
CHAPTER 1 INTRODUCTION

1.1 Orientation and origin

Mango (*Mangifera indica L.*) is one of the most important fruit crop in tropical and subtropical regions of the world under the family of Anacardiaceae and it was originated in South Asia or Malayan archipelago (Salunkhe and Desai, 1984). It had been cultivated for more than 4000 years (Mukherjee, 1949; Candole, 1984; Bose, 1985). Mukherjee (1949) explored that the genus Mangifera originated in Burma, Indochina and Malayan peninsula; but the mango itself had originated in the Assam – Burma region which also included Bangladesh.

1.2 Production status

The mango is a commercial crop in many countries of the South-East Asia *viz.*, India, Pakistan, the Philippines, Indonesia, Malaysia, Thailand, Burma, Sri Lanka and Java.The main mango producing countries of the world are India, Pakistan, Mexico, Brazil, Haiti and the Philippines, of which India being the largest producer (Salunkhe and Desai, 1984).

Mango ranks third among the tropical fruits grown in the world with a total production of 23.87 million tones (FAO, 2006), to which Bangladesh contributed to only 0.64 million tones (BBS, 2006). It is considered as a fruit crop for home consumption. It is also gaining rapid popularity in the Middle-East, South East Africa, South Africa, Florida, Israel and Australia. Among the fruits, Mango ranks first in terms of area and third in terms of production in Bangladesh (BBS, 1999). Bangladesh produces 184 thousand tons of mangoes per annum from 50.60 thousand hectares of land.

1.3 Uses and importance

×

X

Mango is one of the most important, popular and tasteful fruit crop not only in Bangladesh but also in the world owing to its greater utility, characteristics flavor, attractive color, pleasant aroma, delicious taste and nutritional value. It is consumed as fresh ripe and green fruits. Both the green and ripe fruits are also used to make different varieties of processed products like juice, chutney, pickles, jam, jelly etc. For this reason, it is acknowledged as the king of fruits in Bangladesh as well as in other South-East Asian countries (Pursglove, 1972; Shahjahan *et al.*, 1994).

*

+

*

×

Nutritionally, it contains substantial quantity of appreciable β carotene, vitamin C, and dietary fibre (Pal, 1998) as well as soluble sugars and different minerals which are used for good sources of nutrition and readily available and easily assumable in human body (Singh, 1960) and therefore, is capable to prevent many deficiency diseases (Samad *et al.* 1975, Purohit, 1985, Anon., 1962).

An adult individual needs about 2222 kcal at least of which 2.5% kcal should come from fruits dally as the gross food requirements, but he/she obtains only 1% kcal.(Mondal, 2000). In these circumstances, it is not possible to meet the increasing demand of fruits for growth and development of the human body. About 18, 79,000 metric tons of various types of fruits were produced from the 2, 81,000 hectares of land in the year of 1998-99 whereas our annual demand was about 29,83,875 mt. (BBS, 2000). So, normally there is a shortage of fruit production per year in the country. About 10% of the national income comes from various fruits among the income from agricultural crops, which play a vital role in our national economy (Mondal, 2000). Fruit plants in Bangladesh occupy about 3.91% of the total cultivable land providing only 6.5% of the total foods produced in Bangladesh and they also supply 2.05% of GDP (Mondal, 2000). So, fruit must be economically a very important crop grown in Bangladesh. Mango is a major part of the fruit crops occupied in Bangladesh; but it always decays after harvest.

1.4 Nutritional deficiency

Malnutrition and under nutrition have now become an alarming problem of the people of the third world countries affecting their economic and physical development. Protein-energy malnutrition, vitamin and mineral deficiencies are the most serious nutritional disorder in low income groups. Due to these deficiencies, under-weights and high mortality are prevalent in pre- school children and infants. The minimum dietary requirement of fruit per day per person is 115 g, but our availability is only 30-35 g, which is much lower than the recommended daily allowance (Siddiqui and Scanlan, 1995). The per capita availability of fruits is further reduced sometimes due to high level of postharvest losses (Mondal *et al.*, 1995). Postharvest losses can be considerably reduced by applying improved storage technology and prolonging the shelf life of fruits.

de.

1

*

A

1.5 Quantity of postharvest losses

Postharvest losses and deterioration of nutritional quality of fresh fruit are the most important problems in tropical and sub- tropical regions of the world. A huge quantity of nutritious fruits is being markedly deteriorated due to the lack of proper knowledge on post harvest management practices. As a result, people do not have sufficient nutrition from fruits according to their requirements. A considerable amount of fresh fruits goes waste every year through post-harvest decay. The magnitude of post harvest losses in fresh fruit is estimated to be about 5 to 25% in developed countries and 20 to 50% in developing countries (Khader, 1985). Srinivas *et al.* (1996) reported from India that the total post harvest losses of mango cv. Totapuri and Alphonso to be 17.9% (3.5% orchard/field, 4.9% transportation, 4.10% storage and 5.4 retail level) and 14.4% (1.9% orchard /field, 3.7% transportation, 3.7% storage and 5.3% retail level), respectively.

Mushtaq *et al.* (2005) found that total postharvest losses in the marketing Channel's of mango were 11.97% at farm level (Producer/Contractor level), 11.10% at market level (Phariawala and retailer level) and 7.90% at consumer's level.

Quroshi and Meah (1991) found that post harvest losses of mango fruits varied according to variety from 0 to 16.3%, with an average loss of 12.5%. It also depends on transport distance from production site to retail location.

Approximately 30-50% fruits go waste during postharvest handling, storage and ripening (Lashley, 1984). Among the fruits mango manifested high postharvest losses because of its high perishability and climacteric pattern of respiration. The marketability of this perishable fruit is closely linked with the development of suitable technology which reduces the losses at different stages of harvesting and storage condition.

Losses in terms of quality and quantity of fruits occur at all stages in the post harvest system from harvesting to consumption. Reliable statistical data are inadequate especially in Bangladesh to indicate the magnitude of post harvest losses of mango. Singh (1960) reported that the postharvest losses of mango fruit in India due to microbial decay ranged from 20 to 33%. Quality mangoes are produced in north –western part of Bangladesh, of which about 35- 38 % of post harvest losses

the.

are caused due to inefficient handling during its transportation , storage and marketing (Rubbi *et al.*1985). In India, every year 20- 40% of fruits and vegetables are damaged during transportation and storage (Banik, 1995). Hence, the development of technology for reducing postharvest decay is a necessary prerequisite for the promotion of the fruit industry. It is very important not only to produce more but also to save whatever is grown at high production cost.

Lashley (1984) reported the methods of reducing postharvest decay through genetic control of storage life, field and post harvest treatments viz., hot water treatments, wooden crate packaging, corrugated fibre board packaging and plastic films for atmospheric medication. In addition, wax coating, application of fungicide and growth regulators, low temperature storage, oil dipping, use of different chemicals for slowing down the physical activity and ripening process of fruit during the transport and marketing are the commonly used methods to prevent postharvest wastage of mango fruits. Many workers (Dara, *et al.* 1982; Sankat *et al.* 1993; Baez, *et al.* 1993; Feng *et al., 1991;* Inyang and Agbo, 1995) studied the effects of many postharvest treatments with a large number of mango cultivars and observed the extended shelf life.

1.6 Causes of postharvest losses

The postharvest life of any fruit consists of ripening and senescence. After harvest, fruits undergo many physiological and biochemical changes during storage. Apart from those changes, microbial decay also contributes to postharvest losses during ripening and storage. The storage life of a fruit could be prolonged significantly through slowing down the process leading to ripening, and controlling the microbial decay.

The huge amount of important fruit crops are being spoiled due to prevailing temperature, humidity, inappropriate post harvest handling as well as sub- optimal knowledge in the field of postharvest technology after harvesting. This spoilage of fruit is attributed to adverse biochemical changes, namely loss of weight owing to respiration and transpiration, loss of flesh hardness, loss of resistance to different microblal attack and overall devastating deterioration of carbohydrate, protein, lipid, some oxidative enzymes, minerals and nutrient status.

A Introduction

Chapter 1

Sr.

1

3

*

Appropriate production technology, careful harvesting and proper packaging and storage management can contribute to keep the produce quality good. Once a crop is harvested it is impossible to improve its quality. The fruit crops having their high moisture content are inherently more liable to deteriorate especially under tropical conditions. Moreover, they are biologically active and carry out transpiration, respiration, ripening and other biochemical activities, which deteriorate the quality of the produce.

Mango is the staple fruits in Bangladesh being lost from time immemorial should get top priority to reduce markedly and to extend the shelf life by applying different storage treatments. Then, we can use them properly throughout the year round conserving their quality and quantity. As a result, sustainable utility of mango must be increased.

1.7 Postharvest technology

Post harvest loss of fresh mango fruit is one the major problems in the tropic for its high perishability in nature and climacteric pattern in respiration. To extend the postharvest life of mango, its respiration rate should be reduced as per as possible. Ripening of fruits starts after synthesis of ethylene which causes ripening fast. The factors which strongly influence the rate of respiration and to inhibit the synthesis of ethylene are very essential for reduction of losses and extension of shelf life of postharvest mango. Application of different postharvest treatments viz., paraffin coating, perforated polyethylene cover, unperforated polyethylene cover, hot water treatment and low temperature in refrigerator are very much important obstacles to normal respiration of mango fruits. These treatments strongly impede in ethylene synthesis that resulted in low respiration and delay ripening. These materials also reduced the losses and prolonging the shelf life of mango (Tefera et al. 2007; Benitez et al. 2006; Fawaz, 2006; Muy et al. 2004 and Fonseca et al. 2004). In addition, hormonal treatments like Maliec hydrazide (MH) and Gibberellic acid (GA3) and fungicidal treatments like Bavistin DF (BDF) are also excellent ethylene inhibitors. These treatments performed effectively in reduction of postharvest decay, and extension of shelf life of mango (Ranjan et al. 2005; Dhemre and Waskar, 2004; Gautam et al. 2003; Reddy and Haripriya, 2002 and Ahmed and Singh, 2000). Apparently, these treatments deteriorate the qualities of fruits to some

6-

1

6

extent, but the reduction of losses and extension of postharvet life of mango will help to increase the market price in the off seasons which play a good role in the economic development.

Postharvest decay causes a huge amount of economic loss that hampers the total GDP in a country. There are no available research findings regarding the controlling of postharvest decay and extending the shelf life of mango in Bangladesh contest. So, therefore, it is very much essential to conduct an investigation for details information in the development of new technology that will help in delay ripening of the commercial mango varietles. Studies also needed for minimizing the postharvest decay and to extend the shelf life of mango. So, the outcome of this work will help solve the persistent problems and minimize the economic loss in this regard.

1.8 Objectives of the study

The present investigation was undertaken with the following objectives.

- 1. To identify the better one between two mango varieties in terms of nutrient content.
- To study the behavioral pattern of physicochemical properties of postharvest mango in the storage conditions.
- 3. To find out a desirable technology for extension of shelf life of mango.
- To select the best method for reduction of losses and extension of postharvest life of mango.
- 5. To assess the shelf life of selected fruits as influenced by different physical, chemical and hormonal treatments under different storage conditions.



Y

REVIEW OF LITERATURES

CHAPTER 2

REVIEW OF LITERATURES

Many researchers have studied on physicochemical behavior and shelf life of postharvet *Mangifera indica* L. during storage environments using different kinds of storage treatments. A good number of information's regarding the physicochemical changes and shelf life of mango varieties and storage treatments namely, paraffin coating, perforated and unperforated polyethylene cover, hot water, low temperature in refrigerator, different doses of Maleic hydrazide, Gibberellic acid and Bavistin DF have been accomplished in many parts of the world, but unfortunately, sufficient documents in this connection are not available for our local varieties. Some of research findings relevant to the present investigation, so far as possible to collect, are reviewed and stated in this chapter.

2.1 Effect of paraffin coating

3-

1

4

-

A study was investigated by Muy *et al.* (2004) on storage conditions and waxing affect water status and quality of mango. They exhibited that weight loss increased directly proportional to vapor pressure deficit (VPD). Wax application was effective in reducing by 30% the weight loss, but only at high VPD conditions, empiric model were generated to predict shelf life on control and waxed fruit ($r^2 = 0.90$) when the fruit reached 89% RWT and psi p = 0 (Cellular Plasmolysis) the commercial quality was unaffected. Fruits with 84% RWC psi-p< 0 and firmness = 20 N, defined the beginning of commercial quality loss. Psi w was reduced during storage from 1.0 to 2.8 MPa, depending on VPD conditions and wax application. This reduction was partially due to solute accumulation (Degrees Brix) that modified Psis. As degrees Brix increased, Psi – s descended in a lineal relation that can also be used for prediction.

Fonseca *et al.* (2004) conducted an experiment in a laboratory to study the pulp and skin pigments in mango 'Haden' applying with fungicide and wax. They showed that the use of wax reduced the synthesis of carotenoid pigments in the pulp of 'Haden' mango. It also reduced chlorophyll breakdown of the skin, mainly chlorophylla, resulting in a lesser development of the yellow background coloration. On the other hand, benomyl and benzalconic chloride did not affect the synthesis or the breakdown of pigments of Haden mango.

C Review of Literatures

Chapter 2

5

1

*

4

See.

Fonseca *et al.* (2001) investigated a trial using of fungicides and wax in storage and postharvest quality of 'Haden' mango in cold storage to study the physicochemical behaviors. They reported that Haden mangoes were dipped in benomyl (1g/litre) or benzalkonium chloride (2g/litre) with or without clean wax (an emulsion containing 18.5 to 20.5% of camauba wax and acrylic resins mixture) and stored $(13\pm1)^{\circ}$ C and 80-90%RH for 21 days. Partial ripening occurred during storage. Waxing increased the general appearance of the fruits, mainly by maintaining them more turgid. Both fungicides controlled anthracnose development during storage. They also showed partial ripening occurred during storage.

An experiment was conducted by Zambrano *et al.* (1997) to study the change during storage of wax coating mango fruit cv. Palmer and Keitt. The fruits were dipped fully in 1% aqueous suspension of Prolong or Primafresh C (original concentration) and stored at 15°C and 85-95% RH. Three fruits were analyzed, at 3 days intervals, for 18 days. Observation shows that there was no significant effect on external color but coating affected internal colour. Wax treated fruits exhibited reduced weight loss as compared with untreated ones.

Yuniarti and Suhardi (1992) observed prolonged ripening period (11 days) and longer over ripe stage (by 9 days) compared to control when mango fruits were dipped in wax emulsion of concentration 6 or 7%. They also found minimum weight loss and low content of soluble solids in the above mentioned treatments.

Chandra and Pathak (1992) claimed that post harvest anthracnose severity of mango fruits could be reduced by pre or post inoculation treatments with waxol, mustard oil, ABC (Attapulgite Based Clay) dust and phytoalexin-84. Viller and Korsten(1994) found that the hot water dip of mango cv. 'Kent' followed by wax + ¼ prochloraz + Bacillus licheniformis treatments were significantly better in reducing anthracnose disease than hot water treatment on the wax + Bacillus subtilis treatment.

Castrillo and Bermudez (1992) carried out an extensive study on postharvest ripening in wax-coated 'Bocado' mango. The results showed that the delayed ripening effects of wax-coated fruits affected fresh weight loss, exocarp chlorophyll degradation and mesocarp pH change; but it did not affect mesocarp chlorophyll, sugar and starch content.

·

-th

1

4

die.

Mohiuddin *et al.* (1991) made an investigation to observe the effects of different coating materials viz. paraffin, soybean and mustard oil on the shelf life of mango cv. 'Ashwina' with or without prior application of Dithane M-45. They observed that the paraffin coating reduced both weight loss and rotting with concomitant enhanced shelf life.

Ilanggantilake *et al.* (1989) conducted an experiment to study the effects of pretreatment and hypoboric storage life of mango. They found that mangoes precooled at 15^oC temperature followed by low concentration of 0.5% wax emulsion treatment successfully kept up to 30 days in store without deteriorating fruit quality. Parmar and Chundwat (1989) while working with mango cv. Kesar found that post harvest waxol treatment was most effective in delay ripening along impaired fruit quality when stored at ambient temperature for 10 to 27 days. Joardar (1980) reported that wax coating treatment increased the shelf life and significantly prevented the weight loss of mango.

Krishnamaruthy (1989) conducted a study on the effect of coating material on shelf life of mango and quality attributes. The study consisted on coating unripe mature fruits with Tal-Prolong (a product of sucrose esters, glycerides and cellulose) by dipping fruits for min. in 1 % suspension before storage. This treatment delayed ripening by 4-5 days under ambient condition (28-31°C), for 18 days with reduced weight loss and quality. In another study Huddar *et al.* (1989) found that the dipping in Tal-Prolong at 1 % gave the best retention of fruit colour, flavour and texture over a period of 15 days storage of fruits cv. 'Dashehari' and 'Mallika'.

Chhatpar *et al.* (1972) reported that it was possible to store mangoes for about two and a half months at 5-8°C by coating them with wax emulsion. Ilangantilake and Salokhe (1990) observed that chemical waxing treatment was capable of retaining chlorophyll up to 20 days of storage, but injured the skin after 20 days storage when high concentration of wax was used. It also retarded the ripening process of mango after removal to room temperature conditions. Brown color development on the skin during storage which resulted in an undesirable appearance on the samples. Six percent waxol treated 'Kesar' mango fruits were stored at ambient temperature for 10-27 days, and fruit ripeness, % marketable fruit, weight loss, total soluble solids, sugars and titratable acidity were assessed

the

the.

.

T

Jr.

(Parmar and Chundawat, 1989). Waxol was the most effective in delaying ripening but it had undesirable effect on fruit quality.

Thangaraj and Irulappan (1988) reported that the weight loss (2.3%) and the longest shelflife (9.7 days) were obtained with 10% frutox+ cool chamber storage. Waxing increased the shelf life and prevented the weight loss of mango significantly (Joarder, 1981).

2.2 Effect of polyethylene cover

Fawaz (2006) conducted an experiment with postharvest mango cv. Bullock's heart to investigate the physico-chemical characters during storage. He observed that 8% wax emulsion resulted in the lowest weight losses and the highest acidity of the fruits, and individual wrapping of fruits in low density polythene film (LDPF) resulted the highest carotenoid content. Postharvest decay was not observed in fruits dipped in 2% calcium chloride or 85 % wax emulsion. Flesh firmness was highest in fruits dipped in 2 % calcium chloride. Among the treatment, dipping in 4 % wax emulsion resulted in the highest total suger content of the fruits, although the values obtained were lower than those of the control fruits.

Illeperuma and Jayasuriya (2002) carried out an investigation to chalk out the physicochemical properties and shelf life prolonged storage of 'Karuthacolomban' mango using modified atmosphere packaging at low temperature. They showed that the fruits packaged with scavengers had lower percent weight loss and minimum changes in physicochemical properties compared with the fruits packaged without scavengers. The modified atmosphere created in low density polythene bags was effective in delaying ripening of Karuthacolomban mango up to 16 days but ripening was further delayed up to 21 days in the presence of scavengers. Physicochemical and sensory properties, visual quality rating and disease index of modified atmosphere stored mango after ripening were similar to those packaged in perforated low density polyethylene bags of 1:1 surface area to weight ratio (cm^2g^{-1}) with 50 ml of saturated potassium permanganate absorbed on to suitable porous matrices and 2 gm of activated granular charcoal could be recommended to increase storage life at 13°C.

Review of Literatures

Chapter 2

r

--

1

+

Fruits of cv. Himsagar were stored in lined wooden boxes for up to 14 days after treatment with tap water, cold water or an aminoethoxyvinyl glycine (10 ppm) solution. For each treatment, half of the fruits were kept in boxes with KMnO₄ soaked paper shavings. Physiological weight loss and decay loss were minimal in fruits with the cold water treatment + KMnO₄. Fruit quality characteristics were not appreciably affected (Chattapadhyay, 1989). Mondal *et al.* (1995) found a shelf life of mango of 21 days by keeping mango fruits in polybag. Mondal *et al.* (1998) also observed highest shelf life (26.75 days) of mango by polythene wrapping followed by low temperature (12-13°C) storage in another experiment.

Shivarama-Reddy *et al.* (1989) worked with wax coated fruits cv. 'Alphonso' and stored in perforated polyethylene bags (3, 4, 6 or 8% wax emulsion). They demonstrated that the weight loss and the rate of spoilage in storage were reduced over 20 days when the fruits were treated with fungicide (Thiabendazole, Benlate [Benomy] and Bavistin [Carbendazim] at 100, 250 and 500 ppm). The best storage life of mango fruits was found on treatment with thiabendazole (100 ppm) and 6% wax emulsion.

Mature green fruits (cv. Keaw Sawoey) were individually placed in sealed packages in polyethylene film (0.011 mm thick) or pvc film (0.014 mm thick) and stored at 13°C with 78% RH or under ambient conditions (28-33°C, 80% RH). The packaged fruits retained its quality for 6-9 days at ambient temperature and 30-36 days at 13°C, while unpacked fruits rapidly became shrivelled and over ripe (Sornsrivichai *et al.*1989). They noted that pvc film was more effective in delayed ripening than polyethylene film.

Wavhal and Athale (1989) reported that packing in lots of 10 in polyethylene bags (30 \times 40 cm, 100 gauge with 0.2% perforation) reduced weight loss, maintained fruit quality and prolonged shelf-life by 4-8 days, compared with non-wrapped fruits. They also reported that inclusion of a bag of vermiculite (30g) saturated with KMnO₄ in the polythene bags gave a reduction of weight loss and storage disorders.

Various wrappers such as polythene, cellophane, tissue paper and polio film were tried to prolong the storage life of mangoes (Cheema *et al.*, 1939; Mukherjee,

11

Decumentation Section Documentation Section Document No. D.-.2.9.84 Date . 1.218 (.0.9....

1956; Singh, 1960). Among these, polythene proved superior to others. It was found that during six weeks storage of mangoes wrapped in polythene bags, the fruits showed lower physiological losses and chilling injury at 7.2°C (Mukherjee, 1956 and Singh, 1960). Similarly, the efficacy of tissue paper in prolonging the storage life of mangoes was demonstrated at 15.6°C (Cheema *et al.* 1939).

Singh *et al.* (1967) found perforated polythene bags in combination with fungicidal wax coating to increase the storage life. Cellophane wrappers caused a greater retention of total soluble solids. Miller *et al.* (1986) reported that total time required for fruit to reach the soft ripe stage was reduced to 16 days by wrapping them with plastic film after the initiation of ground peel colour development, from 20 days required when wrapping mature green fruits.

Working on 'Nam Dok Mai' mango, Koolpluksee *et al.* (1993) found that storage treatments (polyethylene or polypropylene bags, perforated or none perforated, and with or without ethylene absorbent) reduced off odours, off-flavours, chilling injury and delayed ripening compared with control fruits. Longest storage life occurred in fruits kept in perforated polypropylene bags with or without ethylene absorbent (21 and 23 days respectively), while control fruits had a storage life of < 5 days.

2.3 Effect of hot water treatment

Benitez *et al.* (2006) reported that mango cv. Namdokmai fruits at maturegreen stage were heat treated by dipping in 50 or 55°C water for 5 minutes and stored at 25°C with 90-95% relative humidity. They found fruits as dipped in ambient water served as the control. Softening slowed down in response to the heat treatment. The effect of the two heat treatments did not considerably differ. Concomitant with softening, the heat-treated fruits exhibited reduced pectin methylesterase [pectinesterase] (PME) and polygalacturonase (PG) activities in both peel and pulp tissues. Heat treatment at 55°C generally resulted in lower PME and PG activities than at 50°C. These result indicated that heat effect of ripening-associated was due to inhibition of pectin-degrading enzymes.

Nair *et al.* (2001) conducted a laboratory experiment on the reduction of chilling injury and quality of mango fruits (cv. Kensington pride) using hot water and

T

-

-

+

S.

hot air treatments. They found that heat treatment did not result in a substantial reduction in chilling injury development after 21 and 35 days of storage at 5°C. All heat treatments increased respiration rate, but ethylene production, physiological weight loss during storage and total soluble solids were increased only by some heat treatments.

Baez et al. (2001) reported an experiment on postharvest performance of mango 'Tommy Atkins' using hot water and wax. They explored that Mexico is the largest mango exporter in the world, producing 40 million ton-pound boxes filled with fruits. They also found that hot water-treated fruits lost more weight than untreated fruits and the use of wax had a positive response in the hot water-treated fruits. Respiratory behavior was normal for all treatments with the climacteric peak at five days, except for hot water and without wax fruit treatments. These fruits lost more weight and did not ripen normally. This general behavior was observed in fruit stored at 12°C for 20 days and after marketing condition.

Nyanjage et al. (2001) carried out a laboratory investigation for the determination of postharvest changes in mango fruit cv. Tommy Atkins using hot water treatments and storage temperature. They found that fruit reflectance decreased, whereas chroma and hue algae increased over storage time and also with increase in storage temperature. The yellow colour increased with a rise in storage temperature in hot water treated mango. Soluble solid content of mangoes held at 22°C was highest at 5 days of storage, but decreased subsequently over storage time. Impedance (being a potential non-destructive measure of tissue damage following heat treatment) of all fruits was decreased with Increase In frequency, storage temperature and time In store. The impedance of hot water treatments (hwt) mangoes was lower than that of non-hwt fruits on 8th day after immersion, but recovered almost to control levels on day 4 to 5 or 13°C, but decreased gradually after 5 days at 13°C. Impedance of all mangoes stored at 22°C decreased continuously during storage. Impedance was higher in the inner mesocarp than outer pulp. Impedance of hwt fruits was poorly correlated with soluble solid content and chroma but well correlated with reflectance of fruit pulp at 22°C. Changes in impedance of mangoes were discussed in relation to physiological and biochemical changes that occur during heat treatment and storage.

Review of Literatures

Chapter 2

Dr.

-

AC

1

+

Y

S.

Jacobi *et al.* (2001) carried out an experiment on loss of heat tolerance in 'Kensington' mango fruits following heat treatments. They observed that the greatest reduction in heat tolerance occurred in fruit placed at 22^oC for 16 hrs. compared with the other condition of fruit. These fruit had the highest incidence and severity of skin scalding external cavities starch layer increased F0 values and a lack of recovery in FV/FM ratios after HWT. It was concluded that the loss of heat tolerance in 'Kensington' fruit occurred at a lower rate than the increase in heat tolerance brought about by the 40°C conditioning treatments. Exposure of fruit to 22°C for 24th day or longer accelerated the fruit ripening and induced some protection against heat injury.

There was a finding investigated by Jacobi *et al.* (2000) on the effect of hot air conditioned of 'Kensington' mango fruit on the response to hot water treatment. They showed hot water injuries were reduced and in some cases eliminated by conditioning the fruits at 40°C for 8 h. The conditioning temperature was more important than the duration of the HWT in injury alleviation. Conditioning at 40°C prior to hot water treatment accelerated fruit ripening, increased weight loss, reduced fruit firmness, increased degrees Brix and lowered titratable acidity compared to untreated fruits and fruits receiving other heat treatments. These treated fruits were also more resistant to postharvest diseases.

Gafur *et al.* (1997) conducted a study on extension of postharvest storage life of mango cv. Fazli and Aswina. They reported that hot water treatment $(52\pm2^{\circ}C)$ using 1% CaCl₂ for 5 min. was the most effective in retarding of fruit ripening (ripening was delayed by 5-8 days) and spoilage was also significantly reduced. The shelf life of control fruit was short; fruit exhibited a rapid rate of ripening as evident by a rapid increase in total soluble solid and by a rapid decreased in acidity and ascorbic acid content. In another experiment, Nyanjage *et al.* (1998) found that hot water treatment at 46.5°C for 45 min and then 2 days of intermittent warming resulted in a significantly low incidence of diseases and severity of external injury, softer fruits, higher brix and better general appearance. Kruger *et al.*, (1997) reported from his investigation of cv. 'Heidi' that hot water treatment at 40°C for 15 min gave best results for storage life.

P

*

+

the second

T.

Bugante *et al.*(1997) reported that hot water dipping (53°C for 5 or 10 min.) significantly decreased incidence of diseases of which 10 min dip providing the most effective control of 'Carabao' mango fruits. Preharvest bagging combined with a 10 min, hot water dip significantly decreased the incidence of disease, anthracnose was reduced by 83% and stem end rot by 100%.

A study was conducted by Jacobi *et al.*(1996) with mango fruits cv. 'Kensington' subjected to conditioning treatments with hot air (38 to 40°C) for 0, 4, 8 or 12 h before hot water treatment (40°C for 30 minutes) and storage at 22°C for 9 days. Fruits conditioned for 8 or 12 h before hot water treatment had minimal injuries. It was reported that conditioning with hot air before hot water treatment had the potential to minimize or eliminate heat injuries associated with hot water disinfestations treatment. In another experiment Jacobi and Giles (1997) treated 'Kensington' fruits following combined vapour heat disinfestations and hot water disease control treatments and found that the HW+VHT treatment combined with continuous storage at 22°C resulted in the highest quality fruits, and is recommended for air freight marketing.

Saucedo et al. (1995) carried out a laboratory trial to study the physiology and shelf life of Manila' mangoes following hot water (46.1℃ for 0,80 and 90 min) and stored at 10, 13 and 20°C. They observed the respiration rates, titratable acidity, total soluble solid, weight loss, chilling injury and anthracnose decay. Results showed that these treatments increased the yellow color and weight loss of fruits and reduced chilling injury and anthracnose decay. The best was hot water treatment (46.1°C) for 90 min. In another experiment Carrillo Lopez et al. (1995) treated mango fruits cv. 'Haden' with hot water at 46°C for 90 min and then dipped in semperfresh (0.7, 1.4 or 2.1 %). All fruits were than stored at 13°C for a month. All concentration of semperfresh used delayed ripening. Titratable acidity, firmness and green colour were higher in treated 'Haden' fruits than untreated fruits and weight loss, Brix and PH were lower. Vitamin C content was unaffected in the treated and untreated fruits. Saiman (1995) reported that there were no significant differences between infrared treatment (3 minutes exposure to short-wave) and commercially used 5 minutes hot water bath treatment in their effect on fruit quality. The infrared treatment is much cheaper and faster than hot water treatment.

×

1

+

-

S.

Hermanto and Yuaniarti (1994) showed that hot water treatment at 49°C for 10 min. or 51°C or higher for 5 min. suppressed anthracnose on mangoes inoculated with *C. gloeosporioides*. Hot water treatment at 51 or 53°C for 10 min. was most effective. There was no loss of fruit quality. Hassan *et al.* (1998) found that hot water treatment at 52± 2°C for 5 min. extended shelf life of mango by 5.34 days over control.

Upadhyay *et al.* (1994) conducted an experiment with mature green mangoes cv. 'Red' to evaluate the physicochemical behavior and shelf life indicating dose of 1 KGY after hot water at 55°C for 5 minutes. They reported that irradiation significantly reduced rotting, delayed color development, preserved quality and extended shelf life. A hot water treatment followed by irradiation at 0.3 KGY was found to be the best treatment combination. In an experiment Jacobi and Gowanlock (1995) submerged mature green fruit in hot water at 46°C until the fruit centre reached 45°C, and were than held for 30 minutes. After 7-10 days, the hot water treatment mango fruits were ripe and kept for observation of injury. They reported that the skin injury suggested the disruption of enzyme involved in carbohydrate metabolism.

In Australia, the current recommendation by Johnson *et al.* (1994) for postharvest control of anthracnose (*Colletotrichum gloeosporioides*) in mangoes is a 5- minute heating (52°C) Benomyl dip or 30 second unheated over head spray of Prochloraz. Similar results were also shown by Saaiman and Lonsdale (1994) with combination of prochloraz and imazalil or guazatile. The combination gave improved control of post harvest diseases affecting mango. Hot water (55°C for 2 min.) plus prochloraze (81.0 or 40.5 g **a.i**/100 litre) effectively controlled anthracnose and soft brown rot without being phytotoxic.

Lonsdale *et al.* (1991 a) and Lonsdale (1992 a) conducted a trial with hot water dips and various chemicals. Five minute hot water dip at 55°C plus prochloraze or 5 minute hot water dip followed by ambient temperature (\pm 25°C) dip in prochloraze were found to be phytotoxic particularly at the higher doses, causing burn on the skin surface. Lonsdale *et al.* (1991 b) and Lonsdale (1992 a) found that a mild irradiation (0.75 KGY) in combination with hot water treatment gave effective control of anthracnose and soft brown rot. Feng *et al.* (1991) reported that hot water

7

treatment (52-54°C for 8-10 min.) or treatment with thiabendazole (100 ppm) Controlled mango anthracnose (Glomerella cingulata) effectively. Esquerra and Lizada (1990) conducted a study with 'Carabao' mango fruits

subjected to vapour heat treatment for 10 min. at a pulp temperature of 46°C, and then ripen at 25°C. Vapor heat treatment enhanced peel color quality resulting in a greater portion of treated fruits reaching full yellow color. No adverse effect on the VİSUƏl quality was observed at the tableripe stage. 'Kensington' mango fruits were treated with hot water or vapour heat for 47°C for 7.5 to 30 minutes (Jacobi and Wong, 1992). The hot water treatment shortened fruit softening time and caused extensive internal and external injury. In another observation, Boonraung et al. (1993) found effective control of Phomopsis mangifera by dipping mangoes in hot water at 53°C for 10 minutes.

2.4 Effect of low temperature in refrigerator

Tefera et al. (2007) conducted an experiment with mangoes for extension of shelf life by maintaining induced temperature between 14.33 and 19.26°C and relative humidity between 70.15% and 82.4°C during storage. They found shelf life of mangoes increased from 3 to 28 days as in the evaporative cooling compared to storage at ambient condition. The storage temperature strongly influenced on all postharvest quality parameter tested in mangoes during storage. Higher temperature rapidly deteriorated the physiological and chemical quality of mangoes. Similarly, modified atmosphere packaging positively affected the physiological and chemical quality of mango during storage. It also reduced the physiological weight loss (PWL) and maintained better quality in terms of pH, ascorbic acid and marketability, compared to control during storage. The benefit of the combined effects of post harvest treatments on mangoes included reduction in PWL maintenance of better chemical quality and improvement in the shelf life of mangoes.

Durigan et al. (2004) carried out an investigation on postharvest conservation of 'Tommy Atkins' mango fruit using gamma radiation, wax, hot water and low temperature in refrigeration. They reported shelf life of fruit increased when

×.

×

-

-

S.

temperature in refrigerated storage and improved shelf life after transferring to ambient temperature.

Puttaraju and Reddy (1997) carried out an experiment with mango cv. 'Malika' to chalk out the effect of pre-cooling on the storage life. Their treatments were hydro-cooling (15 or 30 min), running water (15-30 min), room cooling (30 or 60 min.), ice cooling (15 or 30 min), evaporated cooling and control (non-cooling). Cooling under running cold (4-5°C) water for 30 min. retarded ripening significantly, extending the storage life by 3-4 days. In terms of physiological weight loss, delayed ripening and higher fruit firmness ice cooling treatment were most effective but the quality of these fruits were unacceptable due to high rate of spoilage.

Mohammed and Sankat (1995) conducted an experiment with intermittent warning and the cooling cycles of the Julie mango. The fruits were subjected to 2 cyclical cooling and warning regimes for up to 35 days, then kept at ambient temperature (28°C) to ripen. They found that fruits treated to a cycle of 12 h at 5°C followed by 12 h at 16°C showed good ripening quality when stored up to 28 days. Fruits treated to a cycle of 18 h at 5°C followed by 6 h at 16°C have good keeping quality and ripening. They could be stored up to 14 days, but for keeping of longer period reduced shelf life which might be due to chilling injury.

Castrillo and Bermudez (1992) conducted an extensive study, on post harvest ripening in wax coated mango fruits cv. Bocado. The results showed that the delayed ripening of wax coated fruits resulted in fresh weight loss, exocarp chlorophyll degradation and mesocarp pH change, but it did not significantly affect mesocarp chlorophyll, sugar and starch content.

Chattapadhyay (1989) conducted a study on mango fruit cv. 'Himsagar' following treatment with chemicals and cooling; the fruits stored in lined wooden boxes for up to 14 days. He reported that physiological weight loss and decay loss were less in fruits with cold water treatment +KMnO₄. In another study Ilangatileke *et al.*, (1989) found that mangoes precooled to 15°C and treated with a low wax concentration (<0.5%) could be kept for up to 30 days of storage without loss of quality after ripening. An experiment of hydro-cooling of mango cv 'Keitt' was conducted by Oothuyse *et al.*, (1995). Hydro-cooling for 30, 60 or 120 min at 20°C or 5°C was compared with air-cooling at 20°C for 30, 60 or 120 min. Incidence and

*

F

1

+

A.

severity of soft brown rot was reduced by hydro cooling at 5°C for 120 min with extended shelf life of mango accompanied with enhanced taste by hydro cooling at 5°C.

Thangaraj and Irulappan(1988) observed minimum weight loss (2.3%) and the maximum shelf life (9.7 days) of mango fruits, when stored in cool chamber following hot water $(52\pm1)^{0}$ C + wax coating treatments. In an experiment conducted by Wanlapha *et al.* (1980) the mango fruits were treated with hot benomyl solution and wax coated, and then stored in a refrigerator at 10^{0} C temperatures with 85% relative humidity. After three weeks of storage the fruits were transferred from the refrigerator and kept at room temperature (30^{0} C) for ripening. Maximum storage life of about one month was obtained with acceptable quality of fruit at the end of the storage period.

2.5 Effect of Maleic hydrazide (MH)

Gautam et al. (2003) carried out a laboratory trial to investigate the efficiency of postharvest treatments on the shelf life and quality of mango cv. Banganpali using maleic hydrazide (1000 ppm), NAA (500 ppm) hot water at (50±2)°C, AgNO₃ (40 ppm) Ca(NO₃)₂ 1%, wax emulsion coating (6%) storage in carbonate box containing 10 g permanganate-silica gel at room temperature, storage at 15° C with 0.1% Bavistin (carbendazim) and/or storage in zero energy cool chamber (ZECC). They found the treatments significantly retarded ripening compared to the control. Ripening retardation was maximal in fruit treated with wax emulsion coating + ZECC storage, followed by dipping in calcium nitrate +ZECC storage. Both treatments delayed ripening by more than 6 days over the control. The maximum delay in spoilage was observed in fruits treated with permanganate-silica gel +ZECC storage, followed by $AgNO_3+ZECC$ storage. Generally, ZECC storage retarded the time of fruit spoilage irrespective of the treatment combined with it. At the fourth day, the maximum fruit number ripened in the control, while the minimum percent of ripening was observed in calcium nitrate treatment, which was at par with NAA and MH+ ZECC treatments. No ripening was recorded in the following treatments as Bavistin, wax coating both under ambient and ZECC storage. Fruit subjected to various postharvest treatments (Bavistin, wax emulsion coating and NAA) and stored under either ZCC or cold storage conditions recorded lower TSS; TSS: acid ratio; reducing,

5

×

*

*

-

X.

non reducing and total sugars; higher total titratable acidity and ascorbic acid content and extended shelf life compared to the control. Hot water treatment did not exhibit any effect on the quality.

Reddy and Haripriya (2002) conducted a laboratory experiment to test the efficiency of certain treatment using fungicides (carbendazim), growth regulators (GA3, maleic hydrazide, and 2,4,5-T) and subsequent storage in polythene bags with ethylene scrubbers, or the wrapped fruits in wooden boxes on the shelf life and quality of mango cultivars Bangalora and Neelum at room temperature. Among the subjected treatments, GA3 (200 ppm) treated fruits stored in ventilated polythene bags with ethylene absorbent significantly reduced the physiological weight loss, rate of respiration, delayed color development and ripening and had longer shelf life. Besides, the fruits also exhibited better quality on account of its favourable effect on slower increase in total soluble solids, sugars and retaining more acidity thereby rendering them acceptable up to a period of 18 days in Bangalora and 17 days in Neelum. The other treatments with 1000 ppm 2, 4, 5-T and storage in similar condition as above were the best treatment.

Kumar *et al.* (1992) conducted an experiment where they dipped mature firm mango cv. 'Dashehari' fruits in aqueous solutions of 2, 4, 5-T (500, 1000 and 1500 ppm), maleic hydrazide (500, 1000 and 1500 ppm) and calcium nitrate (1.0, 1.5 and 2.0%) for 10 minutes. Then they noted that the quality of fruits improved during storage as TSS, total sugar and reducing sugar contents were increased but acidity as well as ascorbic acid decreased during storage.

Rashid and Farooqui (1984) carried out an experiment with green hard mangoes (cv. Samar Bahisht) and found that fruits dipped in 1000, 1500 and 2000 ppm solution of maleic hydrazide (MH) for 3 minutes following stored at room temperature (23-30°C) delayed ripening compared with controlled fruits.

Srivastava (1967) reported that the addition of growth regulator maleic hydrazide (MH) with wax emulsion, at the rate of 1000 to 1500 ppm reduced the ripening rate of mango. In the contrary, Joarder (1980) concluded that maleic hydrazide (MH) was useful in ripening but failed to retard spoilage in prolonged storage condition.

X

×

4

+

-

X

2.6 Effect of Gibberellic acid (GA3)

Ranjan *et al.* (2005) carried out an experiment on mangoes storage life. They observed reduced physiological weight loss and spoilage of langra mangofruits influenced by the application of calcium salts and GA3. The highest reduction of weight loss and spoilage were obtained with Ca $(NO_3)_2$ at 1 %. Gradual reduction in specific gravity of langra mango was reduced during storage although the effect of treatments was marginal. Fruits treated with 200 ppm GA3 and 1.0 % Ca $(NO_3)_2$ exhibited maximum marketability and delayed ripening of fruits.

Jain *et al.* (2001) conducted an investigation to find out the influence of postharvest treatments (tap water, air dried, bengound, wax emulsion, cool chamber, Ca $(NO_3)_2$ and GA3) and the storage condition on the quality of mango during storage. They showed that the postharvest application of wax emulsion (8%) and calcium nitrate (1%) in combination with cool chamber storage markedly reduced the rate of ripening and helps to retain the quality characteristics of fruits during storage. At the end of the storage period, minimum total soluble solid, total sugar and reducing sugar contents and maximum acidity were recorded for wax emulsion (8%) + cool chamber storage. Maximum ascorbic acid and highest organoleptic score were obtained with calcium nitrate (1%) + cool chamber storage.

Jain and Mukherjee (2001) carried out an experiment with mango cv. Langra to investigate the delay ripening using growth regulator Gibberellic acid (GA3). They recorded GA3 at the rate of 200 or 300±1 mg significantly delayed ripening of mango fruits cv. ''Langra" when stored at 36.15°C maximum and 27.08°C minimum temperature. GA3 treatment retarded the increase in TSS, total sugars, loss in ascorbic acid content and acidity and reduced the spoilage percentage in mango fruits.

Singh *et al.* (2000) conducted an experiment with postharvest mango cv. Langra to assess the quality and shelf life of mango, using different doses of Gibberellic acid (GA3) and plant extracts. They observed that physiological loss in weight an upward trend irrespective of treatments. Neem oil showed minimum physiological loss in weight (7.77%), closely followed by castor oil (8.35%) and GA3 at 1000ppm (9.87%). Fruits in the control showed maximum physiological weight loss (17.28%) on the 12th day of storage. Spoilage percentage showed an increasing

X

+

7-

S.

trend in all the treatments. None of the fruits decayed up to the 6th day of storage at room temperature. Postharvest application of Neem oil showed the minimum decay loss of 9.98% followed by caster oil (13.30%) and GA3 (13.30%). The spoilage organisms associated with decay were *Botryodiplodia theobromae* and *Aspergillus niger*. The infestations of these micro-organisms were prevented by the antiseptic and antifungal action of Neem oil. A sharp fall in firmness in all the treatments was found during the storage period. Yet, edible firmness was retained favorably in fruits treated with neem oil (3.00). Application of GA3 at 500 ppm was recorded the maximum organoleptic rating during storage. Castor oil did not consider due to the obnoxious smell and patchy appearance on the fruits. The total soluble solids increased with increasing period of storage irrespective of treatments. Acidity showed a reverse trend in all the treatments.

Mondal *et al.* (1995) conducted an experiment to study the effect of postharvest treatments on physicochemical changes and shelf life of mango. Seven postharvest treatments *viz.* control, wax coating, keeping mango in cold room and polybag, spraying mango with GA3 and fungicide and keeping mango in poly bag with KMnO₄ were used. Among the treatments, the one with wax coating was found to be the best in extending self life of mango (24.4 days) followed by KMnO₄ treatments (22.4 days). Mondal *et al.*, (1998) also found 21.00 days extension of self life of mango by wax coating in their experiment.

Singh *et al.* (1995) conducted an experiment with GA3 and Ethrel (ethapon) to enhance the ripening and improved the quality and shelf life of mango cv. 'Amrapali'. They found that ethrel at 500 ppm was very effective In enhancing the ripening and improving the quality in terms of TSS, total sugar, ascorbic acid and β -carotene content. Youlin *et al.* (1997) observed that fruit of mango cv. Zihua dipped with growth regulator, 2, 4-D, GA₃ or NAA, each together with prochloraze prolonged shelf life and improved the acceptability of the fruits.

Mendez (1994) conducted an experiment with GA3 at 250 mg/litre and found that fruit ripening was retarded by GA3 application compared with untreated fruit. Mendez (1995) reported that GA3 application was reduced soluble solids contents and percentage weights loss after storage and resulted in higher colour ratings, when the mango trees were sprayed with GA3 2000 ppm 30 days before harvest and

*

N.

-

X

st.

X

the harvested fruits were stored at 12 or 25°C for 25 days. While working on mango fruits cv. 'Dashehari', Khader (1989) found maximum delaying in ripening (3 days) over the control and prolonged shelf life with GA_3 (200 mg/L).

Mangoes cv. 'Violet Flower' were treated with gibberellic acid for 10 minutes and then stored at 20-25°C. These delayed ripening and peel colour development. Furthermore, it significantly reduced weight loss during storage as described by Guoqiang *et al.* (1994).

Singh *et al.* (1995) conducted an experiment with GA3 and Ethrel (ethaphon) to enhance the ripening and improved the quality and shelf life of mango cv. 'Amrapali'. They found that ethrel at 500 ppm was very effective in enhancing the ripening and improving the quality in terms of TSS, total sugar, ascorbic acid and β -carotene content. Youlin *et al.* (1997) observed that fruit of mango cv. Zihua dipped with growth regulator, 2, 4-D, GA3 or NAA, each together with prochloraze prolonged shelf life and improved the acceptability of the fruits.

Kumar and Singh (1993) observed that postharvest spray of GA_3 (50 or 75 ppm) or ethrel (500 ppm) was hastened fruit maturity by 8-11 days and ripening significantly improved fruit quality (TSS, sugar, ascorbic acid and β -carotene concentrations) and reduced spoilage loss during storage.

Khumlert (1992) carried out an experiment to study some physico-chemical changes and physiological characteristics of 'Kheaw Sawoey' mango fruit as influenced by GA3 application. Results showed that GA3 (200 ppm) solution delayed ripening of mango via inhibition of some ethylene mediated ripening changes. The colour of peel and pulp, pulp firmness, soluble solids, titratable acidity, conversion of starch to sugar, solubilization of pectic substances was also delayed. Among the fruit characters studied the delayed in peel colour development was the most prominent for the effect of GA3, both in mature and immature fruits.

Khader (1992) conducted an experiment on ripening and quality of fruits cv. 'Mallika' as influenced by GA₃ and vapour Gard (di-l-p-methene). The fruits were treated with 200mg/L GA₃ or 2.5% vapour Gard or both and stored at an ambient temperature (34-37°C) for 16 days or at 15°C for 20 days. He found that GA₃ alone

Review of Literatures

Chapter 2

7

4

4

E

or with vapour Gard significantly delayed fruit ripening, retarded ascorbic acid degradation in pulp and chlorophyll degradation in the peel, reduced ∞ -amylase and peroxidase activities during storage. He further observed that fruits dipping in vapour Gard alone or with GA₃ reduced percentage weight loss during storage at both temperatures. Dipping fruits in vapour Gard alone or with GA₃ markedly delayed ripening in fruits stored at 15[°] C. Finally he concluded that mango could be successfully stored for up to 20 days at 15[°] C after treating with 200 mg/L. GA₃ +2.5% vapour Gard.

Khader (1991) reported that, GA3 application as a foliar spray to the mango variety 'Dashehari' at concentrations of 100, 200, 300 or 400 mg/L after fruit set in 1988 and 1989 retarded the ripening of fruit up to 6 days of storage under ambient temperatures between $36 \pm 2^{\circ}$ C and $40 \pm 3^{\circ}$ C. Contrastly, with increasing GA3 concentrations, postharvest ripening during the first 6 days was delayed significantly. Kumar and Dhawan (1995) observed that ethrel (5000 ppm) exhibited a slow rate of ripening compared with controlled fruit.

Feungchan (1992) found that the gibberellic acid (GA3) at 5-200 ppm did not affect the development of chlorophyll in mango fruits. However, 100 ppm of GA retarded fruit ripening by up to 73% Sheikh *et al.* (1992) described that storage life of mango could be extended by postharvest application of cycocel (500 ppm) and 2,4-D (50 ppm) and reduced weight loss, skin and pulp colour development and wrinkling throughout the storage period. However, fruits of these treatments showed less organoleptic values during initial storage period as compared with other treatments.

Mature green 'Kesar' mango fruits were dipped in solutions of gibberellic acid 200 ppm, cycocel 500 ppm and Bavistin 1000 ppm in combination with growth regulators. The fruits were stored at ambient temperature for 10-27 days and fruit ripeness, % marketable fruit, weight loss, total soluble solids, sugar and titratable acidity and vitamin C were assessed. Bavistin (1000 ppm) + GA (150 ppm) delayed ripening by about 2 days and gave the highest % marketable fruit after storage (Parmer and Chundawat, 1989).

3L

4

-

R

Murthy and Rao (1982) reported that the ripening of mango fruits cv. 'Alphonso' can be significantly retarded by postharvest treatments with GA3 (250 mg/L), cycocel, alar and menadione bi-sulphite, as judged by the number of ripe fruits during storage at 28° C. They also observed that cycocel treatment retarded ripening in the early part of storage.

Mango fruits ripening could be controlled through the use of chemical, the value for the Brix/acid ratio was low and firmness was high in fruits treated with gibberellic acid and menadione sulphite. While dipping of mango fruits in GA3 (10^{-6} M), indole-3-acetic acid (IAA) (10^{-6} M) or Kinetin (10^{-5} M) solutions delayed ripening (Majumder *et al.*, 1981).

2.7 Effect of Bavistin DF (BDF)

A research work was conducted by Dhemre and Waskar (2004) for the assessment of postharvest losses, shelf life and quality of Kesar mango fruit. They showed the content of TSS, sugar, (total, reducing and non reducing), physiological weight loss (PWL) and rotting percentage increased with the increase in storage period. Fruits treated with waxol + carbendazim, pre-cooled then subjected to cold storage recorded the longest shelf life (50 days) compared to fruits that were not pre-cooled (46 days), and fruits stored in a cool chamber (26 days) or at room temperature (20 days). The rate of increase in PWL and rotting was also slowest in fruits treated with waxol + carbendazim. The organoleptic rating in terms of color, flavor and texture was highest for fruits treated with waxol + carbendazim than stored at room temperature caused by *Aspergillus niger*

Ahmed and Singh (2000) reported a trial in a laboratory for extension storage life of post harvest mango, using GA3, bavistin (carbendazim) and 200 gauge perforated polyethylene (20% vent) bags. They found that the PWL progressively increased as the storage period advanced on the eleventh day of storage, PWL was maximum (18.53%) under the control and minimum (7.48%) GA3+ polythene bag treatment. Higher percentage of TSS, sugar (total, reducing and non reducing) and reduced level of titratable acidity, vitamin C were recorded in all the treatments up to the last day of storage. Fruits treated with GA3+ perforated polythene bag had

X

4

À

X

the lowest acidity (39.84%), highest in TSS (22.15%) and pulp pH. The highest TSS/acid ratio was observed in the control (113.67%) where as GA3+ perforated polythene bag showed the lowest ratio (69.46%).

Wasker and Masalkar (1997) conducted an experiment of hydro-cooling and Bavistin dip with mango fruit 'Kesar', 'Totapuri' and 'Vanraj'. The fruits were harvested with stalks (25 cm) hydrocooled at 12°C, given a postharvest dip of Carbendazim (as Bavistin, 1000 ppm) and were then stored at room temperature (23.5-35.1°C, 47.5-80.3% RH) or in lower cool chamber (22.0-26.1°C, 92-95% RH). The shelf life of 'kesar,' 'Totapuri' and 'Vanraj' were 17, 21 and 19 days respectively when stored at room temperature and in cool chamber 25, 36 and 31 days respectively. Percentage weight loss was lower and organoleptic scores were higher in fruits stored in the cool chamber than in fruits stored at room temperature. The fruits which were not dipped in Carbendazim became infected with Colletotrchum gloeosporioides and Diplodia natalensis. In another experiment Tang et al. (1997) reported that dipping the harvested fruits in the growth regulators 2, 4-D, GA3 or NAA each together with prochloraz increased the efficiency of the fungicides against anthracnose and stem end rot, with increasing storage life of the fruits. Gajbhiye et al. (2000) also found an effective control against stem end rot and anthracnose diseases when the harvested mango fruits treated with 0.1% Carbendazim up to 10 days.

2.8 Changes in skin color

Morphological features, especially skin color is the most obvious change that occurs in many fruits and is frequent the principal criterion followed by the consumers to identify whether the fruit is ripe or unripe. The most common change in case of mango is the loss of green color (Wills *et al.*, 1989). Hayes (1966) stated that the dark green color of unripe mango gradually changes to light green, and then yellow thereafter, in some cases a deep orange. He also stated that a greenish yellow color might be taken to indicate a stage of maturity that will yield a fruit of satisfactory quality.

2.9 Physical changes during storage

2.9.1 Edible portion

There was no available information's in dealing with edible portion of mango. But, Samson (1986) described the edible portion of mango as 60–75 percent of the total weight. Mondal (2000) also stated that edible portion of Langra and Khirshapat was approximately 73.1% and 67.2%, respectively.

2.9.2 Pulp to peel ratio

Little information is available dealing with the changes in pulp peel ratio during storage and ripening of mango though this is an important physical property of any fruit. However, Tripathi *et al.* (1981) conducted an experiment with banana and observed that, pulp to peel ratio was increased during ripening.

2.9.3 Physiological weight loss

Weight loss reduced when mango fruits were treated with wax as reported by Zambrano *et al.* (1997). Manzano *et al.* (1997) reported that 6.2 percent fresh weight loss of mango fruits occurred when stored at 25° C for 20 days. According to Martinez *et al.* (1997) application of wax coating on postharvest mango fruits significantly reduced weight loss by inhibiting transpiration.

2.9.4 Moisture content

From an observation, Srivastava (1967) found that unripe mango fruit contained higher percentage of moisture. Shahiahan *et al.* (1994) reported that moisture content of mature hard mango pulp was 79.95% which Salunkhe and Desai (1984) observed that mango pulp contained 81% moisture.

2.9.5 Dry matter content

-

A

Scientific information about the changes in dry matter content during storage and ripening of mango is not available. However, Paramanik (1995) found that dry matter content in Fazli increased from 17.14 to 28.86% during storage in ambient temperature.

×

Y

-in

2.9.6 Ash content

Srivastava (1967) reported that ash content of mango fruits ranges from 0.26 to 1.66 percent. It shows the presence of calcium, magnesium, copper, manganese, silicon, iron and phosphorus in the edible part of mango. On the other hand, Singh (1968) observed that ash content of mango fruits depend on varieties and he reported that it was 0.26 to 1.16 percent. Ripe mango is a source of calcium, phosphorus and iron (Salunkhe and Desai, 1984).

2.10 Biochemical changes during storage

2.10.1 Vitamin C content

Mango is a good source of vitamin C at early stages of development, which decreases rapidly 5-7 weeks after fruit set as reported by Gofur *et al.* (1994). They also reported that, at 12 weeks after fruit set ascorbic acid content was 105.2, 65.7 and 17.3 mg/100 g in Langra, Ashwini and Fazli varieties, respectively.

The green fruits stored at 10-12° C for 7 weeks had little change in vitamin C content. Maximum portion of vitamin C was lost when the fruits were stored at room temperature (26-30° C). In addition reduction in vitamin C with progress of fruit maturity and ripening was found in cv. Gopalbhog, Khirshapat, Langra and Fazli as described by Shahjahan *et al.* (1994).

Tripathi (1988) noticed that fruit quality and flavour were best at the 6th day of storage. He noted that the percentage of ascorbic acid was also highest at the same day. During transport and storage (at 10°C for up to 32 days) Joshi and Roy (1988) reported a continuous decrease in vitamin C.

2.10.2 Titratable acidity

Titratable acidity was decreased during storage and ripening as reported by Upadhyay and Tripathi (1985). Leon and Lima (1968) and Medlicott *et al.* (1986) also observed similar results. According to them acidity was reduced during later stage of growth on attainment of maturity and ripening.

Shahjahan *et al.* (1994) revealed that acidity of mango was decreased gradually at the time of storage and ripening. They also found that the percent acidity in green and ripe mango cv. 'Fazli' was 0.32 and 0.10, respectively. Acidity was decreased continuously during storage (Joshi and Roy, 1988). Dhalla and Hanson (1988) conducted an experiment to study the effect of prolong treatments on the storage life of mango and found that decrease of titratable acidity during storage was delayed by the treatments. However, titratable acidity was decreased during post harvest period (Rangavalli *et al.* 1993).

Kumar *et al.* (1993) observed that acid concentration of fruits was fallen in the storage. They also claimed that Kinetin application in mango resulted in a drastic decline in the concentration of all acids.

There were no differences in titratable acidity when fruits stored in cold condition at different temperatures as described by Lam and Wong (1988). In another study, O'Hare (1995) reported that titratable acidity was declined slowly when mango fruits were stored at 13° C.

2.10.3 Pulp pH

X

-h

Yuniarti (1980) carried out an experiment in an attempt to investigate pH content of mango fruits during storage and reported that the pH content of 'Arumanis' mango remained in an increasing fashion during storage. Joshi and Roy (1988) reported that there was a steady rise in pH of the fruits of 'Alphonso' mango during storage. The pulp pH of ripe Fazli was 4.64 as observed by Kumar *et al.* (1993). They also found that pH of mango pulp was increased during storage. Salles and Tavares (1999) reported that pH of mango pulp increased during storage in a refrigerated room. The pH of hot water treated fruits was higher than non hot water treated fruits (Zhu *et al.*, 2002)

2.10.4 Total soluble solids (Brix %)

Total soluble solids (Brix %) varied from variety to variety (Jagirdar and Maniyar, 1960; Mollah and Siddique, 1973). They recorded that TSS of mango varieties Fazli and Langra were 7.70 to 14.8% and 12.15 to 18.0% respectively. Hossain and Ahmed (1994) reported 18.3% TSS content in cv. Aswina. Shahjahan *et*

-ster-

×-

Y-

SK.

the.

X

al. (1994) reported from their experiment that TSS content was lower in early harvest and higher in later stage (mature stage) of harvesting of ripening fruits in all varieties. Similar result was also obtained by Lakshminarayana (1975) with Florida mangoes. An assessment made by Simao and Gomes (1996) on the Brix in different parts of the fruit (dorsal, ventral and middle parts) of 10 mango varieties indicated that the dorsal part was sweeter than the other parts.

Absar *et al.* (1993) reported that TSS content increased markedly with advances in maturity. They reported that at the ripened stage Langra showed the highest TSS (22.2%) and Fonia the lowest (16.8%). Tripathi (1988) reported highest percentage of TSS in cv. Gourjeet mango fruit on the 6th day of storage. In another observation, Joshi and Roy (1988) reported that TSS increased initially and declined later on during storage. In an experiment conducted by McCollum *et al.* (1993) in order to study the chilling injury and quality attributes as influenced by heat treatment, fruits of cv. 'Keitt' were kept at 38°C temperature for 0, 24 and 48 hrs before storage and cold temperature at 5°C for 11 days. They reported that TSS was higher in the fruits stored at the higher temperature. In an observation, Upadhyay and Tripathi (1985) reported that TSS increased gradually with ripening in cv. Gourjeet. Tripathi (1988), Ashwani and Dhawan (1995) and Mondal *et al.* (1995) also reported similar findings.

2.10.5 Total sugar content

Tandon and Kalra (1983) claimed an increased in total sugar content after 72 days from fruit set. While working with mango fruits Upadhayay and Tripathi (1985) found that total sugar content were increased gradually, when stored for 6 days at room temperature (32-35°C). The total sugar, sucrose and fructose contents of fruits were increased during storage as reported by Tsuda *et al.* (1999). However, maximum sugar content was observed in mango at full ripe stage as reported by Takuji *et al.* (1997).

Shahjahan *et al.* (1994) carried out an experiment with Gopalbhog, Khirshapat, Langra and Fazll. They found that total sugar contents of these fruits were increased at later stage of harvesting and this sugar content was further increased during storage. Tandon *et al.* (1985) reported that total sugar content was

¥.

1

X

-

X

initially decreased in 'Mallika' and it increased until maturity, whereas in 'Langra' total sugar content was increased throughout the growing period. During transport and storage of mango, a rise followed by decline in total sugar content was found by Joshi and Roy (1988). Total sugar content of mango fruits was the highest on the 6th day of storage (Tripathi, 1988).

2.10.6 Reducing sugar content

Kalra and Tandon (1983) conducted an experiment with mango fruit cv. 'Dashehari' harvested after 85, 90 or 95 days of fruit set and then stored at ambient temperature (maximum / minimum, 37.5/31.5°C) with 39-91% relative humidity. The results showed that the reducing sugar content was higher at each successive harvesting stage. However, reducing sugar content of mango was found to be increased markedly up to 5 days after harvest (Lianni *et al.*, 1994). On the other hand, Tandon *et al.* (1985) found increased reducing sugar content of fruits until harvest.

Tripathi (1988) reported that fully mature fruits stored at room temperature for up to 6 days contained highest reducing sugar percentage on the 6th day of storage. Upadhaya and Tripathi (1985) concluded that reducing sugar content was increased gradually with fruit ripening. In an experiment with mango fruit cv. 'Arumanis' harvested 90 days after anthesis and stored at ambient temperature (26°C and high relative humidity) for 14 days, Yuniarti (1980) noticed that there was no significant differences in sucrose or reducing sugar content. On the contrary, Rangavalli *et al.* (1993) observed that reducing sugar content was gradually increased and reached the highest peak of 7.5 percent. Castrillo *et al.* (1992) also observed that reducing sugar was increased slightly during ripening.

2.10.7 Non-reducing sugar content

Post harvest changes in mango were studied by Rangavalli *et al.* (1993), and found a gradual increase in non-reducing sugar content. Sarker and Mushi (1978) conducted an experiment to investigate on non-reducing sugar content at various stages of fruit development and ripening. They noticed that percent of non-reducing sugar was low at the bud stage. But in case of 'Fazli' this amount started to increase

ji-

×

X

-te

X

gradually followed by a lag period at the stone formation stage. However, nonreducing sugar content reached at the peak of 17.5 percent at ripening stage. They further observed that at the fully ripe stage, the local variety contained less than one third non-reducing sugars as compared with the Fazli and Gopalbhog. On the other hand, ripe mango contained higher percentage of disaccharide (sucrose) and a smaller amount of aldohexoses (glucose and its isomers).

Non-reducing sugar after attaining a peak remains more or less constant (Joshi and Roy, 1988). Tandon *et al.* (1985) reported that fructose content was increased during ripening. Yuniarti (1980) reported that there were no significant differences in non-reducing sugar content in mango fruit when stored at ambient temperature (+ 26°C and high RH) for 14 days.

2.10.8 Crude fibre content

Crude fibre demonstrated a very slight change during ripening that could be attributable to a decrease in insoluble pectin associated with an increase in soluble pectin in the course of ripening as reported by Mathooko (2000). He also stated that crude fibre was perceived in a small quantity but, it significantly decreased during ripening.

2.10.9 Total lipid content

Biochemical composition as fat of mango variety might be differed with the cultivar and the stage of maturity and ripening as narrated by Anon, 1962. Bandyopadhyay and Nair (1990) stated the changes of fatty acid composition of mango pulp, aroma and flavor of gamma irradiated (0.25kGy) and control mature green Alphonso mangoes during ripening at 25–30°C. Ripening of both control and irradiated mangoes was accompanied by changes in glycerides as well as fatty acids.

Crude fat contents showed a slight increase as ripening progressed. The observed slight increase in fat concentration agreed with the general observation that a link exists between lipid content, color and flavor development of the mango during ripening (Gomez-Lim, 1997). A slow change in fat content on extended storage could be due to decreased citrate level, which is believed to be the

d

X

-A

immediate source of acetyl coenzyme A required for biosynthesis of fatty acid and triglyceride (Gomez-Lim, 1997).

2.10.10 Water soluble protein content

Watt and Merril (1963) reported that mature green mango contained 0.7% of protein, but 0.5 to 1.0% of water soluble protein was recorded from Lakshminarayana (1980). Water soluble protein was differed with the cultivar and stage maturity and as well as ripening as described by Anon (1962).

An increase in protein content observed during ripening is in agreement with the findings made in several fruits by Gomez-Lim (1997). He also reported an increase in the amounts of some proteins and enzymes. Mathooko (2000) described a dramatic increase in protein, reflecting the enzyme required for ripening. Peter *et al.* (2007) recommended that protein content increased with the advances of storage duration

2.11 Mineral changes during storage

Peter *et al.* (2007) showed that light changes increasingly in mineral levels (Mg, K, Na, P, and Fe) were obtained on ripening but, there were no significant variations (P > 0.05) in terms of mineral levels among treatments. But, Ca content slightly diminished with the extending of storage period.

Nadkarni (1963) elucidated that iron, calcium and phosphorus contents in 16 cultivars ranged between 0.90 to 3.2 mg, 10 to 20 as well as 10 to 30 mg per 100 g, respectively, but, these minerals components were changed during storage period. Watt and Merrill (1963) reported that ripe mango contained 189 mg per 100 g of fresh pulp.

2.12 Shelf life

The extended shelf life of mango was perceived to be increased during storage period using different postharvest technology as noted by Hossain (1999), Azad (2001) and Barua (2003).



and .

5

¥

1

X

-A

¥

CHAPTER 3 MATERIALS AND METHODS

Altogether four experiments were conducted to achieve the distinct goals and objectives of the research work. The materials used and methodology employed during the period of this investigation are presented in this section.

3.1 Location and time

a

1

X

-the

The investigation including four different experiments was conducted in Protein and Enzyme laboratory of the Department of Biochemistry and Molecular Biology, Rajshahi University, Rajshahi, Soil Resources Development Institute (SRDI), Rajshahi and Bangladesh Council of Scientific and Industrial Research, BCSIR Laboratory, Rajshahi, Bangladesh during the period from January, 2005 to January, 2008.

3.2 Weather data

The maximum and minimum temperature (0 C) both inside and outside of the laboratory experienced and also relative humidity(%) and rainfall (mm) were recorded during the period of investigation (Appendix 1.1)

3.3 Experimental materials

Two mango varieties namely, Langra and Khirshapath were selected as experimental materials. The mango varieties that undertaken for investigation were collected from mango grower of Kansart, Shibgonj Upazila of ChapaiNowabgonj district and Chirghat upazila of Rajshahi district and others material used as postharvest treatments *viz.*, paraffin coating, polyethylene cover, Maleic hydrazide(MH), Gibberellic acid (GA3) and Bavistin DF (BDF) were collected from Rajshahi City market, Rajshahi by requisition. A brief description of above two varieties under study is given in the followings.

Mango varieties

i) Langra

Langra is one of the most popular mango varieties amongst the varieties of the best quality in Bangladesh. The skin color of the variety is light green to light yellow at ripening stage. Mesocarp is yellowish, aromatic, tasteful, sweetest and
Sh.

5

4

X

J.

fibreless. Peel color is light green and stone is small in size. It is on an average of 9.7 cm in length, 7.3 cm in width and 5.2 cm in high as well as average weight of 314.10 gm. The sweetest of this variety is about 19.7%. Edible portion of this type is approximately 73.1%. It is a seasonal fruit. Ripening stages of this variety is the last of June and available in the market in the month of July. Average yield is good (Plate 1).

ii) Khirshapat

Khirshapat is a popular variety produced early in Bangladesh. It is a medium and oval shaped variety having 8.6 cm long, 7.5 cm wide and 6 cm high on an average. The skin color is light yellow and pulp is yellowish. It is very sweet, succulent, attractive flavor and fibreless. The sweetest of this type is approximate 18.4%. This fruit is somewhat wide and hard like and stone is lighter. Edible portion of this variety is 67.2%. Stage of ripening starts from first week of June. Overall, yield is good but irregular (Plate 2)



Plate 1 Photograph of Langra before using postharvest treatments

G.

the second

A

I.



Plate 2 Photograph of Khirshapat before using postharvest treatments

4. Experiments and experimental design

To develop a new approach to increase shelf life, the following experiments on two mango varieties have been adopted in this investigation.

4.1 Experiment 1

Influence of different storage environments on physicochemical changes and shelf life of postharvest mango

4.1.1 Experimental design

The experiment consisted of two factors and was conducted in Randomized Complete Block Design (RCBD) with three replicates.

4.1.2 Treatments

Factor A: Varieties of mango (V) as identified alphabetically below.

Alphabetical symbol	Varieties
V1	Langra
V ₂	Khirshapat

a

I

Ŧ

X

I

Materials and Methods

Factor B: Postharvest storage treatments (T) as identified alphabetically below:

Alphabetical symbol	Treatments
Τo	Control
T ₁	Paraffin coating
T ₂	Perforated polyethylene cover
T ₃	Unperforated polyethylene cover
T₄	Hot water treatments (55±2) ^o C
Ts	Low temperature in refrigerator (4±1) ^o C

4.1.3 Experimental layout

There were thirty six treatment combinations including three replications. Each replicate contained twelve treatment combinations.

	Postharvest storage treatments						
Varieties	Control (T ₀)	Paraffin coating (T ₁)	Perforated polythene cover (T ₂)	Unperforated polythene cover (T ₃)	Hot water treatment (T4)	Refrigerate d condition (T ₅)	
Langra (V ₁)	V ₁ T ₀	V ₁ T ₁	V ₁ T ₂	V ₁ T ₃	V ₁ T ₄	V ₁ T ₅	
Khirshapat (V ₂)	V2T0	V ₂ T ₁	V ₂ T ₂	V ₂ T ₃	V2T4	V ₂ T ₅	

Layout R₁

 R_2

R₃

V ₂ T ₃	V ₁ T ₀	V ₁ T ₅	V ₂ T ₃	V ₁ T ₀	V_1T_3
V ₂ T ₁	V ₁ T ₄	V ₂ T ₀	V ₁ T ₃	V ₁ T ₂	V ₂ T ₃
V ₁ T ₅	V ₁ T ₁	V ₂ T ₂	V ₂ T ₁	V ₂ T ₀	V ₁ T ₄
V ₁ T ₂	V ₁ T ₃	V ₂ T ₄	V ₁ T ₂	V ₂ T ₂	V ₂ T ₄
V ₂ T ₄	V ₂ T ₀	V ₁ T ₄	V ₁ T ₁	V ₁ T ₁	V ₁ T ₅
V ₂ T ₅	V ₂ T ₂	V ₁ T ₀	V ₂ T ₅	V ₂ T ₁	V ₂ T ₅

A

à.

F

×

I

14-

4.1.4 Data collection

Data were collected at three days interval from the starting day of storage (i.e.; 0, 3, 6, 9, 12, 15 etc.) still over ripening.

4.1.5. Methodology

The experiment was laid out in a Randomized Complete Block Design with three replicates. The post harvest treated fruits were assigned at random in each replication. The required numbers of unblemished physically similar, more or less uniform size, shape and color fruits for the experiment were harvested manually from each plant of the varieties, Langra and Khirshapath. The fruits were carefully selected during harvest. The skin of fruits was cleared with the help of a cloth just after harvesting.

4.1.6 Application of storage treatments

The storage treatments used in the experiment were sequentially assigned to the collected fruits. After the application of treatments, the fruits were kept on a brown paper, which was previously laid out in Randomized Complete Block Design and placed on the laboratory floor at ambient condition. Each of the blocks consisted of the experimental type. For each treatment combination of replication, there were six fruits, of which one was kept to record shelf life, changes in weight, color and other external fruit characteristics. The remaining five fruits were preserved in a deep refrigerator (-85⁰ C) at Protein and Enzyme Laboratory in the Department of Biochemistry and Molecular Biology, University of Rajshahi, for recording the data periodically at five different dates (at 3 days interval). Five fruits from each treatment combination of every replicate were chemically analyzed for the determination of the changes in edible portion, juice content of fruits, vitamin C content, total titratable acidity, pulp pH, total soluble solid (TSS), sugar content, crude fibre, total lipid, protein, different minerals etc. To ensure the application of the different storage treatment of the fruits for each variety, the following procedure was accomplished.

I

1

A

-

X

E

+

4.1.7 Preparation of postharvest storage treatments

4.1.7.1 Control (T_o)

Eighteen mangoes of each variety were selected at random from a mango basket and the fruits were kept on a brown paper at ambient condition arranging at random by application.

4.1.7.2 Paraffin coating (T₁)

Solid paraffin wax was made liquid by heating it with the help of an electric heater in a large aluminum pot. Eighteen fruits from both the varieties were taken for wax coating individually. The fruits were treated as per treatment and dipped in it quickly. Care was taken to ensure a uniform coating on all the fruits and kept on brown paper for observation (Plate 4).

4.1.7.3 Perforated polyethylene covers (T₂)

The 19 cm x 15 cm sized polyethylene cover was taken at laboratory room and these were perforated with a scissor in nine equal positions of the bags. Individual fruit was taken into polyethylene cover and the open portion of the bag was sealed with burning. Eighteen fruits of each variety were bagged in this system and then placed on brown paper for next observation.

4.1.7.4 Unperforated polythene covers (T₃)

Unperforated 19 cm x 15 cm sized polyethylene cover was used for this treatment. An individual fruit was taken into polyethylene cover. Eighteen fruits of each variety were kept into it, and the top of the bag was tied by burning and then placed on brown paper for observation.

4.1.7.5 Hot water treatment (T₄)

Tap water was heated in a hot water bath at $(55 \pm 2)^{0}$ C. Fruits were individually dipped into hot water for a period of 5 minutes and then stored at ambient condition on brown paper (Plate 4).

4.1.7.6 Low temperature in refrigerator (T₅)

Fruits were stored in a refrigerated incubator at (4 ± 1) °C. The temperature of the refrigerated incubator was maintained by adjusting the button on the incubator.

4.2 Experiment 2

Influence of different doses of Maleic hydrazide on physicochemical changes and shelf life of mango during storage E

A

The second

1

T



Plate 3 Photograph of Langra ang Khirshapat in basket after collection from farmer's orchard



Plate 4 Photograph of the application of different postharvest treatments especially, paraffin coating and hot water treatment on mangoes

C.

9

F

I

1

4.2.1 Experimental design

Two factors experiment was conducted in Randomized Complete Block Design (RCBD) with three replicates

4.2.2 Treatments

Factor A: Varieties of mango (V) as in experiment 1

Factor B: Maleic hydrazide used as postharvest hormonal treatment (MH)

Alphabetical symbol	Treatments (ppm)
Mo	Control
М 1	200
M 2	400
М 3	600

4.2.3 Experimental layout

 R_1

There were twenty four treatment combinations including three replications. Each replicate contained eight treatment combinations

Maleic hydrazide (MH) solution Varieties 200 ppm (M1) Control (M₀) 400 ppm (M₂) 600 ppm (M₃) Langra (V₁) V₁M₃ V1Mo V1M 1 V1M2 Khirshapat (V₂) V2Mo V2M 2 V_2M_3 M_1V_2

Layout

R₂

 R_3

Į	V ₁ M ₀	V ₂ M ₁		$V_1 M_0$	V ₂ M ₃	V2M1	V_M _
	V2M3	V2M ()		V ₁ M ₃	V ₂ M ₀	V ₁ M ₂	V _I M ₀
ſ	V2M2	V ₁ M ₂		V2 M1	V ₁ M ₁	V ₂ M ₃	V ₁ M ₃
	V ₁ M ₁	V ₁ M ₃	4	V ₂ M ₂	V ₁ M ₂	V ₂ M ₀	V _I M _I

4.2.4 Data collection as in experiment 1

4.2.5 Methodology as in experiment 1

4.2.6 Application of storage treatment as in experiment 1

4.2.7 Preparation of Maleic hydrazide (MH) solution

Materials and Methods

Chapter 3

The solution of Maleic hydrazide (MH) of 200, 400 and 600 ppm were prepared by dissolving 200, 400 and 600 mg of MH in 1 litre of distilled water. For successfully dissolvent of MH, 10 ml of ethanol was taken in a conical flask and thereafter, the required amount of MH was added and mixed with it thoroughly by shaking. Finally, the volume of the solution was made upto 1 litre. The fruits of both varieties were dipped into the solution for a period of 5 minutes. Care was taken to ensure sufficient absorption of MH by the fruits and then they were stored at room temperature on brown paper.

4.3 Experiment 3

Effect of different doses of Gibberellic acid on physicochemical behavior and shelf life of postharvest mango

4.3.1 Experimental design

Two factors experiment was carried out in Randomized Complete Block Design (RCBD) with three replications

4.3.2 Treatments

×

R

£

Factor A: Varieties of mango (V) as in experiment 1

Factor B: Gibberrelic acid used as a postharvest hormonal treatment (GA3)

Alphabetical symbol	Treatments (ppm)
Go	Control
G1	100
G2	200
G3	400

4.3.3 Experimental layout

There were twenty four treatment combinations including three replications. Each replicate contained eight treatment combinations

		Postharvest horm	onal treatment (GA	3)
Varietles	Control (G ₀)	100 ppm(G ₁)	200 ppm(G ₂)	400 ppm (G ₃)
Langra (V ₁)	V1Go	V ₁ G ₁	V ₁ G ₂	V ₁ G ₃
Khirshapat (V ₂)	V ₂ G _o	V ₂ G ₁	V ₂ G ₂	V ₂ G ₃

×

1

X

×

Materials and Methods

out	R ₁	F	R ₂	R	3
V ₁ G ₃	V ₂ G ₃	V ₂ G ₃	V ₁ G ₃	V ₁ G ₀	V ₂ G ₃
V₂G₀	V ₂ G ₁	V ₂ G ₂	V ₁ G ₂	V ₁ G ₁	V ₂ G ₂
V1G2	V ₂ G ₂	V ₂ G ₁	V ₁ G _o	V ₁ G ₂	V ₂ G ₁
V ₁ G ₁	V ₁ G _o	V ₁ G ₁	V ₂ G _o	V ₁ G ₃	V₂G₀

4.3.4 Data collection as in Experiment 1

4.3.5 Methodology as in Experiment 1

4.3.6 Application of storage treatments as in experiment 1

4.3.7 Preparation of Gibberellic acid (GA3) solution

The solution of GA3 of 100, 200, and 400 ppm was prepared by dissolving 100, 200, and 400 mg of GA3 in one litre of distilled water. The fruits of both varieties were dipped into the solution for a period of 5 minutes. Care was taken to ensure sufficient absorption of GA3 by the fruits and then they were stored at room temperature on brown paper.

4.4 Experiment 4

Influence of different doses of Bavistin DF on physiochemical properties and shelf life of mango varieties during storage

4.4.1 Experimental design

The experiment consisted of two factors and was conducted in Randomized Complete Block Design (RCBD) with three replicates

4.4.2 Treatments

Factor A: Varieties of mango (V) as in experiment 1

Factor B: Bavistin DF used as a postharvest fungicidal treatment (BDF)

Alphabetical symbol	Treatments (ppm)
Bo	Control
B1	250
B ₂	500
B ₃	750

×

1

T

t

2

4.4.3 Experimental layout

There were twenty four treatment combinations including three replications. Each replicate contained eight treatment combinations

	Bavistin D	F as a posthar	vest fungicidal	treatment
Varieties	Control (B ₀)	250 (B ₁)	500 (B ₂)	750 (B ₃)
Langra (V1)	V ₁ B _o	V ₁ B ₁	V ₁ B ₂	V ₁ B ₃
Khirshapat (V ₂)	V ₂ B _o	V ₂ B ₁	V ₂ B ₂	V ₂ B ₃

Layout R₁

R₂

 R_3

V ₁ B _o	V ₁ B ₂	V ₂ B ₂	V ₁ B ₃	V ₁ B ₁	V ₁ B ₂
V ₁ B ₁	V ₂ B _o	V ₁ B ₁	V ₂ B ₁	V ₂ B _o	V ₂ B ₂
V ₂ B ₃	V ₁ B ₃	V ₁ B ₂	V ₁ B _o	V ₁ B ₃	V ₁ B _o
V ₂ B ₁	V ₂ B ₂	V ₂ B ₃	V ₂ B _o	V ₂ B ₃	V ₂ B ₁

4.4.4 Data collection as in experiment 1

4.4.5 Methodology as in experiment 1

4.4.6 Application of storage treatment as in experiment 1

4.4.7 Preparation of Bavistin DF (BDF) solution

The solution of BDF of 250, 500, and 750 ppm were prepared by dissolving 250, 500 and 750 mg of BDF in one litre of distilled water. The fruit of both the varieties were dipped into the BDF solution for a period of 5 minutes. Care was taken to ensure enough quantity of BDF being absorbed by the fruits and stored at ambient condition on brown paper.

4.5 Organoleptic evaluation

Organoleptic evaluation of fresh mango pulp at full ripe stage during storage was made by a taste panel with a group of 10 male and female undergraduate and post graduate students at third year B.Sc.(Hons), M.Sc and M.Phil. levels. The opinion of all students was collected on taste and quality in a predesigned format questionnaire. Then the results have been presented both in percentage and in acceptable scores.

×

1

1

X

2

5.0 Parameters studied

The following parameters were studied for investigation of the experiments:

5.1. External fruit features

External fruit character such as skin color was recorded visually. Changes in skin color of the fruits were also observed by using a standard color chart.

5.2 Physicochemical parameters

Mango fruits from each treatment combination were selected at random from both the varieties and replications at an interval of three days such as 0, 3, 6, 9, and 12 of storage for physicochemical analysis. Care was taken while collecting the fruits from storage. Biochemical analysis was accomplished mostly in Protein & Enzyme Lab, of the Department of Biochemistry and Molecular Biology, University of Rajshahi, Soil Resources Development Institute (SRDI) Shyampur, Rajshahi and Bangladesh Council of Scientific and Industrial Research Laboratory, BCSIR, Rajshahi, Bangladesh.

The physicochemical parameters were estimated through the methods adopted from the manual of analysis of fruit and vegetable products by Rangana (1979), Analytical methods: soil, water, plant material, fertilizer by Leif Peterson (2002) and Biochemical Methods for Agricultural Sciences by Sadasivan and Manickhan (1992).

5.2.1 Edible portion of mango

At first, the weight of fruit was taken with an electric balance. Then, the fruit was peeled with a sharp knife and the inner stone was removed. The remaining fruit pulp (edible portion) was weighed. Finally, the percentage of edible portion was measured with the following formula.

Edible portion $=\frac{WP}{TW} \times 100$

Where,

WP = Weight of mango pulp

TW= Total weight of fruit

5.2.2 Determination of pulp to peel ratio

Pulp was removed from the fruits at an interval of three days during ripening. After separation, the weight of peel and pulp was taken by using an electrical

F

X

X

balance and the pulp to peel ratio was calculated. The pulp was then used for another chemical analysis.

5.2.3 Physiological weight loss of mango fruit

A specific fruit from each treatment combination of each block was taken and individually weighed using electrical balance and sometimes rough balance and then kept for ripening. The process was continued still ripening at three days interval. Percent total weight loss was calculated by using the following formula:

% Weight loss (WL)
$$= \frac{IW - FW}{IW} \times 100$$

Where,

% WL = Physiological weight loss

IW = Initial fruit weight and

FW = Final fruit weight

5.2.4 Percent moisture content of pulp

Ten g of fruit pulp from each treatment combination of each replication was taken in a porcelain crucible (which was previously cleaned, heated at 100° C, cooled and weighed). The crucible was then placed in an electrical oven at 80° C for 72 hours until the weight became constant. It was then cooled in a desiccator and weighed again.

Calculation

% Moisture =
$$\frac{IW - FW}{IW} \times 100$$

Where,

IW = Initial weight of pulp

FW = Final weight of oven dried pulp

5.2.5 Percent dry matter content of pulp

Percent dry matter content was calculated by using the data obtained during moisture estimation using the following formula

%Dry matter = 100 - % moisture content

5.2.6 Percent ash content of pulp

The oven-dried sample from 5.2.4 above was ashed in a muffle furnace at 600° C for six hours after initial pre-ashing at 200° C and percent ash was calculated as follows

-

1

x

% Ash =
$$\frac{A}{I} \times 100$$

Where,

A= Weight of ash and

I = Initial weight of pulp

5.2.7 Estimation of vitamin C content of mango pulp

Vitamin C of mango pulp was estimated by the titrimetric method as stated by Bessey and King (1933)

1. Reagents

The following reagents were used for estimation of vitamin C

i) 3% Metaphosphoric acid (HPO₃)

Three g of metaphosphoric acid was dissolved in 80 ml of acetic acid and made volume up to 100 ml with distilled water.

ii) Standard vitamin C solution (0.1 mg/ml)

Ten mg of pure vitamin C was dissolved in 3% metaphosphoric acid and made volume up to 100 ml with this acid.

iii) Dye solution

Two hundred mg of 2, 6-dichlorophenol indophenols and 210 mg of sodium bicarbonate were dissolved in distilled water and made volume up to 100 ml .The solution was then filtered.

2. Procedure

The following steps were followed.

Ten ml of standard vitamin C solution was taken in a conical flask and titrated it with the dye solution.

Five g of mango pulp were cut into small pieces and homogenized well with the 3% metaphosphoric acid (approximately 20 ml) and filtered it through double layer of muslin cloth. The filtrate was centrifuged at 3,000 rpm for 10 min. and the clear supernatant was titrated with 2, 6-dichlorophenol indophenols solution. The amount of vitamin C present in the extract was determined by comparing with the titration result of standard vitamin C solution.

Calculation

Quantity of vitamin C content (mg/ 100 g) of fresh mango pulp

1

1

4

A

= Quantity of vitaminCobtained Weight of mango pulp ×100

5.2.8 Titrable acidity in mango pulp

Titratable acidity of mango pulp was determined by the method of Ranganna (1979)

1. Reagents

The following reagents were used for determination of titrable acidity.

i) Standard NaOH solution (0.1 N)

ii) 1% Methyl red

2. Extraction of mango juice

Ten g of mango pulp were taken in a 100 ml beaker and then it was homogenized with distilled water in a blender. The blended materials was then filtered and transferred to a 100 ml volumetric flask and the volume was made up to the mark with distilled water.

3. Titration Procedure

Ten ml of pulp solution was taken in conical flask. Two to three drops of Phenolphthalein indicator was added and then the conical flask was shaken vigorously. It was then filtrated immediately with 0.1 N NaOH solutions from a burette till a permanent pink color will be appeared. The volume of NaOH solution required for titration was recorded. Percent titratable acidity was calculated by using the following formula:

% Titratable acidity = $\frac{T \times N \times V1 \times E}{V2 \times W \times 1000} \times 100$

Where,

T = Titre

N = Normality of NaOH

V1 = Volume made up

E = Equivalent weight of acid

V2 = Volume of extract

W = Weight of sample

A.

1

J.

X

5.2.9 Pulp pH of mango

1. Preparation of standard buffer solution

pH 7 and pH 4 buffer tablets (BDH chemicals Ltd., Poole, England) were dissolved in water and made up to the mark of 100 ml with distilled water.

2. Extraction of fruit juice

For the determination of pulp pH, 1 g of fresh pulp was taken in a mortar with pestle. Then, the pulp was crushed thoroughly and homogenized well with 10 ml of distilled water. Finally, the extract was filtrated through two layers of muslin cloths.

3. Procedure

The electrode assembly of pH meter was dipped into the standard buffer solution of pH 7 taken in a clean and dry beaker. The temperature correction knob was set to 28°C and the fine adjustment was made by asymmetry potentially knob to pH 7. After washing with distilled water the electrode assembly was dipped into a solution of standard pH 4 and adjusted to the required pH by fine asymmetry potential knob. The electrode assembly was raised, washed twice with distilled water, and then rinsed with mango juice and finally it was dipped into the juice of mango for recording the pH of the extract.

5.2.10 Total soluble solids (% Brix) of mango pulp

Total soluble solid content of mango pulp was estimated by using Abbe Refractometer. A drop of mango juice squeezed from the fruit pulp on the prism of the Refractometer. Percent TSS was obtained from the direct reading of the instrument. Temperature correction was done using the methods as described by Ranganna (1979).

5.2.11 Determination of total sugar content of mango pulp

Total sugar content of mango pulp was determined calorimetrically by the Anthrone method (Jayaraman, 1981).

1. Reagents

Following reagents were used for determination of total sugar:

- A. Anthrone reagent: The reagent was prepared by dissolving 2 g of anthrone in one litre of concentrated H_2SO_4 .
- B. Standard glucose solution: A standard solution of glucose was prepared by dissolving 10 mg of glucose in 100 ml of distilled water.

2. Extraction of sugar from mango pulp

Extraction of sugar from mango pulp was done by the following method (Loomis and Shull, 1937). Four g of mango pulp was cut into small pieces and immediately plunged into boiling ethyl alcohol and was allowed to boil 5 to 10 minutes (5 to 10 ml of alcohol was used per gram of pulp). The extract was cooled and crushed thoroughly in a mortar with pestle. Then the extract was filtered through two layers of muslin cloths and the ground tissue was re-extracted for three minutes in hot 80% alcohol, using 2 to 3 ml of alcohol per gram of tissue. The second extraction was ensured complete removal of alcohol suluble substances. The extract was cooled and passed through two layers of muslin cloths and the ground tissue was re-extracted for three minutes in hot 80% alcohol, using 2 to 3 ml of alcohol per gram of tissue. The second extraction was ensured complete removal of alcohol suluble substances. The extract was cooled and passed through two layers of muslin cloth. Both of the

The volume of the extract was evaporated to about 25 %(1/4) of the volume over a steam bath and cooled. This reduced volume of the extract was transferred to a 100 ml volumetric flask and it was made up to the mark with distilled water.

3. Procedure

4

Tr.

X

Aliquot of 1 ml of pulp extract was pipetted into test tubes and 4 ml of the anthrone reagent was added to each of this solution and mixed well. Glass marbles were placed on top of each test tube to prevent loss of water through evaporation. Then the tubes were placed in a boiling water bath for 10 minutes and then cooled. A reagent blank was prepared by taking 1 ml of water and 4 ml of anthrone reagent in a tube and treated similarly. The absorbance of blue green solution was measured at 680 nm in a colorimeter.

A standard curve of glucose was prepared by taking 0.0, 0.1, 0.2, 0.4, 0.6, 0.8 and 1.0 ml of standard glucose solution in different test tubes containing 0.0, 10, 20, 40, 60, 80, and 100 µg of glucose respectively, and the volume was made up to 1 ml with distilled water. Then 4 ml of anthrone reagent was added to each test tube and mixed well. All these solutions were treated similarly as described above. The absorbance was measured at 680 nm using the blank containing 1 ml of water and 4 ml of another reagent.

The amount of total sugar present in the extract was calculated from the standard curve of glucose (Fig. 1). Finally, the percentage of total sugar was determined by using the following formula.

2

1

5

X

% Total sugar (g/ 100g of mango) = $\frac{\text{Quantity of sugar obtained}}{\text{Weightof sample}} \times 100$

5.2.12 Determination of reducing sugar of mango pulp

Reducing sugar content of mango was determined by Dinitrosalicylic acid method (Miller, 1972).

1. Reagents

i) Dinitrosalicylic acid (DNS) solution: Simultaneously, 1.0 g of DNS, 200 mg of crystalline phenol and 50 mg of sodium sulphite were placed in a beaker and mixed with 100 ml of 1% NaOH by stirring. When it was needed to store, sodium sulphite was added just before use.

ii) 40 % solution of Rochelle salt

It was prepared by dissolving 40 g of sodium potassium tartarate with distilled water and made up to the mark in 100 ml of volumetric flask.

2. Extraction of sugar from mango pulp

The same procedure for extraction of sugar from mango pulp was followed as described in 5.2.11.



Fig. 1 Standard curve of glucose for estimation of total sugar

X

X

3

X

3. Procedure

Aliquot of 3 ml of the extract was pipetted into a test tube and 3 ml of DNS regent was added to each of this solution and mixed well. The test tubes were heated for 5 minutes in a boiling water bath. After the color has developed, 1 ml of 40 % Rochelle salt was added when the contents of the tubes were still warm. The test tubes were then cooled under a running tap water. A reagent blank prepared by taking 3 ml of distilled water and 3 ml of DNS reagent in a tube and treated similarly. The absorbance of the solution was measured at 575 mm in a colorimeter. The amount of reducing sugar was calculated from the standard curve of glucose (Fig. 2). The percentage of reducing sugar present in the mango pulp was

determined by using the following formulae:

% Reducing sugar (g/100 g of sample)

 $= \frac{\text{Amount of reducing sugar obtained}}{\text{Weight of sample}} \times 100$





5.2.13 Estimation of non-reducing sugar

Non-reducing sugar content of mango pulp was calculated by using the following formulae:

X

3

x

% Non-reducing sugar = % total sugar - % reducing sugar.

5.2.14 Estimation of crude fibre of mango pulp

Crude fibre of mango pulp was estimated following the procedure as given in the Biochemical Methods for Agricultural Sciences by Sadasivam and Manickam (1992)

Reagents

- A. H_2SO_4 solution (0.255 N): 1.25 g of concentrated sulphuric acid was dilluted to 100 ml of distilled water.
- B. NaOH solution (0.313 N): 1.25 g of sodium hydroxide was diluted to 100 ml of distilled water.
- C. Alcohol.

Procedure

- 1. Two g of oven dried ground material of fruit pulp which was less than 1% of fat content previously estimated was taken in a beaker. The beaker was made up to the mark of 200 ml with 1.25% H₂SO₄ solution. The mixture was boiled for 30 minutes having bumping chips for smooth boiling keeping the volume constant by addition of distilled water at frequent interval.
- 2. After boiling, the mixture was filtered through muslin cloth and the residue was washed with boiling water until washings were no longer acidic.
- The material/residue was transferred to the same beaker and 200 ml of 1.25% NaOH solution was added in it carefully and boiled for 30 minutes.
- Again, the solution was filtered through muslin cloth and was washed with 25 ml of 1.25% H₂SO₄ solution, three times of 50 ml portion of distilled water and 25 ml of alcohol.
- 5. The residue after filtration was again transferred to pre weighed porcelain crucible (W_c) and was dried at $130^0 \pm 2^0$ C for 2 hrs. in an oven and the dried material was cooled in a desiccators and it was weighed in a electrical balance(W_d)
- 6. Finally, the crucible was then ignited for 30 minutes at $600^{0}\pm15^{0}$ C in a muffle furnace for ashing and it was cooled in desiccators and reweighed (W_a).

Calculation

% crude fibre in ground sample

Materials and Methods

Chapter 3

X

1

1º

x

= Loss in weight on ignition (Wd-Wc)- (Wa-Wc) Weight of the sample

5.2.15 Determination of total lipid content of mango pulp

Total lipid content of fruit pulp was determined by the method of Bligh and Dyer (1959).

Reagents

A mixture of chloroform and ethanol was prepared at the ratio of 2:1 V/V.

Procedure:

About 2 gm of fruit pulp was taken in a mortar and grinded it properly adding there 20 ml of distilled water. The grinded juice was transferred to a separating funnel and 60 ml of chloroform and ethanol mixture was added. The mixture was mixed thoroughly. It was then kept overnight at room temperature in the dark. At the end of this period, 40 ml of chloroform and 40 ml of water were further added and mixed. Generally, three layers were found at this time. A clear lower layer of chloroform containing the entire lipid, a colored aqueous layer of ethanol with all water soluble materials and a thick pasty interphase were seen.

The chloroform layer was carefully collected in a pre-weighed beaker and then placed on a steam bath for evaporation. After evaporation of the chloroform, the weight of beaker was determined again. The difference in between the weights gave the amount of the lipid.

Calculation:

Percent of lipid content (g/100 g of fruit pulp)

 $= \frac{\text{Amount of lipid obtained}}{\text{Weight of fruit pulp}} \times 100$

5.2.16 Estimation of water soluble protein content of mango pulp

Water soluble protein content of mango pulp was determined following the method of Lowry *et a*l. (1951)

1. Required reagents

Following reagents were used for the estimation of protein

A. 2% Na₂CO₃ solution in o.1N NaOH.

- B. 0.5% copper sulphate in 1% sodium potassium tartarate.
- C. Folin Ciocalteau Reagent (FCR): (FCR: H₂O=1:1).

K

10

×

D. Protein standard: 100µg/ml in distilled water. A standard solution of egg albumin was prepared by dissolving 100 µg of egg albumin in 1 ml of distilled water

2. Extraction of mango juice

Five g of mango pulp was taken in a mortar with a pestle and it was grinded and homogenized with distilled water. The blended and homogenized materials was then filtered with double layer muslin cloth to a 100 ml beaker and then it was transferred to a 100 ml volumetric flask and the volume was made up to the mark with distilled water.

3. Procedure

1. Reagent A and B were mixed in a ratio of 50:1 and diluted reagent C were prepared carefully and FCR solution was used just before taking spectrophotometric reading.

2. Aliquot of 1 ml was taken in test tube duplicately. Five ml of A and B mixture was added to each of the test tube. After ten minutes, 0.5ml of FCR solution was added to each test tube. Finally, the reading was taken in a spectrophotometer at 650 nm after 30 minutes. A graph was drawn by plotting conc. against absorbance and from the graph the amount of protein was calculated in the supplied sample (Fig. 3).



Fig. 3 Standard curve of protein for estimation of water soluble protein

k

1

A

5.2.17 Determination of different minerals

Different important minerals of mango pulps were determined following the procedure as described in Analytical Methods (Petersen, 2002).

Preparation of plant sample

Procedure of drying

The cleaned porcelain crucibles were placed in an oven at 105[°] C for overnight. The crucibles were allowed to cool in a desiccator's and these were weighed. The mango pulps were collected with spatula and put into crucible and again weighed. The crucibles were placed in the oven at 105[°] C for 24 hours. Then, the crucibles were allowed to cool in a desiccator and weighed. The crucibles were again placed in the oven at 105[°] for 2 hours. These were cooled in a desiccator and weighed again. Drying, cooling and weighing were accomplished repeatedly until the weight became constant. The dried pulps were stored in airtight plastic container. The moisture content was calculated in the sample.

Procedure of grinding

The dried plant material was ground in a mortar with pestle. These were further kept in an oven at 105° C for overnight due to absorption of moisture in the time of grinding for keeping the weight constant.

5.2.17.1 Determination of phosphorus content

Ground mango pulp was digested and P was released by digestion with nitric acid and it was determined by spectrophotometry.

Reagents

A. HNO_3 (68%)

- B. Ammonium molybdate ascorbic acid
- C. Diluted HNO₃ 1: 100

Twenty ml of 68% HNO₃ was transferred to 2000 ml volumetric flask and made the volume with distilled water and mixed well.

Digestion procedure

1. Ground pulp material (0.3 g) was taken into digestion tube. The two remaining tubes were blanks. Five ml of 68% nitric acid was added to each of all the 40 tubes. The content was mixed in each tube and was kept the tubes overnight. The tubes were placed in the digester and the tubes were covered with the exhaust

manifold. The temperature was set at 125° C. The digester was turned on and the digestion was continued for 4 hours after boiling started. It was observed that no tubes became dry.

2. After cooling, the digestion mixture was transferred with distilled water to a 200 ml volumetric flask. The flask was made up to the mark with distilled water and mixed well. It was filtered on a dry filter into a dry bottle which could be closed with a screw cap. The filtrate was kept in the closed bottle and used for estimation of Phosphorus.

Measurement of phosphorus

Five ml of diluted filtrate was transferred to a 50 ml volumetric flask. Water was added to approx. 30 ml and was mixed well and then, 10 ml of ammonium molybdate- ascorbic acid solution was added and made volume with water and was mixed well. After 15 minutes, the absorbance was measured on a spectrophotometer at 890 nm. Sometimes, the procedure was done repeatedly due to higher absorbance of the highest standard solution using a smaller amount of filtrate. In this case, 1:100 diluted HNO₃ was added to the volumetric flask to make the total volume of 1: 100 diluted HNO₃ and filtrate equal to 5 ml.

When the content of P was very high, the filtrate was diluted further before transfer to the 50 ml flask. The dilution was made up to the mark. After transfer of 5 ml diluted filtrate to the 50 ml volumetric flask, 5 ml 1:100 diluted HNO₃ and water to aprox. 30 ml were added. Then 10 ml ammonium molybdate ascorbic acid was added, the 50 ml volumetric flask was made to volume with water and the absorbance was measured at 890 nm after 15 minutes.

Calculation

-it

Amount of P was determined by the following formula:-

mg per kg mango pulp= $\frac{a \times 25000}{b \times c}$

Where,

a = mg P per litre.

b = ml diluted filtrate transferred into the 50 ml volumetric flask for determination of P

c = g of plant material taken into the digestion tube.

1

4

Jr.

A

5.2.17.2 Determination of potassium content

Powder form of mango pulp was digested and K was released by digestion with nitric acid and it was determined by flame photometry.

Reagents

- A. HNO₃ (68%)
- B. Diluted HNO₃1: 100

Twenty ml of 68% HNO₃ was transferred to 2000 ml volumetric flask and made the volume with distilled water and mixed well.

Digestion procedure

Same as phosphorus

Measurement of potassium

The diluted filtrate (10 ml) was transferred into a 50 ml volumetric flask using a pipette. The flask was made up to the volume with water and mixed well. The content of K was measured by flame photometry. When the reading was found to be higher than the reading of the highest standard solution, further dilution was made. In this case, 1:100 diluted HNO₃ was added to the volumetric flask to make the total volume of 1:100 diluted HNO₃ and filtrate was equal to 10 ml.

Calculation

Amount of K was determined by the following formula:-

mg per kg mango pulp= $\frac{a \times 25000}{b \times c}$

Where,

a = mg of K per litre.

b = ml diluted filtrate transferred into the 50 ml volumetric flask for determination of K

c = g of plant material taken into the digestion tube.

5.2.17.3 Determination of calcium content

Ground mango pulp was digested and Ca was released by digestion with nitric acid and it was determined by atomic absorption spectrophotometer.

Reagents

- A. HNO₃ (68%)
- B. Diluted HNO₃1: 100

1

4

T:

*

-th

Twenty mI of 68% HNO₃ was transferred to 2000 ml volumetric flask and made the volume with distilled water and mixed well.

iii) LaCl₃ solution

The 435 g of LaCl₃ 7H₂O was weighed into a beaker. One hundred ml of 5 M HNO_3 and 400 ml water were added in it. The salt was heated gently until it was dissolved. After cooling, 300 ml of more 5 M HNO_3 was added and the solution was transferred to 5 l volumetric flasks. It was made to volume with water and mixed. The solution contained 3.25% of La.

Digestion procedure

Same as phosphorus

Measurement of calcium

Twenty ml of diluted filtrate was transferred into a 50 ml volumetric flask using a pipette. The LaCl₃ solution (5 ml) was added to make a volume with water and mixed well. The content of Ca was measured by atomic absorption spectrometer. When the reading was found to be higher than the reading of the highest standard solution, a larger dilution was made. In this case, 1:100 diluted HNO₃ was added to the volumetric flask to make the total volume of 1:100 HNO₃ and filtrates was equal to 20 ml.

Calculation

Amount of Ca was determined by the following formula:-

mg per kg mango pulp= $\frac{a \times 25000}{b \times c}$

Where,

a = mg of Ca per litre.

b = Amount of filtrate transferred into the 50 ml volumetric flask for determination of Ca

c = g of plant material taken into the digestion tube

5.2.17.4 Determination of magnesium content

Ground material of mango pulp was digested and Mg was released by digestion with nitric acid and it was determined by atomic absorption spectrophotometer.

Reagents

x

1

A

Same as calcium

Digestion procedure

Same as phosphorus

Measurement of Magnesium

Diluted filtrate (5 ml) was transferred into a 50 ml volumetric flask using a pipette. The LaCl₃ solution (5 ml) was added to make a volume with water and mixed well. The content of Mg was measured by atomic absorption spectrometer. When the reading was found to be higher than the reading of the highest standard solution, a larger dilution was made. In this case, 1:100 diluted HNO₃ was added to the volumetric flask to make the total volume of 1:100 diluted HNO₃ and filtrate was equal to 5 ml.

Calculation

Amount of Mg was determined by the following formula:

mg per kg mango pulp=
$$\frac{a \times 25000}{b \times c}$$

Where,

a = mg of Mg per litre.

b = ml of diluted filtrate transferred into the 50 ml volumetric flask for determination of Mg

c = g of plant material taken into the digestion tube.

5.2.17.5 Determination of copper content

Powder form of mango pulp was digested and Cu was released by digestion with nitric acid and it was determined by atomic absorption spectrophotometer.

Digestion procedure

Same as phosphorus

Measurement of copper

The content of Cu was measured by atomic absorption spectrometer (AAS) directly in the undiluted filtrate.

Calculation

Amount of Cu was determined by the following formula:

mg per kg mango pulp= $\frac{d \times 200}{c}$

A

4

*

1

Where,

d = mg of Cu per litre.

c = g of ground pulp material taken into the digestion tube.

5.2.17.6 Determination of iron content

Ground mango pulp material was digested and Fe was released by digestion with nitric acid and it was determined by atomic absorption spectrophotometry.

Digestion procedure

Same as phosphorus

Measurement of iron

The content of Fe was measured by atomic absorption spectrometer (AAS) directly in the undiluted filtrate.

Calculation

Amount of Fe was determined by the following formula:

mg per kg mango pulp=
$$\frac{d \times 200}{c}$$

Where,

d = mg of Fe per litre.

c = g of ground pulp material taken into the digestion tube.

5.2.17.7 Determination of manganese content

Ground material of mango pulp was digested and Mn was released by digestion with nitric acid and it was determined by atomic absorption spectrophotometry.

Digestion procedure

Same as phosphorus

Measurement of manganese

The content of Mn was measured by atomic absorption spectrometer directly in the undiluted filtrate.

Calculation

Amount of Mn was determined by the following formula:

mg per kg mango pulp=
$$\frac{d \times 200}{c}$$

Where,

A.

-

*

1

d = mg of Mn per litre.

c = g of ground pulp material taken into the digestion tube.

5.2.17.8 Determination of zinc content

Ground mango pulp material was digested and Zn was released by digestion with nitric acid and it was determined by atomic absorption spectrophotometry.

Digestion procedure

Same as phosphorus

Measurement of zinc

The content of Zn was measured by atomic absorption spectrometer directly in the undiluted filtrate.

Calculation

Amount of Zn was determined by the following formula:

mg per kg mango pulp=
$$\frac{d \times 200}{c}$$

Where,

d = mg of Zn per litre.

c = g of ground pulp material taken into the digestion tube.

5.3 Shelf life of mango

Shelf life (days) of mango as influenced by different post harvest treatments and varieties was calculated by counting the days required to ripen fully with retaining optimum marketing and eating qualities.

6.0 Statistical analysis

The collected data was statistically analyzed by analysis of variance method. The means of different parameters was compared using DMRT as described by Gomez and Gomez (1984).

GUAPTER FOUR

1

-

S.

-

*

1

RESULTS AND DISCUSSION

CHAPTER 4

RESULTS AND DISCUSSION

This chapter deals with the observation of four experiments with graphical representation and discussion of the results obtained from different experiments. The results of all the experiments on external feature, physicochemical parameters and shelf life of two mango varieties influenced by different treatments as in experiment 1:different storage treatments, experiment 2: different doses of Maleic hydrazide (MH) solution, experiment 3: different doses of Gibberellic acid (GA3) solution and experiment 4: different levels of Bavistin DF (BDF) solution as well as their combined effects are presented in the following Tables 1.1–1.15, 2.1-2.19, 3.1- 3.25 and 4.1-4.25 and in the Figures 1.1–1.46, 2.1-2.33, 3.1-3.22 and 4.1-4.22, respectively, for clear interpretation and good understanding. The brief analysis of variance of data in respect of various parameters studied are shown in Appendices 1.1-1.14, 2.1-2.12, 3.1-3.12 and 4.1- 4.12.

7.0 Experiment 1

A

ant-

4

X

Influence of different storage environments on physicochemical changes and shelf life of postharvest mango

7.1 Changes in skin color

Morphological features, especially skin color is the most obvious change that occurs in many fruits and is frequent the principal criterion followed by the consumers to identify whether the fruit is ripe or unripe. The most common change in case of mango is the loss of green color (Wills *et al.*, 1989).

Postharvest treatments strongly influenced the skin color of both the fruits *viz.*, Langra and Khirshapat (Table 1.1). Both varieties demonstrated the original green color at the initial stage of harvesting. At 3^{rd} day, Langra and Khirshapat developed a trace in yellow at control (T₀) but, they retained green color at T₁, T₃, andT₅ treatments. Langra also exhibited trace in yellow at T₂ and light green at T₄ treatments while Khirshapat developed light green at T₂ and T₄ treatments.

At 6^{th} day, Langra was found to be greenish yellow at T₀ while, light green was noticed from T₁ and T₃ treatments. But it developed yellowish green color at T₂

Chapter 4: Experiment 1

1

a

4

1

and T_4 treatment and remained green at T_5 treatment. On the other hand, Khirshapat showed yellowish green at T_0 , but, it developed greenish yellow at T_2 and T_4 , and light green at T_1 and T_3 treatments. It also remained green at T_5 treatments.

After 9 days of storage, Langra was found to be deep yellow and yellowish green at T_0 , T_1 and T_3 as well as greenish yellow at T_2 and T_4 treatments, respectively but, it still retained its original green color at T_5 treatments having slightly shrinkage. On the other hand, Khirshapat showed yellow at T_0 and T_2 treatments, yellowish green at T_1 and T_4 treatments and yellowish green at T_3 treatments. Further, it also retained its original green color with slightly shrinkage at T_5 .

After 12 days of storage, Langra was found to be yellow with black spot at T_0 , yellowish green at T_1 , T_3 and T_4 , deep yellow at T_2 treatment, but, it retained green color at T_5 treatment showing shrinkkage. On the other hand, Khirshapat was also found yellow with light spot at T_0 , and T_2 ; greenish yellow and yellowish green at T_1 , T_3 and T_4 but, it still retained its original green color having shrinkage condition at T_5 treatment.

After 15 days of storage, Langra showed no existence of T_0 , T_2 and T_4 and retained yellow color at T_1 and T_3 treatment while, it developed light green color at T_5 treatment having heavy shrinkage condition. On the other hand, Khirshapat showed completely rotten condition at T_0 , T_2 and T_4 treatment but, it modified into yellow color at T_1 and still retained its original green color at T_5 treatment having heavy shrinkage condition. Regarding the skin color development, harvest of mango at due time was found to be desirable. Changes in skin color are partially supported by the findings of Azad (2001) and Hossain (1999).

7.2 Changes in physical characters during storage environments

Various parameters regarding to the physical changes of mango are presented and discussed in the following sub-headings.

7.2.1 Edible portion of mango

Edible portion is one of the most important considerations of mango for the consumers as well as for the researchers. This portion is called pulp in case of mango. Varieties of mango were found to be significant in relation to edible portion

Chapter 4: Experiment 1

X

1

at different days after storage (Appendix 1.2). The edible portion of mango was recorded to be increased gradually with the passing of time of storage (Fig. 1.1). At 12th day the highest edible portion (72.65%) of mango was obtained from Langra and lowest (71.60) was reported from Khirshapat. Present investigation was supported by the findings of Azad (2001).

Different storage treatments in respect of edible portion were also found to be highly significant at different days after storage (Appendix 1.2). Edible portion of mango among the means of storage treatments increased gradually with different days after storage (Fig. 1.3). At 9th day, control (T_0) exhibited the highest percentage of edible portion (76.34%), whereas at this time the lowest percentage (67.50%) was obtained from low temperature in refrigerated condition (4±1)°C. At 12th day, edible portion decreased slightly due to its rotten situation at T_0 , at the same time another treatments showed increased level of edible portion of mango (Fig. 1.3). These results are also supported by the findings of Hossain (1999).

Combined effects of varieties and storage treatments on edible portion were not significant at various days after storage (Appendix 1.2). But, edible portion was increasing at all the days after storage. At each day, the increasing trend was high with the combination of Langra (V₁) and control (T₀). After 9th day, edible portion was decreased at V₁T₀ and V₂T₀ combinations, but other treatment combination showed augmented situation in this time. At 12th day, the maximum edible portion (75.82%) was obtained from the treatment combination of V₁T₀ followed by others. The lowest (68.12%) was recorded from the treatment combination of Khirshapat and low temperature in refrigerator (Table 1.2). This situation occurred by the treatments combination was possibly due to inhibition of ethylene synthesis at low temperature in refrigerator but no obstacle in ethylene synthesis in control situation.

7.2.2 Pulp to peel ratio

Variation in pulp to peel ratio between two varieties was found to be highly significant in all days of storage (Appendix 1.2). The ratio was increased for the time being at different days in both the varieties. In all the days, Langra exhibited better performance in pulp to peel ratio compared to Khirshapat (Fig. 1.2). At 12th day,

Chapter 4: Experiment 1

X

or

AL.

-

X

Langra produced higher ratio (8.21) whereas Khirshapat produced lower (8.00). These results are in conformity with the findings of Azad (2001). This happened possibly due to genetically dissimilarities between varieties.

Differences between storage treatments were also found highly significant in case of pulp to peel ratio in all the storage period (Appendix 1.2). In all the cases, control treatments showed more ratio followed by others. The increasing trend of this ratio was the same up to 6th day at T_o treatment, thereafter, this trend slightly decreased due to rotten situation. On the other hand, the other treatments showed more or less similar increasing trend of the ratios. At 12th day, T_o treatment produced the highest ratio (10.69) followed by hot water treatment (10.07), perforated polyethylene cover (9.05), unperforated polyethylene cover (6.98), paraffin coating (6.53) and low temperature in refrigerator (5.3). The lowest ratio was observed from refrigerator due to suspension of ethylene synthesis at low temperature but, control treatment produced maximum ratio due to higher synthesis of ethylene. So, the preservative value of refrigerator was much better than the other treatment. But in this system, ripening did not occur after taking to normal environments from refrigerator.

Combined effects of varieties and different storage treatments were found to be significant in terms of pulp to peel ratio at different days after storage except at initial stage (Appendix 1.2 and Table 1.2). Pulp to peel ratio was increased at the space of time in all days of storage. But, it was found that the treatment combination of V_1T_0 produced wider ratio compared to the other treatment combination. The highest ratio (10.95) was recorded from the treatment combination of V_1T_0 at 12th day and the lowest (5.15) was recorded from the combination of V_2T_5 at 12 DAS. These results are in partial agreements with the findings of Hossain (1999) and Azad (2001).

7.2.3 Physiological weight loss

Varieties exhibited highly significant physiological weight loss (PWL) at different days after storage (Appendix 1.3 and Table 1.4). At each day of storage, Khirshapat (V_2) gradually lost comparatively more PW as compared to Langra with advancing of storage time (Fig. 1.5). The highest (12.88%) and lowest (12.00%) of

1

me

÷.

1





Fig. 1.1 Edible portion of mango pulp in varieties at different days after storage



Fig. 1.3 Edible portion of mango pulp as influenced by storage treatments at different days after storage. Vertical bars represent LSD at 0.05 level

Figures represent

- $V_1 = Langra$ $V_2 = Khirshapat$ $T_0 = Control$
- T₁=Paraffin coating T₂=Perforated polyethylene cover T₃=Unperforated polyethylene cover

Fig. 1.2 Pulp to peel ratio of mango in varieties at different days after storage



Fig. 1.4 Pulp to peel ratio of mango as influenced by storage treatments at different days after storage. Vertical bars represent LSD at 0.05 level

 T_4 =Hot water treatments $(55\pm2)^{\circ}C$ T_5 = Low temperature in refrigerator $(4\pm1)^{\circ}C$ Table: 1.1 Changes in skin color of two mango varieties (viz.,Langra and Khirshapat) as influenced by different postharvest treatments during storage at ambient condition

1

1

-

T

	Varieties	Treatments	Days after storage							
	varieties		Initial	3	6	9	12	15		
-		To	Green	Trace in yellow	Greenish yellow	Deep yellow	Black spotted yellow	_		
		T ₁	Green	Green	Light green	Greenish Yellowish	Greenish yellow	_		
		T ₂	Green	Trace in yellow	Yellowish green	Greenish yellow	Deep Yellow	_		
	V ₁	T ₃	Green	Green	Light green	Yellowish green	Yellow	Yellow		
		T₄	Green	Light green	Yellowish green	Greenish yellow	Yellow	_		
		Ts	Green	Green	Green	Green (slight shrinkage)	Green (shrinkage)	Light green (deep shrinkage)		
	V ₂	To	Green	Trace in yellow	Yellowish green	Yellow	Light spotted yellow			
		T ₁	Green	Green	Light green	Greenish yellow	Greenish yellow	_		
68		T ₂	Green	Light Green	Greenish yellow	Yellow	yellow	_		
		T ₃	Green	Green	Light green	Yellowish green	Greenish yellow	Yellow		
		T₄	Green	Light Green	Greenish yellow	Greenish yellow	Yellowish green	_		
		Ts	Green	Green	Green	Green (slight shrinkage)	Green (shrinkage)	Green (deep shrinkage)		

Treatments

 $V_1 = Langra$ $V_2 = Khirshapat$ $T_0 = Control$ 4)

A

T₁=Paraffin coating T_2 =Perforated polyethylene cover T_3 =Unperforated polyethylene cover

A

 T_4 =Hot water treatments (55±2) $^{\rm 0}$ C and T_5 = Low temperature in refrigerator $~(4\pm1)^{\rm 0}$ C

4)

A

69

A)

Treatments combination	Edible portion (%)at different days				Pulp to peel ratio at different days					
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12
V ₁ T ₀	67.64 ab	71.22 a	75.25 a	76.85 a	75.82 a	4.42 a	6.85 a	8.85 a	9.95 a	10.95 a
V ₁ T ₁	66.15 bc	66.95 c-e	67.78 ef	68.48 d	70.12 de	4.25 a-c	4.42 f	4.92 h	5.52 i	6.32 i
V ₁ T ₂	68.52 a	70.02 ab	71.35 b	72.65 b	74.68 a-c	4.30 a-c	4.89 d	6.35 e	7.65 e	9.15 e
V ₁ T ₃	68.60 a	69.52 ab	70.68 bc	71.15 b	72.87 c	4.20 bc	4.48 ef	5.22 g	6.22 g	7.13 g
V ₁ T ₄	66.78 a-c	68.38 b-d	70.18 b-d	71.98 b	73.95 a-c	4.28 a-c	5.25 c	7.13 c	8.62 c	10.23 c
V1T5	65.85 bc	66.45 de	67.05 f	67.68 d	68.48 ef	4.35 ab	4.32 f	4.75 h	5.12 j	5.45 j
V ₂ T ₀	66.74 a-c	70.24 ab	74.24 a	75.83 a	74.84 ab	4.12 c	6.23 b	8.35 b	9.45 b	10.43 b
V_2T_1	65.83 bc	66.62 de	67.32 ef	68.02 d	69.52 d-f	3.83 de	4.12 g	4.32 i	5.94 h	6.74 h
V ₂ T ₂	67.33 a-c	68.83 bc	70.13 b-d	71.43 b	73.43 bc	3.78 de	4.65 e	6.15 f	7.45 f	8.95 f
V_2T_3	66.68 a-c	67.59 c-e	68.38 d-f	69.22 cd	70.92 d	3.89 de	4.33 f	5.13 g	5.92 h	6.82 h
V ₂ T ₄	65.56 c	67.16 c-e	68.96 c-e	70.76 bc	72.76 c	3.92 d	5.08 c	6.82 d	8.32 d	9.92 d
V ₂ T ₅	65.52 c	66.13 e	66.72 f	67.32 d	68.12 f	3.72 e	4.12 g	4.25 i	4.82 k	5.15 k
Level of significance	NS	NS	NS	NS	NS	NS	**	**	***	***
CV%	1.56	1.51	1.43	1.45	1.46	2.55	2.13	1.73	1.47	1.28

Table 1.2 Combined effects of varieties and different storage treatments on edible portion and pulp to peel ratio of postharvest mango

7

*)

-

1

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents

$V_1 = Langra$	T ₁ =Paraffin coating	T₄=Hot water treatments (55±2 ⁰ C	** indicate at 1% level
$V_2 = Khirshapat$	T ₂ =Perforated polyethylene cover	T_5 = Low temperature in Refrigerator $(4\pm1)^0$ C	***indicates at 0.1%
$T_0 = Control$	T ₃ =Unperforated polyethylene cover	 Indicates at 5% level 	NS non significant
7

N'S

et

PWL were recorded from Khirshapat and Langra at 12th day, respectively. The results also revealed that total PWL progressively increased with the increase in storage period. The findings also indicated that Langra showed better performance in respect of PWL as compared to Khirshapat. Water loss through lenticels seems to be the probable cause of physiological weight loss in the fruits during storage. Lower lenticels density in Langra facilitated lesser water loss leading to minimum total weight loss (Azad, 2001). Singh *et al.* (2000) also reported more or less the similar findings.

Variation was also found highly significant among the means of postharvest storage treatments on PWL (Appendix 1.3). At different days after storage, it revealed that control (T_o) treatment was faster in PWL followed by other treatments (T_4 , T_2 , T_5 and T_1) and the mangoes treated with T_3 treatment were slower in PWL. It was also indicated that T_o , T_4 and T_2 treatments showed PWL more or less in the same rate whereas, T_5 , T_1 and T_3 treatment, followed more or less same rate of losses (Fig. 1.7). At 12th day, the maximum PWL (19.88%) was obtained in T_o and minimum (8.20%) in T_3 . These phenomena happened seems to be protected by polyethylene cover around the fruits which hindered in PWL. These results are in partial coincided with the findings of Hossain (1999) while, these result are strongly supported by the findings of Fawaz (2006) and Illeperuma and Jayasuriya (2002)

The combined effect of varieties and different storage treatments exhibited highly significant on PWL at different days of storage (Appendix 1.3). At different days of storage, it showed that different treatment combination showed PWL gradually with advancing of storage time. At 12^{th} day, there was found the maximum PWL (20.50%) at V₂T₀ and the minimum (4.20%) at V₁T₃ (Table 1.3). It also revealed that Langra lost minimum amount of water owing to using polythene cover followed by the other treatment combination. These results are in partial supported by the findings of Azad (2001) and Hossain (1999).

7.2.4 Moisture content

Varieties showed highly significant moisture content at different days after storage (Appendix 1.3). At different days, the results indicated that moisture content increased with the passing of time of storage (Fig. 1.6). This increasing rate was

Int:

more or less similar from 3 to 9 days and thereafter, it decreased slightly due to starting rotten. Figure 1.6 denoted that each day of storage, Khirshapat produced more moisture as compared to Langra, the maximum (85.99%) and minimum (83.89%) was recorded from V_2 and V_1 at 12^{th} day. Shajahan *et al*, (1994) observed different results in this regard and the moisture content increased in Langra than Khirsapat. These results are in agreement with the findings of Azad (2001). This variation might be due to genetical, location, weather effect, soil quality or maturity of the fruit.

Variation among the means of storage treatments on moisture content was highly significant at different days after storage (Appendix 1.3). At different days, moisture content increased gradually with space of time but, this increasing point was the maximum in control treatment at 6th day, in T₄ at 9th day and in T₁, T₂, and T₃ at 12th day (Fig. 1.8). T_o treatment produced the highest moisture content (85.88%) at 9th day, and thereafter, it decreased up to 12th day and the lowest (82.73%) was recorded from T₅ at 9th day. The increasing trend of moisture content from initial to 9 days might be due to a metabolic activities and osmotic pressure inside the mango fruit as well as this decrease might be due to suspending metabolic activities resulting in rotten and drying.

The combined effect of varieties and different storage treatments in respect of moisture content was found to be significant at initial and 3^{rd} day but, non significant from 6th days to onward (Appendix 1.3). At different days, moisture content increased with the augmentation of storage time. The treatment combination of V₂T_o, V₂T₄ and V₂T₃ produced the highest moisture content (86.92%, 86.86% and 86.87%) at 6, 9 and 12th day, respectively. In this storage period, the lowest values (81.75%, 81.94% and 81.75%) were observed from the treatment combination of V₁T₅ (Table 1.3). These treatment combinations provided slightly lesser amount of moisture content from 6 and 9 days. The moisture content increasing from initial to other days might be due to metabolic activities and osmotic principles and decreasing in a certain days might be due to suspending metabolic activities resulting in drying,

Results and Discussion





Fig. 1.5 Physiological weight loss of mango between varieties at different days after storage



Fig. 1.6 Moisture content of mango pulp between varieties at different days after storage



Fig. 1.7 Physiological weight loss of mango as influenced by storage treatments at different days after storage. Vertical bars represent LSD at 0.05 level

Figures represent

-

.

 $V_1 = Langra$ $V_2 = Khirshapat$ $T_0 = Control$

T₁=Paraffin coating T₂=Perforated polyethylene cover T₃=Unperforated polyethylene cover

Fig.1.8 Moisture content of mango pulp as influenced by storage treatments at different days after storage. Vertical bars represent LSD at 0.05 level

 T_4 =Hot water treatments $(55\pm2)^{\circ}$ C T_5 = Low temperature in refrigerator $(4\pm1)^{\circ}$ C *

73

4

Treatments combination	Physiol	ogical weigh	t loss (%) al	t different days		Moisture con	ntent (%) a	t different da	iys
Varieties × Treatments	3	6	9	12	Initial	3	6	9	12
V ₁ T ₀	6.35 b	9.50 b	15.65 b	19.25 b	81.23 e	82.65 e	84.85 d	84.53 f	83.68 g
V ₁ T ₁	1.75 i	3.20 h	4.25 h	4.85 h	81.56 d	81.85 g	82.38 i	83.11 i	83.86 g
V ₁ T ₂	3.50 f	8.30 d	13.75 d	17.30 d	81.55 d	82.36 f	83.27 g	84.38 f	84.84 e
V ₁ T ₃	1.20 k	2.20 j	3.15 j	4.20 i	81.35 e	81.98 g	82.72 h	83.65 h	84.65 f
V_1T_4	4.60 d	8.85 c	14.55 c	18.60 c	81.65 d	82.45 f	83.63 f	84.87 e	84.52 f
V ₁ T ₅	1.80 i	3.75 f	5.85 f	7.80 f	81.22 e	81.42 h	81.75 j	81.94 j	81.75 h
V ₂ T ₀	7.25 a	10.60 a	16.80 a	20.50 a	83.75 a	85.17 a	86.92 a	86.62 b	85.76 d
V ₂ T ₁	2.30 h	3.40 g	4.50 g	5.65 g	83.65 a	84.07 c	84.64 e	85.37 d	86.12 c
V ₂ T ₂	4.45 e	8.75 c	14.50 c	18.45 c	83.28 bc	84.45 b	85.36 c	86.46 b	86.92 a
V ₂ T ₃	1.50 j	2.60 i	3.45 i	4.75 h	83.45 b	84.22 c	84.94 d	85.87 c	86.87 a
V2T4	5.20 c	9.50 b	15.65 b	19.30 b	83.35 bc	84.46 b	85.64 b	86.86 a	86.52 b
V ₂ T ₅	2.50 g	4.30 e	6.50 e	8.60 e	83.25 c	83.45 d	83.72 f	83.94 g	83.76 g
Level of significance	***	***	***	***	***	**	NS	NS	NS
CV%	1.48	1.67	1.05	0.84	1.05	0.13	0.12	0.12	0.12

Table1.3 Combined effects of variety and different storage treatments on physiological weight loss and moisture content of postharvest mango

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents

V1 = LangraT1=Paraffin coatingT4=Hot water treatments $(55\pm 2\ ^0 C)$ V2 = KhirshapatT2=Perforated polyethylene coverT5= Low temperature in Refrigerator $(4\pm 1)^0 C$ T0 = ControlT3=Unperforated polyethylene cover* indicates at 5% level

** indicate at 1% level
***indicates at 0.1% NS non significant

Results and Discussion

transpiration and evaporation. Moisture content increased in the present study is in partial agreement with the findings of El-Mahmoudi and Eisawi (1968) in banana. 7.2.5 Dry matter content

₹.

The variation in varieties means on dry matter content of mango pulp demonstrated highly significant at different days after storage (Appendix 1.4). At different days of storage, dry matter content decreased gradually with the advancement towards storage time (Fig. 1.9). It showed that Langra gave comparatively more dry matter content comparing to Khirshapat. At initial stage, Langra provided the maximum (18.56%) while Khirshapat provided the minimum (16.55%) but at 12th day, Langra provided minimum (16.12%) while Khirshapat gave 14.00 % of dry matter. These results are in partial agreement with the findings of Hassain (1991).

Different storage treatment adopted in this study on dry matter content of mango exhibited significant effect at different days after storage (Appendix 1.4). At different days, dry matter content decreased gradually with the passing of time of storage (Fig. 1.11). It was also noticed that control treatment produced the lowest dry matter (14.11%) at 6th day whereas, the highest dry matter (17.27%) was observed in mangos that kept in the refrigerator at this period and thereafter, it increased to onward. At 12th day, dry matter content has found to be 17.24% and 14.12% for storage at refrigeration and perforated polyethylene cover, respectively (Fig. 1.11).

The combined effect of varieties and different storage treatments on dry matter content of mango pulp was significant at initial and 3rd day but, nonsignificant at 6, 9 and 12th day (Appendix 1.4). At different days, dry matter content decreased gradually up to 9th day and then, it increased slightly from control at 12th day. At initial day, the highest dry matter content (18.78%) was noticed from the treatment combination of V1T5 which was statistically similar with the combination of V_1T_3 and V_1T_0 . In this level, the lowest value (16.25%) was observed from the treatment combination of V_2T_0 which was also statistically similar with the combination of V₂T₁. At 12th day, the highest value (18.25%) was recorded from the combination of V_1T_5 and the lowest value (13.08%) was recorded from V_2T_0 (Table

1.4). The decrease in dry matter content with increasing storage period may be due to breaking down of the complex carbohydrates into simple molecules and H_2O as well as adding water through osmotic process and metabolic activities. The decrease in dry matter content of mango pulp is not in agreement with the findings of Hossain (1999).

7.2.6 Ash content

Variation in varieties means on ash content of mango pulp was highly significant at different days after storage (Appendix 1.4). At different days, values of ash content reduced gradually with the passing of storage time (Fig. 1.10). It also revealed that Langra accumulated comparatively more ash than Khirshapat at all the days. Higher (1.07%) ash content was accumulated from Langra at initial level whereas Khirshapat accumulated lower quantity (0.95%).

Different storage treatments exhibited significant variation in ash content of mango at different days except at initial level (Appendix 1.5). It was demonstrated from the Fig. 1.12 that ash content was affected by different storage treatments and increased slightly than control at 3^{rd} day in all the treatments, except at T₂ treatment but, it decreased after this stage. At 6^{th} day, T₅ treatment produced highest (0.99%) where as the lowest ash (0.81%) was found in T₀ treatment. At 12th day, the maximum (0.99%) ash content was gathered from T₅ and minimum (0.81%) was accumulated from perforated polythene cover storage.

The combined effect of varieties and different storage treatments was found to be non significant on ash content of mango at various days after storage (Appendix 1.4 and Table 1.4). It may suggest from the present data that ash content showed good correlation with the dry matter content.

7.3 Changes in biochemical properties of mango during storage environments

Various changes relating to biochemical properties of mango fruits are presented and discussed under the following sub-headings.

7.3.1 Vitamin C content

Variation in between varieties means on vitamin C content of mango pulp was noticed to be highly significant at different days after storage (Appendix 1.5). At various days after storage, it was observed that Langra showed better performance X

5

-

-1







Fig. 1.11 Dry matter content of mango pulp as influenced by storage treatments at different days after storage. Vertical bars represent LSD at 0.05 level

Figures represent

- V₁ = Langra V₂ = Khirshapat
- $T_0 = Control$
- T_1 =Paraffin coating T_2 =Perforated polyethylene cover T_3 =Unperforated polyethylene cover



Fig. 1.10 Effect of varieties on ash content of mango pulp at different days after storage



Fig. 1.12 Ash content of mango pulp as influenced by storage treatments at different days after storage. Vertical bars represent LSD at 0.05 level

 T_4 =Hot water treatments (55±2)^oC T_5 = Low temperature in refrigerator (4±1)^oC

Treatments combination	Dr	Dry matter content (%) at different days					Ash content(%) at different days			
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12
V ₁ T ₀	18.77 a	17.35 d	15.15 g	15.47 e	16.32 b	1.07 a	0.99 d	0.87 de	0.88 d	0.93 b
V_1T_1	18.38 b	18.15 b	17.62 b	16.89 b	16.14 b	1.05 a	1.04 ab	1.01 b	0.97 b	0.92 b
V ₁ T ₂	18.45 b	17.64 c	16.73 d	15.62 e	15.16 d	1.06 a	1.01 b-d	0.96 c	0.89 d	0.87 c
V ₁ T ₃	18.65 a	18.02 b	17.28 c	16.35 c	15.35 c	1.08 a	1.03 a-c	0.99 b	0.93 c	0.88 c
V ₁ T ₄	18.35 b	17.55 c	16.37 e	15.13 f	15.48 c	1.05 a	1.00 cd	0.94 c	0.86 de	0.89 c
V ₁ T ₅	18.78 a	18.58 a	18.27 a	18.06 a	18.25 a	1.08 a	1.06 a	1.04 a	1.03 a	1.04 a
V ₂ T ₀	16.25 e	14.83 h	13.08 j	13.38 i	14.21 e	0.93 b	0.85 g	0.75 g	0.76 g	0.81 d
V ₂ T ₁	16.35 e	15.93 f	15.36 f	14.63 g	13.88 f	0.94 b	0.91 f	0.88 d	0.84 e	0.79 de
V ₂ T ₂	16.72 cd	15.55 g	14.64 h	13.54 i	13.08 h	0.96 b	0.89 f	0.84 ef	0.77 g	0.75 f
V ₂ T ₃	16.55 d	15.78 f	15.06 g	14.13 h	13.13 h	0.95 b	0.90 f	0.86 de	0.81 f	0.76 ef
V ₂ T ₄	16.65 cd	15.54 g	14.36 i	13.14 j	13.48 g	0.94 b	0.89 f	0.82 f	0.75 g	0.77 ef
V ₂ T ₅	16.75 c	16.55 e	16.28 e	16.06 d	16.24 b	0.96 b	0.95 e	0.93 c	0.92 c	0.93 b
Level of significance	***	**	NS	NS	NS	NS	NS	NS	NS	NS
CV%	0.59	0.62	0.66	0.68	0.73	2.07	2.17	2.29	2.40	2.42

4

Table 1.4 Combined effects of varieties and different storage treatments on dry matter and ash content of postharvest mango

1

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents

rubrerepresents		
$V_1 = Langra$	T ₁ =Paraffin coating	T_4 =Hot water treatments (55±2 ° C
$V_2 = Khirshapat$	T ₂ =Perforated polyethylene cover	T ₅ = Low temperature in Refrigerator
T ₀ =Control	T ₃ =Unperforated polyethylene cover	 indicates at 5% level

4)

+

A

77

±2 ° C ** indicate at 1% level gerator (4±1)° C NS non significant 如

×

×

4

in accumulating vitamin C as compared to Khirshapat (Fig. 1.13). It also indicated that quantity of vitamin C decreased with advancing of storage period. The decreasing trend was higher from initial to 6^{th} day, and thereafter it was slightly slower. At the initial stage, Langra produced higher amount (151.23 mg/100 g) of vitamin C whereas Khirshapat showed lower amount (56.00 mg/100 g). At 12^{th} day, Langra also showed wider amount of vitamin C (23.84 mg/100 g) and Khirshapat gave the lower amount (17.35 mg/100 g) (Fig. 1.13). These results also elucidated that vitamin C content gradually reduced with the advancement of storage period in case of both the varieties. It might be due to augmentation of ethylene synthesis resulting breaking down of ascorbic acid. The results of the present study are in agreement with the findings of Azad (2001), Shyamalamma (1995), Gofur *et al.* (1994) and Absar *et al.* (1993).

Variation among the treatments means on vitamin C content was highly significant at different days after storage (Appendix 1.5). At initial day, green mango treated with low temperature in refrigerator $(4\pm1)^0$ C produced the highest amount (105.50 mg/100 g) of vitamin C which was statistical similar with the fruit treated with unperforated polyethylene cover (T₃) and untreated mango (T_o). After initial level, it decreased gradually with the passing of storage time (Fig. 1.14). It was found that decreasing trend was faster with untreated fruit but, it showed slower with the fruit in low temperature at refrigerated condition. At 12th day, it ranged between 12.97 to 30.87 mg per 100 g of pulp obtained from control (T_o) and treated with low temperature in refrigerator. The decreasing trend in vitamin C content from both the treated and untreated mangoes at different storage period might be due to oxidation of ascorbic acid and low temperature in refrigerator might be possibly due to causing delay ripening resulting in lower oxidation of vitamin C. These results are supported by the findings of Hossain (1999).

The combined effect of varieties and different storage treatments exhibited significant variation in respect of vitamin C content at different days after storage (Appendix 1.5 and Table 1.5). At various days of storage, it was found that vitamin C content decreased with the advancement of storage period. The quantity of vitamin C ranged between 52.50 to 155.30 mg per 100 g of fresh mango pulp at initial day

×

The

X





Fig. 1.13 Effect of varieties on vitamin C content of mango pulp at different days after storage

Fig. 1.14 Vitamin C content of mango pulp as influenced by storage treatments at different days after storage. Vertical bars represent LSD at 0.05 level

Figures represent	
$V_1 = Langra$	Τ1=
$V_2 = Khirshapat$	$T_2 =$
$T_0 = Control$	T3=

T₁=Paraffin coating T₂=Perforated polyethylene cover T₃=Unperforated polyethylene cover T₄=Hot water treatments (55±2)⁰C T₅= Low temperature in refrigerator (4±1)⁰ C

of storage with the combination of V_2T_4 and V_1T_5 . These quantities reduced gradually after initial day. At 12th day, it also ranged between 10.33 to 35.80 mg per 100 g, with the treatment combination of V_2T_0 and V_1T_5 . The maximum (35.80 mg/100 g) was obtained from the treatment combination of V_1T_5 whereas the lowest (10.33 mg/100 g) was recorded from the treatment combination of V_2T_0 .

7.3.2 Titratable acidity

Variation in between varieties in relation to titratable acidity was noticed to be highly significant at different days after storage (Appendix 1.6). At various days of storage, Langra showed higher titratable acidity as compared to Khirshapat (Fig. 1.15). Titratable acidity decreased with the passing of storage period. The decreasing trend was very faster from initial to 6th day, and thereafter, this trend was slightly slower from 6 to 12th day. At initial day, the highest (4.31%) was recorded in Langra

Freatments combination		Vitamin C cor	ntent (mg/100 g) at different day	/S
Varieties × Treatments	Initial	3	6	9	12
V ₁ T ₀	150.60 bc	110.70 e	45.60 f	24.80 g	15.60 b
V ₁ T ₁	152.30 b	120.40 b	75.50 b	36.50 b	27.35 b
V ₁ T ₂	149.60 cd	114.60 d	55.30 e	28.60 e	19.20 f
V ₁ T ₃	151.40 bc	118.30 c	62.40 d	30.70 d	21.60 e
V ₁ T ₄	148.20 d	119.70 bc	68.60 c	33.80 c	23.50 d
V_1T_5	155.30 a	125.80 a	95.60 a	58.20 a	35.80 a
V ₂ T ₀	58.60 e	30.60 i	15.60 I	15.60 k	10.33 j
V ₂ T ₁	55.20 f	40.50 fg	28.80 h	26.30 f	21.65 e
V ₂ T ₂	54.70 f	35.80 h	20.30 k	18.60 j	12.83 i
V ₂ T ₃	59.40 e	38.75 g	22.45 j	21.30 i	15.50 h
V ₂ T ₄	52.50 g	39.40 g	25.60 i	23.40 h	17.85 g
V ₂ T ₅	55.61 f	42.20 f	35.55 g	30.85 d	25.93 c
Level of significance	***	*	***	***	***
CV%	1.01	1.34	1.62	0.36	1.06

Table 1.5 Combined effects of varieties and different storage treatments on vitamin C content of postharvest mango

1)

+

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents			
$V_1 = Langra$	T ₁ =Paraffin coating	T_4 =Hot water treatments (55±2 $^{\circ}$ C	** indicate at 1% level
V ₂ = Khirshapat	T ₂ =Perforated polyethylene cover	$T_5 = Low temperature in Refrigerator (4\pm1)^{0} C$	***indicates at 0.1%
T ₀ =Control	T ₃ =Unperforated polyethylene cover	 indicates at 5% level 	NS non significant

08

1

+

.

fr.

L

à-

X

whereas the lowest (3.55%) was recorded from Khirshapat. At 12th day, the highest (0.53%) was obtained from Langra while Khirshapat produced the lowest amount (0.45%). The decreasing trend of titratable acidity at storage period was also reported by Upadhyay and Tripathi (1985), Leon and Lima (1968) and Medlicott *et al.* (1986). According to them, acidity reduced during storage growth on attainment of maturity and ripening. It also might be due to genetical dissimilarities between the varieties.

Different storage treatments used in this investigation on titratable acidity demonstrated significant variation among the means at various days after storage (Appendix 1.6). At various days of storage, titratable acid content declined very sharply from initial to 6^{th} day, and then, it declined steadily from 9^{th} day to onward (Fig. 1.17). In all the cases of storage period, the wider titratable acidity (4.01%, 2.59%, 1.43%, 1.15% and 0.96%) was recorded at low temperature in refrigerator from initial to 12^{th} day, followed by 3.90%, 1.23%, 0.53%, 0.30% and 0.17% in untreated (T_o) mangoes, respectively. These results are partially supported by the findings of Tefera *et al.* (2007).

The combined effect of varieties and storage treatments on titratable acidity of mango pulp was observed to be significant at different days after storage except at 3^{rd} and 6^{th} days (Appendix 1.6). At different days after storage, there was found decreasing trend of titratable acidity with the advancement of storage period (Table 1.6). At initial day, the highest concentration (4.42%) was noticed from the treatment combination of V₁T₅ which was statistically identical with the combination of V₁T₃ and the lowest acid content (3.48%) was reported from the treatment combination of V₂T₄. At 12th day, titratable acid concentration ranged between 0.15 to 1.10%. In this stage, the highest (1.10%) amount was recorded from the treatment combination of V₁T₅, whereas the lowest (0.15%) was recorded from the treatment combination of V₂T₀. This occurrence might be possible due to inhibition of acid oxidation at low temperature in refrigerator and genetical variation in between varieties.

7.3.3 Pulp pH

K

3-

X

Variation in between the means of varieties was highly significant on pulp pH of mango at different days after storage (Appendix 1.6). At various days of storage, there was found an increasing trend of pulp pH with the passing of storage period (Fig. 1.16). In all the storage period, pulp pH showed more amount in Khirshapat as compared to Langra. Higher pH (3.72) was obtained from green Khirshapat at initial day whereas lower (3.60) was reported from green Langra. At 12th day, the highest pulp pH (5.83) was recorded from Khirshapat and the lowest value (5.67) was noticed from Langra, respectively. Increasing trend of pulp pH was also observed by Yuniarti (1980), Kumar *et al.* (1993) and Shahjahan *et al.* (1994). This phenomenon might be due to oxidation of acid during storage resulting in higher pulp pH and genetical dissimilarities between the varieties.

The postharvest treatments imposed to the mango fruits exhibited significant variation in pulp pH at different days after storage (Appendix 1.6). An increasing trend of pulp pH was noticed from different treated and untreated fruits from initial to onwards (Fig. 1.18). Pulp pH was wider in control at all stages of storage followed by the other treatments; especially fruits treated with low temperature in refrigerator gave the lowest trend of augmentation of pulp pH. pH value of mango pulp was the maximum (6.95) from control whereas the fruits treated with low temperature in refrigerator gave the lowest (4.45) value at 12th day (Fig. 1.18). These results are strongly supported by the findings of Tefera *et al.* (2007).

The combined influence of varieties and imposed storage treatments regarding pulp pH was found to be significant at different days of storage except, initial and 3^{rd} day. There was showing an increasing trend of pulp pH from various treatment combinations at different days of storage (Table 1.6). At different days of storage, the treatment combination of V_2T_0 denoted the highest pH value (7.00) which was statistical similar with the combination of V_1T_0 and the lowest pH value(4.40) was recorded from the treatment combination of V_1T_5 . It was also statistically identical with the combination V_2T_5 . This phenomenon might be possible due to low temperature in refrigerator which hindered in acld oxidation that caused delay ripening. Further, low temperature in refrigerator may cause delay ripening

K.

2

4

X



Fig. 1.15 Effect of varieties on titratable acidity of mango pulp at different days after storage



Fig. 1.17 Titratable acidity of mango pulp as influenced by storage treatments at different days after storage. Vertical bars represent LSD at 0.05 level

Figures represent

$V_1 = Langra$	$I_1 = P$
V ₂ = Khirshapat	$T_2 = P$
$T_0 = Control$	$T_3 = U$





Fig. 1.16 Effect of varieties on pH of mango pulp at different days after storage



Fig. 1.18 Pulp pH of mango as influenced by storage treatments at different days after storage. Vertical bars represent LSD at 0.05 level

 T_4 =Hot water treatments $(55\pm 2)^{\circ}$ C T_5 = Low temperature in refrigerator $(4\pm 1)^0$ C

reatments	Т	ïtratable a	cidity (%)	at different	days	Pulp pH at different days				
ombination	1.1									
arieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12
V ₁ T ₀	4.25 bc	1.50 d	0.60 hi	0.35 h	0.18 g	3.70 ab	4.80 a	5.70 b	6.80 a	6.90 a
V ₁ T ₁	4.30 b	2.45 b	1.20 bc	0.87 c	0.65 c	3.60 ab	3.80 fg	4.10 h	4.40 g	5.10 g
V ₁ T ₂	4.20 c	1.82 c	0.75 f-h	0.45 g	0.29 f	3.70 ab	4.30 cd	4.80 d	5.30 d	6.20 c
V ₁ T ₃	4.35 ab	2.02 c	0.92 d-f	0.62 e	0.42 e	3.60 ab	3.90 ef	4.30 fg	4.70 f	5.40 f
V ₁ T ₄	4.32 b	2.30 b	1.05 cd	0.75 d	0.53 d	3.50 b	4.20 d	4.60 e	5.20 d	6.00 d
V ₁ T ₅	4.42 a	2.83 a	1.60 a	1.35 a	1.10 a	3.50 b	3.60 h	3.80 i	4.00 h	4.40 h
V ₂ T ₀	3.55 ef	0.95 f	0.45 i	0.24 i	0.15 g	3.80 a	4.90 a	5.90 a	6.90 a	7.00 a
V ₂ T ₁	3.65 d	1.99 c	0.95 de	0.67 e	0.62 c	3.70 ab	3.90 ef	4.20 gh	4.60 f	5.00g
V2T2	3.50 ef	1.25 e	0.57 hi	0.34 h	0.26 f	3.80 a	4.40 bc	4.90 d	5.50 c	6.30 c
V ₂ T ₃	3.52 ef	1.45 de	0.72 gh	0.44 g	0.37 e	3.60 ab	4.00 e	4.40 f	4.90 e	5.60 e
V₂T₄	3.48 f	1.49 d	0.84 e-g	0.55 f	0.49 d	3.80 a	4.50 b	5.10 c	5.80 b	6.60 b
V ₂ T ₅	3.60 de	2.35 b	1.25 b	0.95 b	0.82 b	3.60 ab	3.70 gh	3.90 i	4.10 h	4.50 h
Level of significance	*	NS	NS	***	***	NS	NS	*	**	***
CV%	1.33	6.89	11.46	4.10	2.38	2.85	2.50	2.24	2.01	1.81

Table 1.6 Combined effects of varieties and different storage treatments on titratable acidity and pulp pH of postharvest mango

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents

 $V_1 = Langra$ T₁=Paraffin coating $V_2 =$ Khirshapat T_2 =Perforated polyethylene cover T₀ = Control

4

4)

T₄=Hot water treatments (55±2 ° C T_5 = Low temperature in Refrigerator $(4\pm1)^0$ C ***indicates at 0.1% T₃=Unperforated polyethylene cover * indicates at 5% level

** indicate at 1% level NS non significant

40

*

84

 (\mathbf{v})

4

7

·F

2

resulting in low acid degradation achieving low pulp pH, but, Khirshapat without treatment showed highest pulp pH due to none inhibition of acid degradation.

7.3.4 Total soluble solid (Brix %) content

Statistically highly significant variation was observed in total soluble solid content between two varieties at different days after storage (Appendix 1.7). The results indicated that TSS content of mango pulp augmented gradually with the passing of storage period. The augmenting trend was more or less similar and it increased faster from initial to 6th day thereafter, it increased slower from 6th day to onward (Fig. 1.19). At every stage of storage period, Khirshapat gave better performance in TSS accumulation than Langra. The highest TSS value (18.00%) was recorded from Khirshapat and the lowest (17.04%) was recorded from Langra at 12th day. The increase of TSS during storage in mango was reported by Singh (1968). Absar *et al.* (1993) reported that TSS was increased with maturity of mango fruit. But, they found highest TSS in Langra. Mollah and Siddique (1973) reported that TSS value varied from cultivar to cultivar. These might be possible due to genetically differences between varieties.

Different storage treatments induced with the postharvest mangoes in this study demonstrated significant variation in respect of TSS content at different days after storag (Appendix 1.7). At different days after storage, it was noticed that TSS accumulation increased with the advancement of storage time. It also indicated that TSS content sharply increased from untreated mangoes at initial to 6th day and then, it decreased markedly at 6 to 12th day (Fig. 1.21). Another storage treatments *viz.*, hot water treatment also increasingly produced TSS from initial to 9th day and thereafter, it decreased sharply. The mangoes treated with perforated polyethylene cover also gave more or less increasing trend from initial to onward. But, the fruits treated with low temperature in refrigerator produced TSS very steady motion at various days of storage. The highest (20.67, 19.68 and 20.67%) TSS accumulation was obtained from control, hot water and perforated polyethylene cover at 6, 9 and 12th day whereas, the lowest (7.99, 9.59 and 11.09%) was obtained from control.

et al. (1994) showed that TSS content increased markedly up to 5 days after harvest without using any treatment. These happened possibly due to ripening condition resulting in maximizing TSS gathering in control and low temperature in refrigerator retardated in ethylene synthesis that caused delay ripening and ultimately in lower TSS accumulation. It also elucidated that TSS gathering is strongly related to ripening and it caused decrease owing to becoming rotten.

The combined effect of varieties and implied storage treatments on TSS exhibited non significant at initial, 3 and 6th days but, significant at 9 and 12th days (Appendix 1.7 and Table 1.7). There was found an increasing behavior of TSS at different days (Table 1.7). The highest accumulation (21.13, 20.12 and 21.12%) was recorded from the treatment combination of V_2T_0 , V_2T_4 and V_2T_2 at 6, 9 and 12th days of storage, whereas the lowest value (7.56, 9.25 and 10.75%) was recorded from the treatment of V_1T_5 , respectively.

7.3.5 Total sugar content

22

*

Variation in between both the varieties means on total sugar content was found to be significant at different days after storage (Appendix 1.7). It was noted that total sugar content increased markedly with the passing of time at different days after storage (Fig. 1.20). This increasing tendency was markedly from initial to 9 days at both the varieties and then, it increased steadily. At all days of storage Khirshapat produced comparatively more quantity than that of Langra. At initial day, Khirshapat produced the highest (6.19%) whereas, Langra gave the lowest (5.78%). At 12 days after storage Khirshapat produced the highest quantity (17.62%) and the lowest (17.17%) was recorded from Langra. Upadhyay and Tripathi (1985) reported that total sugar content increased gradually during storage for 6 days at room temperature. Sugar content increased during ripening (Srivastava, 1967). These results are in conformity with the findings of Shahjahan *et al.* (1994). Tsuda *et al.* (1999) also found the similar result. The increase in total sugar might be due to conversion of complex starch or carbohydrate into simple compound like sucrose, fructose, galactose etc.

A Results and Discussion

Different storage treatments applied to this experiment on total sugar content of mango pulp showed significant variation at different days after storage except at initial day (Appendix 1.7 and Table 1.7). At different days after storage, it indicated that total sugar content increased markedly with the advancement of storage period (Fig. 1.22). The increasing tendency was very fast in untreated mango followed by other treatments. It also showed excellent performance in sugar formation as compared to T_4 , T_2 , T_3 , T_1 and T_5 treatments at 6th day. The maximum quantity of total sugar (11.06, 17.43 and 20.43%) was recorded from control at 3, 6 and 9th days and in which 20.70% was recorded from hot water treatment at 12th day, but the minimum (6.91, 8.06, 9.67 and 11.94%) was recorded from low temperature in refrigerator. Puttaraju and Reddy (1997) stated in their findings that low temperature in most effective in delay ripening but the quality of these fruits were unacceptable due to high rate of spoilage. Mondal et al. (1995) also reported the increasing trend of total sugar content at the later stage of storage. The increasing trend of total sugar at untreated mangos might be due to breaking down of complex carbohydrate into simple compound but, low temperature in refrigerator made delay ripening at storage period resulting in lower conversion of complex compound into simple molecules.

The combined effect of varieties and adopted storage treatments in this study in terms of total sugar content of mango pulp exhibited non significant variation at different days after storage. It indicated that total sugar content increased with the passing of storage time at different days after storage. The maximum quantity of total sugar (20.78 and 20.84%) was received from the treatment combination of V_2T_0 and V_2T_4 at 9 and 12th day, whereas the minimum (9.47% and 11.57%) was received from the treatment combination of V_1T_5 .

7.3.6 Reducing sugar content

Te

r

Analysis of variance showed that varieties had significant effect on reducing sugar content of mango pulp at different days after storage (Appendix 1.8). There was found an increasing trend of reducing sugar with the passing of storage time Table 1.8). It was also noticed that Khirshapat produced comparatively more reducing sugar than Langra at different days after storage. (The highest (6.51%) quantity of this sugar was achieved from Khirshapat while, the lowest (6.33%) was





Fig. 1.19 Effect of varieties on TSS of mango pulp at different days after storage



Fig. 1.21 TSS of mango pulp as influenced by storage treatments at different days after storage. Vertical bars represent LSD at 0.05 level

Figures represent $V_1 = Langra$ $V_2 = Khirshapat$ $T_0 = Control$

1

¥

 T_1 =Paraffin coating T_2 =Perforated polyethylene cover T_3 =Unperforated polyethylene cover

Fig. 1.20 Effect of varieties on total sugar content of mango pulp at different days after storage



Fig. 1.22 Total sugar content of mango pulp as influenced by storage treatments at different days after storage. Vertical bars represent LSD at 0.05 level

 $\begin{array}{l} T_4 = Hot \mbox{ water treatments } (55 \pm 2)^0 \mbox{C} \\ T_5 = Low \mbox{ temperature in refrigerator } (4 \pm 1)^0 \mbox{ C} \end{array}$

10

68

1.1	
- 14	
2	

4)

4

 \mathbf{Y}_{i}

45

¥1

Table 1.7 Combined effects of varieties and different storage treatments on total soluble solid and total sugar content of postharvest mango

Treatments combination		TSS content (%) at different days					TSS content (%) at different days Total sugar content (%) at different days				t days
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12	
V ₁ T ₀	5.82 b	12.13 b	20.22 b	18.75 d	17.25 h	5.85 b	10.86 b	17.27 b	20.12 b	19.15 c	
V ₁ T ₁	5.63 bc	6.62 j	10.12 h	13.63 i	16.83 i	5.82 b	7.32 h	9.33 j	12.55 j	15.55 f	
V ₁ T ₂	5.74 b	8.74 f	13.21 e	17.72 e	20.22 b	5.75 b	9.25 e	13.05 f	17.12 f	19.12 c	
V ₁ T ₃	5.51 c	7.05 i	11.12 g	15.13 g	18.72 e	5.78 b	7.82 g	10.32 h	13.85 h	17.05 d	
V ₁ T ₄	5.73 b	9.24 e	14.24 d	19.24 c	18.45 f	5.72 b	9.78 d	14.07 d	18.79 d	20.56 a	
V ₁ T ₅	5.65 bc	6.43 k	7.56 j	9.25 k	10.75 k	5.77 b	6.67 i	7.87	9.47 1	11.57 h	
V ₂ T ₀	6.72 a	13.05 a	21.13 a	19.62 b	18.13 g	6.25 a	11.25 a	17.58 a	20.78 a	19.77 b	
V ₂ T ₁	6.56 a	7.52 h	11.23 g	14.73 h	18.23 g	6.22 a	7.72 g	9.72 i	12.94 i	15.93 e	
V ₂ T ₂	6.62 a	9.62 d	14.11 d	18.62 d	21.12 a	6.18 a	9.67 d	13.48 e	17.42 e	19.57 b	
V ₂ T ₃	6.53 a	8.12 g	12.15 f	16.15 f	19.75 c	6.12 a	8.12 f	10.62 g	14.13 g	17.32 d	
V ₂ T ₄	6.64 a	10.13 c	15.12 c	20.12 a	19.32 d	6.21 a	10.13 c	14.48 c	19.15 c	20.84 a	
V ₂ T ₅	6.53 a	7.35 h	8.42 i	9.92 j	11.42 j	6.15 a	7.14 h	8.25 k	9.86 k	12.31 g	
Level of significance	NS	NS	NS	*	***	NS	NS	NS	NS	NS	
CV%	1.70	1.24	0.79	0.65	0.59	1.74	1.18	1.46	0.67	1.00	

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents $V_1 = Langra$

T₀ =Control

T₁=Paraffin coating

 T_4 =Hot water treatments (55±2 $^{\circ}$ C $T_5 =$ Low temperature in Refrigerator $(4\pm 1)^0$ C ***indicates at 0.1%

** indicate at 1% level NS non significant

R

24

T

achieved from Langra at 12th day of storage (Table 1.8). These results are in agreement with the report of Upadhyay and Tripathi (1985). Casttrillo *et al.* (1992) elucidated that reducing sugar increased during storage period. Khirshapat producing comparatively more reducing sugar might be due to genetical variation in both the varieties.

Storage treatments imposed to this study exhibited significant variation in terms of reducing sugar content of mango pulp at different storage period (Appendix 1.8). The results showed that reducing sugar of mango pulp was augmented gradually at different days after storage. It also denoted that untreated mangoes showed better performance in producing reducing sugar as compared to other treatments. Again, it was found from control that reducing sugar content increased gradually up to 9th days of storage and then, it decreased due to starting decomposition. At 12th day, the maximum (7.13%) amount of reducing sugar was attained from hot water treatment and the lowest (4.91%) was attained from the fruit treated with low temperature in refrigerator. Increase in reducing sugar during storage was stated by Tripati (1988). Lower increasing trend of reducing sugar content treated with low temperature in refrigerator might be probably due to delay ripening resulting in lower conversion of carbohydrates into simples molecules.

The combined effect of varieties and implied storage treatments of mango pulp demonstrated significant variation in respect of reducing sugar content of mango pulp at 9 and 12 days of storage and non significant at initial, 3 and 6th days of storage (Appendix 1.8). It denoted that reducing sugar content increased gradually at three days interval up to 9th days of storage thereafter, it decreased. At 9 and 12th days, the maximum (7.43 and 7.22%) quantity was recorded from the treatment combination of V₂T₀ and V₂T₄ whereas, the lowest (3.38 and 4.88%) was reported from the treatment combination of V₂T₅, respectively (Table 1.8).

7.3.7 Non reducing sugar content

The variation in between the varieties means was highly significant in terms of non-reducing sugar content at different storage days (Appendix 1.8). There exhibited an increasing trend of non reducing sugar content at different days of storage. At all days after storage, it was noticed that Khirshapat showed better

p.

4

÷.

performance in achieving of non reducing sugar content as compared to Langra (Table 1.9). At 12th day the highest (11.06%) amount of non reducing sugar was obtained from Khirshapat and the lowest amount (10.84%) was obtained from Langra. These results are in conformity with the findings of Ali and Mazhar (1960). They reported that non reducing sugar content of ripe fruits was 11.20%. The result obtained from this experiment might be due to varietal differences.

Different storage treatments adopted in this trial had the profound significant variation on non reducing sugar content of mango at different days after storage except at initial day (Appendix 1.8). It was observed that non reducing sugar content of mango pulp increased sharply at various days. It also indicated that untreated fruits showed better performance in achieving comparatively more non reducing sugar followed by the other treatments. This increasing trend decreased at 9th day due to rotten situation. Lower increasing trend was observed from the fruits treated with low temperature in refrigerator. Hot water treatment showed better performance in accumulation of non reducing sugar content at 12th day. The highest value (13.15%) was recorded from hot water treatment which was statistically at par with control and the lowest value (6.27%) was recorded from low temperature in refrigerator at 12th day. These events might be possibly due to low temperature in refrigerator that inhibited fruit ripening resulting in smaller accumulation of non reducing sugar. These results are in agreement with the reports of Singh (1968), Joshi and Roy (1988) and Rangavalli et al. (1993). They found a gradual increase in non reducing sugar content of untreated fruit.

The analysis of variance of the combined effect of varieties and induced storage treatments had non significant effect on non reducing sugar content of mango at different days after storage except at 9th day (Appendix 1.8). It was noticed that an increasing trend of non reducing sugar was found from different treatment combination at various days of storage (Table 1.9). At 9th day, the highest (13.35%) quantity of non reducing sugar was recorded from the treatment combination of V₂T₀ which was statistically at par with V₂T₄ whereas; the lowest (6.09%) was recorded from the treatment combination of V₁T₅.

Treatments		Reducing sugar (%) at different days					Non-reducing sugar (%) at different days			
Variety	Initial	3	6	9	12	Initial	3	6	9	12
V1	1.61 b	2.19 b	3.65 b	4.94 b	6.33 b	4.18 b	6.43 b	8.28 b	10.38 b	10.84 b
V2	1.78 a	2.37 a	<u>3.83 a</u>	5.09 a	6.51 a	4.41 a	6.62 a	8.52 a	10.63 a	11.06 a
Level of significance	***	***	***	***	***	***	***	***	***	***
Treatments										
To	1.77 a	3.37 a	6.56 a	7.35 a	6.71 c	4.29	7.69 a	10.70 a	13.10 a	12.75 b
T ₁	1.65 bc	1.85 e	2.76 e	4.02 e	6.15 e	4.37	5.67 e	6.77 e	8.73 d	9.60 e
T ₂	1.72 ab	2.22 c	3.42 c	5.04 c	7.03 b	4.24	7.24 c	9.85 c	12.23 b	12.32 c
T ₃	1.68 bc	1.98 d	3.07 d	4.43 d	6.59 d	4.28	5.99 d	7.41 d	9.56 c	10.60 d
T₄	1.71 a-c	2.51 b	4.11 b	5.83 b	7.13 a	4.26	7.45 b	10.18 b	13.15 a	13.57 a
Ts	1.64 c	1.75 f	2.56 f	3.40 f	4.91 f	4.32	5.11 f	5.51 f	6.27 e	6.87 f
Level of significance	**	***	***	***	***	NS	***	***	***	***

Table 1.8 Changes of reducing and non reducing sugar content between mango varieties and influenced by different storage treatments

4

+

4

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

40

92

1

4

Table represents $V_1 = Langra$ $T_1 = Paraffin coating$ $T_4 = Hot water treatments (55 \pm 2 ° C)** indicate at 1% level<math>V_2 = Khirshapat$ $T_2 = Perforated polyethylene cover<math>T_5 = Low temperature in Refrigerator (4 \pm 1)^{\circ} C$ ** indicates at 0.1% $T_0 = Control$ $T_3 = Unperforated polyethylene cover* indicates at 5% levelNS non significant$

4/1

93

4

1

4.1

+

4

r

Reducing sugar (%) at different days Treatments combination Non-reducing sugar (%) at different days 9 3 Varieties × Treatments Initial 3 6 12 Initial 6 9 12 V_1T_0 1.65 d-g 3.25 b 6.46 b 7.26 b 6.57 de 4.21 b-d 7.61 ab 10.48 b 12.86 b 12.58 c 9.53 f V_1T_1 1.55 g 1.75 hi 2.65 h 3.92 i 6.02 g 4.27 a-c 5.57 q 6.68 g 8.63 f 6.93 c 7.13 d 9.73 d 12.17 c 12.19 d V_1T_2 1.62 e-g 2.12 e 3.32 e 4.95 e 4.11 cd 9.53 d V_1T_3 1.88 fg 2.95 g 6.53 e 4.22 b-d 5.94 ef 7.37 e 10.52 e 1.58 fg 4.32 g 1.66 d-f 2.46 c 4.06 c 7.03 b 4.06 d 7.32 c 10.03 c 13.01 b 13.53 a V₁T₄ 5.78 c V_1T_5 1.58 fg 1.68 i 2.48 i 3.38 j 4.88 h 4.19 b-d 4.99 i 5.39 i 6.09 h 6.69 h V₂T₀ 1.88 a 3.48 a 6.65 a 7.43 a 6.85 c 4.37 ab 7.77 a 10.93 a 13.35 a 12.92 b 9.66 f V_2T_1 1.75 b-d 1.95 f 2.86 q 4.12 h 6.27 f 4.47 a 5.77 f 6.86 f 8.82 e 2.32 d 12.29 c 12.45 c V₂T₂ 1.82 ab 3.52 d 5.13 d 7.12 b 4.36 ab 7.35 c 9.96 c V_2T_3 1.78 bc 2.08 e 3.18 f 4.54 f 6.65 d 4.34 ab 6.04 e 7.44 e 9.59 d 10.67 e V₂T₄ 1.75 b-d 2.55 c 4.15 c 5.87 c 7.22 a 4.46 a 7.58 b 10.33 b 13.28 a 13.62 a V₂T₅ 1.70 с-е 1.82 gh 7.05 g 2.63 h 3.42 j 4.93 h 4.45 a 5.23 h 5.62 h 6.44 q Level of significance * ** * NS NS NS NS NS NS NS CV% 2.28 0.95 3.07 1.39 1.04 0.81 2.39 1.60 1.24 0.99

Table 1.9 Combined effects of varieties and different storage treatments on reducing and non reducing sugar content of postharvest mango

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents

 $V_1 = Langra$ T_1 =Paraffin coating $V_2 = Khirshapat$ $T_2 = Perforated polyethylene cover$ $T_0 = Control$

T₄ =Hot water treatments (55±2 ° C $T_5 =$ Low temperature in Refrigerator $(4\pm 1)^0$ C T₃=Unperforated polyethylene cover * indicates at 5% level

** indicate at 1% level ***indicates at 0.1% NS non significant

1

F.

+

T

7.3.8 Crude fibre content

Crude fibre is very important for man and animals. It contains very little amount in mango, but it protects the human body from catastrophic diseases like colon cancer. It is not assimilated by human body but, it strongly help in digestion of food and making stool in the intestine.

The analysis of variance of mango varieties used in this experiment showed significant variation on crude fiber content of mango pulp at different storage period (Appendix 1.9). There was found a decreasing trend of crude fiber from both the varieties at different days of storage (Fig. 1.23). It also denoted that Langra showed better performance in accumulation of crude fiber. At initial day, Langra gave the highest (1.30%) quantity of crude fiber as compared to Khirshapat (1.18%). The amount of crude fiber decreased gradually from initial to onward at both the varieties. At 12th day, the highest (0.75%) was noted from Langra whereas; the lowest (0.65%) was noted from Khirshapat. There were no available research findings regarding crude fiber content of mango in the scientific literature. The decrease in crude fiber content with the advancement of storage period might be due to breaking down of cellulose and lignin into smaller compound.

Various storage treatments applied to this trial had significant effect on crude fiber content of mango pulp at different days after storage (Appendix 1.9). There exhibited that crude fiber content decreased gradually with the passing of storage period (Fig. 1.25). At all days, it was found that crude fiber content was comparatively higher at the fruit treated with low temperature in refrigerator followed by other treatments. At initial day, the maximum amount of crude fiber (1.28%) was reported from T₀ treatment which was statistically identical with T₂ and T5 treatment and the lowest (1.22%) was reported from T₁ treatment which was also statistical identical with T₃, and T₄ treatment. At 12th day, the maximum (0.90%) was recorded from T₅ treatment where as the lowest (0.42%) recorded from T₀ treatment. Crude fiber decreased in mango pulp at the advancement of storage period was reported by Mathooko (2000).

The combined effect of varieties and different storage treatments demonstrated the significant variation in respect of crude fiber content of mango at

t.

The.

different days of storage period (Appendix 1.9). It was observed that the crude fiber content of mango pulp decreased gradually towards time of storage period (Table 1.10). It indicated that the treatment combination of V_1T_5 gave better performance in achieving comparatively more crude fiber content at all storage time except at initial day. At 12th day, the highest (0.96%) quantity of crude fiber was gathered from the treatment combination of V_1T_5 and the lowest (0.36%) was gathered from the treatment combination of V_2T_0 .

7.3.9 Total lipid content

The analysis of variance of varieties had significant effect on lipid content in mango at different days after storage (Appendix 1.9).The results denoted that an increasing trend of lipid content was noted in mango pulp at different days (Fig. 1.24). In all the days, it indicated that Langra produced higher amount of lipid content as compared to Khirshapat. At 12th day, the highest (0.72%) amount of lipid was noticed from Langra and the lowest (0.65%) was noticed from Khirshapat (Table 1.16). This event might be due to the genetically dissimilarities between two varieties.

The difference between postharvest storage treatments used in this study in terms of lipid content was highly significant at different days after storage (Appendix 1.9). There was found an increasing trend of lipid content with the passing of storage period at various days (Fig. 1.26). It was also noticed that control treatment produced comparatively higher amount of lipid followed by T_1 , T_2 , T_3 , T_4 and T_5 treatment, from initial to 9 days; and then, it decreased due to becoming rotten situation. At 9th day, the highest (0.80%) amount of lipid was obtained from control whereas; the lowest (0.38%) was obtained from the fruits treated with low temperature in refrigerator. At 12th day, perforated polyethylene cover treated fruits gave the highest (0.78%) amount of lipid which was statistically similar with the fruits treated with hot water treatment and the lowest (0.48%) was produced from the fruits treated with low temperature in refrigerator. These occurrences might be due to the low temperature in refrigerator which caused delay ripening resulting in lower production of lipid content.

The interaction effects of varieties and induced storage treatment in this trial were highly significant in respect of lipid content of mango pulp at different days

X

T

*

-

-



Fig. 1.23 Effect of varieties on crude fibre of mango pulp at different days after storage



Fig. 1.25 Crude fibre of mango pulp as influenced by storage treatments at different days after storage. Vertical bars represent LSD at 0.05 level

Figur	es i	repr	resent
V	lan	ara	

V ₁ = Langra	1 ₁ =Paraffin coating
V ₂ = Khirshapat	T ₂ =Perforated polyethylene cover
$T_0 = Control$	T ₃ =Unperforated polyethylene cover



Fig. 1.24 Effect of varieties on total lipid content of mango pulp at different days after storage



Fig. 1.26 Total lipid content of mango pulp as influenced by storage treatments at different days after storage. Vertical bars represent LSD at 0.05 level

 T_4 =Hot water treatments $(55\pm2)^{\circ}C$ T_5 = Low temperature in refrigerator $(4\pm1)^{\circ}C$

Treatments combination	Crude fibre (%) at different days					Lipid content (%) at different days				
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12
V ₁ T ₀	1.34 a	1.13 de	0.93 f	0.72 g	0.48 h	0.18 a-c	0.47 a	0.77 a	0.83 a	0.72 cd
V ₁ T ₁	1.25 cd	1.23 ab	1.15 ab	1.03 ab	0.92 b	0.17 b-d	0.25 cd	0.35 d-f	0.53 de	0.71 cd
V ₁ T ₂	1.28 bc	1.17 b-d	1.04 d	0.88 de	0.71 e	0.19 a-c	0.29 bc	0.42 c	0.67 c	0.80 ab
V ₁ T ₃	1.30 a-c	1.21 bc	1.11 bc	0.98 bc	0.84 c	0.15 d-f	0.24 c-e	0.36 c-f	0.56 d	0.74 c
V ₁ T ₄	1.32 ab	1.15 c-e	0.96 ef	0.78 f	0.58 g	0.20 ab	0.33 b	0.49 b	0.76 b	0.81 a
V ₁ T ₅	1.33 ab	1.28 a	1.18 a	1.06 a	0.96 a	0.21 a	0.26 cd	0.34 d-f	0.42 f	0.52 f
V ₂ T ₀	1.21 de	0.96 g	0.76 h	0.56 h	0.36 i	0.15 d-f	0.45 a	0.72 a	0.77 b	0.67 de
V ₂ T ₁	1.18 e	1.10 ef	1.05 cd	0.93 cd	0.81 c	0.14 d-g	0.22 de	0.32 f	0.49 e	0.64 e
V ₂ T ₂	1.19 e	1.09 ef	0.96 ef	0.82 ef	0.65 f	0.16 c-e	0.26 cd	0.39 c-e	0.63 c	0.75 bc
V ₂ T ₃	1.16 e	1.11 d-f	1.01 de	0.88 de	0.74 d	0.13 e-g	0.20 de	0.33 ef	0.53 de	0.69 c-e
V ₂ T ₄	1.15 e	1.05 f	0.85 g	0.67 g	0.47 h	0.11 g	0.24 c-e	0.40 cd	0.67 c	0.72 cd
V ₂ T ₅	1.16 e	1.14 de	1.07 cd	0.96 c	0.84 c	0.12 fg	0.18 e	0.25 g	0.33 g	0.43 g
Level of significance	**	*	*	***	***	***	***	***	***	**
CV%	1.87	1.84	2.07	1.33	1.50	6.90	3.71	2.40	1.76	1.42

Table 1.10 Combined effects of varieties and different storage treatments on crude fibre and lipid content of postharvest mango

k.

+

1

K

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents

4

1

97

÷.

$V_1 = Langra$	T_1 =Paraffin coating	T₄ =Hot water treatments (55±2 ° C	** indicate at 1% level
$V_2 = Khirshapat$	T ₂ =Perforated polyethylene cover	$T_s = Low temperature in Refrigerator (4 \pm 1)^{\circ} C$	***indicates at 0.1%
$T_0 = Control$	T ₃ =Unperforated polyethylene cover	 indicates at 5% level 	NS non significant

Se.

K.

+

*

after storage (Appendix 1.9). There was found an increasing trend of lipid content in mango pulp with the advancement of time of storage (Table 1.10). It also denoted that lipid content was produced comparatively more quantity from the treatment combination of V_1T_0 at initial to 9th day and then, it decreased due to becoming rotten situation. At this time, lower quantity was reported from the treatment combination of V_2T_5 . At 12th day, the highest (0.81%) quantity of lipid was recorded from the treatment combination of V_1T_4 whereas; the lowest (0.43%) was recorded from the treatment combination of V_2T_5 .

7.3.10 Water soluble protein content

The analysis of variance of mango varieties demonstrated highly significant variation in respect of water soluble protein content (WSPC) in mango at different days after storage (Appendix 1.10). It was found an increasing trend of WSPC at time being passed (Fig. 1.27). It also exhibited that Langra was better in WSPC accumulation as compared to Khirshapat at all stages of storage. At 12th day, the highest (1.20%) accumulation of WSPC was reported from Langra whereas the lowest (1.08%) was reported from Khirshapat. This might be possible due to the seed protein of mango extended to pulp areas during ripening stage.

Various storage treatments imposed to this investigation in terms of WSPC exhibited significant variation at different days after storage (Appendix 1.10). Various results obtained from biochemical analysis showed an increasing trend of WSPC in mango pulp with the advancement of storage period (Fig. 1.29). It was also denoted that WSPC was synthesized comparatively more amount in untreated fruits followed by the fruits treated with T_4 , T_2 , T_3 , T_1 and T_5 treatment, respectively. The increasing trend of WSPC in control was sharp from initial to 6 days, thereafter, its increasing trend declined up to 9th day and then, it totally decreased due to becoming rotten condition. At the same time, the increasing trend of WSPC from the fruit treated with low temperature in refrigerator was very slower due to delay ripening. At 9th day, the highest (1.37%) quantity of WSPC was obtained from control whereas; the lowest (0.76%) was noted from low temperature in refrigerator.

3

+

-

The results of the present investigation were supported by the findings of Gomez-lim (1997) and Anon (1962).

The combined effect of varieties and implied storage treatments in this experiment were found to be significant in respect of WSPC at different days after storage (Appendix 1.10). The results obtained from the investigation demonstrated that WSPC increased markedly in the treatment combination of V_1T_0 from initial to 9 days and then, it decreased due to beginning rotten situation whereas; the lower increasing trend was found from the treatment combination of V_2T_5 , respectively. At 12th day, the highest (1.40%) quantity of WSPC was noticed from the treatment combination of V_1T_4 whereas; the lowest (0.87%) was noticed from the treatment combination of V_2T_5 (Table 1.11).

7.4 Changes in minerals of mango during storage environments

Different changes regarding minerals of mango fruits are presented and discussed in the following sub-headings.

7.4.1 Phosphorus content

Variation between varieties means in terms of P content was found to be highly significant at different days after storage (Appendix 1.10). There exhibited a steadily increasing trend of P content in both the varieties with the advancement of storage period (Fig. 1.28). It also indicated that Langra produced comparatively more quantity of P as compared to Khirshapat. The increasing trend of P in Langra was more or less similar up to 9 days thereafter, its trend declined due to becoming rotten condition. At 12th day, higher (23.08 mg/100 g) quantity of P content was recorded from Langra and lower (20.30 mg/100 g) was recorded from Khirshapat. Better performance in increased level of P content from Langra might be due to genetical dissimilarities. It also elucidated that the increase of P content in mango pulp was intimately related to ripening during storage. These results are in partially supported by the findings of Nadkarni (1963) worked with 16 cultivars. He found the range between 10-30 mg/100 g.

Difference between various storage treatments adopted to this trial in relation to P content were highly significant at different days after storage (Appendix 1.10).

SF

K

4

It was observed that P content increased steadily at various days after storage. It also indicated that P content from control was increased at slower rate from initial to 6 days and then; it increased more slowly rate than its previous trend. After 9^{th} day, it declined possibly due to becoming decay (Fig. 1.30). At 12^{th} day, the highest (23.43 mg/100 g) quantity of P content was reported from the fruit treated with hot water treatment and the lowest value (19.65 mg/100 g) was obtained from the fruit treated with low temperature in refrigerator. Lower amount of P content from the fruits treated with low temperature in refrigerator might be due to the delay ripening which resulted in lower accumulation of P content and keeping the quality best. The results of the present investigation were strongly supported by the reports of Peter *et al.* (2007).

The combined effect of varieties and imposed storage treatments exhibited significant variation on P content of mango pulp at different days after storage except at initial day (Appendix 1.10). There was found an increasing trend of P content with the passing of storage period (Table 1.11). It also indicated that P content increased gradually from initial to 9 days with the treatment combination of V_1T_0 and then, it decreased due to becoming spoilage. In this time, P content showed lower increasing trend from the treatment combination of V_2T_5 . At 12^{th} day, the highest (24.49 mg/100 g) quantity was recorded from the treatment combination of V_1T_4 whereas; the lowest (18.25 mg/100 g) was recorded from the treatment combination of V_2T_5 . It also denoted that Langra with low temperature in refrigerator was found to be the best in keeping the quality preservation followed by the other treatment combinations.

7.4.2 Potassium content

The analysis of variance of mango varieties in terms of potassium (K) content was found to be highly significant at different days after storage (Appendix 1.11). There was found an increasing trend of K content in both the varieties with the passing of storage period at different days after storage (Fig. 1.31). In all the storage period, Langra was noticed as a more producer of K as compared to Khirshapat. It was also observed that increasing trend of K stopped at 9th day. At this period, the highest (0.30%) K was recorded from Langra and the lowest (0.27%) was recorded from Khirshapat. These phenomena might be probably due to rotten

Se

X-

4

-



Fig. 1.27 Effect of varieties on water soluble protein content of mango pulp at different days after storage



Fig. 1.28 Effect of varieties on phosphorus content of mango pulp at different days after storage



Fig. 1.29 Water soluble protein content of mango pulp as influenced by storage treatments at different days after storage. Vertical bars represent LSD at 0.05 level

Figures represent	
$V_1 = Langra$	T ₁ =Paraffin coating
$V_2 = Khirshapat$	T ₂ =Perforated polyethylene cover
$T_0 = Control$	T ₃ =Unperforated polyethylene cover





 T_4 =Hot water treatments (55±2)^oC T_5 = Low temperature in refrigerator (4±1)^oC

Treatments combination	Water	soluble pr	otein cont	ent (%) at o	lifferent days	Phosphorus (mg/100 g) different days				
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12
V ₁ T ₀	0.65 a	0.92 a	1.31 a	1.45 a	1.36 ab	19.75 a	21.87 a	24.29 a	24.62 a	23.13 c
V ₁ T ₁	0.62 ab	0.65 de	0.72 ef	0.84 f	1.04 g	19.45 cd	19.98 d	20.52 e	21.36 f	22.62 d
V ₁ T ₂	0.66 a	0.72 c	0.85 d	1.05 d	1.32 bc	19.55 bc	20.46 c	21.43 c	22.66 c	24.22 b
V ₁ T ₃	0.55 cd	0.66 d	0.76 e	0.92 e	1.12 f	19.35 de	20.08 d	20.84 d	21.83 e	22.98 c
V ₁ T ₄	0.58 bc	0.74 c	0.92 c	1.17 c	1.40 a	19.65 ab	20.85 b	22.38 b	24.23 b	24.49 a
V ₁ T ₅	0.55 cd	0.62 f	0.69 fg	0.79 fg	0.94 h	19.25 e	19.58 e	19.92 g	20.44 g	21.06 g
V ₂ T ₀	0.52 c-e	0.78 b	1.18 b	1.28 b	1.22 de	16.85 f	18.95 f	21.38 c	21.72 e	20.21 h
V ₂ T ₁	0.48 e	0.54 h	0.62 h	0.74 gh	0.94 h	16.35 ij	16.86 j	17.39 j	18.22 j	19.47 i
V ₂ T ₂	0.52 c-e	0.58 g	0.73 ef	0.93 e	1.18 ef	16.65 g	17.55 h	18.52 h	19.79 h	21.35 f
V ₂ T ₃	0.49 de	0.56 g h	0.66 gh	0.82 f	1.02 g	16.45 hi	17.18 i	17.94 i	18.89 i	20.12 h
V ₂ T ₄	0.48 e	0.62 ef	0.82 d	1.05 d	1.26 cd	16.58 gh	18.75 g	20.28 f	22.13 d	22.38 e
V ₂ T ₅	0.51 de	0.53 h	0.62 h	0.72 h	0.87 i	16.22 j	16.62 k	16.96 k	17.48 k	18.25 j
Level of significance	***	**	*	***	*	NS	***	***	***	***
CV%	1.83	1.58	1.70	1.19	1.83	0.58	0.55	0.52	0.49	0.48

Table 1.11 Combined effects of varieties and different storage treatments on water soluble protein and phosphorus content of postharvest mango

4

Y

4

×

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents

21

4.

$V_1 = Langra$	T_1 =Paraffin coating	T₄=Hot water treatments (55±2 ⁰ C	** indicate at 1% level
V ₂ = Khirshapat	T ₂ =Perforated polyethylene cover	T_5 = Low temperature in Refrigerator $(4\pm1)^{\circ}$ C	***indicates at 0.1%
T ₀ =Control	T ₃ =Unperforated polyethylene cover	 indicates at 5% level 	NS non significant

102

Ν

1-

1

x

situation which esulted in metabolic activities suspension. Langra was the better producer possibly due to genetical dissimilarities between the two varieties. At storage period, increase of K content might be probably due to transmission of K from stone and peel to pulp of mango.

Storage treatments applied to this experiment in respect of K content of mango had highly significant at different days after storage (Appendix 1.11). It indicated that K content increased gradually with the advancement of storage time. But, K content from the untreated fruits decreased after 9th day whereas, other treatments namely, T_4 , T_2 , T_3 , T_1 and T_5 retained its increasing with days (Fig. 1.33). In this period there was found a very lower trend of K increasing at the fruit treated with low temperature in refrigerator. At 9th day, the maximum (0.32%) of K content was noticed from the untreated fruits whereas; the lowest (0.24%) was noticed from the fruits treated with low temperature in refrigerator. Lower dissemination of K content in the fruits treated with low temperature in refrigerator might be possible due to delay ripened causing lower transmission of K.

The combined effect of implied varieties and induced storage treatments in relation to K content of mango pulp was found to be non significant at different days of storage (Appendix 1.11). It denoted that K content of mango pulp from different treatment combination increased gradually with passing of storage time (Table 1.12). But, only the treatment combination of V_1T_0 gave the highest decreasing trend after 9 days. At 9th day, the highest (0.33%) quantity of K was recorded from the treatment combination of V_1T_0 whereas; the lowest (0.23%) was recorded from the treatment combination of V_2T_5 .

7.4.3 Calcium content

The analysis of variance of imposed varieties to this experiment in terms of Ca content demonstrated highly significant variation at different days after storage (Appendix 1.11). There was found an increasing trend of Ca with the passing of storage time from both the varieties (Fig. 1.32). It also denoted that Khirshapat showed better performance in achieving of Ca content over Langra. At 12th day the highest (21.71%) quantity of Ca was obtained from Khirshapat and the lowest (19.97%) was recorded from Langra. This might be due to genetical variation

F.

X

1

4

x

between two varieties. These results are in partially supported by the report of Nadkarni (1963), when he worked with different varieties of mango.

Different storage treatments on Ca of mango pulp exhibited significant variation at different days after storage (Appendix 1.11). At various days, Ca content increased gradually with the passing of storage time (Fig. 1.34). It also elucidated that Ca content in control increased sharply from initial to 6th day and then, it increased smoothly and thereafter, it decreased due to becoming rotten. At the same time, Ca content from the fruits treated with low temperature in refrigerator increased very smoothly with the passing of time. At 9th day, the highest (24.40 mg/100 g) quantity of Ca was obtained from untreated fruits whereas; the lowest (14.50 mg/100 g) was obtained from the fruits treated with low temperature in refrigerator. These events might be possible due to delay ripening caused in lower transmission of Ca content from peel and stone to pulp of mango, but, these results are not supported by the findings of Peter *et al.* (2007).

The combined effect of varieties and imposed storage treatments in terms of Ca content of mango pulp showed non significant variation at different storage treatments (Appendix 1.11). There was found a light increasing trend of Ca content of mango pulp from various treatment combinations (Table 1.12). After 9th day, the maximum (25.27 mg/100 g) quantity of Ca was recorded from the treatment combination of V₂T₀ and the minimum (13.64 mg/100 g) was recorded from the treatment treatment combination of V₁T₅.

7.4.4 Magnesium content

Significant variation was observed in mg content between the varieties at different days after storage (Appendix 1.12). It was noticed that Mg content increased gradually with the passing of storage period (Fig. 1.35). It also indicated that Mg content increased steadily from initial to 9th day, thereafter, it decreased slightly due to rotten situation. At 9th day, the highest (17.85 mg/100 g) was recorded in Langra and the lowest (16.66 mg/100 g) was recorded in Khirshapat. This occurrence might be possible due to genetical dissimilarities between two varieties. There were no available research articles relating to Mg content in the scientific literature.

F

I.

4

×





Fig. 1.31 Effect of varieties on potassium content of mango pulp at different days after storage



Fig. 1.33 Potassium content of mango pulp as influenced by storage treatments at different days after storage. Vertical bars represent LSD at 0.05 level

Figures represent

 $V_1 = Langra$ $V_2 = Khirshapat$ $T_0 = Control$

T₁=Paraffin coating pat T₂=Perforated polyethylene cover T₃=Unperforated polyethylene cover

Fig. 1.32 Effect of varieties on calcium content of mango pulp at different days after storage



Fig. 1.34 Calcium content of mango pulp as influenced by storage treatments at different days after storage. Vertical bars represent LSD at 0.05 level

 T_4 =Hot water treatments $(55\pm2)^{\circ}C$ T_5 = Low temperature in refrigerator $(4\pm1)^{\circ}C$
Treatments Potassium content (%) at different days Calcium content (mg/100 g) at different days combination 3 6 9 12 3 9 12 Varieties × Treatments Initial Initial 6 D 0.27 a 0.30 a 0.33 a 0.31 ab 22.12 b 23.52 b 20.13 h V₁T₀ 0.32 a 10.75 d-f 15.32 c 19.62 i 17.39 i V_1T_1 0.23 b-d 0.25 d-f 0.27 d-f 0.28 cd 0.29 bc 10.65 ef 12.16 i 14.28 j 22.05 d 10.82 de 19.92 f V_1T_2 0.24 bc 0.28 a-c 0.30 a-c 0.31 ab 0.32 a 13.34 q 16.45 q 0.22 cd 0.27 b-d 18.34 h 20.63 g V_1T_3 0.29 b-d 0.30 bc 0.31 ab 10.72 d-f 12.75 h 15.25 i V₁T₄ 0.25 b 18.45 d 21.15 e 22.23 c 0.29 ab 0.31 ab 0.32 ab 0.33 a 10.85 d 14.34 e 15.16 k V_1T_5 0.22 cd 0.23 f-h 0.24 gh 0.25 ef 0.26 de 10.62 f 10.92 j 12.13 13.64 k 0.23 cd 0.26 c-e 0.28 c-e 0.30 bc 0.28 cd 12.58 ab 17.14 a 23.86 a 25.27 a 21.78 e V₂T₀ 16.07 h 21.29 f V_2T_1 0.19 e 0.22 gh 0.24 gh 0.25 ef 0.26 de 12.45 a-c 13.95 f 19.18 g 21.73 d 23.86 b 0.22 cd 0.27 de 0.28 cd 12.62 a 15.12 d 18.24 e V_2T_2 0.24 e-q 0.26 e-q 20.05 f 22.36 c V_2T_3 0.21 d 0.23 f-h 0.25 fg 0.26 de 0.27 cd 12.42 bc 14.45 e 16.95 f 0.21 d 0.25 d-f 0.27 d-f 0.28 cd 12.53 a-c 16.08 b 20.18 c 22.87 c 24.11 a V₂T₄ 0.28 cd V₂T₅ 0.18 e 0.21 h 0.22 h 0.23 f 0.24 e 12.36 c 12.65 h 13.85 k 15.36 j 16.86 j NS NS Level of significance NS NS NS NS NS NS NS NS CV% 4.72 3.84 0.60 0.52 0.50 4.12 3.70 3.64 0.90 0.74

Table 1.12 Combined effects of varieties and different storage treatments on potassium and calcium content of postharvest mango

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents

X

4

106

4

 $V_1 = Langra$ $T_1 = Paraffin coating$ $V_2 = Khirshapat$ $T_2 = Perforated polyethylene cover$

 T_0 = Control T_3 =Unperforated polyethylene cover * indicates at 5% level

 T_4 =Hot water treatments (55±2 ° C T_5 = Low temperature in Refrigerator (4±1) ° C * indicates at 5% level ** indicate at 1% level
***indicates at 0.1%
NS non significant

4

*

K

Results and Discussion

Chapter 4: Experiment 1

*

*

2

I

X

Different storage treatments induced in this trial had highly significant variation on Mg content of mango pulp at different days after storage (Appendix 1.12). There was found an increasing trend of Mg content of mango pulp with the advancement of storage period (Fig. 1.37). It also showed that Mg content from untreated and hot water treated fruit augmented very markedly from initial to 6 days and from initial to 9 days respectively and then, it decreased very sharply. The highest (18.28, 18.15 and 17.93 mg/100 g) quantity of Mg content was obtained from control, hot water treatment and perforated polyethylene cover at 6, 9 and 12 days of storage whereas, the lowest (16.24, 16.55 and 16.94 mg/100 g) was obtained from fruit treated with low temperature in refrigerator, respectively. There might have a relation of augmentation of Mg content of mango pulp to ripening event. The results of the present experiment are inconformity with the findings of Peter *et al.* (2007).

Different varieties used and storage treatments imposed in this experiment in terms of Mg content of mango pulp exhibited significant variation at various days after storage except at 9 and 12^{th} days (Appendix 1.12). There was found an increasing trend of Mg content of mango pulp with the passing of storage time (Table 1.13). At 6th day, the highest (18.92 mg/100 g) quantity of Mg was observed from the treatment combination V₁T₀ whereas; the lowest (15.53 mg/100 g) was recorded from the treatment combination of V₂T₅.

7.4.5 Copper content

Variation between varieties in relation to Cu content of mango pulp was highly significant at different days after storage (Appendix 1.12). It revealed that Cu content of mango pulp decreased gradually with the passing of storage period (Fig. 1.36). It also denoted that Cu content of Khirshapat was comparatively more as compared to Langra. At initial day higher (0.36 mg/100 g) amount was recorded from Khirshapat and lesser amount (0.33 mg/100 g) was recorded from Langra. At 12th day, Khirshapat gave the highest (0.20 mg/100 g) amount of Cu content and the lowest (0.18 mg/100 g) was obtained from Langra. These phenomena might be possible due to genetical dissimilarities between the varieties. There were no available research reports regarding Cu content of mango at the scientific literature,

*

X

1

I

×

-

but, the results of the present study revealed that Cu content diminished during storage.

The postharvest treatments used in the present study in relation to Cu content of mango pulp were observed to be highly significant at different days after storage (Appendix 1.12). It explored that Cu content decreased gradually with the advancement of storage period at various days of storage. The decreasing trend of untreated fruits was much higher than the treated fruits with T_4 , T_2 , T_3 , T_1 and T_5 . At 12^{th} day, the highest quantity of Cu (0.26 mg/100 g) was recorded from the fruits treated with low temperature in refrigerator whereas; the lowest (0.14 mg/100 g) was recorded from control (Fig. 1.38). This means low temperature in refrigerator was much better treatment followed by the others. Its preservative quality was also found to be the best.

The combined effect of used varieties and imposed storage treatments in terms of Cu content of mango exhibited non significant variation at various days after storage except at initial day (Appendix 1.12). It indicated that Cu content from different treatment combination decreased with the advancement of storage duration. At 12^{th} day, the highest (0.27 mg/100 g) was obtained from the treatment combination of V₂T₅, and the lowest (0.13 mg/100 g) was obtained from the treatment treatment combination of V₁T₀ (Table 1.13).

7.4.6 Iron content

The analysis of variance of varieties on iron (Fe) content of mango pulp showed highly significant at different days after storage (Appendix 1.13). There was found an increasing trend of Fe content with the passing of storage duration. The increasing trend was more or less similar from initial to 9th day and thereafter, it increased very smoothly. At 12th day, the highest (4.24 mg/100 g) quantity was noticed from Langra and the lowest (3.18 mg/100 g) was noticed from Khirshapat (Fig. 1.39). These results are in partially supported by the findings of Nadkarni (1963), when he worked with different varieties of mango fruits.

There was found significant variation due to the effect of storage treatments on Fe content of mango pulp at various days after storage (Appendix 1.13).

2

4

×

2

T

9





Fig. 1.35 Effect of varieties on magnesium content of mango pulp at different days after storage







Fig. 1.37 Magnesium content of mango pulp as influenced by storage treatments at different days after storage. Vertical bars represent LSD at 0.05 level

Fig. 1.38 Copper content of mango pulp as influenced by storage treatments at different days after storage. Vertical bars represent LSD at 0.05 level

Figures represent
V1 = Langra

vi – Langra
$V_2 = Khirshanat$
T Control
$I_0 = Control$

 T_1 =Paraffin coating pat T_2 =Perforated polyethylene cover T_3 =Unperforated polyethylene cover T_4 =Hot water treatments (55±2)^o C T_5 = Low temperature in refrigerator (4±1)^o C

Treatments combination	Magn	iesium con	tent (mg/1	00 g) at diffe	erent days	Сор	per conten	t (mg/100	g) at differ	rent days
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12
V ₁ T ₀	16.85 a	17.95 a	18.92 a	17.82 c	16.72 h	0.35 b-d	0.27 f	0.23 f	0.18 h	0.13 h
V ₁ T ₁	16.35 e	16.62 d	16.95 f	17.43 e	17.94 c	0.32 e	0.31 b-d	0.29 bc	0.26 bc	0.22 bc
V ₁ T ₂	16.45 de	16.88 c	17.35 d	18.21 b	18.52 a	0.33 de	0.29 d-f	0.25 d-f	0.21 e-g	0.16 e-f
V ₁ T ₃	16.55 cd	16.75 cd	17.14 e	17.65 cd	18.24 b	0.32 e	0.30 c-e	0.27 cd	0.23 de	0.18 de
V ₁ T ₄	16.75 ab	17.25 b	17.92 b	18.73 a	17.94 c	0.34 c-e	0.28 ef	0.24 ef	0.19 gh	0.14 gh
V ₁ T ₅	16.65 bc	16.72 cd	16.94 f	17.25 f	17.65 d	0.33 de	0.32a-c	0.30 ab	0.28 ab	0.24 b
V ₂ T ₀	15.52 f	16.64 d	17.65 c	16.55 h	15.49 j	0.38 a	0.29 d-f	0.25 d-f	0.20 f-h	0.14 gh
V ₂ T ₁	15.36 f	15.58 g	15.92 i	16.39 h	16.92 g	0.36 bc	0.33 ab	0.31 ab	0.28 ab	0.24 b
V ₂ T ₂	15.45 f	15.89 f	16.38 h	17.03 g	17.34 e	0.37 ab	0.31 b-d	0.27 cd	0.22 ef	0.17 ef
V ₂ T ₃	15.15 g	15.65 g	16.04 i	16.55 h	17.15 f	0.35 b-d	0.32 a-c	0.29 bc	0.25 cd	0.20 cd
V ₂ T ₄	15.45 f	16.08e	16.75 g	17.58 de	16.79 gh	0.36 a-c	0.30 c-e	0.26 de	0.21 e-g	0.15 f-h
V ₂ T ₅	15.15 g	15.32 h	15.53 j	15.84 i	16.23 i	0.35 b-d	0.34 a	0.32 a	0.30 a	0.27 a
Level of significance	***	*	*	NS	NS	NS	NS	NS	NS	NS
CV%	0.65	0.63	0.61	0.60	0.63	3.00	3.41	3.18	4.44	5.58

Table 1.13 Combined effects of varieties and different storage treatments on magnesium and copper content of postharvest mango

F

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Tabl e represents		
$V_1 = Langra$	T_1 =Paraffin coating	T ₄ =Hot water treatments (5
V ₂ = Khirshapat	T ₂ =Perforated polyethylene cover	T ₅ = Low temperature in Ref
T ₀ =Control	T ₃ =Unperforated polyethylene cover	* indicates at 5% level

4

15)

110

4)

55±2 ° C ** indicate at 1% level frigerator (4±1)° C ***indicates at 0.1% NS non significant

+

X

*

-

X

24

Th-

G.

It denoted that Fe content increased very sharply from initial to 6 days and then, it decreased markedly at control. But, Fe content of fruit from the other treatments such as T_1 , T_2 , T_3 , T_4 and T_5 increased gradually whereas, T_5 treatment showed least increasing tendency at all storage duration (Fig. 1.41). At 12th day, the highest (4.71 mg/100 g) quantity of Fe was noticed from the fruits treated with hot water treatment whereas; the lowest (2.28 mg/100 g) was noticed from the untreated mangoes. The low increasing trend of the fruits treated with low temperature in refrigerator was possibly due to the delay ripening which resulted in lower increasing trend. These results are in partially agreement with the findings of Peter *et al.* (2007).

There was non significant variation in respect of Fe content of mango pulp at different days after storage as influenced by the treatment combination of varieties and implied storage treatments (Appendix 1.13). An increasing trend of Fe content was also noticed from initial to 6 days thereafter, it decreased rapidly with the treatment combination of V_1T_0 whereas; the lowest increasing trend was observed with the treatment combination of V_2T_5 (Table 1.14). At 6th day, the highest (5.24 mg/100 g) quantity of Fe was reported from the treatment combination of V_2T_5 (Table 1.14). At 6th day, the highest (5.24 mg/100 g) quantity of Fe was reported from the treatment combination of V_2T_5 (Table 1.14).

7.4.7 Manganese content

The manganese (Mn) content of mango pulp was observed to differ significantly in both the varieties at different days after storage (Appendix 1.13). It was found that Mn content increased gradually with the increase of storage period (Fig. 1.40). It also revealed that Langra was better in the performance of Mn accumulation as compared to Khirshapat. At 12th day, the highest (1.21 mg/100 g) quantity of Mn was recorded from Langra whereas; the lowest (1.06 mg/ 100 g) was recorded from Khirshapat. There were no available research reports relating to Mn content in the scientific literature.

Different storage treatments had also highly significant effect on Mn content of mango pulp at different days after storage (Appendix 1.13). It was found that Mn content increased gradually from initial to 6 days, and then, it decreased markedly in control. On the other hand, it increased from initial to 6 days thereafter, it decreased

*

X

Y

*

jk-

2

in control. At the same time, very lower increasing trend of Mn was noticed from the fruit-treated with low temperature in refrigerator (Fig. 1.42). At 12th day, Mn content ranged from 0.89 to 1.31 mg per 100 g of fresh pulp of mango. The maximum (1.31 mg/100 g) was recorded from perforated polyethylene cover treated fruits which was statistical identical with the fruits treated with unperforated polythene cover. The minimum (0.89 mg/100 g) was found from untreated fruits.

The combined effects of varieties and used different storage treatments in terms of Mn content of mango pulp were found to be significant at various days after storage except at 6, 9 and 12^{th} days (Appendix 1.13). There was found an increasing trend of Mn content in different treatment combination with the increase of storage duration (Table 1.14). At 12^{th} day, the highest (1.38 mg/100 g) was reported from the treatment combination of V₁T₂, and the lowest (0.92 mg/100 g) was obtained from the treatment combination of V₂T₅.

7.4.8 Zinc content

Variation in respects of zinc (Zn) content of mango pulp due to the effect of varieties showed highly significant at different days after storage (Appendix 1.14). It was observed that Zn content decreased markedly with the increasing of storage duration from both the varieties (Fig. 1.43). It also revealed that Khirshapat noticed the signal of better performance in achieving comparatively more Zn content over Langra. Zn content ranged between 1.31 to 1.49 mg per 100 g of mango pulp at initial day and 0.79 to 0.97 mg per 100 g at 12th day. The highest (1.49 and 1.31 mg/100 g) was recorded in Khirshapat and the lowest (0.97 and 0.79 mg/100 g) was recorded from Langra at initial and 12th days. These events might be possible due to genetical dissimilarities.

Different postharvest storage treatments caused significant differences in Zn content of mango pulp at different days after storage except at 3rd day (Appendix 1.14). There was found a decreasing trend of Zn content with the increase of storage period from the fruits treated with different storage treatment. It also explored that the decreasing trend was very high in control and very low in the fruits treated with low temperature in refrigerator (Fig. 1.45). At 12th day, the maximum (1.06 mg/100 g) was noticed from the fruits treated with low temperature in refrigerator which was 1.38 mg per 100 g of fresh mango pulp at initial level and the lowest (0.61 mg/100 g) was noticed from control which was 1.45 mg per 100 g of mango pulp at initial

F

X

1

u

4

Results and Discussion





Fig. 1.39 Effect of varieties on iron content of mango pulp at different days after storage



Fig. 1.41 Iron content of mango pulp as influenced by storage treatments at different days after storage. Vertical bars represent LSD at 0.05 level

Figures represent $V_1 = Langra$ $V_2 = Khirshapat$ $T_0 = Control$

T1=Paraffin coating T2=Perforated polyethylene cover T3=Unperforated polyethylene cover

Fig. 1.40 Effect of varieties on manganese content of mango pulp at different days after storage



Fig. 1.42 Manganese content of mango pulp as influenced by storage treatments at different days after storage. Vertical bars represent LSD at 0.05 level

 T_4 =Hot water treatments $(55\pm2)^{0}$ C T_5 = Low temperature in refrigerator $(4\pm1)^{0}$ C

	Treatments	Ir	Iron content (mg/100 g) at different days					Manganese content (mg/100 g) at different da			
	combination						1				
Vari	ieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12
	V ₁ T ₀	2.65 a	3.74 a	5.24 a	4.69 a	2.74 g	0.68 a	1.02 a	1.37 a	1.22 bc	0.98 f
	V_1T_1	2.35 bc	2.68 e	3.11 f	3.66 d	4.18 c	0.66 bc	0.77 d	0.92 ef	1.08 d-f	1.25 bc
	V ₁ T ₂	2.55 a	3.08 c	3.72 d	4.48 b	5.12 a	0.64 d	0.82 b	1.03 cd	1.25 ab	1.38 a
	V_1T_3	2.49 ab	2.92 cd	3.44 e	4.09 c	4.72 b	0.65 cd	0.79 c	0.96 de	1.14 cd	1.32 ab
	V ₁ T ₄	2.59 a	3.32 b	3.94 c	4.66 a	5.25 a	0.67 ab	0.82 b	1.08 c	1.33 a	1.19 cd
	V ₁ T ₅	2.25 c	2.38 f	2.63 g	2.98 f	3.42 e	0.62 e	0.72 e	0.84 fg	0.98 fg	1.12 de
	V ₂ T ₀	1.65 d	2.75 de	4.25 b	3.72 d	1.82 i	0.49 g	0.83 b	1.18 b	1.03 e-g	0.79 g
	V ₂ T ₁	1.25 f	1.58 h	2.02 i	2.57 g	3.09 f	0.52 f	0.63 h	0.78 g	0.94 g	1.11 de
	V ₂ T ₂	1.45 e	1.97 g	2.62 g	3.38 e	4.02 c	0.45 hi	0.68 f	0.82 fg	1.12 c-e	1.24 bc
	V ₂ T ₃	1.52 de	1.88 g	2.42 h	3.07 f	3.74 d	0.44 i	0.65 g	0.82 fg	1.01 fg	1.19 cd
	V ₂ T ₄	1.59 de	2.28 f	3.12 f	3.84 d	4.16 c	0.46 h	0.69 f	0.95 de	1.22 bc	1.08 e
	V ₂ T ₅	1.12 f	1.26 i	1.48 j	1.83 h	2.22 h	0.42 j	0.52 i	0.64 h	0.77 h	0.92 f
	Level of significance	NS	NS	NS	NS	NS	***	***	NS	NS	NS
	CV%	5.32	4.19	3.26	2.91	2.81	1.86	1.55	5.72	4.77	4.60

Table 1.14 Combined effects of varieties and different storage treatments on iron and manganese content of postharvest mango

148

7

12

R.

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents

\$

1

4)

$V_1 = Langra$	T ₁ =Paraffin coating	T₄=Hot water treatments (55±2 ° C	** indicate at 1% level
$V_2 = Khirshapat$	T ₂ =Perforated polyethylene cover	T_5 = Low temperature in Refrigerator $(4\pm1)^{\circ}$ C	***indicates at 0.1%
$T_0 = Control$	T ₃ =Unperforated polyethylene cover	 indicates at 5% level 	NS non significant

n-

2

4

the

*

day. Zn content decreased in the storage period was possibly due to transmission from pulp to stone and metabolic activities of stored mango may be depressed or suppressed to the presence of Zn content. There was no available research articles regarding Zn content of mango pulp in the scientific literature, but, the present results elucidated that Zn content decreased with the advancement of storage duration.

The combined effect of varieties and imposed storage treatments in terms of Zn content of mango demonstrated non significant variation at different days after storage (Appendix 1.14). There was also found the decreasing trend of Zn content of mango pulp from various treatment combinations (Table 1.15).

7.5 Shelf life

Highly significant difference was found from varieties on shelf life of mango (Appendix 1.14). The results revealed that shelf life ranged between 15.22 to 16.44 days (Fig. 1.44). Longer shelf life (16.44 days) was obtained in Khirshapat and the slightly shorter (15.22 days) was obtained in Langra (Fig. 1.44). Variation in shelf life between varieties might be possible due to genetical. These results indicated that Khirshapat was better in preservation of mango.

Difference between imposed storage treatments in this experiment in terms of shelf life of mango exhibited highly significant (Appendix 1.14). The shelf life of mango ranged between 7.83 to 36.50 days in different storage treatments (Fig. 1.46). The longest shelf life (36.50 days) was recorded from the fruits treated with low temperature in refrigerator followed by the shelf life of the fruits treated with paraffin coating (14.17 days), hot water treatment (12.00 days), perforated polyethylene cover (11.00 days) whereas the shortest shelf life (7.17 days) was recorded from untreated fruit (Fig. 1.46). The longest shelf life obtained from the fruits treated with low temperature in refrigerator might be possible due to Suppression or inhibition of physiological and biochemical activities in responding for slower senescence of harvested fruits and consequently led to the longest shelf life, but, mangoes were found deeply shrinkage condition. The results are inconformity with the findings of Tefera *et al.* (2007), Durigan *et al.* (2004) also found the similar results. Wanlapha *et al.* (1980) observed the longest shelf life of mango where they worked with mango under preservation at low temperature in refrigerator.

2.

T.

N.

1

2

F

Y





Fig.1.43 Effect of varieties on zinc content of mango pulp at different days after storage



Fig. 1.45 Zinc content of mango pulp as influenced by storage treatments at different days after storage. Vertical bars represent LSD at 0.05 level

Figures represent	
$V_1 = Langra$	T_1 =Paraffin coating
$V_2 = Khirshapat$	T ₂ =Perforated polyethylene cover
$T_0 = Control$	T ₃ =Unperforated polyethylene cover

Fig. 1.44 Effect of varieties on shelf life of mango



Fig. 1.46 Shelf life of mango as influenced by storage treatments. Vertical bars represent LSD at 0.05 level

 T_4 =Hot water treatments $(55\pm2)^{\circ}C$ T_5 = Low temperature in refrigerator $(4\pm1)^{\circ}C$

Results and Discussion

P

T

F

2

Y

Results and Discussion

The combined effect of varieties and induced storage treatments on shelf life of mango was found to be highly significant (Appendix 1.14). The shelf life ranged between 7.33 to 38.00 days from various treatment combinations. The longest shelf life (38.00 days) was noticed from the treatment combination of V_2T_5 , whereas the shortest (7.33 days) was noticed from V_1T_0 (table 1.15), but, mangos treated with low temperature in refrigerator were noticed deeply shrinkage condition. 1

118

12

4

r.

Treatments combination Zinc content (mg/100 g) at different days Shelf life Varieties × Treatments Initial 3 6 9 12 Total days 1.35 cd 0.97 e 7.33 f V₁T₀ 1.19 b 0.75 i 0.52 j V_1T_1 1.28 d 1.21 b 1.14 cd 1.04 d-f 13.67 cd 0.88 g V₁T₂ 0.96 fg 1.32 d 1.23 b 1.11 cd 0.78 h 10.67 e V_1T_3 1.27 d 1.22 b 1.12 cd 0.91 f 13.33 d 1.01 e-g V₁T₄ 1.33 d 1.20 b 1.04 de 0.86 h 0.67 i 11.33 e 1.28 d V_1T_5 1.25 b 1.18 bc 1.09 c-e 0.98 d 35.00 b V₂T₀ 1.55 a 8.33 f 1.35 a 1.13 cd 0.92 gh 0.69 i 1.45 ab 1.42 a V₂T₁ 1.34 a 1.22 ab 1.08 b 14.67 c V₂T₂ 1.48 ab 1.38 a 1.26 ab 1.12 b-d 0.94 e 11.33 e V_2T_3 1.43 bc 1.39 a 1.28 a 1.17 a-c 1.04 c 13.67 cd V₂T₄ 1.52 ab 1.36 a 1.30 a 1.12 cd 0.92 ef 12.67 d V₂T₅ 1.48ab 1.43 a 1.36 a 1.26 a 1.14 a 38.00 a Level of significance NS NS NS NS ** * CV% 3.73 4.00 4.39 4.99 2.37 4.36

Table 1.15 Combined effects of varieties and different storage treatments on zinc content and shelf life of postharvest mango

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

1

Table represents

 $V_1 = Langra$ T_1 =Paraffin coating V_2 = Khirshapat T_2 = Perforated polyethylene cover T₀ =Control

T₄=Hot water treatments (55±2 ° C $T_5 =$ Low temperature in Refrigerator $(4 \pm 1)^{\circ}$ C T₃=Unperforated polyethylene cover * indicates at 5% level

** indicate at 1% level ***indicates at 0.1% NS non significant

8.0 Experiment 2

3

X

T

N.

h

4

.

Influence of different doses of Maleic hydrazide on physicochemical changes and shelf life of mango during storage

8.1 Changes in skin color

External features, especially peel color is the most distinct change that occurs in many fruits and is frequent the major criterion followed by the consumers to identify whether the fruit is ripe or unripe. The most common change in case of mango is the loss of green color (Wills *et al.* 1989).

Different doses of Maleic hydrazide (MH) solution strongly affected the skin color of both the fruits namely, Langra and Khirshapat (Table 2.1). Both the varieties showed the original green color at the initial stage of harvesting. At 3^{rd} day, Langra turned in to a light green color at all doses except M₃ treatment, but Khirshapat retained its original green color.

At 6^{th} day, Langra was noticed to be yellowish green at control (M₀), 200 ppm (M₁) and 400 ppm (M₂) treatment and it held its original green color at 600 ppm (M₃) of MH solution. On the other hand, Khirshapat developed a trace in yellow color at M₀, but it retained its original green color at M₁, M₂ and M₃ treatments.

At 9th day, Langra gave yellow color at control, yellowish green at M_1 and M_2 as well as light green at M_3 treatment, respectively. Khirshapat demonstrated greenish yellow at control treatment, trace in yellow at M_1 and M_2 treatments. But, it also retained its original green color at M_3 treatment.

At 12th day of storage, Langra was reported as blackish yellow at control and yellow, greenish yellow and trace in yellow at M_1 , M_2 and M_3 treatments, respectively. On the other hand, Khirshapat was also noticed as deep yellow color at M_0 , greenish yellow, yellowish green and trace in yellow at M_1 , M_2 and M_3 treatments. After 15 days of storage, Langra exhibited no existence at control, blackish yellow, yellow and greenish yellow at M_1 , M_2 and M_3 treatments, respectively. Khirshapat showed completely rotten condition at M_0 but showed yellow, greenish yellow and M_1 , M_2 and M_3 treatments, respectively.

8.2 Changes in physical characters during storage environments

X

x

P

¥.

2

4

T

Different parameters relating to the physical changes of mango are presented and elucidated in the following sub-headings.

8.2.1 Physiological weight loss

Varieties showed highly significant in terms of PWL at different days after storage (Appendix 2.1). At each day, Khirshapat gradually showed comparatively more PWL as compared to Langra with the passing of storage duration (Fig. 2.1). The maximum (10.67%) and the minimum (9.78%) PWL were observed from Khirshapat and Langra at 12th day, respectively. The results also revealed that total PWL progressively increased with the increase of storage duration. The findings also indicated that Langra showed better performance in respect of PWL as compared to Khirshapat. Water loss through lenticel seems to be the probable cause of physiological weight loss in the fruits during storage. Lower lenticel density in Langra facilitated lesser water loss leading to minimum total weight loss (Azad, 2001). Singh *et al.* (2000) also reported more or less the identical findings.

Variation of the means as influenced by different doses of Maleic hydrazide (MH) showed highly significant on PWL (Appendix 2.1). At different days after storage, it revealed that the untreated fruit was very fast in PWL compared to M_1 , M_2 and M_3 treatments. It also denoted that untreated fruit showed in PWL sharply as compared to M_3 treatment (Fig. 2.3). At 12th day, the maximum PWL (11.65%) was recorded M_0 and minimum (9.19%) in M_3 treatment. These phenomena happened seems to be inhibited by 600 ppm of MH solution retarded the fruit ripening. These results are in partial coincided with the findings of Reddy and Haripriya (2002).

The combined effect of varieties and different doses of MH showed highly statistical significant on PWL at different days after storage (Appendix 2.1). At different days, it revealed that various treatment combinations showed in PWL gradually with the increase of storage duration. At 12^{th} day, there exhibited the maximum PWL (12.00%) at V₂M₀ and the minimum (8.75%) at V₁M₃ treatment combination. It also explored that Langra lost the minimum amount of water with 600 ppm (M₃) treatment followed by the other treatment combinations.

120

3

L

P

F.

7

8.2.2 Moisture content

The analysis of variance of imposed varieties showed highly significant on moisture content at different days (Appendix 2.1). At different days after storage, the results denoted that moisture content increased with the passing of storage duration (Fig. 2.2). The increasing trend was more or less similar from initial to 9 days and thereafter, the decrease in increasing tendency might be due to rotten. Figure 2.2 also revealed that on all the experimental days, Khirshapat showed more moisture comparing to Langra. Higher (88.63%) and the minimum (85.86%) were obtained from V₂ and V₁ at 12th day. Shajahan *et al.*, (1994) observed different results in this regard and the moisture content increased in Langra than Khirsapat. But, these results are in agreement with the findings of Azad (2001). This variation might be possible due to genetical, location, weather effect, soil quality or maturity of the fruit.

Variation among the means of different doses of MH solution on moisture content was observed to be highly significant at different days after storage except at initial day (Appendix 2.1). At different day, moisture content was increased gradually with the extension of storage duration (Fig. 2.4). Untreated fruit produced the highest moisture content (87.88%) at 9th day and thereafter; it decreased, whereas the lowest (85.82%) was recorded from M₃ treatment.

The combined effect of varieties and different doses of MH solution in terms of moisture content exhibited significant variation at different days (Appendix 2.1). At different days, moisture content increased with the passing of storage period. The treatment combination of V_2M_0 , V_2M_1 and V_2T_2 produced the maximum moisture content (*viz.*, 89.10, 89.25 and 89.70%) at 6, 9 and 12 days, respectively. In these storage periods, the lowest values (83.60, 84.40 and 85.20%) were observed from the treatment combination of V_1M_3 (Table 2.2). These treatment combinations provided slightly lesser amount of moisture content from 6 and 9th days. The increase of moisture content from initial to consequent days might be possible due to metabolic activities and osmotic principles and decreasing in a certain day might be due to suspending metabolic activities resulting in drying, transpiration and evaporation. Moisture content increased in the present study is in partial agreement with the findings of El-Mahmoudi and Eisawi (1968) in banana.

X

T

5

F

+

F

A











Fig. 2.3 Physiological weight loss of mango as influenced by different doses of Maleic hydrazide (MH) solution at different days after storage. Vertical bars represent LSD at 0.05 level

Graph represents

N

$V_1 = Langra$	M ₁ =200 ppm of MH solution
$V_2 = Khirshapat$	M ₂ =400 ppm of MH solution
$M_0 = Control$	$M_3 = 600 \text{ ppm of MH solution}$

Fig. 2.4 Moisture content of mango pulp as influenced by different doses of Maleic hydrazide (MH) solution at different days after storage. Vertical bars represent LSD at 0.05 level





Days after storage Varieties Treatments Initial 3 6 9 12 15 Mo Light green Yellowish green Yellow Blackish yellow Green Yellow Blackish yellow M₁ Green Light green Yellowish green Yellowish green M₂ Yellow V_1 Green Light green Yellowish green Yellowish green Greenish yellow M_3 Green Trace in yellow Greenish yellow Green Green Light green Mo Green Green Trace in yellow Greenish yellow Deep yellow Yellow M₁ Green Green Green Trace in yellow Greenish yellow M₂ Green Green Green Trace in yellow Yellowish green Greenish yellow V₂ Yellowish green Trace in yellow M₃ Green Green Green Green

Table 2.1 Changes in skin color of two mango varieties as influenced by different doses of Maleic hydrazide solution during storage at ambient condition

Yel.

1

×

٣.

Table 2.2 Combined effects of varieties and different doses of Maleic hydrazide solution on physiological weight loss of mango and moisture content of mango pulp during storage at ambient condition

Treatments combination	Physio	Physiological weight loss (%) at different days				Moisture content (%) at different days			
Varieties × Treatments	3	6	9	12	Initial	3	6	9	12
V ₁ M ₀	6.80 b	8.30 b	9.70 b	11.30 b	82.60 b	84.90 d	86.40 d	86.12 e	85.26 f
V ₁ M ₁	5.50 d	7.00 d	8.40 e	10.00 e	82.50 b	83.50 e	84.50 e	86.50 e	86.08 e
V ₁ M ₂	4.50 f	6.00 f	7.45 g	9.05 g	82.40 b	83.20 ef	84.00 f	84.90 f	86.90 d
V ₁ M ₃	4.25 g	5.75 g	7.15 h	8.75 h	82.30 b	82.90 f	83.60 g	84.40 g	85.20 f
V ₂ M ₀	7.50 a	9.00 a	10.40 a	12.00 a	85.30 a	87.60 a	89.10 a	88.82 b	87.96 c
V ₂ M ₁	6.45 c	7.95 c	9.35 c	10.95 c	85.23 a	86.25 b	87.25 b	89.25 a	88.82 b
V ₂ M ₂	5.60 d	7.00 d	8.50 d	10.10 d	85.17 a	86.00 bc	86.80 c	87.80 c	89.70 a
V ₂ M ₃	5.12 e	6.62 e	8.02 f	9.62 f	85.10 a	85.75 c	86.45 cd	87.25 d	88.05 c
Level of significance	*	*	***	***	NS	NS	NS	NS	NS
CV %	1.57	0.99	0.61	0.56	0.38	0.25	0.24	0.26	0.23

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents

-1)

(*)

*

$V_1 = Langra$	M ₁ =200 ppm of MH solution	* indicates at 5% level	NS means non significant
$V_2 = Khirshapat$	M ₂ =400 ppm of MH solution	** indicate at 1 % level	
M ₀ =Control	$M_3 = 600 \text{ ppm of MH solution}$	*** indicate at 0.1% level	

Je-

Ĩ

-

X

m

4

T

8.2.3 Dry matter content

The variation in varieties means in respect of dry matter content demonstrated highly significant at different days (Appendix 2.2). At different days, dry matter decreased gradually with the advancement of storage duration (Fig. 2.5). The results indicated that Langra produced comparatively more dry matter comparing to Khirshapat. At initial day, Langra gave higher (17.55%) dry matter whereas; Khirshapat produced lesser (14.78%) and at 12th day, Langra contributed the highest (14.13%) whereas, Khirshapat gave the highest (11.37%). These results are in partially supported by the findings of Hossain (1991).

Different doses of MH imposed in this study on dry matter content of mango had highly significant at different days after storage (Appendix 2.2). At different days of storage, dry matter content decreased gradually with the passing of storage duration (Fig. 2.7). It also noticed that M_3 treatment gave the highest dry matter (14.98%) at 6th day whereas; the lowest dry matter (12.25%) was recorded from control.

The combined effect of varieties and different doses of MH solution on dry matter content of mango showed non significant at different days after storage (Appendix 2.2). At different days, dry matter content decreased gradually up to 9th day and then, it increased slightly from control at 12th day. At initial day, the highest dry matter content (17.70%) was noticed from the treatment combination of V₁M₃ which was statistically identical with the combination of V₁M₀ whereas; the lowest (14.70%) was recorded from the combination of V₂M₀ which was also statistically similar with the combination of V₂M₁, V₂M₂ and V₂M₃. At 12th day, the highest value (14.80%) was recorded from the treatment combination of V₁M₃ and the lowest value (10.30%) was recorded from V₂M₂ (Table 2.3). The decrease in dry matter with the increasing storage period might be possible due to breaking down the complex carbohydrates into simple molecules and H₂O as well as adding water through osmotic process and metabolic activities. The decreasing of dry matter content of mango pulp was not supported by the findings of Hossain (1999).

8.2.4 Ash content

Variation in varieties means in respect of ash content of mango pulp demonstrated significant at different days after storage (Appendix 2.2). At different days, values of ash content reduced gradually with the passing of storage duration (Fig. 2.6). The results denoted that Langra gave comparatively more ash than Khirshapat at all days. Higher (0.98%) ash was achieved from Langra at initial day where Khirshapat achieved (0.85%) and again, the maximum achievement was 0.78% from Langra at 12th day whereas, the minimum achievement (0.65%) from Khirshapat.

Different doses of MH had non significant variation on ash content of mango at different days (Appendix 2.2). The results demonstrated from the figure 2.8 that ash content influenced by different doses of MH decreased slightly from M_3 and markedly from control.

The combined effect of varieties and different doses of MH were found nonsignificant variation in terms of ash content of mango at various days after storage (Appendix 2.2 and Table 2.3).There was no available research findings relating to ash content of mango. It also denoted that ash content was more related to dry matter content. The results of the present study revealed that ash content depended upon the quantity of dry matter achieved.

8.3 Changes in biochemical properties of mango during storage environments

Various changes regarding biochemical properties of mango pulp are presented and discussed under the following sub-headings.

8.3.1 Vitamin C content

1

1

1

Variation between varieties means in respect of vitamin C content of mango pulp was found to be highly statistically significant at different days after storage (Appendix 2.3). At various days, the results denoted that Langra showed better performance in accumulation of vitamin C as compared to Khirshapat (Fig. 2.9). It also revealed that quantity of vitamin C decreased with the increase of storage period. The decreasing trend was observed higher from initial to 6th days and thereafter, it was slightly slower. At the initial day, Langra gave the highest (132.13 mg/100 g) amount of vitamin C whereas; Khirshapat contributed the lowest amount

5

Č.

Results and Discussion

V2



Fig. 2.5 Effect of varieties on dry matter content of mango pulp at different days after storage



-M1 -M2

-M0

3

0.9

0.8

0.7

0.6

0.5

0.4

0.3

0.2 0.1

0

Initial

9

12

M3







6

Days after storage

9

12

Graph represents	
$V_1 = Langra$	M ₁ =200 ppm of MH solution
$V_2 = Khirshapat$	M ₂ =400 ppm of MH solution
$M_0 = Control$	M ₃ =600 ppm of MH solution

2

A Results and Discussion

(42.98 mg/100 g). At 12th day, Langra again exhibited higher (16.03 mg/100 g) amount of vitamin C and Khirshapat produced the lesser (8.55 mg/100 g) amount (Fig. 2.9). These results explored that vitamin C content gradually reduced with the increase of storage duration in both the varieties. It might be possible due to augmentation of ethylene synthesis resulting in oxidation of ascorbic acid. The results of the present study are in agreement with the findings of Azad (2001), Shyamalamma (1995), Gafur *et al.* (1994) and Absar et al. (1993).

Variation among the means of different doses of MH in terms of vitamin C content showed highly statistical significant at different days after storage (Appendix 2.3). At initial day, green mangoes treated with 600 ppm (M₃) gave the highest (89.35 mg/100 g) amount of vitamin C while, the lowest (85.56 mg/100 g) was recorded from M₁ treatment. After initial day, it decreased gradually with the passing of storage time (Fig. 2.10). It also indicated that decreasing trend was faster in control, and slower in M₃ treated fruit. At 12th day, vitamin C content ranged between 8.10 to 16.60 mg per 100 g of mango pulp was obtained from control and M₃ treated fruit, respectively. The decrease in vitamin C content in both treated and untreated mangoes at different storage period might be possible due to oxidation of ascorbic acid and M₃ dose was possibly causing delay ripening resulting in lower oxidation in vitamin C. These results are in supported by the findings of Hossain (1999). Gautam *et al.* (2003) also found the similar findings.

The combined effect of varieties and different doses of MH solution demonstrated significant variation in respect of vitamin C at different days after storage (Appendix 2.3 and Table 2.4). At various days during storage, the results were observed that vitamin C content decreased with the advancement of storage period. The quantity of vitamin C ranged between 40.92 to 133.60 mg per 100 g of fresh mango pulp at initial stage was noticed from the combination of V₂M₁ and V₁M₂ and in which V₁M₂ was statistically at par with V₁M₀ and V₁M₃. These quantities were reduced gradually after initial day. It denoted at 12th day that vitamin C content ranged between 4.30 to 20.40 mg per 100 g of fresh mango pulp was achieved from the treatment combination of V₂M₀ and V₁M₃ (Table 2.4).

127

Results and Discussion





Fig. 2.9 Effect of varieties on vitamin C content of mango pulp at different days after storage

Fig. 2.10 Vitamin C co	ntent of mango pulp as
influenced by different	doses of MH solution at
different days after	storage. Vertical bars
represent LSD at 0.05	evel

Graph represents	
$V_1 = Langra$	M ₁ =200 ppm of MH solution
$V_2 = Khirshapat$	M ₂ =400 ppm of MH solution
$M_0 = Control$	$M_3 = 600 \text{ ppm of MH solution}$

8.3.2 Titratable acidity

2

Variation between varieties means on titratable acidity was statistically highly significant at different days after storage (Appendix 2.4). At various days, Langra exhibited higher titratable acidity as compared to Khirshapat (Fig. 2.11). Titratable acidity decreased with the increase of storage period. The decreasing trend was very fast from initial to 3rd day and thereafter, its trend was slightly slower. At initial level the highest (3.79%) was noticed from Langra whereas the lowest (2.53%) was noticed from Khirshapat. At 12th day, the highest (0.33%) was recorded from Langra whereas Khirshapat gave the lowest amount (0.26%). The decreasing trend of titratable acidity at storage period was reported by Upadhyay and Tripathi (1988) and Leon and Lima (1968) and Medlicott *et al.* (1986) also found the similar results. According to them, acidity was reduced during storage growth on attainment of maturity and ripening. It also might be due to genetical dissimilarities between the varieties.

Treatments combination	Dry matter content (%) at different days				Ash content(%) at different days					
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12
V1M0	17.40 c	15.10 d	13.60 d	13.88 c	14.74 a	0.97	0.87 ab	0.78 ab	0.76 ab	0.77 ab
V ₁ M ₁	17.50 bc	16.50 c	15.50 c	13.50 d	13.92 b	0.98	0.93 a	0.88 a	0.76 ab	0.75 ab
V ₁ M ₂	17.60 ab	16.80 b	16.00 b	15.10 b	13.07 c	0.99	0.94 a	0.90 a	0.86 a	0.76 ab
V ₁ M ₃	17.70 a	17.10 a	16.40 a	15.60 a	14.80 a	0.99	0.96 a	0.92 a	0.88 a	0.84 a
V ₂ M ₀	14.70 d	12.40 g	10.90 g	11.18 g	12.04 d	0.84	0.72 b	0.65 b	0.62 b	0.65 ab
V2M1	14.75 d	13.75 f	12.75 f	10.75 h	11.18 e	0.84	0.79 ab	0.74 ab	0.63 b	0.61 b
V ₂ M ₂	14.80 d	14.00 ef	13.20 e	12.20 f	10.30 f	0.85	0.80 ab	0.76 ab	0.71 ab	0.62 b
V ₂ M ₃	14.85 d	14.25 e	13.55 d	12.75 e	11.95 d	0.85	0.82 ab	0.78 ab	0.74 ab	0.70 ab
Level of significance	NS	NS	NS	NS	NS	NS	NS	NS	NS	NS
CV%	0.65	1.01	0.90	1.06	0.99	11.61	12.01	12.59	13.79	13.89

Table 2.3 Combined effects of varieties and different doses of Maleic hydrazide solution on dry matter and ash content of postharvest mango during storage at ambient condition

129

Table 2.4 Combined effects of varieties and different doses of Maleic hydrazide solution on vitamin C content of postharvest mango during storage at ambient condition

Treatments combination		Vitami	n C content (mg/100 g	g) at different days	
Varieties × Treatments	Initial	3	6	9	12
V ₁ M ₀	132.50 a	85.53 d	38.20 d	19.30 d	11.90 e
V ₁ M ₁	130.20 b	112.30 c	58.60 c	34.50 c	14.30 c
V ₁ M ₂	133.60 a	116.60 b	63.20 b	39.20 b	17.50 b
V ₁ M ₃	132.20 a	120.30 a	68.80 a	42.30 a	20.40 a
V ₂ M ₀	42.30 d	24.35 h	12.55 h	7.80 h	4.30 h
V ₂ M ₁	40.92 d	30.20 g	16.30 g	9.70 g	7.20 g
V ₂ M ₂	42.20 d	32.55 f	18.40 f	12.50 f	9.90 f
V ₂ M ₃	46.50 c	35.50 e	21.60 e	13.60 e	12.80 d
Level of significance	**	***	***	***	*
CV%	1.21	1.16	0.57	0.86	1.08

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents

 $V_1 = Langra$ $V_2 = Khirshapat$ M₀ =Control

2

M₁=200 ppm of MH solution * indicates at 5% level M₂=400 ppm of MH solution $M_3 = 600 \text{ ppm of MH solution}$

NS means non significant

1.1

** indicate at 1 % level *** indicate at 0.1% level

D Results and Discussion

Different doses of MH solution used in this investigation on titratable acidity showed significant variation among the means at various days after storage except initial day (Appendix 2.4). At various days, titratable acid content declined very sharply from initial to 3^{rd} day and then, it declined steadily (Fig. 2.13). In all storage period, the higher titratable acidity (1.26, 0.96, 0.76, and 0.45%) was noted from M₃ treatment from initial to 12^{th} day, followed by 0.70%, 0.43%, 0.24% and 0.12% in untreated mangoes, respectively (Fig. 2.13). These results are in partially supported by the findings of Tefera *et al.* (2007), but, these results are in agreement with the findings of Kumar *et al.* (1992).

The combined effect of varieties and different doses of MH in relation to titratable acidity of mango pulp were found to be non significant at different days after storage except 3^{rd} day (Appendix 2.4 and Table 2.5). At different days, there was observed a decreasing trend of titratable acid content with the advancement of storage period (Table 2.5). At 3^{rd} day, the highest (1.32%) concentration was recorded from the treatment combination of V₁M₃ which was statistically similar with the combination of V₁M₂, V₁M₁ and V₂M₃ and the lowest acid concentration (0.50%) was recorded from the treatment combination of V₂M₀ (Table 2.5). These occurrences might be possible due to retardation of acid oxidation at V₁M₃ treatment combination and showed genetical variation in between varieties.

8.3.3 Pulp pH

C.

F.

V

The analysis of variance in between the varieties demonstrated statistical highly significant on pulp pH of mango at different days after storage (Appendix 2.4). At various days, there exhibited an increasing trend of pulp pH with the passing of storage period (Fig. 2.12). In each storage period, pulp pH indicated more acedic quantity more from Khirshapat compared with Langra. The highest pulp pH (3.55) was noticed from green Khirshapat at initial day whereas; the lowest (3.47) was noticed from green Langra. At 12th day, the highest pH (6.20) was obtained from Khirshapat and the lowest value (6.10) was obtained from Langra. Increasing trend of pulp pH was also observed by Yunlartl (1980), Kumar *et al.* (1993) and Shahjahan *et al.* (1994). This phenomenon might be perhaps due to oxidation of acid during

3

a)

1

2

X

storage resulting in higher pulp pH and also exhibiting genetical dissimilarities between varieties.

Different doses of MH adopted to this trial showed significant variation in pulp pH at different days after storage (Appendix 2.4). An increasing trend of pulp pH was observed from different treated and untreated fruits at various days of storage (Fig. 2.14). Pulp pH was higher in control at all stages of storage over the fruits treated with M_1 , M_2 and M_3 treatments. pH value of mango pulp was the highest (6.95) in the control whereas the fruits treated with M_3 treatment produced the lowest (5.65) value at 12th day (Fig. 2.14). The results of the present investigation are in partially supported by the findings of Reddy and Hariprya (2002).

The combined effect of varieties and different doses of MH subjected to this study regarding pulp pH were found to be non significant at different days after storage (Appendix 2.4 and Table 2.5). There was showing an increasing trend of pulp pH from various treatment combinations at different days (Table 2.5). At 12^{th} day, the highest (7.0) pH value was recorded from the treatment combination of V₂M₀ and the lowest (5.60) was recorded from the treatment combination of V₁M₃, respectively.





Fig. 2.11 Effect of varieties on titratable acidity of mango pulp at different days after storage

Figures represent V1 = Langra V2 = Khirshapat

Fig. 2.12 Effect of varieties on pulp pH of mango at different days after storage

4

F



Fig. 2.13 Titratable acidity of mango pulp as influenced by different doses of MH solution at different days after storage. Vertical bars represent LSD at 0.05 level

Fig. 2.14 Pulp pH of mango as influenced by different doses of MH solution at different days after storage. Vertical bars represent LSD at 0.05 level

12

Graph represents	
$V_1 = Langra$	M ₁ =200 ppm of MH solution
V2 = Khirshapat	M ₂ =400 ppm of MH solution
$M_0 = Control$	$M_3 = 600 \text{ ppm of MH solution}$

8.3.4 Total soluble solid (Brix %) content

Statistically highly significant variation was noticed in connection with TSS content between two varieties at different days after storage (Appendix 2.5). The results denoted that TSS content of mango pulp increased gradually with the advancement of storage period. The increasing trend was faster from initial to 6th day thereafter; it increased slower (Fig. 2.15). At initial to 6th day, Khirshapat exhibited better performance in TSS accumulation than Langra. But, after 6 days, Langra showed better than Khirshapat and it again, lower than Khirshapat at 12th day (Fig. 2.15). At 12th day, the highest (16.63%) TSS quantity was obtained from Khirshapat and the lowest (16.43%) was obtained from Langra. Increased in the percentage of TSS during storage of mango was reported by Singh (1968). Absar *et al.* (1993) reported that TSS was increased with maturity of mango fruit. But, they

4

E.

5

1

5

found highest TSS in Langra. Mollah and Siddique (1973) reported that TSS value varied from cultivar to cultivar. These might be possible due to genetical differences between varieties.

Different doses of MH subjected to the postharvest mangoes in this study were observed significant variation in respect of TSS content at different days after storage (Appendix 2.5). At different days of storage, it was noticed that TSS accumulation increased with the advancement of storage time. It also denoted that TSS content was markedly increased in untreated mangoes from initial to 6th day and then, it decreased significantly (Fig. 2.17). Another treatment viz., M_1 also gave increased TSS from initial to 9th day and thereafter, it decreased sharply. The mangoes treated with M₂ treatment also showed more or less similar increasing trend from initial to 12th day. But, the fruits treated with M₃ produced TSS changes very steady motion at various days after storage. The highest (19.65, 18.90 and 19.15%) TSS accumulation was obtained from M_0 , M_1 and M_2 at 6, 9 and 12^{th} day respectively, whereas, the lowest (8.65, 12.15 and 16.15%) was obtained from M_3 treatments. These results are in partially supported by the findings of Jacobi et al. (2000). Lianni et al. (1994) showed TSS content was increased markedly up to 5 days after harvest without using any treatment. This happened possibly due to ripening condition resulting in maximize the TSS gathering in control and 600 ppm of MH (M₃) hindered in ethylene synthesis that caused delay ripening and ultimately in lower TSS accumulation. It also elucidated that TSS gathering is strongly related to ripening and it caused decrease owing to rotten.

The combined effect of varieties and induced different doses of MH on TSS content demonstrated significant at different days after storage except initial and 3^{rd} day (Appendix 2.5 and Table 2.6). There was found an augmentig behavior of TSS content at different days of storage (Table 2.6). The highest accumulation (20.00, 19.10 and 19.70%) was noticed from the treatment combination of V₁M_o, V₁M₁ and V₁M₂ at 6, 9 and 12th days, whereas; the lowest value (8.40, 11.90 and 15.90%) was noticed from the treatment combination.

A.

SE

X

R

8.3.5 Total sugar content

Variation in between both the varieties means in respect of total sugar content of mango pulp was observed to be significant at different days after storage (Appendix 2.5). The results indicated that total sugar content (TSC) increased markedly with the passing of storage duration (Fig. 1.16). This increasing trend was more or less sharp from initial to 9th days in both the varieties, thereafter; it increased steadily. At all days of storage Khirshapat gave comparatively more quantity of TSC than Langra. At initial day, Khirshapat had the highest (5.98%) whereas Langra had the lowest (5.50%). At 12th day, Khirshapat accumulated the highest quantity (19.27%) and the lowest (18.80%) was recorded from Langra. Upadhyay and Tripathi (1985) reported that total sugar content was augmented gradually, when stored for 6 days at room temperature. Sugar content increased during ripening (Srivastava, 1967). These results are in conformity with the findings of Shahjahan *et al.* (1994). Tsuda *et al.* (1999) also found the similar results. The increase in TSC might be due to conversion of complex starch or carbohydrate into simple compound like sucrose, Fructose, galactose etc.

Different storage treatments applied to the investigation on total sugar content of mango pulp demonstrated significant variation at different days after storage except initial level (Appendix 2.5). At different days, it denoted that TSC increased markedly with the advancement of storage period (Fig. 2.18). The increasing trend was very fast in untreated mango followed by other treatments namely M_1 , M_2 and M_3 , respectively. The maximum quantity of TSC (10.60, 19.00) and 21.00%) was reported from untreated mangoes at 3, 6 and 9 days, whereas; the minimum (7.38, 9.38 and 11.88%) was recorded from M_3 treatment. Mondal *et* al. (1995) reported the increasing trend of total sugar content at the later stage of storage. Total sugar content of mango pulp was the highest on the 6th day of storage (Tripathi, 1988). The results of the present study are in agreement with the findings of Reddy and Haripriya (2002) and Kumar et al. (1992). The increasing trend of total sugar at untreated might be possible due to breaking down of complex carbohydrate into simple compound (sucrose, fructose etc.) but, M₃ treatment made delay ripening at storage period resulting in lower conversion of complex compound into simple molecules.

134

-2

G.

7

5

X

9





Fig. 2.15 Effect of varieties on total soluble solid of mango pulp at different days after storage





Fig. 2.17 Total soluble solid of mango pulp as influenced by different doses of MH solution at different days after storage. Vertical bars represent LSD at 0.05 level

Graph represents	
$V_1 = Langra$	M_1 =200 ppm of MH solution
$V_2 = Khirshapat$	M ₂ =400 ppm of MH solution
$M_0 = Control$	M ₃ =600 ppm of MH solution



Fig. 2.18 Total sugar content of mango pulp as influenced by different doses of MH solution at different days after storage. Vertical bars represent LSD at 0.05 level



.

i.a.

۲

he ;

6

1

Treatments combination	Titratable acidity (%) at different days					mbination Titratable acidity (%) at different days pulp pH at different days					1
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12	
V ₁ M ₀	3.75 a	0.90 d	0.52 d	0.28 e	0.13 e	3.57 a-c	4.20 ab	5.30 a	6.30 a	6.90 a	
V ₁ M ₁	3.80 a	1.15 a-c	0.80 bc	0.53 c	0.30 c	3.50 bc	4.00 cd	4.60 b	5.30 b	6.10 b	
V ₁ M ₂	3.78 a	1.25 ab	0.92 ab	0.68 b	0.38 b	3.40 c	3.80 ef	4.30 cd	4.90 c	5.80 cd	
V ₁ M ₃	3.83 a	1.32 a	1.05 a	0.82 a	0.50 a	3.40 c	3.60 g	4.10 e	4.60 d	5.60 e	
V ₂ M ₀	2.45 b	0.50 e	0.33 e	0.20 e	0.10 e	3.70 a	4.30 a	5.40 a	6.40 a	7.00 a	
V ₂ M ₁	2.52 b	1.05 cd	0.71 c	0.39 d	0.23 d	3.60 ab	4.10 bc	4.70 b	5.40 b	6.20 b	
V ₂ M ₂	2.55 b	1.15 bc	0.80 bc	0.55 c	0.32 c	3.50 bc	3.90 de	4.40 c	5.00 c	5.90 c	
V ₂ M ₃	2.60 b	1.20 a-c	0.87 b	0.70 b	0.40 b	3.40 c	3.70 fg	4.20 de	4.70 d	5.70 de	
Level of significance	NS	*	NS	NS	NS	NS	NS	NS	NS	NS	
CV%	3.18	8.49	10.75	10.30	10.05	2.56	2.56	1.33	1.69	1.72	

Table 2.5 Combined effects of varieties and different doses of Maleic hydrazide solution on titratable acidity and pulp pH of postharvest mango during storage at ambient condition

17

136

40

Table 2.6 Combined effects of varieties and different doses of Maleic hydrazide solution on total soluble solid and total sugar content of postharvest mango during storage at ambient condition

Treatments combination	tments combination TSS content (%) at different days Total sugar content (%) at different						at different da	iys		
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12
V1Mo	4.20 d	11.20 b	20.00 a	17.00 c	13.50 g	5.60 c	10.35 b	18.75 b	20.75 b	19.76 e
V ₁ M ₁	4.10 de	6.60 e	13.60 c	19.10 a	16.10 e	5.50 cd	8.07 d	13.67 d	18.97 d	20.98 b
V ₁ M ₂	4.00 ef	6.00 f	9.00 ef	15.20 e	19.70 a	5.50 cd	7.70 e	10.90 f	14.50 f	19.60 e
V ₁ M ₃	3.90 f	5.90 f	8.40 g	12.40 g	16.40 d	5.40 d	7.05 g	9.15 h	11.65 h	14.85 g
V ₂ M ₀	5.30 a	12.30 a	19.30 b	16.30 d	12.80 h	6.10 a	10.85 a	19.25 a	21.25 a	20.26 c
V ₂ M ₁	5.20 ab	7.70 c	13.20 d	18.70 b	19.20 b	6.00 a	8.57 c	14.17 c	19.47 c	21.48 a
V2M2	5.10 bc	7.10 d	9.10 e	14.10 f	18.60 c	6.00 a	8.25 d	11.45 e	14.95 e	20.05 d
V ₂ M ₃	5.00 c	7.00 d	8.90 f	11.90 h	15.90 f	5.80 b	7.50 f	9.60 g	12.10 g	15.30 f
Level of significance	NS	NS	***	***	***	NS	NS	NS	NS	NS
CV%	2.31	1.31	0.84	0.65	0.63	1.85	1.24	0.79	0.62	0.56

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents $V_1 = Langra$ $V_2 = Khirshapat$

Mo =Control

 M_1 = 200 ppm of MH solution M_2 = 400 ppm of MH solution M_3 = 600 ppm of MH solution * indicates at 5% level NS means non significant ** indicate at 1 % level

*** indicate at 0.1% level

The combined effect of varieties and imposed different doses of MH solution in this study in terms of total sugar content of mango pulp demonstrated non significant variation at different days after storage (Appendix 2.5 and Table 2.6). The results indicated that total sugar content was increased with the passing of storage time (Table 1.13). At 12th day the maximum (21.48%) quantity of TSC was achieved from the treatment combination of V₂M₁ whereas, the minimum (14.85%) was achieved from the treatment combination of V₁M₃.

8.3.6 Reducing sugar Content

1

S.

X

1

4

Analysis of variance exhibited that varieties had significant effect on reducing sugar content of mango pulp at different days after storage (Appendix 2.6). There was observed an increasing trend of reducing sugar with the passing of storage period (Table 2.7). It also noticed that Khirshapat gave comparatively more reducing sugar than Langra at different days of storage. The highest (5.38%) quantity of this sugar was received from Khirshapat whereas; the lowest (5.08%) was achieved from Langra at 12th day of storage (Table 1.7). These results are in agreement with the report of Upadhyay and Tripathi (1985). Casttrillo *et al.* (1992) elucidated that reducing sugar was increased during storage period. Khirshapat producing comparatively more reducing sugar might be possible due to genetical variation in both the varieties.

Different doses of MH solution imposed to this study showed significant variation in respect of reducing sugar content of mango pulp at different storage period (Appendix 2.6). It was found that reducing sugar of mango pulp was augmented gradually at different days after storage. It also denoted that untreated mangoes were noticed better in producing of reducing sugar as compared to other treatments. There was found from control that reducing sugar content increased gradually up to 9 days and then, It might be affected due to decomposition. At 12th day, the maximum (6.35%) amount of reducing sugar was attained from M₁ treatment and the lowest (4.80%) was attained from the fruit treated with M₃ treatment. Increase in reducing sugar during storage was stated by Tripati (1988) and Kumar et al. (1992). Lower increasing trend of reducing sugar content treated

4

E

X.

5

X

1

Y

with M_3 treatment might be possible due to delay ripening resulting in lower conversion of carbohydrates into simple's molecules.

The combined effect of varieties and implied different doses MH solution of mango pulp showed non significant variation in relation to reducing sugar content of mango pulp at different days (Appendix 2.6). It denoted reducing sugar content increased gradually at three days interval up to 9th day thereafter, it decreased from the treatment combination of V₂M₀ (Table 2.8). At 12th day, the highest (6.5%) quantity was noticed from the treatment combination of V₂M₀.

8.3.7 Non reducing sugar content

The variation in between the varieties means had highly significant in respect of non-reducing sugar content at different days after storage (Appendix 2.6). There showed an increasing trend of non reducing sugar content at different days. At all days, it was observed that Khirshapat was better in achieving of non reducing sugar content as compared to Langra (Table 2.8). At 12th day, the highest (13.90%) amount of non reducing sugar was attained from Khirshapat and the lowest (13.72%) amount was attained from Langra. These results are in conformity with the report of Ali and Mazhar (1960). They reported that non reducing sugar content of ripe fruits was 11.20%. The result obtained from the investigation might be possible due to varietals dissimilarities.

Different doses of MH solution subjected to this trial gave the profound variation on non reducing sugar content of mango at different days after storage except initial level (Appendix 2.6). It was noticed that non reducing sugar content of mango pulp increased markedly at various days. It also denoted that untreated mango exhibited better performance in accumulation of more quantity of non reducing sugar followed by other treatment. This increasing trend decreased at 9th day due to putrified situation. Lower increasing trend was found from the fruit treated with M₃ treatment. At 12th day the highest result (15.21%) was recorded from control and lowest value (11.02%) was recorded from M₃ treatment. These events might be possible due to the M₃ treatment impeded ethylene synthesis of

R

4

5

A

I

mango resulting in delay ripening and little **a**mount of non reducing sugar achievement. These results are in agreement with the reports of Singh (1968), Joshi and Roy (1988) and Rangavalli *et al.* (1993). They found a gradual increase in non reducing sugar content of untreated fruit.

The analysis of variance of the combined effect of varieties and applied different doses of MH solution were found to be non significant in terms of non reducing sugar content of mango pulp at different days (Appendix 2.6). It revealed that an increasing trend of non reducing sugar was noticed from different treatment combination at various days (Table 2.8). At 12th day, the highest (15.31%) quantity of non reducing sugar was recorded from the treatment combination of V₂M₀ whereas; the lowest (10.95%) was recorded from the treatment combination of V₁M₃, respectively.

8.3.8 Crude fibre content

The analysis of variance of mango varieties imposed in this investigation exhibited significant variation on crude fibre content of mango pulp at different days after storage (Appendix 2.7). There was found a decreasing trend of crude fibre from both the varieties at different days after storage (Fig. 2.19). It also indicated that Langra demonstrated better performance in accumulation of crude fibre. At initial day, Langra produced the highest (1.29%) quantity of crude fibre as compared to Khirshapat (1.16%). The amount of crude fibre decreased gradually with the increase of storage period from both the varieties. At 12th day, the highest (0.54%) quantity of crude fiber was noticed from Langra whereas; the lowest (0.41%) was noticed from Khirshapat. There were no available research findings regarding crude fiber content of mango in the scientific literature but, the present study revealed that crude fibre decreased with the advance of storage duration. The decrease in crude fiber content with the advancement of storage period might be possible due to breaking down of cellulose and lignin into smaller compound.

Various doses of MH solution used to this trial was found to be significant effect on crude fiber content of mango pulp at different days after storage except initial day (Appendix 2.7). The results explored that crude fibre content decreased

Treatments	1	Reducing	, sugar (%) at	different days		Non-reducing sugar (%) at different days				
Variety	Initial	3	6	9	12	Initial	3	6	9	12
V ₁	1.33 b	2.05 b	3.63 b	4.95 b	5.08 b	4.17 b	6.24 b	9.49 b	11.52 b	13.72 b
V ₂	1.60 a	2.35 a	3.93 a	5.25 a	5.38 a	4.39 a	6.44 a	9.69 a	11.69 a	13.90 a
Level of significance	***	***	***	***	***	***	***	***	**	**
Treatments										
Mo	1.58 a	3.10 a	6.10 a	6.30 a	4.80 c	4.30	7.50 a	12.90 a	14.70 a	15.21 a
M1	1.44 ab	2.05 b	3.65 b	6.05 b	6.35 a	4.31	6.27 b	10.27 b	13.17 b	14.88 b
M ₂	1.47 ab	1.90 c	3.00 c	4.60 c	5.70 b	4.28	6.08 c	8.18 c	10.13 c	14.13 c
M ₃	1.38 b	1.75 d	2.35 d	3.45 d	4.05 d	4.23	5.53 d	7.03 d	8.43 d	11.02 d
Level of significance	*	***	***	***	***	NS	***	***	***	***

Table 2.7 Changes of reducing and non reducing sugar content in mango varieties and influenced by different doses of Maleic hydrazide solution at ambient condition during storage

2

Table 2.8 Combined effects of varieties and different doses of Maleic hydrazide solution on reducing and non reducing sugar content of postharvest mango during storage at ambient condition

Treatments combination Reducing sugar (%) at different days Non-reducing sugar (%) at different days 3 6 9 12 Initial 3 6 9 12 Initial Varieties × Treatments 4.15 c 7.40 b 14.60 b V_1M_0 1.45 b-d 2.95 b 5.95 b 6.15 b 4.65 f 12.80 b 15.11 b V_1M_1 1.30 de 1.90 d 3.50 d 5.90 c 6.20 b 4.20 bc 6.17 d 10.17 d 13.07 d 14.78 c V₁M₂ 1.32 c-e 1.75 ef 2.85 f 4.45 e 5.55 d 4.18 c 5.95 e 8.05 f 10.05 e 14.05 d V_1M_3 1.25 e 1.60 f 2.20 h 3.30 g 3.90 h 4.15 c 5.45 f 6.95 g 8.35 f 10.95 e V₂M₀ 1.70 a 3.25 a 6.25 a 6.45 a 4.95 e 4.44 a 7.60 a 13.00 a 14.80 a 15.31 a V₂M₁ 1.58 ab 2.20 c 3.80 c 6.50 a 4.42 a 6.37 c 10.37 c 13.27 c 14.98 b 6.20 b V₂M₂ 4.38 ab 14.20 d 1.62 ab 2.05 cd 3.15 e 4.75 d 5.85 c 6.20 cd 8.30 e 10.20 e V_2M_3 1.50 bc 1.90 de 3.60 f 4.30 a-c 5.60 f 8.50 f 11.10 e 2.50 g 4.20 g 7.10 g Level of significance NS CV% 0.77 7.24 4.82 2.81 2.08 2.03 2.48 1.67 1.11 0.91

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

7

Table represents

(2)

140

V ₁ = Langra	M ₁ =200 ppm of MH solution
$V_2 = Khirshapat$	M ₂ =400 ppm of MH solution
$M_0 = Control$	M ₃ =600 ppm of MH solution

A.

* indicates at 5% level ** indicate at 1 % level

NS means non significant

1

+

*** indicate at 0.1% level

1

2

2

1

1

~

gradually with the passing of storage period (Fig. 2.21). At all days, it was found that crude fiber content was comparatively higher at the fruit treated with M₃ treatment followed by other treatments. At 3^{rd} day, the maximum (1.15%) amount of crude fiber was obtained from M₃ treatment and the lowest (0.79%) was noticed from M₀ treatment. At 12^{th} day, the maximum (0.65%) was mentioned from M₃ treatment whereas the lowest (0.39%) was mentioned from M₀ treatment (Fig. 2.21). Crude fibre exhibited a very slight change during ripening as stated by Mathoko (2000).

The combined effect of varieties and different doses of MH solution were observed the non significant variation in terms of crude fiber content of mango pulp at different days after storage (Appendix 2.7). It indicated that crude fibre decreased gradually with the increase of storage period (Table 2.9). It also indicated that the treatment combination of V₁M₃ was better in achieving crude fibre content at all storage times. At 9th day, the highest (0.90%) quantity of crude fiber was obtained from the treatment combination of V₁M₃ and the lowest (0.37%) was gathered from the treatment combination of V₂M_o, which was statistically at par with V₂M₁ and V₁M_o respectively.

8.3.9 Total lipid content

The analysis of variance of varieties was significant on lipid content in mango pulp at different days after storage (Appendix 2.7). An increasing trend of lipid content was obtained in mango pulp at different days (Fig. 2.20). At all days of storage, it denoted that Langra gave higher amount of lipid as compared to Khirshapat. At 12th day, the highest (0.67%) amount of lipid was recorded from Langra and the lowest (0.63%) was recorded from Khirshapat. These events might be possible due to the genetically dissimilarities between two varieties.

Different doses of MH solution used in this study in terms of lipid content exhibited highly significant at different days after storage (Appendix 2.7). It was found an increasing trend of lipid content with the passing of storage period at various days after storage (Fig. 2.22). It also indicated that control treatment produced comparatively higher amount of lipid followed by M_1 , M_2 and M_3 treatments from initial to 9 days; and then, it decreased due to becoming decayed situation. At 9th day, the highest (0.75%) amount of lipid content was reported from control whereas; the lowest (0.45%) was reported from the fruit treated with M_3 treatment.

141
-

4

a.

a

A

A

At 12^{th} day, M₁ treated fruit had the highest (0.73%) amount of lipid and the lowest (0.55%) was noticed from the fruit treated with M3 treatment, respectively. These occurrences might be possible due to M₃ treatment caused delay ripening resulting in lower production of lipid content.

The interaction effects of varieties and induced various doses of MH solution in this trial were found to be non significant in respect of lipid content of mango pulp at different days after storage (Appendix 2.7). It was observed an increasing trend of lipid content in mango pulp with the advancement of storage duration (Table 2.9). It also denoted that lipid content was obtained comparatively more quantity from the treatment combination of V₁M₀ at initial to 9th day and then, it decreased due to starting of decomposition. At this time, lower quantity was noticed from the treatment combination of V₂M₃. At 12th day, the highest (0.76%) quantity of lipid was reported from the treatment combination of V₁M₁ whereas; the lowest (0.54%) was reported from the treatment combination of V₂M₃, respectively.





Fig. 2.19 Effect of varieties on crude fibre of mango pulp at different days after storage

Graph represents $V_1 = Langra$ $V_2 = Khirshapat$

Fig. 2.20 Effect of varieties on lipid content of mango pulp at different days after storage

*

4

Ch.

A

2

x





Fig. 2.21 Crude fibre of mango pulp as influenced by different doses of MH solution at different days after storage. Vertical bars represent LSD at 0.05 level

Fig. 2.22 Lipid content of mango pulp as influenced by different doses of MH solution at different days after storage. Vertical bars represent LSD at 0.05 level

Graph represents $V_1 = Langra$ $V_2 = Khirshapat$ $M_0 = Control$

M₁=200 ppm of MH solution M₂=400 ppm of MH solution M₃ =600 ppm of MH solution

8.3.10 Water soluble protein content

The analysis of variance of mango varieties exhibited significant variation in respect of water soluble protein content (WSPC) in mango pulp at different days after storage except initial, 3rd and 12th days (Appendix 2.8). It was found an increasing trend of WSPC with becoming extended time of storage duration. It also explored that Langra was better in WSPC accumulation as compared to Khirshapat at all stages of storage. At 9th day, the highest (1.01%) accumulation of WSPC was reported from Langra whereas the lowest (0.92%) was reported from Khirshapat (Table 2.10). These might be possible due to the seed protein of mango disseminated to pulp areas during ripening stages and it also might have genetical variation in between two varieties.

Different doses of MH solution imposed to this investigation in terms of WSPC showed significant variation at different days except initial and 12th day (Appendix 2.8). Various results obtained from biochemical analysis was noticed an increasing

Treatments combination		Crude fibre (%) at different days Lipid content (%) at different days									
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12	
V ₁ M ₀	1.25 ab	0.85 e	0.50 e	0.47 de	0.47 cd	0.20 a	0.39 a	0.68 a	0.77 a	0.67 b	
V ₁ M ₁	1.27 ab	1.07 bc	0.82 c	0.52 d	0.47 cd	0.19 ab	0.34 c	0.44 b	0.70 bc	0.76 a	
V ₁ M ₂	1.29 a	1.14 ab	0.94 b	0.74 b	0.51 bc	0.17 b-d	0.27 d	0.37 c	0.47 d	0.69 b	
V ₁ M ₃	1.30 a	1.20 a	1.05 a	0.90 a	0.70 a	0.16 cd	0.25 de	0.35 cd	0.45 de	0.55 d	
V ₂ M ₀	1.12 c	0.72 f	0.40 f	0.37 e	0.32 e	0.19 ab	0.37 b	0.66 a	0.72 b	0.62 c	
V ₂ M ₁	1.14 c	0.94 de	0.69 d	0.39 e	0.31 e	0.18 a-c	0.33 c	0.43 b	0.69 c	0.69 b	
V ₂ M ₂	1.17 bc	1.02 cd	0.82 c	0.62 c	0.39 de	0.16 cd	0.26 de	0.36 cd	0.46 de	0.68 b	
V ₂ M ₃	1.20 a-c	1.10 a-c	0.95 b	0.80 b	0.60 b	0.15 d	0.24 e	0.34 d	0.44 e	0.54 d	
Level of significance	NS	NS	NS	NS	NS	NS	NS	NS	**	***	
CV%	4.36	5.28	6.88	8.82	12.15	6.06	3.38	2.34	1.72	1.63	_

Table 2.9 Combined effects of varieties and different doses of Maleic hydrazide solution on crude fibre and lipid content of postharvest mango during storage at ambient condition

7

2.3

2

2

144

+

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

7

Table represents			12
$V_1 = Langra$	M ₁ =200 ppm of MH solution	* indicates at 5% level	NS means non significant
$V_2 = Khirshapat$	M ₂ =400 ppm of MH solution	** indicate at 1 % level	-
$M_0 = Control$	$M_3 = 600 \text{ ppm of MH solution}$	*** indicate at 0.1% level	

×

 \mathbf{x}_{i}

A

T

trend of WSPC in mango pulp with the advancement of storage period (Fig. 2.23). It also revealed that WSPC was accumulated more in untreated fruit followed by the fruit treated with M_1 , M_2 and M_3 , treatments, respectively. The increasing trend of WSPC in control was very distinct from initial to 6 days thereafter, its increasing trend declined due to spoilage. At the same time, the increasing trend of WSPC from the fruit treated with M_3 treatment was very slower due to delay ripening. At 9th day, the highest (1.21%) accumulating of WSPC was recorded from control which was statistically identical with M_1 and the lowest (0.72%) was recorded from M_3 treatment, respectively. It elucidated that extension of WSPC synthesis was strongly depended upon fruit ripening. An increase in protein content observed during ripening is in agreement with the findings of Gomez-Lim (1997).

The combined effect of varieties and imposed different doses of MH solution in this experiment demonstrated non significant variation in respect of WSPC at different days after storage (Appendix 2.8). The results obtained from the investigation indicated that WSPC increased markedly in the treatment combination of V₁M₀ from initial to 9th day and then, it decreased due to becoming putrefied condition whereas, the lower increasing trend was noticed from the treatment combination of V₂M₃. At 9th day, the highest (1.24%) quantity of WSPC was counted from the treatment combination of V₁M₀ whereas; the lowest (0.68%) was counted from the treatment combination of V₂M₃.

8.4 Changes in minerals of mango during storage environments

Different changes regarding minerals of mango fruits are presented and discussed in the following sub-headings.

8.4.1 Phosphorus content

Variation in between varieties means in respect of P contents showed highly significant at different days after storage (Appendix 2.8). It was noticed a steadily increasing trend of P contents in both the varieties with the increase of storage period (Table 2.16). It also indicated that Langra gave comparatively more quantity of P as compared to Khirshapat. The increasing trend of P in Langra was more or less similar up to 9 days thereafter, its trend declined due to becoming spoilage of fruits.

x

X

X

At 12th day, the highest (23.52 mg/100 g) quantity of P was obtained from Langra and the lowest (21.65 mg/100 g) was obtained from Khirshapat. Better performance of increasing trend of P content in Langra might be due to genetical dissimilarities. It also elucidated that the increase of P content in mango was intimately related to ripening during storage. These results are in partially supported by the findings of Nadkarni (1963) when he worked with 16 cultivars. He found the ranged between 10-30 mg/100 g among cultivars. Peter *et al.* (2007) also showed the similar findings.

Different doses of MH solution adopted to this trial in terms of P content exhibited highly significant at different days after storage (Appendix 2.8). The results were noticed that P content augmented steadily at various days after storage. It also revealed that P content in control increased at slow motion from initial to 6th day thereafter; it increased at very slow rate than its previous trend. After 9th day, it declined possibly due to becoming hackneyed situation of fruit (Fig. 2.24). At 12th day, the highest (23.64 mg/100 g) quantity of P content was noticed from the fruit treated with M₁ treatment and the lowest value (21.67 mg/100 g) was noticed from the fruit treated with M3 treatment. Lower amount of P content from the fruit treated with M3 treatment might be possible due to delay ripening resulting in lower accumulation of P content and keeping the quality best for preservation. There were no available research articles regarding P changes as influenced by MH solution in the scientific literature but, the results of the present research revealed that P content was changed during storage.

The combined effect of varieties and imposed various doses of MH solution showed significant variation on P content of mango pulp at different Phosphorus (Appendix 2.8). It was found an increasing trend of P content with the increasing of storage period (Table 2.11). It also denoted that P content increased gradually from initial to 9th day with the treatment combination of V₁M₀ and then, it decreased due to starting of fruit bad situation. In this time, P content demonstrated lower increasing trend from the fruit treated with the treatment combination of V₂M₃. At 12th day, the highest (24.62 mg/100 g) quantity of P was recorded from the treatment combination of V₁M₁ whereas; the lowest (20.35 mg/100 g) was recorded from the treatment combination of V₂M₃. It also revealed that Langra along with M₃

treatment was found the best combination in keeping the quality good in preservation followed by other treatment combinations. These treatment combinations strictly impeded the ripening of mango fruits during storage.

8.4.2 Potassium content

×

Se.

A

A.

Analysis of variance of mango varieties in terms of K content showed highly significant at different days after storage (Appendix 2.9). It was observed an increasing trend of K content in both the varieties with the increase of storage period at different days during storage (Table 2.12). In all the storage period, Langra was found as more producer of K content as compared to Khirshapat. It was also noticed that the increasing trend of K content was suspended at 9th day of storage. At this period, the highest (0.28%) of K was obtained from Langra and the lowest (0.25%) was recorded from Khirshapat (Table 2.12). These events might be possible due to becoming spoilage of mango fruit resulting in lower metabolic activities. Langra was better producer of K content possibly due to genetical dissimilarities between varieties. At different storage period, K content increased might be possible due to transmission of K from stone and peel to pulp of mango.

Different doses of MH solution applied to this experiment in respect of K content of mango were highly significant at different days after storage (Appendix 2.9). The results indicated that K content augmented gradually with the passing of storage time. But, K content from the untreated fruit decreased at 9th day, whereas other treatments *viz.*, M_1 , M_2 and M_3 retained their increasing trends (Fig. 2.25). In this period, there was found a very lower increasing trend of K in the fruit treated with M_3 treatment. At 9th day, the maximum (0.29%) of K content was recorded from the untreated fruit whereas, the lowest (0.25%) was recorded from the fruit treated with M_3 treatment which was statistical similar with M_2 treatment. Lower dissemination of K content in the fruit treated with M_3 treatment might be possible due to delay ripening causing lower transmission of K resulting in keeping quality good.

The combined effect of varieties and induced different doses of MH solution in relation to K content of mango pulp demonstrated non significant at different days after storage (Appendix 2.9). It indicated that K content of mango pulp from different treatment combination increased gradually with the passing of storage time

(Table 2.13). But, only the treatment combination of V_1M_0 gave the highest decreasing trend at 9th day. In this period the highest (0.30%) quantity of K was reported from the treatment combination of V_1M_0 which was statistical identically with V_1M_1 whereas, the lowest (0.23%) was reported from the treatment combination of V_2M_3 .

8.4.3 Calcium content

3h

St.

14

×

Analysis of variance of imposed varieties to this investigation in terms of Ca content showed highly significant at different days after storage (Appendix 2.9). The results were observed an increasing trend of Ca content with the advancement of storage time from both the varieties (Table 2.12). It also indicated that Khirshapat was better in achieving of Ca content followed by Langra. At 12th day, the highest (22.70 mg/100 g) quantity of Ca was recorded from Khirshapat and the lowest (20.75 mg/100 g) was recorded from Langra. These events might be possible due to genetical variation in between two varieties. These results are in partially supported by the report of Nadkarni (1963).

Different doses of MH solution in respect of Ca content of mango pulp demonstrated significant variation at different days after storage (Appendix 2.9). At various days during storage, Ca content augmented gradually with the advancement of storage time (Fig. 2.26). It also elucidated that Ca content from control increased markedly from initial to 6th day and then, it increased smoothly thereafter, it decreased due to starting decay. At the same time, Ca content from the fruit treated with M₃ treatment increased very smoothly. At 9th day, the highest (24.53 mg/100 g) quantity of Ca was recorded from untreated fruit whereas; the lowest (16.30 mg/100 g) was recorded from the fruit treated with M₃ treatment. These phenomena might be possible due to delay of ripening caused in lower transmission of Ca content from peel and stone to pulp of mango. The results of the present investigation are not supported by the findings of Petter *et al.* (2007).

The combined effect of varieties and imposed different doses of MH solution in terms of Ca content of mango pulp exhibited significant variation at different storage duration (Appendix 2.9). The results was noticed an increasing trend of Ca content of mango pulp from various treatment combinations (Table 2.13). At 9th day, the highest (25.39 mg/100 g) quantity of Ca was obtained from the treatment

the.

2h

×

1

*



Fig. 2.23 Effect of different doses of MH solution on water soluble protein content of mango pulp at different days after storage. Vertical bars represent LSD at 0.05 level



Fig. 2.25 Potassium content of mango pulp as influenced by different doses of MH solution at different days after storage. Vertical bars represent LSD at 0.05 level

 M_2 =400 ppm of MH solution M_3 =600 ppm of MH solution



Fig. 2.24 Effect of different doses of MH solution on phosphorus content of mango pulp at different days after storage. Vertical bars represent LSD at 0.05 level



Fig. 2.26 Calcium content of mango pulp as influenced by different doses of MH solution at different days after storage. Vertical bars represent LSD at 0.05 level

Results and Discussion

Treatments	Wate	r soluble prot	tein content	(%) at diffe	rent days		Phosphor	us (mg/100 g) at different da	ys
Variety	Initial	3	6	9	12	Initial	3	6	9	17
V ₁	0.56 a	0.71 a	0.89 a	1.01 a	1.19 a	19.42 a	20.43 a	21.89 a	23.20 a	23.52 a
V ₂	0.49 b	0.62 b	0.77 b	0.92 b	1.13 b	16.15 b	17.83 b	19.30 b	20.72 b	21.65 b
Level of significance	NS	NS	*	*	NS	***	***	***	***	***

Table 2.10 Changes of water soluble protein and phosphorus content in postharvest mango varieties during storage at ambient condition

13

Table 2.11 Combined effects of varieties and different doses of Maleic hydrazide solution on water soluble protein and phosphorus content of postharvest mango during storage at ambient condition

Treatments combination	Water	soluble prote	ein content (%) at differe	ent days		Phosphorus	(mg/100 g)) different d	ays
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12
V ₁ M ₀	0.54	0.85 a	1.21 a	1.24 a	1.22	19.86 a	21.88 a	24.38 a	24.78 a	23.38 b
V ₁ M ₁	0.57	0.69 a-c	0.83 b	1.18 a	1.24	19.66 b	20.46 c	21.96 c	24.42 b	24.62 a
V ₁ M ₂	0.55	0.67 a-c	0.78 bc	0.88 bc	1.20	19.25 c	19.95 d	20.98 d	22.12 d	23.08 c
V ₁ M ₃	0.58	0.62 bc	0.73 bc	0.75 cd	1.10	18.92 d	19.43 e	20.24 e	21.48 e	22.98 c
V ₂ M ₀	0.50	0.77 ab	1.12 a	1.18 a	1.19	16.55 e	20.95 b	23.45 b	23.95 c	22.45 e
V2M1	0.51	0.60 bc	0.70 bc	1.05 ab	1.17	16.25 f	17.36 f	18.88 f	21.32 e	22.65 d
V ₂ M ₂	0.48	0.57 bc	0.64 bc	0.75 cd	1.12	15.84 g	16.54 g	17.55 g	19.05 f	21.13 f
V ₂ M ₃	0.45	0.55 c	0.60 c	0.68 d	1.05	15.94 g	16.48 g	17.32 h	18.55 g	20.35 g
Level of significance	NS	NS	NS	NS	NS	*	***	***	***	***
CV%	17.65	15.95	12.63	11.01	9.26	0.60	0.55	0.52	0.48	.47

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents

 $V_1 = Langra$ $V_2 = Khirshapat$ $M_0 = Control$ $M_1=200$ ppm of MH solution $M_2=400$ ppm of MH solution $M_3=600$ ppm of MH solution * indicates at 5% level NS means non significant

** indicate at 1 % level

*** indicate at 0.1% level

Treatments		Potassiu	ım (%) at d	ifferent days	;		Calcium (m	ng/100 g) at	different da	ys
Variety	Initial	3	6	9	12	Initial	3	6	9	12
V.	0.23 a	0.25 a	0.27 a	0.28 a	0.28 a	10.65 b	13.02 b	16.26 b	19.45 b	20.75 b
V ₂	0.18 b	0.19b	0.22 b	0.25 b	0.25 b	12.47 a	14.95 a	18.19 a	21.39 a	22.70 a
evel of significance	***	***	***	***	***	***	***	***	***	***

7

Table 2.12 Changes of potassium and calcium content between postharvest mangoes during storage at ambient condition

Table 2.13 Combined effects of varieties and different doses of Maleic hydrazide solution on potassium and calcium content of postharvest mango during storage at ambient condition

Treatments combination		Potassium c	ontent (%) a	at different da	ays	Cal	cium conten	t (mg/100	g) at differe	ent days
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12
V ₁ Mo	0.24 a	0.26 a	0.28 a	0.30 a	0.27 a-c	10.85 e	15.45 b	22.18 b	23.68 c	20.18 f
V, M,	0.23 ab	0.25 ab	0.27 ab	0.28 ab	0.29 a	10.65 f	13.15 e	16.25 d	21.82 d	22.88 c
V ₁ M ₂	0.21 c	0.24 a-c	0.26 a-c	0.27 bc	0.28 ab	10.35 g	11.82 f	13.67 g	16.88 g	21.42 e
V ₁ M ₃	0.22 bc	0.23 bc	0.25 b-d	0.26 b-d	0.27 a-c	10.75 ef	11.65 f	12.92 h	15.42 h	18.53 g
VaMa	0.19 d	0.22 c	0.24 c-e	0.27 bc	0.25 cd	12.55 b	17.16 a	23.89 a	25.39 a	21.89 d
V2M1	0.18 de	0.19 d	0.23 de	0.25 с-е	0.26 b-d	12.82 a	15.42 b	18.53 c	24.09 b	25.15 a
V2M2	0.17 e	0.18 d	0.22 e	0.24 de	0.25 cd	12.35 c	13.86 c	15.72 e	18.92 e	23.45 b
V2M3	0.19 d	0.17 d	0.20 f	0.23 e	0.24 d	12.15 d	13.35 d	14.62 f	17.17 f	20.29 f
Level of significance	NS	NS	NS	NS	NS	***	***	***	**	**
CV%	5.21	4.88	4.35	4.04	4.02	0.92	0.76	0.62	0.52	0.49

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents	
$V_1 = 1 \text{ and } ra$	M ₁ =200 ppm
$V_2 = Khirshapat$	M ₂ =400 ppm
Ma =Control	M ₃ =600 ppn

10

200 ppm of MH solution 400 ppm of MH solution 600 ppm of MH solution

* indicates at 5% level NS means the NS mean

*** indicate at 0.1% level

NS means non significant

K

7

1

combination of V_2M_0 and the lowest (15.42 mg/100 g) was obtained from the treatment combination of V_1M_3 , respectively.

8.4.4 Magnesium content

Highly significant variation was found in Mg content between the varieties means at different days after storage (Appendix 2.10). The results were observed that Mg content increased gradually with the passing of storage period (Table 2.14). It also reveled that Mg content increased steadily from initial to 9th days thereafter, it decreased slightly due to becoming gone bad of the fruit situation. At 9th day, the highest (17.61 mg/100 g) was reported from Langra and the lowest (16.98 mg/100 g) was recorded from Khirshapat. This phenomenon might be due to genetical dissimilarities between the varieties. There were no available research findings relating to Mg content during storage in the scientific literature. The results of the present study revealed that Langra was batter in Mg accumulation over Khirshapat but, it has also increased during storage.

Different doses of MH solution induced in this trial exhibited highly significant variation in relation to Mg content of mango pulp at different days after storage (Appendix 2.10). It was noticed a smooth increasing trend of Mg content of mango pulp with the advancement of storage period (Fig. 2.27). It also explored that Mg content from control increased gradually from initial to 6th day and then, it decreased very sharply. The highest (17.90, 17.74 and 18.28 mg/100 g) quantity of Mg was recorded from control, M₁ and M₂ treatment at 6, 9 and 12th day whereas, the lowest (16.30, 17.04 and 17.77 mg/100 g) was recorded from the fruit treated with M₃ treatment, respectively. The augmentation of Mg content in mango pulp might be related to starting of ripening.

The combined effect of varieties used and various doses of MH solution imposed in this experiment in terms of Mg content of mango pulp showed significant variation at various days except 3 and 9 (Appendix 2.10). The results noticed an increasing trend of Mg content of mango pulp with the increase of storage time (Table 2.15). At 6th day, the highest (18.25 mg/100 g) quantity of Mg was attained

from the treatment combination of V_1M_0 whereas; the lowest (15.75 mg/100 g) was recorded from the treatment combination of V_2M_3 , respectively.

8.4.5 Copper content

ж.

×

Variation in between varieties in terms of Cu content of mango pulp showed highly significant at different days after storage (Appendix 2.10). The results revealed that Cu content of mango pulp decreased gradually with the advancement of storage period (Table 2.14). It also denoted that Cu content of Khirshapat was more as compared to Langra. At initial day, higher (0.35 mg/100 g) content of Cu was recorded in Khirshapat and lower (0.32mg/100 g) was recorded in Langra. At 12th day, Khirshapat produced the highest (0.20 mg/100 g) amount of Cu and the lowest (0.18 mg/100 g) was recorded from Langra (Table 2.14). These events might be possible due to genetical dissimilarities between the varieties. There were no available research reports regarding Cu content of mango at the scientific literature but, the present research findings revealed that Cu content reduced gradually with the advance of storage duration.

Different doses of MH solution used in the present study in relation to Cu content of mango pulp exhibited highly significant variation at different days after storage except initial day (Appendix 2.10). The results indicated that Cu content decreased gradually with the increase of storage period at various days. The decreasing trend from control had higher than the fruit treated with M_1 , M_2 and M_3 treatment respectively. At 12^{th} day, higher quantity of Cu (0.24 mg/100 g) was obtained from the fruit treated with M_3 whereas, lower (0.12 mg/100 g) was obtained from control (Fig. 2.28). It meant M_3 treatment was much better than the other treatments in conservation of Cu content. So, the present findings elucidated that M_3 treatment was the best treatment in preservation of mango.

The combined effect of varieties and imposed different doses of MH solution in terms of Cu content of mango were found to be non significant variation at various days after storage except initial and 12th day (Appendix 2.10). The results denoted that Cu content in different treatment combination decreased with the increase of storage duration. At 12th day, the highest (0.26 mg/100 g) was recorded from the

treatment combination of V_2M_3 , and the lowest (0.11 mg/100 g) was recorded from the treatment of V_1M_0 (Table 2.15).

8.4.6 Iron content

AL.

K

The analysis of variance of varieties in relation to Fe content of mango pulp was found to be highly significant at different days after storage (Appendix 2.11). The results were observed an increasing trend of Fe content with the advancement of storage duration. The increasing trend of Fe content showed more or less similar from initial to 9th day thereafter, it decreased slightly. At 9th day, the highest (4.62 mg/100 g) quantity of Fe was recorded from Langra and the lowest (3.81 mg/100 g) was recorded from Khirshapat. These results are in partially supported by the findings of Nadkarni (1963). These events might be possible due to genetical dissimilarities between the varieties.

Highly significant variation was found due to the effect of different doses of MH solution on Fe content of mango pulp at various days after storage (Appendix 2.11). The results indicated that Fe content increased markedly from initial to 6th day and then, it decreased markedly at control and similarly M₁ treated fruits decreased of Fe after 9th days. But, Fe content obtained from the other fruits treated with M₂ and M₃ treatment increased gradually up to 12^{th} day (Fig. 2.29). At 12^{th} day, the highest (5.03 mg/100 g) quantity of Fe was recorded from the fruit treated with M₂ treatment whereas; the lowest (3.07 mg/100 g) was recorded from the untreated fruit. A smaller quantity of Fe obtained from untreated fruits was possibly due to decaying resulted in lower amount. On the other hand Fe content gradually augmented in M₃ treatment containing their good quality. These results are partially supported by the findings of Peter *et al.* (2007).

There was non significant variation in respect of Fe content of mango pulp at different days after storage as influenced by the combined effect of varieties and implied different doses of MH solution (Appendix 2.11). These results indicated that an increasing trend of Fe content was found from initial to 6^{th} day thereafter, it decreased very fast from the treatment combination of V₁M₀ whereas the lowest increasing trend was noticed in the treatment combination of V₂M₃, respectively (Table 2.17). At 6^{th} day, the highest (5.75 mg/100 g) quantity of Fe was recorded

A.

Yr.

from the treatment combination of V_1M_0 and the lowest (1.88 mg/100 g) was recorded from the treatment combination of V_2M_3 .

8.4.7 Manganese content

Manganese Mn content of mango pulp was found to be differed significantly in both the varieties mean at different days after storage (Appendix 2.11). The results noticed that Mn content increased gradually with the advancement of storage period (Table 2.16). It also explored that Langra showed better in the accumulation of Mn content as compared to Khirshapat. At 9th day, the highest (1.16 mg/100 g) quantity of Mn was reported from Langra whereas; the lowest (0.96 mg/100 g) was reported from Khirshapat, respectively (Table 2.16). It also explored that Mn content did not increase after 9th day. There were no available research reports relating to Mn content in the scientific literature. The present research findings revealed that Langra was better than Khirshapat in Mn content accumulation.

Different doses of MH solution were highly significant on Mn content of mango pulp at different days after storage (Appendix 2.11). The results noticed that Mn content augmented gradually from initial to 6^{th} day, and then it reduced markedly in control. On the other hand, it increased from initial to 9^{th} day thereafter; it reduced in the fruit treated with M₁ treatment. At the same time, very lower increasing trend of Mn content was recorded from the fruit treated with M₃ treatment (Fig. 2.30). At 12^{th} day, Mn content ranged between 0.89 to 1.21 mg per 100 g of fresh pulp of mango. The maximum (1.21 mg/100 g) was noticed from M₂ treated fruits and the minimum (0.89 mg/100 g) was recorded from control.

The combined effect of varieties and used different doses of MH solution in respect of Mn content of mango pulp were found to be non significant at various days after storage (Appendix 2.11). The results observed an increasing trend of Mn content in different treatment combination with the increase of storage duration (Table 2.17) At 6th day, the highest (1.32 mg/100 g) was obtained from the treatment combination of V₁M₀, and the lowest (0.62 mg/100 g) was obtained from the treatment combination of V₂M₃, (Table 2.17). At 12th day, the highest (1.32 mg/100 g) was noticed from the treatment combination of V₁M₂ and the minimum (0.92 mg/100 g) was noticed from the treatment combination of V₂M₃.

-

2

×

2

X



Fig. 2.27 Effect of different doses of MH solution on magnesium content of mango pulp at different days after storage. Vertical bars represent LSD at 0.05 level



Fig. 2.29 Iron content of mango pulp as influenced by different doses of MH solution at different days after storage. Vertical bars represent LSD at 0.05 level

Graph represents $M_0 = Control$ $M_1 = 200 \text{ ppm of MH solution}$

 M_2 =400 ppm of MH solution M_3 =600 ppm of MH solution

MD 🔳 M1 M2 MB 0.4 0.35 Copper content (mg/100 g) 0.3 0.25 0.2 0.15 0.1 0.05 0 Initial 6 9 3 12 Days after storage

Fig. 2.28 Effect of different doses of MH solution on copper content of mango pulp at different days after storage. Vertical bars represent LSD at 0.05 level





Treatments Magnesium content (mg/100 g) at different days Copper content (mg/100 g) at different days Initial Variety 12 3 6 12 Initial 3 6 9 9 17.59 b 16.95 a 0.32 b 0.18 b V₁ 16.37 a 16.74 a 17.61 a 0.28 b 0.24 b 0.21 b V₂ 15.39 b 15.98 b 16.42 a 16.98 b 16.53 b 0.35 a 0.27 a 0.24 a 0.20 a 0.31 a *** *** *** *** *** *** *** *** Level of significance *** ***

Table 2.14 Changes of magnesium and copper content in postharvest mango varieties during storage at ambient condition

1

Table 2.15 Combined effects of varieties and different doses of Maleic hydrazide solution on magnesium and copper content of postharvest mango du	Iring
storage at ambient condition	

Treatments combination	Magn	esium conte	nt (mg/100	g) at differe	ent days	Сор	per content	(mg/100 g) at differen	t days
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12
V ₁ M ₀	16.88 a	17.28 a	18.25 a	17.15 cd	15.22 g	0.32 cd	0.26 e	0.21 e	0.16 e	0.11 e
V ₁ M ₁	16.63 b	16.98 b	17.82 b	18.13 a	16.25 e	0.33 b-d	0.28 c-d	0.24 cd	0.21 d	0.17 d
V1M2	16.22 c	16.55 c	17.45 c	17.88 b	18.45 a	0.31 d	0.27 de	0.25 cd	0.23 cd	0.20 c
V ₁ M ₃	15.75 d	16.15 d	16.85 d	17.26 cd	17.89 c	0.32 cd	0.29 cd	0.26 bc	0.25 bc	0.22 bc
V ₂ M ₀	15.65 de	16.45 c	17.55 c	16.62 f	14.79 h	0.35 ab	0.29 cd	0.23 de	0.18 e	0.13 e
V2M1	15.48 ef	16.13 d	16.25 e	17.35 c	15.55 f	0.34 a-c	0.30 bc	0.26 bc	0.23 cd	0.17 d
V ₂ M ₂	15.32 f	15.88 e	16.12 e	17.13 d	18.12 b	0.36 a	0.32 ab	0.28 b	0.26 b	0.23 b
V ₂ M ₃	15.12 g	15.45 f	15.75 f	16.82 e	17.65 d	0.35 ab	0.33 a	0.31 a	0.29 a	0.26 a
Level of significance	**	NS	***	NS	*	*	NS	NS	NS	*
CV%	0.72	0.65	0.62	0.66	0.63	3.17	3.63	4.16	4.69	5.69

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents

$V_1 = Langra$	M ₁ =200 ppm of MH solution	* indicates at 5% level
$V_2 = Khirshapat$	M ₂ =400 ppm of MH solution	** indicate at 1 % level
M ₀ =Control	$M_3 = 600 \text{ ppm of MH solution}$	*** indicate at 0.1% lev

NS means non significant

1

"he

e at 1 % level ate at 0.1 % level

157

Treatments	Ir	on content	(mg/100 g)	at different	days	M	anganese o	content (mg/10	00 g) at differen	t days
Mariabi	Initial	3	6	9	12	Initial	3	6	9	12
Variety	2.34 a	2.96 a	3.97 a	4.62 a	4.58 a	0.60 a	0.79 a	1.02 a	1.16 a	1.16 a
	1.55 b	2.18 b	3.16 b	3.81 b	3.77 b	0.42 b	0.60 b	0.83 b	0.96 b	0.96 b
	***	***	***	***	***	***	***	***	***	***

Table 2.16 Changes of iron and manganese content in postharvest mango varieties during storage at ambient condition

100

-

Table 2.17 Combined effects of varieties and different doses of Maleic hydrazide solution on iron and manganese content of postharvest mango during storage at ambient condition

reatments combination	I	ron content	Mang	Manganese content (mg/100 g) at different days						
Varieties × Treatments	Initiał	3	6	9	12	Initial	3	6	9	12
1/ M-	2.68 a	3.65 a	5.75 a	5.25 a	3.46 d	0.62 a	0.94 a	1.32 a	1.22 b	0.99 d
	2.35 b	3.05 b	4.12 c	5.35 a	4.78 b	0.63 a	0.85 b	1.08 c	1.35 a	1.22 b
V 1111	2.22 bc	2.72 c	3.25 d	4.22 c	5.43 a	0.58 b	0.73 d	0.89 d	1.11 d	1.32 a
V 1M3	2.12 c	2.43 d	2.74 e	3.65 d	4.66 b	0.55 c	0.65 e	0.78 e	0.94 f	1.09 c
VaMa	1.85 d	2.88 bc	4.95 b	4.46 b	2.68 e	0.45 d	0.75 c	1.12 b	1.02 e	0.79 f
V ₂ M.	1.65 e	2.35 d	3.38 d	4.58 b	3.96 c	0.42 e	0.64 e	0.88 d	1.14 c	1.01 d
V-Ma	1.42 f	1.92 e	2.42 f	3.43 e	4.63 b	0.38 f	0.53 f	0.69 f	0.89 g	1.10 c
V 2 ¹¹ 2	1.28 f	1.58 f	1.88 g	2.78 f	3.79 c	0.41 e	0.48 g	0.62 g	0.78 h	0.92 e
	NS	NS	NS	NS	NS	NS	NS	NS	NS	NS
CV%	5.45	4.12	2.98	2.52	2.54	4.20	3.05	2.30	2.01	2.01

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents $V_1 = Langra$

158

* indicates at 5% level ** indicate at 1 % level

*** indicate at 0.1% level

NS means non significant

7.

- $V_2 = Khirshapat$ Mo =Control
- M_1 =200 ppm of MH solution M_2 =400 ppm of MH solution M_3 =600 ppm of MH solution

8.4.8 Zinc content

Variation in respect of Zn content of mango pulp due to the effect of varieties demonstrated highly significant at different days after storage (Appendix 2.12). The results indicated that Zn content decreased markedly with the advancement of storage duration from both the varieties (Table 2.18). It also revealed that Khirshapat was found to provide more Zn quantity as compared to Langra. Zn content ranged between 1.31 to 1.54 mg per 100 g of mango pulp at initial day and 0.76 to 0.89 mg per 100 g at 12th day. The highest (1.54 and 0.89 mg/100 g) was obtained from Khirshapat and the lowest (1.31 and 0.76 mg/100 g) was obtained from Langra at initial and 12th day, respectively. There were no available research findings relating to Zn content in the scientific review.

Different doses of MH solution caused significant variation in Zn content of mango pulp at different days after storage (Appendix 2.12). The results observed a decreasing trend of Zn content of mango pulp with the advancement of storage period from the fruit treated with different doses of MH solution. It also denoted that the decreasing trend was very high from control and very low from the fruit treated with M₃ treatment (Fig. 2.31). At 12th day, the maximum (0.99 mg/100 g) was obtained from the fruit treated with M₃ treatment and the lowest (0.63 mg/100 g) was obtained from control. Zn content of mango pulp decreased in the storage period was possibly due to either the transmission of Zn from pulp to stone and peel or Zn content of mango pulp may be depressed or suppressed as influenced by metabolic activities during storage.

The combined effect of varieties and imposed different doses of MH solution in terms of Zn content of mango showed non significant variation at different days after storage (Appendix 2.12). There exhibited the decreasing trend of Zn content of mango pulp from various treatment combinations (Table 2.19).

8.5 Shelf life

Highly significant variation was observed from varieties on shelf life of mango (Appendix 2.12). The longest shelf life (13.17 days) was recorded from Khirshapat and the shortest (11.83 days) was recorded from Langra (Fig. 2.32). Variation in

shelf life between varieties might be possible due to genetical. The present findings from the research revealed that Khirshapat was better than Langra in preservation.

Imposed different doses of MH solution in this investigation in terms of shelf life of mango showed highly significant (Appendix 2.12). The shelf life of mango ranged between 8.00 to 15.50 days from different storage treatments (Fig 2.23). The longest shelf life (15.50 days) was noticed from the fruit treated with M₃ treatment followed by the shelf life of the fruits treated with M₂ (13.67 days), and M₁ (12.83 days), treatments whereas; the shortest shelf life (8.00 days) was noticed from control, respectively (Fig. 2.24). The longest shelf life obtained from the fruit treated with M₃ treatment might be possibly due to suppression or depression of physiological and biochemical activities which is responsible for slower senescence of harvested fruits and consequently led to the longest shelf life. The results are in conformity with the findings of Tefera *et al.* (2007), Gautam *et al.* (2003) also found the similar results.

The combined effects of varieties and induced different dosses of MH solution had non significant effect in terms of shelf life of mango (Appendix 2.12). The shelf life ranged between 7.33 to 16.00 days from various treatment combinations. The longest shelf life (16.00 days) was obtained from the treatment combination of V_2M_3 , whereas the shortest (7.33 days) was obtained from V_1M_0 .





Fig. 2.31 Effect of different doses of MH solution on zinc content of mango pulp at different days after storage. Vertical bars represent LSD at 0.05 level

Fig. 2.32 Effect of varieties on shelf life of mango

A

X

A.

£

2.





Graph represents

$V_1 = Langra$	M ₁ =200 ppm of MH solution
$V_2 = Khirshapat$	M ₂ =400 ppm of MH solution
$M_0 = Control$	$M_3 = 600 \text{ ppm of MH solution}$

Treatments	Zinc content (mg/100 g) at different days							
Variety	Initial	3	6	9	12	Total days		
V1	1.31 b	1.20 b	1.07 b	0.92 b	0.76 b	11.83 b		
V ₂	1.54 a	1.32 a	1.20 a	1.05 a	0.89 a	13.17 a		
Level of significance	***	***	***	***	***	***		

*

H

Table 2.18 Changes of zinc content in varieties and shelf life of postharvest mango during storage at ambient condition

*

+

Y

Table 2.19 Combined effects of varieties and different doses of Maleic hydrazide solution on zinc content and shelf life of postharvest mango during storage at ambient condition

Treatments combination		Zinc conte	nt (mg/100 g) at	different days		Shelf life
Varieties × Treatments	Initial	3	6	9	12	Total days
V ₁ M ₀	1.30 ef	1.15 f	0.98 f	0.78 g	0.53 f	7.33 e
V ₁ M ₁	1.28 f	1.18 e	1.03 e	0.87 f	0.70 e	12.33 c
V1 M2	1.32 de	1.22 d	1.11 d	0.99 d	0.85 c	12.67 c
V_1M_3	1.33 d	1.24 d	1.15 c	1.05 c	0.94 b	15.00 ab
V ₂ M ₀	1.52 bc	1.28 c	1.12 d	0.93 e	0.73 d	8.67 d
V ₂ M ₁	1.51 c	1.31 b	1.16 c	0.99 d	0.84 c	13.33 c
V ₂ M ₂	1.54 b	1.33 b	1.22 b	1.09 b	0.95 b	14.67 b
V ₂ M ₃	1.58 a	1.37 a	1.28 a	1.17 a	1.03 a	16.00 a
Level of significance	NS	NS	NS	NS	**	NS
CV%	1.47	1.68	1.88	2.16	2.58	5.76

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents

$V_1 = Langra$	M ₁ =200 ppm of MH solution	* indicates at 5% level	NS means non significant
$V_2 = Khirshapat$	$M_2 = 400$ ppm of MH solution	** indicate at 1 % level	
M ₀ =Control	$M_3 = 600 \text{ ppm of MH solution}$	*** indicate at 0.1% level	

Jer'

9.0 Experiment 3

Effect of different doses of Gibberellic acid on physicochemical behavior and shelf life of postharvest mango

9.1 Changes in skin color

Different doses of Gibberellic acid (GA3) strongly influenced the skin color of both the fruits namely Langra and Khirshapat (Table 3.1). Both varieties demonstrated the original green color at the initial day of harvesting. At 3^{rd} day, Langra developed into a trace in yellow color at control (G₀) and light green color at 100 ppm (G₁) and 200 ppm (G₂) as well as held green color at 400 ppm (G₃) treatment. Khirshapat showed trace in yellow color at control, light green color at G₁ and G₂ treatment. But, it retained its original green color at G₃ treatment.

At 6th day, Langra was noticed yellow at control (G_0), yellowish green at 100 ppm (G_1) and 200 ppm (G_2) while light green was observed from 400 ppm (G_3) treatment. On the other hand, Khirshapat developed yellowish green, trace in yellow and yellowish green color at G_0 , G_1 , and G_2 treatments, respectively but, it retained its original green color at G_3 treatment.

At 9th day, Langra gave deep yellow color at control, greenish yellow at G_1 and G_2 as well as trace in yellow color at G_3 treatments. Khirshapat showed yellow color at control, greenish yellow and trace in yellow at G_1 and G_2 treatments. But, it also retained its original green color at G_3 treatment.

After 12^{th} day of storage, Langra was found blackish yellow at control and yellow at G₁ and yellowish green at G₂ and G₃ treatments. On the other hand, Khirshapat was also found blackish yellow color at G₀, yellow, yellowish green and trace in yellow at G₁, G₂ and G₃ treatments, respectively. After 15 days of storage, Langra showed no existence at control and G₁ treatment yellow and greenish yellow at G₂ and G₃ treatments, respectively. Khirshapat completely denoted as putrefied condition at G₀, and G₁, treatment and yellow and yellowish green was noticed from G₂ and G₃ treatments, respectively.

				Day	/s after storage		
Varieties	Treatments	Initial	3	6	9	12	15
	Go	Green	Trace in yellow	Yellow	Deep yellow	Blackish yellow	-
	G1	Green	Light green	Yellowish green	Greenish yellow	Yellow	_
V ₁	G₂	Green	Light green	Yellowish green	Greenish yellow	Yellowish green	Yellow
	G3	Green	Green	Light green	Trace in yellow	Yellowish green	Greenish yellow
	Go	Green	Trace in yellow	Yellowish green	Yellow	Light spotted yellow	_
	G1	Green	Light green	Trace in yellow	Greenish yellow	Yellow	_
V ₂	G₂	Green	Light green	Yellowish green	Trace in yellow	Yellowish green	Yellow
	G3	Green	Green	Green	Green	Trace in yellow	Yellowish green

Table 3.1 Changes in skin color of two mango varieties as influenced by different doses of Gibberellic acid during storage at ambient condition

>.

X

×

71

	-	
ί	,	٦
1		
-	H	^
		04

1.10

Table indicates $V_1 = Langra$ G

The

 V_1 = Langra G_0 =Control V_2 = Khirshapat G_1 =100 ppm of GA3

 $G_2 = 200 \text{ ppm of GA3}$ $G_3 = 400 \text{ ppm of GA3}$ A.

10-

A

5

p-

S.

9.2 Changes in physical characters during storage environments

Different parameters subjected to the physical changes of mangoes are presented and elucidated in the following sub-headings.

9.2.1 Physiological weight loss

Varieties had highly significant variation in relation to PWL at different days after storage (Appendix). At each day, Khirshapat (V₂) gradually showed more PWL as compared to Langra with the increase of storage duration (Table 3.2). The highest (10.62%) and lowest (9.72%) of PWL were reported from Khirshapat and Langra at 12th day, respectively. The results explored that total PWL progressively augmented with the advancement of storage duration. The findings also elucidated that Langra showed better performance in respect of PWL as compared to Khirshapat. Water loss through lenticel seems to be the probable cause of physiological weight loss in the fruits during storage. Lower lenticel density in Langra facilitated lesser water loss leading to minimum total weight loss (Azad, 2001). Singh *et al.* (2000) also reported more or less the same findings.

Analysis of variance of mango in terms of PWL as influenced by different doses of Gibberellic acid (GA3) showed highly significant variation at different days after storage (Appendix 3.1 and). At different days, it explored that control treatment was very faster in PWL compared to G_1 , G_2 and G_3 treatments. At 12^{th} day, the maximum PWL (11.64%) was noticed at G_0 and minimum (9.09%) at G_3 treatment (Fig. 3.1). These phenomena happened might be possibly due to 400 ppm of GA3 solution reduced the metabolic activities of mango resulting in lower PWL. These results are in partial coincided with the findings of Reddy and Haripriya (2002).

The combined effect of varieties and different doses of GA3 solution demonstrated significant variation in PWL at different days after storage (Appendix 3.1). At different days, it denoted that various treatments combination affected in PWL gradually with the advancement of storage duration. At 12^{th} day, there showed the maximum PWL (11.97%) at V₂G₀ and the minimum (8.65%) at V₁G₃. It also

1

indicated that langra lost the minimum amount of water along with G_3 treatment followed by the other treatment combinations.

9.2.2 Moisture content

The analysis of variance of imposed varieties exhibited highly significant variation in moisture content at different days after storage except initial day (Appendix 3.1). At different days, it interpreted that moisture content augmented with the passing of storage duration. The increasing trend was more or less the similar from initial to 9th day and thereafter, it reduced the increasing trend due to decay. It also illustrated that each day of storage, Khirshapat gave more moisture comparing to Langra. The highest (87.85%) and the lowest (85.85%) were recorded from V₂ and V₁ at 12th day, respectively (Table 3.2). Shajahan *et al.* (1994) observed different results in this regard and the moisture content increased in Langra than Khirsapat. But, these results are in agreement with the findings of Azad (2001). This variation might be possible due to genetical, location, weather effect and soil quality or maturity of the fruit.

Variation among the means of different doses of GA3 solution in relation to moisture content was noticed to be significant at different days after storage except initial day (Appendix 3.1). At different days of storage, moisture content increased in a continuous stream with the increase of storage duration. The last increased point was observed from control and G_1 treatment at 6 and 9th day (Fig. 3.2) whereas; G_2 and G_3 treatments produced increasingly onward. Untreated fruit gave the highest moisture content (87.40%) at 9th day and the lowest (85.30%) was recorded from G_3 treatment. The increasing trend of moisture content from initial to 6th day might be possibly due to metabolic activities and osmotic pressure inside the mango fruit as well as its decreased might be due to suppression of metabolic activities resulting in decaying and drying.

The combined effect of varieties and different doses of GA3 solution in respect of moisture content showed non significant variation at different days after storage (Appendix 3.1). At different days of storage, moisture content was added in mango with the increase of storage period. The treatment combinations of V_2G_0 , V_2G_1 and

1

*

F

S.

 V_2G_2 provided the maximum moisture content (88.40%, 88.40% and 88.80%) at 6, 9 and 12th days, respectively (Table 3.3). In this storage period, the lowest values (83.50%, 84.30% and 85.10%) were added from the treatment combination of V_1G_3 (Table 3.3). The increase of moisture content from initial to consequent day might be possible due to metabolic activities and osmotic principles and decreasing in a certain day might be due to suppression metabolic activities resulting in drying, transpiration and evaporation. Moisture content increased in the present study is in partially agreement with the findings of EI-Mahmoudi and Eisawi (1968) In banana.

9.2.3 Dry matter content

The variation in varieties means in relation to dry matter content showed highly significant at different days after storage (Appendix 3.2). At each day, dry matter decreased successively with the passing of storage duration. It indicated that Langra provided comparatively more dry matter comparing to Khirshapat. At initial day, Langra produced higher (17.65%) dry matter whereas, Khirshapat gave lower (15.64%) and at 12th day, Langra received the maximum (14.20%) whereas Khirshapat received 12.15 % (Table3.4). These results are in partially supported by the findings of Hassain (1991).

Different doses of GA3 solution imposed in this study in terms of dry matter content of mango pulp showed highly significant variation at different days after storage (Appendix 3.2). At different days of storage, dry matter content reduced continuously with the advancement of storage duration (Fig. 3.3). It also denoted that G₃ treatment contributed the highest dry matter (15.50%) at 6th day whereas; the lowest dry matter (12.65%) was reported from control.

The combined effect of varieties and different doses of GA3 solution on dry matter content of mango pulp exhibited non significant variation at different days after storage (Appendix 3.2). At initial day, the highest dry matter content (17.80%) was obtained from the treatment combination of V_1G_3 which was statistically similar with the treatment combination of V_1G_0 , V_1G_1 and V_1G_2 whereas; the lowest (15.40%) was obtained from the treatment combination of V_2G_0 which was also statistical similar with the combinations of V_2G_1 and V_2G_2 . At 12th day, the highest value

(14.90%) was noticed from the treatment combination of V_1G_3 and V_1G_0 and the lowest value (11.20%) was noticed from V_2G_2 (Table 3.5). The successively decrease in dry matter content with the advancement of storage period might be possibly due to breaking down the complex carbohydrates into simple molecules and H₂O as well as adding water through osmotic process and metabolic activities. The decreasing of dry matter content of mango pulp was not supported by the findings of Hossain (1999).

9.2.4 Ash content

F

S

Variation in varieties means in respect of ash content of mango pulp exhibited significant variation at different days after storage except 3rd day (Appendix 3.2). At different days of storage, values of ash content decreased continuously with the passing of storage duration. It was noticed that langra produced comparatively more ash than Khirshapat at all days of storage. Higher (1.01%) ash content was recorded from Langra at initial day whereas Khirshapat gave (0.90%) and again, higher (0.81%) was gathered from Langra and lesser achievement (0.70%) was gathered from Khirshapat.

Different doses of GA3 solution were identified as non significant variation in respect of ash content of mango pulp at different days after storage (Appendix 3.2). It indicated that ash content influenced by different doses of GA3 decreased slightly from G_3 treatment and markedly from control (Fig. 3.4).

The combined effect of varieties and different doses of GA3 solution were reported to be non-significant variation in terms of ash content of mango at various days after storage (Appendix 3.2 and Table 3.5).There was no available research findings regarding ash content of mango. It also exposited that ash content was intimately associated with dry matter content. The results of the present study elucidated that ash content decreased in relation to dry matter content.

9.3 Changes in biochemical properties of mango during storage environments

Various changes regarding to blochemical properties of mango fruits as influenced by GA3 are presented and discussed under the following sub-headings.

A

-

£



Fig. 3.1 Effect of different doses of Giberellic acid (GA3) on physiological weight of mango at different days after storage. Vertical bars represent LSD at 0.05 level



Fig. 3.3 Dry matter content of mango pulp as influenced by different doses of Gibberellic acid (GA 3) at different days after storage. Vertical bars represent LSD at 0.05 level

Graphs represe	ent
$G_0 = Control$	
$G_1 = 100 \text{ ppm}$	of GA3 solution

1





Fig. 3.2 Effect of different doses of Gibberelic acid (GA3) on moisture content of mango pulp at different days after storage. Vertical bars represent LSD at 0.05 level



Fig. 3.4 Ash content of mango pulp as influenced by different doses of Gibberellic acid (GA 3) at different days after storage. Vertical bars represent LSD at 0.05 level

 $G_2 = 200 \text{ ppm of GA3 solution}$ $G_3 = 400 \text{ ppm of GA3 solution}$

Treatments	Physiol	ogical weight	loss (%) at o	different days		Moisture	e content (%)	at different day	s
Variety (V)	3	6	9	12	Initial	3	6	9	12
V1	5.15 b	6.65 b	8.13 b	9.72 b	82.43 b	83.53 b	84.53 b	85.36 b	85.80 b
V ₂	6.05 a	7.78 a	9.02 a	10.62 a	84.36 a	85.62 a	86.68 a	87.40 a	87.85 a
Level of significance	***	***	***	***	***	***	***	***	***

1

Table 3.2 Changes of physiological weight loss and moisture content in mango varieties during storage at ambient condition

 \mathbf{A}

Table 3.3 Combined effects of varieties and different doses of Gibberellic acid solution on physiological weight loss and moisture content of postharvest mango during storage at ambient condition

Treatments combination	Physiolog	gical weight lo	oss (%) at diff	ferent days	Moisture content (%) at different days					
Varieties × Treatments	3	6	9	12	Initial	3	6	9	12	
V ₁ G ₀	6.70 b	8.20 b	9.65 b	11.30 b	82.50 b	84.80 c	86.30 b	85.93 d	85.10 f	
V ₁ G ₁	5.40 d	6.90 c	8.35 d	9.98 d	82.40 b	83.40 d	84.40 d	86.40 c	86.20 e	
V ₁ G ₂	4.35 f	5.85 d	7.40 f	8.96 f	82.63 b	83.10 de	83.90 e	84.80 e	86.80 d	
V_1G_3	4.15 g	5.65 d	7.10 g	8.65 g	82.20 b	82.80 e	83.50 f	84.30 f	85.10 f	
V ₂ G ₀	7.40 a	8.96 a	10.35 a	11.97 a	84.60 a	87.23 a	88.40 a	88.10 a	87.30 c	
V ₂ G ₁	6.30 c	7.90 b	9.30 c	10.90 c	84.40 a	85.40 b	86.40 b	88.40 a	88.20 b	
V ₂ G ₂	5.50 d	7.05 c	8.40 d	10.07 d	84.25 a	85.05 bc	86.40 b	86.80 b	88.80 a	
V ₂ G ₃	5.00 e	7.22 c	8.03 e	9.52 e	84.20 a	84.80 c	85.50 c	86.30 c	87.10 cd	
Level of significance	***	*	*	**	NS	NS	NS	NS	NS	
CV%	1.24	2.89	0.85	0.71	0.41	0.28	0.25	0.23	0.23	

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

4

Table represents $V_1 = Langra$ $V_2 = Khirshapat$ $G_0 = Control$ $G_1 = 100 \text{ ppm of GA3 solution}$

Y

 $G_2 = 200 \text{ ppm of GA3 solution}$ * $G_3 = 400 \text{ ppm of GA3 solution}$ * indicates at 5% level ** indicate at 1 % level

*** indicate at 0.1% level NS means non -significant

*

Treatments		Dry matter content (%) at different days						Ash content (%) at different days				
Variety (V)	Initial	3	6	9	12	Initial	3	6	9	12		
V ₁	17.65 a	16.48 a	15.48 a	14.65 a	14.20 a	1.01 a	0.96	0.90 a	0.85 a	0.81 a		
V ₂	15.64 b	14.46 b	13.38 b	12.61 b	12.15 b	0.90 b	0.86	0.79 b	0.74 b	0.70 b		
Level of significance	***	***	***	***	***	*	NS	*	*	*		

A

64.5

14

N

Table 3.4 Dry matter and ash content of mango pulp changes in varieties during storage at ambient condition

Table 3.5 Combined effects of varieties and different doses of Gibberellic acid solution on dry matter and ash content of postharvest mango during storage at ambient condition

Treatments combination		Ory matter o	ontent (%)	at different of	lays	Ash content (%) at different days				
Varieties × Treatments	Initial	nitial 3 6 9 12					3	6	9	12
V ₁ G ₀	17.50 a	15.20 c	13.70 e	14.10 c	14.90 a	1.00	0.91	0.81 ab	0.81 ab	0.80
V ₁ G ₁	17.60 a	16.60 b	15.60 c	13.60 d	13.80 b	1.01	0.95	0.90 a	0.80 ab	0.77
V ₁ G ₂	17.70 a	16.90 ab	16.10 b	15.20 b	13.20 c	1.01	0.97	0.93 a	0.88 ab	0.78
V_1G_3	17.80 a	17.20 a	16.50 a	15.70 a	14.90 a	1.02	0.99	0.95 a	0.91 a	0.87
V ₂ G ₀	15.40 c	13.10 e	11.60 f	11.93 f	12.70 d	0.88	0.84	0.70 b	0.67 b	0.70
V ₂ G ₁	15.60 bc	14.60 d	13.60 e	11.60 f	11.80 e	0.90	0.85	0.80 ab	0.70 ab	0.67
V ₂ G ₂	15.75 bc	14.95 c	13.80 e	13.20 e	11.20 f	0.91	0.87	0.81 ab	0.78 ab	0.68
V ₂ G ₃	15.80 b	15.20 c	14.50 d	13.70 d	12.90 cd	0.92	0.88	0.85 ab	0.80 ab	0.76
Level of significance	NS	NS	NS	NS	NS	NS	NS	NS	NS	NS
CV%	1.18	1.27	1.40	1.45	1.53	10.57	13.86	11.98	13.73	14.07

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents $V_1 = Langra$

V₂ = Khirshapat $G_0 = Control$

 $G_2 = 200 \text{ ppm of GA3 solution}$ *** indicate at 0.1% level $G_3 = 400 \text{ ppm of GA3 solution}$ NS means non -significant * indicates at 5% level ** indicate at 1 % level

 $G_1 = 100 \text{ ppm of GA3 solution}$

×

*

*

Y

x

9.3.1 Vitamin C content

Differences between two varieties in relation to vitamin C content of mango pulp were highly significant at different days after storage (Appendix 3.3). At various days of storage, it expounded that Langra showed better performance in synthesis of vitamin C as compared to Khirshapat (Table 3.6). It also narrated that quantity of vitamin C diminished with the advancement of storage period. At the initial day, Langra accumulated the highest (131.65 mg/100 g) quantity of vitamin C whereas; Khirshapat gave the lowest (45.99 mg/100 g). At 12th day, Langra again, was notified the highest (17.18 mg/100 g) producer of vitamin C and Khirshapat was the lowest (9.25 mg/100 g) producer (Table 3.6). These results annotated that vitamin C content successively came down with the augmentation of storage duration in both the varieties. It might be probably due to rising of ethylene synthesis resulting in oxidation of ascorbic acid. The results of the present study are in agreement with the findings of Azad (2001), Shyamalamma (1995), Gofur *et al.* (1994) and Absar *et al.* (1993).

Variation among the means of different doses of GA3 solution in terms of vitamin C content exhibited highly statistical significant at different days after storage (Appendix 3.3). At initial day, green mangoes treated with G₃ dose accumulated the highest (89.43 mg/100 g) amount of vitamin C whereas; the lowest (87.78mg/100 g) was accumulated in the untreated mango but, the lowest value was statistically at par with G₁ treatment. After initial level, it diminished in a continuous stream with the passing of storage time (Fig. 3.5). It also explored that diminishing rate was faster in control, but it was slower in G₃ treated fruit. At 12th day, vitamin C content ranged between 8.70 to 17.70 mg per 100 g of pulp obtained from control (T₀) and G₃ treated fruit, respectively. The reduction of vitamin C content in both treated and untreated mangoes at different storage period might be possible due to oxidation of ascorbic acid and G₃ dose was possibly causing delay ripening which resulted in lower oxidation of vitamin C. These results are in supported by the findings of Hossain (1999). Jain and Mukherjee (2001) also reported the similar results.

The combined effect of varieties and different doses of GA3 solution showed significant variation in respect of vitamin C content at different days after storage

X

10

3

X

except initial day (Appendix 3.3 and Table 3.7). At various days during storage, it was notified that vitamin C content fell away with the rising of storage period. The quantity of vitamin C ranged between 4.90 to 21.50 mg per 100 g of fresh mango pulp at 12^{th} day in the treatment combination of V₁G₃ and V₂G₀, respectively (Table 3.7).

9.3.2 Titratable acidity

Variation in between varieties means in terms of titratable acidity was observed to be highly significant at different days after storage (Appendix 3.4). At various days during storage, Langra showed higher titratable acid content as compared with Khirshapat. Titratable acidity fell off with the passing of storage period. The diminishing trend was very fast from initial to 3rd day and thereafter, its trend was comparatively slower (Table 3.8). At initial day, the highest (3.88%) was derived from Langra whereas the lowest (2.60%) was derived from Khirshapat. At 12th day, the highest (0.35%) was reported from langra whereas; Khirshapat produced the lowest amount (0.29%). The abating trend of titratable acidity at storage period was reported by Upadhyay and Tripathi (1985), Leon and Lima (1968) and Medlicott *et al.* (1986). According to them, acidity was reduced during storage growth on attainment of maturity and ripening. The results of the present investigation might be possibly due to genetical dissimilarities between two varieties.

Different doses of GA3 solution used in this investigation in terms of titratable acidity exhibited significant variation among the means at various days after storage (Appendix 3.4). At various days of storage, titratable acid content came down very sharply from initial to 3 days and then, it came down steadily (Fig. 3.6). In all the storage period, higher titratable acidity (3.29, 1.30, 1.00, 0.79 and 0.49%) was noted at G₃ treatment from initial to 12^{th} days followed by 3.15, 0.75, 0.45, 0.26 and 0.14% from untreated mangoes, respectively (Fig. 3.6). These results are in conformity with the findings of Gain and Mukherjee (2001) and Khumlert (1992) also found the similar results.

The combined effect of varieties and different doses of GA3 solution in relation to titratable acidity of mango pulp exhibited significant variation at different days after storage except 9th and 12th days (Appendix 3.4 and Table 3.9). At different days of storage, there showed a diminishing trend of titratable acid content

with the expansion of storage period (Table 3.9). At 6^{th} day, the highest (1.10%) quantity was obtained from the treatment combination of V₁G₃ which was statistical similar with the combination of V₁G₂ and the lowest acid concentration (0.35%) obtained from the treatment combination of V₂G₀ (Table 3.9). These occurrences might be probably due to the reduction of acid oxidation at V₂G₃ combination and, to have genetical variation in between varieties.

9.3.3 Pulp pH

r

X

The analysis of variance in between the varieties showed significant in respect of pulp pH of mango at different days after storage except initial and 12th day (Appendix 3.4 and Table 3.8). At various days of storage, there noticed an increasing trend of pulp pH with the rising of storage period. In each storage period, pulp pH exposited more in Khirshapat compared with Langra. Higher pulp pH (4.55) was notified from Khirshapat at 3rd day whereas; lower (4.30) was identified from Langra. At 12th day, the highest pH (6.80) was identified with Khirshapat and the lowest value (6.70) was identified with Langra. The growing up trend of pulp pH was also observed by Yuniarti (1980), Kumar *et al.* (1993) and Shahjahan *et al.* (1994). This phenomenon might be possible due to oxidation of acid during storage resulting in higher pH and also might have genetical dissimilarities between varieties.

Different doses of GA3 solution imposed to this trial demonstrated significant variation in pulp pH at different days after storage (Appendix 3.4). The rising trend of pulp pH was found from different treated and untreated fruits at various days of storage (Fig. 3.7). Pulp pH was higher in control at all stages of storage period followed by the fruits treated with G_1 , G_2 and G_3 treatments. The pH of mango pulp was the highest (7.05) in control whereas the fruits treated with G_3 treatment gave the lowest (6.45) value at 12th day (Fig. 3.7). The results of the present investigation of GA3 solution at G_3 treatment retarded the loss of acid oxidation resulting in lower pH value. These results are in partially supported by the report of Jain and Mukherjee (2001).

The combined effect of varieties and different doses of GA3 solution implied to this study in pulp pH were noticed to be non significant at different days after storage (Appendix 3.4). There was the indication of an enhancing trend of pulp pH from various treatment combinations at different days of storage (Table 3.9). At 12th

X

day, the highest (7.10) pH value was obtained from the treatment combination of V_2G_0 and the lowest (6.4) was obtained from the treatment of combination of V_1G_3 .

9.3.4 Total soluble solid (Brix %) content

Statistically highly significant variation was found in TSS content between two varieties at different days after storage (Appendix 3.5). The results showed that TSS content of mango pulp enlarged continuously with the rising of storage period. The increasing trend was faster from initial to 6th day thereafter; it increased slower. From initial to 6th day, Khirshapat was performed better in TSS accumulation than Langra. But, after 6th day, Langra performed better than Khirshapat up to 12th day. At 9th day, the highest (16.28%) TSS quantity was noticed from Khirshapat and the lowest (16.03%) was noticed from Langra (Table 3.10). Increased in the percentage of TSS during storage in mango pulp was reported by Singh (1968). Absar *et al.* (1993) reported that TSS was increased with maturity of mango fruit, but, they found highest TSS in Langra. Mollah and Siddique (1973) reported that TSS value varied from cultivar to cultivar. These might be possible due to genetically differences between varieties.

Different doses of GA3 solution subjected to the postharvest mangoes in this study were observed to be significant variation in respect of TSS content at different days after storage (Appendix 3.5). At different days of storage, it narrated that TSS accumulation augmented with the expansion of storage duration. The results also illustrated that TSS content was sharply grown up from untreated mangoes from initial to 6th day and then, it fell off significantly (Fig. 3.8). The other treatment such as G1 also increasingly provided TSS from initial to 9th day and thereafter, it decreased sharply. Mango fruits treated with G2 also produced more or less similar increasing trend from initial to 12th day. But, the fruits treated with G₃ dose provided very steady rate in TSS accumulation at various days (Fig. 3.8). The highest (19.25, 19.25 and 19.30%) accumulation of TSS was derived from $G_0,\,G_1$ and G_2 treatment at 6, 9 and 12th days whereas, the lowest (10.20, 13.00 and 16.00%) was derived from G_3 treatment, respectively. The results of the present study are strongly supported by the findings of Jain and Mukherjee (2001). These happened possibly due to ripening condition resulting in maximizing TSS gathering in control and 400 ppm of GA3 solution retarded in ethylene synthesis that caused delay ripening and

×

Y

×





Fig. 3.5 Effect of different doses of GA3 on vitamin C content of mango pulp at different days after storage. Vertical bars represent LSD at 0.05 level



Fig. 3.7 Pulp pH of mango as influenced by different doses of GA3 at different days after storage. Vertical bars represent LSD at 0.05 level



 $G_2 = 200 \text{ ppm of GA3 solution}$ $G_3 = 400 \text{ ppm of GA3 solution}$





Fig. 3.8 Effect of different doses of GA3 on total soluble solid content of mango pulp at different days after storage. Vertical bars represent LSD at 0.05 level

Treatments		Vitami	n C content (mg/100	g) at different days	
Variety (V)	Initial	3	6	9	12
V ₁	131.65 a	111.48 a	64.40 a	35.23 a	17.18 a
V ₂	45.99 b	34.38 b	19.49 b	11.90 b	9.25 b
evel of significance	***	***	***	***	***

1

4

Table 3.6 Changes of vitamin C content of mango pulp in varieties during storage at ambient condition

+

Table 3.7 Combined effects of varieties and different doses of Gibberellic acid solution on vitamin C content of postharvest mango pulp during storage at ambient condition

Treatments combination Varieties × Treatments	Vitamin C content (mg/100 g) at different days				
	Initial	3	6	9	12
V ₁ G ₀	130.80 a	87.51 c	40.30 d	20.50 d	12.50 e
V ₁ G ₁	131.50 a	115.50 b	67.53 c	35.40 c	15.20 c
V ₁ G ₂	131.90 a	120.60 a	72.20 b	40.20 b	19.50 b
V ₁ G ₃	132.50 a	122.30 a	77.55 a	44.80 a	21.50 a
V ₂ G ₀	44.80 c	25.60 g	13.66 h	8.30 h	4.90 h
V ₂ G ₁	45.50 bc	34.30 f	18.30 g	10.80 g	7.80 g
V ₂ G ₂	47.30 b	37.40 e	21.40 f	13.20 f	10.40 f
V ₂ G ₃	46.35 bc	40.20 d	24.60 e	15.30 e	13.90 d
Level of significance	NS	***	***	***	***
CV%	1.22	1.44	2.61	3.82	1.28

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents $G_2 = 200 \text{ ppm of GA3 solution}$ *** indicate at 0.1% level $V_1 = Langra$ $G_2 = 200 \text{ ppm of GA3 solution}$ *** indicate at 0.1% level $V_2 = Khirshapat$ $G_3 = 400 \text{ ppm of GA3 solution}$ *** indicate at 0.1% level $G_0 = Control$ * indicates at 5% levelNS means non significant $G_1 = 100 \text{ ppm of GA3 solution}$ ** indicate at 1 % level

A.
Treatments	Titratable acidity (%) at different days						Pulp pH at different days						
Variety (V)	Initial	3	6	9	12	Initial	3	6	9	12			
V1	3.88 a	1.20 a	0.86 a	0.61 a	0.35 a	3.50	4.30 b	5.30 b	5.93 b	6.70			
V ₂	2.60 b	1.03 b	0.71 b	0.49 b	0.29 b	3.58	4.55 a	5.55 a	6.18 a	6.80			
Level of significance	***	***	***	***	***	NS	***	***	***	NS			

~

4

Table 3.8 Changes of titratable acidity and pulp pH of postharvest mango pulp between varieties during at ambient condition

X

Table 3.9 Combined effects of varieties and different doses of Gibberellic acid solution on titratable acidity and pulp pH of postharvest mango during storage at ambient condition

Treatments combination		Titratable a	cidity (%) a	t different o	lays	Pulp pH at different days					
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12	
V ₁ G ₀	3.80 b	0.95d	0.55 e	0.30 e	0.15 g	3.60 ab	4.60 ab	5.60 ab	6.90 a	7.00 ab	
V ₁ G ₁	3.90 ab	1.20 bc	0.85 c	0.58 c	0.33 e	3.50 ab	4.30 de	5.30 de	5.80 cd	6.80 b-d	
V ₁ G ₂	3.82 b	1.28 ab	0.95 b	0.72 b	0.40 c	3.50 ab	4.20 ef	5.20 ef	5.60 e	6.60 d-f	
V_1G_3	3.98 a	1.35 a	1.10 a	0.85 a	0.52 a	3.40 b	4.10 f	5.10 f	5.40 f	6.40 f	
V ₂ G ₀	2.50 d	0.55 e	0.35 f	0.22 f	0.12 h	3.70 a	4.70 a	5.70 a	7.00 a	7.10 a	
V ₂ G ₁	2.60 d	1.10 c	0.75 d	0.42 d	0.25 f	3.60 ab	4.60 ab	5.60 ab	6.10 b	6.90 a-c	
V ₂ G ₂	2.70 c	1.20 bc	0.85 c	0.60 c	0.35 d	3.50 ab	4.50 bc	5.50 bc	5.90 c	6.70 с-е	
V_2G_3	2.60 cd	1.25 ab	0.90 bc	0.73 b	0.45 b	3.50 b	4.40 cd	5.40 cd	5.70 de	6.50 ef	
Level of significance	**	***	*	NS	NS	NS	NS	NS	NS	NS	
CV%	1.65	4.82	4.07	5.80	5.26	2.95	2.40	1.96	1.73	1.85	

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

4.

Table represents $V_1 = Langra$ $V_2 = Khirshapat$ Go = Control

4

 $G_2 = 200 \text{ ppm of GA3 solution}$ $G_3 = 400 \text{ ppm of GA3 solution}$ * indicates at 5% level $G_i = 100 \text{ ppm of GA3 solution}$

** indicate at 1 % level

*** indicate at 0.1% level NS means non significant

178

×

X.

-

X

ultimately in lower TSS accumulation. It also explained that TSS gathering is strongly related to ripening and it caused decrease owing to decaying.

The combined effect of varieties and imposed different doses of GA3 solution on TSS content were found to be significant at different days after storage except initial and 3^{rd} day (Appendix 3.5). The result was expounded a rising behavior of TSS content at different days after storage (Table 2.11). The highest accumulation (19.50, 19.30 and 19.50) was derived from the treatment combination of V₁G₀, V₁G₁ and V₁G₂ at 6, 9 and 12th days, whereas; the lowest value (10.30, 12.90 and 15.90) was identified from the treatment combination of V₂G₃, respectively.

9.3.5 Total sugar content

Highly significant variation was noticed between both the varieties means in terms of total sugar content of mango pulp at different days after storage (Appendix 3.5). The results interpreted that TSC accumulated successively with the rising of storage duration. The accumulating trend was more or less sharp from initial to 9th day in both the varieties, thereafter; it grew up slightly slower. At all days of storage Khirshapat contributed more accumulation of of TSC than Langra. At initial day, Khirshapat gave the highest (5.80%) whereas; Langra produced the lowest (5.34%). At 12th day, Khirshapat produced the highest quantity (18.80%) and the lowest (18.35%) was obtained from Langra. Upadhyay and Tripathi (1985) reported that total sugar content was augmented gradually, when stored for 6 days at room temperature. Sugar content increased during ripening (Srivastava, 1967). These results are in conformity with the findings of Shahjahan *et al.* (1994). Tsuda *et al.* (1999) also found the similar results. The increase in TSC might be due to conversion of complex starch or carbohydrate into simple compound like sucrose, Fructose, galactose etc.

Different doses of GA3 solution implied to the investigation on total sugar content of mango pulp exhibited significant variation at different days after storage (Appendix 3.5). At different days of storage, the results noticed that TSC augmented markedly with the rising of storage period (Fig. 3.9). The developing trend was very fast in untreated mango followed by other treatments *viz.*, G_1 , G_2 and G_3 , respectively. The highest quantity of TSC (20.95% and 20.60%) was recorded in control and G_1 treated mangoes at 9 and 12th days, whereas; the lowest (11.64%

Ja.

-

and 14.60%) was recorded at G_3 treatment. The findings of the present investigation are inconformity with the reports of Jain and Mukherjee (2001) and Singh *et al.* (1995). The enhancing trend of total sugar at untreated mangoes might be possible due to breaking down of complex carbohydrate into simple compound but, G_3 treatment made delay ripening at storage period resulting in lower conversion of complex compound into simple molecules.

The combined effect of varieties and used different doses of GA3 solution in this study in relation to total sugar content of mango pulp showed non significant variation at different days after storage (Appendix 3.5). These results noted that total sugar content accumulated successively with the rising of storage period (Table 3.11). At 12th day the maximum (20.80%) quantity of TSC was achieved from the treatment combination of V₂G₁ whereas, the minimum (14.40%) was achieved from the treatment combination of V₁G₃ (Table 3.11).

9.3.6 Reducing sugar content

Analysis of variance demonstrated significant variation on reducing sugar content of mango pulp at different days after storage except at 6th day (Appendix 3.6). There was noticed an enhancing trend of reducing sugar with the expanding of storage period (Table 3.12). It also stated that Khirshapat was better in achieving of reducing sugar than Langra at different days of storage. The highest (5.18%) quantity of this sugar was notified from Khirshapat whereas; the lowest (4.86%) was notified from Langra at 12th days of storage (Table 3.12). These results are in agreement with the report of Upadhyay and Tripathi (1985). Casttrillo et al. (1992) elucidated that reducing sugar was increased during storage period. Khirshapat producing comparatively more reducing sugar might be possibly due to genetical variation in both the varieties.

Different doses of GA3 solution applied to this study were found to be significant in respect of reducing sugar content of mango pulp at different storage period except initial day (Appendix 3.6). The results narrated that reducing sugar of mango pulp was grown up continuously at different days after storage. It also revealed that untreated mangoes were performed better in accumulating of reducing sugar as compared to the other treatments. Control treatment was recorded as more successive producer of reducing sugar up to 9th day and then, it fell off due to

1

Y

1

4

starting spoilage. At 12^{th} day, the maximum (6.18%) amount of reducing sugar was recorded at G₁ and the lowest (3.78%) was recorded at G₃ treatment (Table 3.12). The results of the present investigation are in conformity with the findings of Jain and Mukherjee (2001). Lower changing trend of reducing sugar content treated with G₃ treatment might be possibly due to delay ripening which resulted in lower conversion of carbohydrates into simple's molecules.

The combined effect of varieties and used different doses of GA3 solution of mango pulp exhibited non significant variation in terms of reducing sugar content of mango pulp at different days after storage except 6th day (Appendix 3.6). The results elucidated that reducing sugar content augmented continuously at three days interval up to 9th day thereafter, it abated from the treatment combination of V₂G₀ (Table 3.13). At 12th day, the highest (6.35%) quantity was obtained from the treatment from V₁G₃.

9.3.7 Non reducing sugar content

The variation between the varieties means exhibited highly significant in respect of non-reducing sugar content at different days after storage (Appendix 3.6). There demonstrated an enlarging trend of non reducing sugar content at different days of storage. At all days, it was noticed that Khirshapat was much better than Langra in achieving of non reducing sugar content (Table 3.12). At 12th day, higher (13.63%) amount of non reducing sugar was recorded from Khirshapat and the lowest (13.46%) amount was recorded from Langra. These results are in conformity with the report of Ali and Mazhar (1960). They reported that non reducing sugar content of ripe fruits was 11.20%. The result obtained from the investigation might be possible due to varietals dissimilarities.

Different doses of GA3 solution subjected to this trial produced the significant variation in respect of non reducing sugar content of mango pulp at different days (Appendix 3.6). The results stated that non reducing sugar content of mango pulp grew up markedly at various days. It also mentioned that untreated fruits were notified better in accumulation of more quantity of non reducing sugar followed by other treatments. This growing up trend was continued up to 9th day and then, it fell off owing to becoming hackneyed. Lower rising trend was noted from the fruit treated with G₃ treatment. The highest result (15.20%) was obtained from control

and the lowest value (10.82%) was obtained from G_3 treatment (Table 3.12). These events might be probably due to G_3 treatment resisted ethylene synthesis of mango pulp resulted in delay ripening and little amount of non reducing sugar deposition. These results are in partially agreement with the reports of Khumlert (1992).

The combined effect of varieties and used different doses of GA3 solution showed non significant variation in terms of non reducing sugar content of mango pulp at different days after storage (Appendix 3.6). There was exposition that a hastening trend of non reducing sugar was notified from different treatment combinations at various days of storage (Table 3.13). At 9th day, the highest (14.80%) quantity of non reducing sugar was derived from the treatment combination of V₂G₀ whereas; the lowest (8.20%) was derived from the treatment combination of V₁G₃.

9.3.8 Crude fibre content

Sil.

×

-de-

4

The analysis of variance of mango varieties imposed in this investigation was found to be significant in respect of crude fibre content of mango pulp at different days after storage (Appendix 3.7). The results noticed a growing up trend of crude fibre from both the varieties at different days of storage. It also indicated that Langra was notified better in performing of crude fibre accumulation. At initial day, Langra gave the highest (1.29%) quantity of crude fiber as compared to Khirshapat (1.18%). The amount of crude fiber diminished successively with the increase of storage period from both the varieties. At 12th day, the highest (0.56%) quantity of crude fiber was recorded from Langra whereas; the lowest (0.41%) was reported from Khirshapat (Table 3.14). These results are in partially conformity with the findings of Peter *et al.* (2007). The fall down of crude fibre content with the rising of storage period might be possible due to metabolic activities resulting in hydrolysis of cellulose and lignin into simple molecules.

Various doses of GA3 solution implied to this trial had significant effect on crude fiber content of mango pulp at different days after storage except initial day (Appendix 3.7). The results illustrated that crude fiber content came down successively with the hastening of storage period (Fig. 3.10). At all days of storage, it was notified that crude fiber content was comparatively higher at the fruit treated with G_3 treatment followed by the other treatments. At 3rd day, the maximum

X

-

do

4

(1.19%) amount of crude fiber was reported from G_3 treatment which was statistically at par with G_1 and G_2 treatments and the lowest (0.91%) was reported from G_0 treatment. At 12th day, the maximum (0.62%) was obtained from G_3 whereas; the lowest (0.42%) was obtained from G_1 treatment which was also statistically at par with G_0 and G_2 (Table 3.14). The diminishing trend of crude fiber influenced by G_3 treatment might be possible due to interruption of ripening resulted in lower declining trend of crude fiber content.

The combined effect of varieties and different doses of GA3 solution were found to be non significant variation in terms of crude fiber content of mango pulp at different days after storage (Appendix 3.7). The results interpreted that crude fibre fell off gradually with the hastening of storage period (Table 3.15). It also elucidated that the treatment combination of V₁G₃ was better in receiving of crude fiber content at all storages duration. At 6th day, the highest (1.13%) quantity of crude fiber was noted from the treatment combination of V₁G₃ which was statistically at par with V₁G₂ and V₂G₃ and the lowest (0.41%) was noted from the treatment combination of V₂G₀.

9.3.9 Total lipid content

The analysis of variance of varieties was highly significant in terms of lipid content in mango pulp at different days after storage (Appendix 3.7). It noticed an enhancing trend of lipid content in mango pulp with the hastening of storage duration. At all days of storage, Langra was observed higher amount of lipid producer comparing to Khirshapat. At 12th day, higher (0.70%) amount of lipid was noticed from Langra and lower (0.65%) was noticed from Khirshapat (Table 3.14). These phenomenon might be possible due to the genetically dissimilarities between two varieties.

Different doses of GA3 solution subjected to this study in terms of lipid content were found to be highly significant at different days after storage (Appendix 3.7). It was exposited that a growing up trend of lipid content was found with the rising of storage period at various days of storage (Fig. 3.11). It also illustrated that control treatment accumulated comparatively higher amount of lipid followed by G₁, G₂ and G₃ treatments from initial to 9th days of storage; and then, it fell off due to starting of decay. At 9th day, the highest (0.80%) amount of lipid content was

obtained from control whereas; the lowest (0.52%) was obtained from the fruit treated with G_3 treatment. At 12^{th} day, G_1 treated fruit gave the highest (0.77%) amount of lipid and the lowest (0.62%) was noted from the fruit treated with G3 treatment. These occurrences might be possible due to G_3 treatment caused delay ripening which resulted in lower production of lipid content

The interaction effects of varieties and imposed various doses of GA3 solution in this trial demonstrated significant variation in respect of lipid content of mango pulp at different days after storage except initial and 3rd day (Appendix 3.7). The results narrated that an increasing trend of lipid content in mango pulp was visible with the hastening of storage duration (Table 3.15). It also explained that the treatment combination of V₁G₀ produced more quantity of lipid at initial to 9th day and then, it came down due to starting in deterioration of fruits. At this time, lower quantity was reported from the treatment combination of V₂G₃. At 9th day, the highest (0.84%) quantity of lipid was reported from the treatment combination of V₁G₀ whereas; the lowest (0.51%) was reported from the treatment combination of V₂G₀, which was statistically at par with V₂G₃. The decrease behavior at V₂G₀ might be perhaps due to spoilage situation of mango fruits and lower in lipid accumulation at V₂G₃ might be due to delay ripening.

9.3.10 Water soluble protein content

1

to

*

The analysis of variance of mango varieties showed significant variation in respect of WSPC in mango pulp at different days after storage except 12th day (Appendix 3.8). There showed an increase trend of WSPC with the advancement of storage duration. It also revealed that Langra was better in WSPC synthesis as compared to Khirshapat at all stages of storage. At 9th day, the highest (1.00%) synthesis of WSPC was noticed from Langra whereas; the lowest (0.88%) was noticed from Khirshapat (Table 3.16). These events were possible due to some of seed protein of mango might have to be disseminated to pulp portion during ripening and complex metabolic activities.

Different doses of GA3 solution subjected to this investigation in terms of WSPC exhibited significant variation at different days after storage except at initial and 12th day (Appendix 3.8). Various results derived from the biochemical analysis was notified an augmenting trend of WSPC in mango pulp with the expansion of

100

1

de.

×

*

storage period (Fig. 3.12). It also denoted that WSPC was achieved more in untreated fruit followed by the fruit treated with G_1 , G_2 and G_3 , treatments, respectively. The growing up trend of WSPC in control was very sharp from initial to 6^{th} day thereafter; it declined due to deterioration of fruit condition. At the same time, the increasing trend of WSPC from the fruit treated with G_3 treatment was very slow due to delay ripening. At 9th day, the highest (1.21%) accumulation of WSPC was obtained from control which was statistically identical with G_1 treatment and the lowest (0.69%) was obtained from G_3 treatment which was also statistically at par with G_2 treatment, respectively. It explained that extension of WSPC synthesis was strongly depended upon the fruit ripening.

The combined effect of varieties and applied different doses of GA3 solution in this experiment showed non significant variation in respect of WSPC at different days after storage (Appendix 3.8). The results reported from the studies elucidated that WSPC grew up gradually from initial to 9th day with the treatment combination of V₁G₀ then, it abated due to putrefied condition of fruits whereas, the lower increasing trend was notified from the treatment combination of V₂G₃. At 9th day, the highest (1.25%) quantity of WSPC was derived from the treatment combination of V₁G₀ whereas; the lowest (0.65%) was derived from the treatment combination of V₂G₃, which was also statistically at par with V₁G₁ and V₂G₀, which was also statistically at par with V₂G₂ and V₁G₃ (Table 3.17).

9.4 Changes in mineral contents of mango during storage environments

Different changes regarding minerals of mango fruits are presented and interpreted in the following sub-headings.

9.4.1 Phosphorus content

Variation between varieties means in relation to P contents demonstrated highly significant variation at different days after storage (Appendix 3.7). It notified very slow rising trend of P contents in both the varieties with the expansion of storage period (Table 3.16). It also denoted that Langra provided more quantity of Phosphorus as compared to Khirshapat. The growing up trend of P in Langra was more or less similar up to 9th day thereafter, its trend abated due to spoilage of fruits. At 12th day, higher (23.34 mg/100 g) quantity was noticed from Langra and lower (20.69 mg/100 g) was noticed from Khirshapat. Better performance in rising

×

Se.

1

X

1

A

Results and Discussion





Fig. 3.9 Effect of different doses of GA3 on total sugar content of mango pulp at different days after storage. Vertical bars represent LSD at 0.05 level







Fig. 3.11 Lipid content of mango pulp as influenced by different doses of GA3 at different days after storage. Vertical bars represent LSD at 0.05 level

 $\begin{array}{l} \mbox{Graphs represent} \\ \mbox{G}_0 = \mbox{Control} \\ \mbox{G}_1 = 100 \mbox{ ppm} \mbox{ of GA3 solution} \end{array}$

Fig. 3.12 Effect of GA3 on water soluble protein content of mango pulp at different days after storage. Vertical bars represent LSD at 0.05 level

 $G_2 = 200 \text{ ppm of GA3 solution}$ $G_3 = 400 \text{ ppm of GA3 solution}$

Water soluble protein content (%)

X

T.

1

-de-

×.

A

trend of P content in Langra might be probably due to genetical dissimilarities. It also elucidated that the expansion of P content in mango was intimately associated with ripening during storage. These results are in partially supported by the findings of Nadkarni (1963) when he worked with 16 cultivars and found the range between 10-30 mg/100 g among the cultivars.

Different doses of GA3 solution imposed to this trial in relation to P content showed highly significant at different days after storage (Appendix 3.8). The results stated that P content grew up steadily at various days. It also revealed that P content in control expanded in slow motion from initial to 6th day thereafter; it augmented in further slow motion than its previous trend. After 9th day, it declined possibly due to purified situation of fruit (Fig. 3.13). At 12th day, the maximum (23.04 mg/100 g) quantity of P was noted from the fruit treated with G₁ and the minimum value (21.03 mg/100 g) was noted from the fruit treated with G₃ treatment. Lower quantity from the fruit treated with G₃ treatment might be probably due to delay ripening which resulted in lower accumulation of P content. The results of the present research are partially supported by the findings of Peter *et al.* (2007).

The combined effect of varieties and used various doses of GA3 solution exhibited significant variation on P content of mango pulp at different days after storage except at initial and 3^{rd} day (Appendix 3.8). It observed a growing up trend of P content with the expansion of storage period (Table 3.17). It also revealed that P content accumulated gradually from initial to 9^{th} day from the treatment combination of V₁G₀ and then, it decreased due to bad fruit situation. In this time, it was found lower changing trend from the fruit treated with the treatment combination of V₂G₃. At 12th day, the highest (24.44 mg/100 g) quantity was notified from the treatment combination of V₁G₁ whereas; the lowest (19.65 mg/100 g) was notified from the treatment combination of V₂G₃. It also revealed that Langra along with G₃ treatment was recommended as the best treatment combination in keeping good preservation quality followed by the other treatment combinations. This treatment combination strictly impeded the ripening of mango fruits during storage.

Treatments		TSS conte	ent (%)at dif	ferent days		Total sugar content (%) at different days						
Variety (V)	Initial	3	6	9	12	Initial	3	6	9	12		
V1	5.30 b	9.28 b	13.78 b	16.28 a	16.35 a	5.34 b	8.08 b	12.80 b	16.08 b	18.35 b		
V ₂	6.35 a	10.33 a	14.28 a	16.03 b	16.05 b	5.80 a	8.58 a	13.28 a	16.58 a	18.80 a		
Level of significance	***	***	***	***	***	***	***	***	***	***		

4)

4

Y

Table 3.10 Changes of total soluble solid and total sugar content of postharvest mango pulp in varieties during storage at ambient condition

1

Table 3.11 Combined effects of varieties and different doses of Gibberellic acid solution on total soluble solid and total sugar content of postharvest mango during storage at ambient condition

Treatments combination		TSS conte	ent (%) at o	different da	ys	Total sugar content (%) at different days					
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12	
V ₁ G ₀	5.50 d	12.50 b	19.50 a	16.50 b	13.50 f	5.50 cd	10.20 b	18.80 b	20.70 b	19.80 d	
V ₁ G ₁	5.30 e	9.30 d	13.30 d	19.30 a	16.30 c	5.40 d	7.90 d	13.20 d	18.40 d	20.40 b	
V ₁ G ₂	5.20 e	8.20 e	12.20 e	16.20 c	19.50 a	5.30 de	7.30 e	10.30 f	13.80 f	18.80 f	
V ₁ G ₃	5.20 e	7.10 f	10.10 g	13.10 e	16.10 d	5.17 e	6.90 f	8.90 h	11.40 h	14.40 h	
V ₂ G ₀	6.50 a	13.50 a	19.00 b	16.00 d	13.00 g	6.00 a	10.70 a	19.20 a	21.20 a	20.20 c	
V ₂ G ₁	6.40 ab	10.40 c	14.50 c	19.20 a	16.20 cd	5.80 b	8.30 c	13.60 c	18.80 c	20.80 a	
V ₂ G ₂	6.30 bc	9.33 d	13.30 d	16.00 d	19.10 b	5.70 bc	7.90 d	10.90 e	14.40 e	19.40 e	
V ₂ G ₃	6.20 c	8.10 e	10.30 f	12.90 f	15.90 e	5.68 bc	7.40 e	9.40 g	11.90 g	14.80 g	
Level of significance	NS	NS	***	**	**	NS	NS	NS	NS	NS	
CV%	1.82	1.17	0.76	0.49	0.50	2.05	1.27	0.79	0.76	0.57	

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

F

$G_2 = 200 \text{ ppm of GA3 solution}$	*** indicate at 0.1% level
$G_3 = 400 \text{ ppm of GA3 solution}$	NS means non significant
 indicates at 5% level 	
** indicate at 1 % level	
	G_2 = 200 ppm of GA3 solution G_3 = 400 ppm of GA3 solution * indicates at 5% level ** indicate at 1 % level

\$

1)

Treatments		Reducing suga	r content (%)	at different	days	Non-reducing sugar content (%) at different days					
Variety (V)	Initial	3	6	9	12	Initial	3	6	9	12	
Vı	1.29 b	1.96 b	3.82	4.81 b	4.86 b	4.03 b	6.11 b	9.29 b	11.26 b	13.46 b	
V ₂	1.66 a	2.35 a	3.83	5.13 a	5.18 a	4.13 a	6.23 a	9.43 a	11.48 a	13.63 a	
Level of significance	***	***	NS	***	***	*	*	**	***	**	
Treatments (G)											
Go	1.55	3.05 a	6.05 a	6.25 a	4.75 c	4.20 a	7.40 a	12.90 a	14.70 a	15.20 a	
G1	1.48	2.03 b	4.14 b	5.98 b	6.18 a	4.10 ab	6.08 b	9.93 b	12.63 b	14.43 b	
G2	1.43	1.83 c	2.88 c	4.38 c	5.38 b	4.05 b	5.78 c	7.73 c	9.73 c	13.73 c	
G3	1.45	1.73 c	2.23 d	3.28 d	3.78 d	3.98 b	5.43 d	6.88 d	8.43 d	10.82 d	
Level of significance	NS	***	***	***	***	*	***	***	***	***	

Table 3.12 Changes of reducing and non reducing sugar content of postharvest mango pulp between varieties and influenced by different doses of Gibberellic acid solution during storage at ambient condition

7

F.

4)

12

Yh

Table 3.13 Combined effects of varieties and different doses of Gibberellic acid solution on reducing and non reducing sugar content of postharvest mango pulp during storage at ambient condition

189

in the

41

Treatments combination	R	Reducing suga	r content (%)	at different	days	Non-reducing sugar content (%) at different days					
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12	
V ₁ G ₀	1.40 b	2.90 b	5.90 a	6.10 b	4.60 f	4.10 ab	7.30 b	12.80 b	14.60 b	15.10 b	
V ₁ G ₁	1.30 bc	1.80 ef	4.63 b	5.80 c	6.00 b	4.05 b	6.10 c	9.90 c	12.60 c	14.40 c	
V ₁ G ₂	1.25 bc	1.65 fg	2.65 de	4.15 e	5.15 d	4.00 b	5.65 e	7.65 d	9.65 d	13.65 d	
V_1G_3	1.20 c	1.50 g	2.10 e	3.20 f	3.70 g	3.97 b	5.40 f	6.80 e	8.20 f	10.70 f	
V ₂ G ₀	1.70 a	3.20 a	6.20 a	6.40 a	4.90 e	4.30 a	7.50 a	13.00 a	14.80 a	15.30 a	
V ₂ G ₁	1.65 a	2.25 c	3.65 c	6.15 b	6.35 a	4.15 ab	6.05 cd	9.95 c	12.65 c	14.45 c	
V ₂ G ₂	1.60 a	2.00 d	3.10 cd	4.60 d	5.60 c	4.10 ab	5.90 d	7.80 d	9.80 d	13.80 d	
V ₂ G ₃	1.69 a	1.95 de	2.35 de	3.35 f	3.85 g	3.98 b	5.45 f	6.95 e	8.65 e	10.95 e	
Level of significance	NS	NS	*	NS	NS	NS	NS	NS	*	NS	
CV%	7.20	4.80	11.14	2.13	2.11	2.63	1.72	1.13	0.93	0.78	

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents $V_1 = Langra$ $V_2 = Khirshapat$ $G_0 = Control$

 $G_3 = 400 \text{ ppm of GA3 solution}$

* indicates at 5% level

 $G_1 = 100 \text{ ppm of GA3 solution}$

 $G_2 = 200 \text{ ppm of GA3 solution}$

*** indicate at 0.1% level NS means non significant

Treatments		Crude fib	re (%) at di	fferent days	5		t different days	erent days		
Variety(V)	Initial	3	6	9	12	Initial	3	6	9	12
V ₁	1.29 a	1.17 a	0.90 a	0.73 a	0.56 a	0.19 a	0.37 a	0.52 a	0.68 a	0.70 a
V ₂	1.18 b	0.98 b	0.71 b	0.54 b	0.41 b	0.18 b	0.35 b	0.49 b	0.63 b	0.65 b
evel of significance	*	***	***	***	***	***	***	***	***	***

Table 3.14 Changes of crude fibre and lipid content of mango pulp between varieties during storage at ambient condition

7

Table 3.15 Combined effects of varieties and different doses of Gibberellic acid solution on crude fibre and lipid content of postharvest mango pulp during storage at ambient condition

Treatments combination		Crude fibr	e (%) at di	fferent days		Lipid content (%) at different days						
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12		
V ₁ G ₀	1.28	1.00 cd	0.60 d	0.55 c	0.52 a	0.22 a	0.45 a	0.74 a	0.84 a	0.69 c		
V ₁ G ₁	1.29	1.16 a-c	0.86 b	0.56 c	0.51 a	0.20 b	0.39 c	0.54 c	0.76 b	0.80 a		
V ₁ G ₂	1.30	1.22 ab	1.02 ab	0.82 ab	0.57 a	0.19 bc	0.33 e	0.43 e	0.58 d	0.68 cd		
V_1G_3	1.30	1.28 a	1.13 a	0.98 a	0.63 a	0.17 cd	0.30 f	0.38 g	0.53 e	0.63 e		
V ₂ G ₀	1.16	0.81 d	0.41 e	0.36 d	0.34 b	0.20 b	0.43 b	0.67 b	0.75 b	0.60 f		
V ₂ G ₁	1.17	0.97 cd	0.67 cd	0.37 d	0.32 b	0.18 c	0.37 d	0.52 d	0.69 c	0.74 b		
V ₂ G ₂	1.18	1.03 bc	0.83 bc	0.63 c	0.38 b	0.16 d	0.31 f	0.41 f	0.56 d	0.66 d		
V ₂ G ₃	1.20	1.10 a-c	0.95 ab	0.80 b	0.60 a	0.16 d	0.28 g	0.36 h	0.51 e	0.61 ef		
Level of significance	NS	NS	NS	NS	NS	NS	NS	**	***	***		
CV%	8.59	9.90	13.03	14.72	14.69	5.53	2.99	2.11	1.64	1.53		

*** indicate at 0.1% level

NS means non significant

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents

 $G_0 = Control$

- $V_1 = Langra$ $V_2 = Khirshapat$
- $G_2 = 200 \text{ ppm of GA3 solution}$ $G_3 = 400 \text{ ppm of GA3 solution}$

* indicates at 5% level

 $G_1 = 100 \text{ ppm of GA3 solution}$

4)

** indicate at 1 % level

¥.

41

×

X

L

R.

A.

X

9.4.2 Potassium content

Imposed varieties in terms of K content exhibited highly significant variation at different days after storage (Appendix 3.9). It noticed a rising trend of K content in both the varieties with the advancement of storage period at different days (Table 3.18). In all the storage period, Langra was identified as higher producer of K content comparing to Khirshapat. It also indicated that the rising trend of K content was stopped at 9th day. At this period, higher (0.29%) amount of K was noted from Langra and lower (0.27%) was noted from Khirshapat (Table 3.18). This event might be possible due to spoilage of mango fruits which resulted in lower metabolic activities. Langra was better producer of K content possibly due to genetical dissimilarities between varieties. At storage period, K content increased might be possible due to transmission of K from stone and peel to pulp of mango.

Different doses of GA3 solution implied to this experiment in terms of K content of mango were noticed significant at different days after storage (Appendix 3.9). The results narrated that K content developed in a continuous stream with the increase of storage period. But, K content from the untreated fruit fell away at 9th day, whereas other treatments such as G_1 , G_2 and G_3 treatment retained their increasing behavior (Fig. 3.14). In this period, it was found a very lower changing up trend of K content in the fruit treated with G_3 treatment. At 9th day, the maximum (0.30%) of K content was manifested from the untreated fruit whereas; the lowest (0.26%) was manifested from the fruit treated with G_3 treatment which was statistically at par with G_2 treatment. Lower quantity of K in the fruits treated with G_3 treatment might be possible due to delay ripening that caused lower transmission of K content. These results are in partially supported by the findings of Peter *et al.* (2007). The results of the present investigation revealed that K content of mango pulp was grown up slightly in the storage period. But, G_3 treatment interrupted its growing up trend during storage period.

The combined effect of varieties and subjected to different doses of GA3 solution in relation to K content of mango pulp exhibited non significant variation at different days after storage (Appendix 3.9). The results indicated that K content of mango pulp from different treatment combination developed gradually with the increasing of storage time (Table 3.19). But, K content from the treatment

combination of V_1G_0 had diminished at 9th day. In this period, the highest (0.31%) quantity of K was notified from the treatment combination of V_1G_0 which was statistically identical with V_1G_1 and V_2G_0 whereas; the lowest (0.25%) was notified from the treatment combination of V_2G_3 (Table 3.19).

9.4.3 Calcium content

A

X

N.

X

Analysis of variance of applied varieties to this investigation in terms of Ca exhibited highly significant variation at different days after storage (Appendix 3.9). It manifested an increasing trend of Ca content with the rising of storage time from both the varieties (Table 3.18). It also narrated that Khirshapat was performed better in achieving of Ca content comparing to Langra. At 12th day, higher (22.39 mg/100 g) quantity of Ca was perceived from Khirshapat and lower (20.32 mg/100 g) was perceived from Langra. This event might be possible due to genetical variation between two varieties. These results are in partially supported by the report of Nadkarni (1963).

Different doses of GA3 solution in relation to Ca content of mango pulp exhibited highly significant variation at different days after storage (Appendix 3.9). At various days during storage, Ca content expanded successively with the rising of storage time (Fig. 3.15). It also annotated that Ca content in control augmented sharply from initial to 6th day and then, it increased smoothly thereafter, it declined due to decay. At the same time, Ca content from the fruit treated with G₃ treatment accumulated very smoothly. At 9th day, the highest (24.39 mg/100 g) quantity of Ca was noticed from untreated fruit whereas; the lowest (15.81 mg/100 g) was noticed from the fruit treated with G₃ treatment. This phenomenon caused by G₃ treatment might be possible due to delay ripening that caused in lower transmission of Ca content from peel and stone to pulp of mango. But, the results of the present investigation are not supported by the findings of Peter *et al.* (2007).

The combined effect of varieties and different doses of GA3 solution in respect of Ca content of mango pulp demonstrated highly significant variation at different days after storage (Appendix 3.9). It showed an increasing trend of Ca content of mango pulp from various treatment combinations (Table 3.19). At 9th day, the highest (25.28 mg/100 g) quantity of Ca was manifested from the treatment combination of V₂G₀ and the lowest (14.96 mg/100 g) was manifested

Y

x

from the treatment combination of V_1G_3 , respectively. The results of the presented investigation revealed that the treatment combination of V_1G_3 was better in mango preservation.

9.4.4 Magnesium content

Highly significant variation was noticed in Mg content between the varieties means at different days after storage (Appendix 3.10). The results denoted that Mg content grew up gradually with rising of storage period (Table 3.20). It also indicated that Mg content accumulated steadily from initial to 9th day thereafter, it abated slightly due to starting of bad the fruit situation. At 9th day, higher (17.44 mg/100 g) was notified from Langra and lower (16.81 mg/100 g) was notified from Khirshapat. These phenomena might be probably due to genetical dissimilarities between the varieties. There were no available research findings in terms of Mg content during storage in the scientific literature but, the results of the present study revealed that Langra was batter in Mg accumulation and that progressively increase during storage.

Different doses of GA3 solution imposed in this trial showed highly significant in relation to Mg content of mango pulp at different days after storage (Appendix 3.10). The results denoted that a smooth accumulating trend of Mg content of mango pulp with the rising of storage period (Fig. 3.16). It also explained that Mg content from control enlarged gradually from initial to 6th day and then, it diminished very sharply. The highest (18.02 and 17.58 mg/100 g) quantity of Mg was derived from control and G₂ treatment at 6 and 9th days whereas, the lowest (16.18 and 16.57 mg/100 g) was derived from the fruit treated with G₃ treatment, respectively. The growing up situation of Mg content in mango pulp might have been related to starting of ripening. Peter *et al.* (2007) showed a slight change of Mg content during storage period.

The combined effect of varieties and various doses of GA3 solution were found to be significant variation in terms of Mg content at various days after storage except initial day (Appendix 3.10). It perceived an increasing trend of Mg content of mango pulp with the augmentation of storage time (Table 3.21). At 6th day, the highest (18.59 mg/100 g) quantity of Mg was recorded from the treatment combination of V₁G₀ whereas; the lowest (15.78 mg/100 g) was recorded from the

5

X

X

x









Fig. 3.14 Potassium content of mango pulp as influenced by different doses of GA3 at different days after storage. Vertical bars represent LSD at 0.05 level



Fig. 3.15 Calcium content of mango pulp as influenced by different doses of GA3 at different days after storage. Vertical bars represent LSD at 0.05 level

 $\begin{array}{l} Graphs \ represent \\ G_0 = \ Control \\ G_1 = \ 100 \ ppm \ of \ GA \ 3 \ solution \end{array}$

Fig. 3.16 Effect of different doses of GA3 on magnesium content of mango pulp at different days after storage. Vertical bars represent LSD at 0.05 level

 $G_2 = 200 \text{ ppm of GA 3 solution}$ $G_3 = 400 \text{ ppm of GA 3 solution}$

Treatments	Water	soluble pr	otein con	tent (%) a	t different days	Phosphorus (mg/100 g) at different days					
Variety	Initial	3	6	9	12	Initial	3	6	9	12	
Vı	0.56 a	0.70 a	0.86 a	1.00 a	1.18	19.00 a	20.10 a	21.59 a	22.93 a	23.34 a	
V ₂	0.46 b	0.60 b	0.74 b	0.88 b	1.11	16.34 b	17.46 b	18.89 b	20.27 b	20.69 b	
evel of significance	**	*	*	*	NS	***	***	***	***	***	

Table 3.16 Changes of water soluble protein and phosphorus content of mango pulp in varieties during storage at ambient condition

4)

2

h

Table 3.17 Combined effects of varieties and different doses of Gibberellic acid solution on water soluble protein and phosphorus content of postharvest mango pulp during storage at ambient condition

Treatments combination	Water s	Water soluble protein content (%) at different days					Phosphorus (mg/100 g) different days					
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12		
V ₁ G ₀	0.55 ab	0.88 a	1.20 a	1.25 a	1.23 a	19.45 a	21.55 a	24.33 a	24.53 a	23.05 c		
V ₁ G ₁	0.56 a	0.68 bc	0.80 b	1.16 ab	1.22 a	19.25 b	20.15 b	21.56 b	24.13 b	24.44 a		
V ₁ G ₂	0.58 a	0.65 bc	0.75 bc	0.86 cd	1.18 ab	18.83 c	19.63 c	20.52 c	21.95 c	23.45 b		
V ₁ G ₃	0.53 ab	0.60 bc	0.70 bc	0.73 de	1.08 ab	18.45 d	19.05 d	19.95 d	21.12 e	22.42 d		
V ₂ G ₀	0.50 ab	0.75 ab	1.10 a	1.16 ab	1.18 ab	16.82 e	18.93 d	21.44 b	21.84 c	20.34 g		
V ₂ G ₁	0.48 ab	0.58 bc	0.67 bc	1.00 bc	1.15 ab	16.53 f	17.44 e	18.83 e	21.34 d	21.64 e		
V ₂ G ₂	0.45 ab	0.55 c	0.62 bc	0.70 de	1.10 ab	16.26 g	17.12 f	18.15 f	19.56 f	21.13 f		
V ₂ G ₃	0.40 b	0.50 c	0.57 c	0.65 e	1.00 b	15.75 h	16.35 g	17.12 g	18.32 g	19.65 h		
Level of significance	NS	NS	NS	NS	NS	NS	NS	*	*	**		
CV%	14.80	14.22	13.32	11.30	9.28	0.60	0.56	0.66	0.49	0.48		

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents

Y

195

X

$V_1 = Langra$	$G_2 = 200 \text{ ppm of GA3 solution}$	*** indicate at 0.1% level
V ₂ = Khirshapat	$G_3 = 400 \text{ ppm of GA3 solution}$	NS means non significant
$G_0 = Control$	* indicates at 5% level	
$G_1 = 100 \text{ ppm of GA3 solution}$	** indicate at 1 % level	

N

Treatments		Potassium (%) at different days Calcium (mg/100 g) at different d							different days	days	
Variety(V)	Initial	3	6	9	12	Initial	3	6	9	12	
V ₁	0.24 a	0.25 a	0.27 a	0.29 a	0.29 a	10.53 b	12.92 b	16.06 b	19.14 b	20.32 b	
V ₂	0.22 b	0.23 b	0.26 b	0.27 b	0.27 b	12.42 a	14.81 a	17.94 a	21.17 a	22.39 a	
evel of significance	***	* **	**	***	***	***	***	***	***	***	

Table 3.18 Changes of potassium and calcium content of mango pulp between varieties during storage at ambient condition

164

Table 3.19 Combined effects of varieties and different doses of Gibberellic acid solution on potassium and calcium content of postharvest mango pulp during storage at ambient condition

Treatments combination		Potassium content (%) at different days					Calcium content (mg/100 g) at different days					
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12		
V ₁ G ₀	0.25 a	0.27 a	0.30 a	0.31 a	0.28 bc	10.65 d	15.35 b	22.09 b	23.49 b	19.88 f		
V ₁ G ₁	0.24 ab	0.25 b	0.28 b	0.30 ab	0.31 a	10.75 d	13.11 d	16.12 d	21.52 c	22.48 c		
V ₁ G ₂	0.23 bc	0.24 bc	0.26 bc	0.28 b-d	0.29 ab	10.45 e	11.78 e	12.48 g	16.58 e	21.01 e		
V_1G_3	0.22 c	0.23 b-d	0.25 c	0.27 с-е	0.28 bc	10.25 f	11.45 f	12.56 h	14.96 f	17.89 h		
V ₂ G ₀	0.23 bc	0.25 ab	0.28 ab	0.29 a-c	0.26 c	12.45 b	17.06 a	23.82 a	25.28 a	21.68 d		
V ₂ G ₁	0.22 bc	0.23 b-d	0.26 bc	0.27 с-е	0.28 bc	12.65 a	15.25 b	18.22 c	23.63 b	24.65 a		
V ₂ G ₂	0.21 cd	0.22 cd	0.25 c	0.26 de	0.27 bc	12.35 bc	13.75 c	15.45 e	19.12 d	23.56 b		
V ₂ G ₃	0.20 d	0.21 d	0.24 c	0.25 e	0.26 c	12.22 c	13.16 d	14.26 f	16.65 e	19.68 g		
Level of significance	NS	NS	NS	NS	NS	NS	**	*	***	***		
CV%	4.71	4.47	4.00	3.81	3.59	0.92	0.77	0.62	0.53	0.50		

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

X

Table represents $V_1 = Langra$ $G_2 = 200 \text{ ppm of GA3 solution}$ $V_2 = Khirshapat$ $G_3 = 400 \text{ ppm of GA3 solution}$ $G_0 = Control$ * indicates at 5% level $G_1 = 100 \text{ ppm of GA3 solution}$ ** indicate at 1 % level

X

*** indicate at 0.1% level NS means non significant

Y

4.1

.4

T

yé.

X

16

末

treatment combination of V₂G_{3.}

9.4.5 Copper content

Variation between varieties in relation to Cu of mango pulp exhibited highly significant at different days after storage (Appendix 3.10). The results indicated that Cu content of mango pulp came down gradually with the rising of storage period (Table 3.20). It also stated that Cu content of Khirshapat was more as compared to Langra. At initial day, higher (0.36 mg/100 g) content of Cu was noticed from Khirshapat and lower (0.33 mg/100 g) was noticed from Langra. At 12th day, Khirshapat had higher (0.20 mg/100 g) amount of Cu and lower (0.18 mg/100 g) amount was noticed from Langra (Table 3.20). These events might be possible due to genetical dissimilarities between the varieties. There were no available research reports in respect of Cu content of mango at the scientific literature. The results of the present studies revealed that green mangoes contained more copper comparing to stored mangoes and Khirshapat was better in copper accumulation than Langra. The present research findings also explored that Cu content abated gradually during storage period.

Different doses of GA3 solution subjected to the present study in terms to Cu content of mango pulp showed highly significant variation at different days after storage (Appendix 3.10). The results narrated that Cu content came down successively with the rising of storage period at various days. The abating trend from control was higher than the fruit treated with G_1 , G_2 and G_3 , treatment respectively (Fig. 3.17). At 12^{th} day, higher quantity of Cu (0.25 mg/100 g) was derived from the fruit treated with G_3 treatment whereas; lower (0.13 mg/100 g) was derived from control. It meant that G_3 treatment was much better than the other treatments in conservation of Cu. So, the present findings explored that G_3 treatment was much better in mango preservation and reduced the losses of Cu content.

The combined effect of varieties and imposed different doses of GA3 solution in respect of Cu content of mango pulp were perceived to be non significant variation at various days after storage (Appendix 3.10). The results narrated that Cu content in different treatment combination diminished with the rising of storage duration. At 12th day, the highest (0.26 mg/100 g) was recorded from the treatment combination

2

y.

X

×

x

of V_2G_3 , and the lowest (0.12 mg/100 g) was recorded from the treatment combination of V_1G_0 (Table 3.21).

9.4.6 Iron content

The analysis of variance of varieties in respect of Fe content of mango pulp was perceived to be highly significant at different days after storage (Appendix 3.11). There showed an expanding trend of Fe content with the passing of storage duration. The expanding trend of Fe content had more or less similar from initial to 9th day thereafter, it diminished slightly. At 12th day, higher (4.57 mg/100 g) quantity of Fe was derived from Langra and lower (3.74 mg/100 g) was derived from Khirshapat. These results are in partially supported by the findings of Nadkarni (1963).

Highly significant variation was perceived due to the effect of different doses of GA3 solution in terms of Fe content in mango pulp at various days after storage (Appendix 3.11). The results denoted that Fe content accumulated markedly from initial to 6th day and then, it fell off significantly at control and similarly G₁ treated fruits abated of Fe after 9th day. But, Fe content notified from the other fruits treated with G₂ and G₃ treatments augmented continuously up to 12th day (Fig. 3.18). At 12th day, the maximum (5.10 mg/100 g) quantity of Fe was noticed from the fruit treated with G₂ treatment whereas; the lowest (2.71 mg/100 g) was noticed from the untreated fruit, respectively. Little quantity of Fe obtained from untreated fruits was possibly due to starting of decay which resulted in lower amount. On the other hand, G₃ treated fruits was better because of its lower accumulation of Fe content with the advancement of storage duration.

The combined effect of varieties and different doses of GA3 solution were found to be non significant in terms of Fe content of mango pulp at different days after storage (Appendix 3.11). The results stated growing up trend of Fe content from initial to 6th day thereafter, it abated very fast in the treatment combination of V₁G₀ whereas; the lowest growing up trend was obtained from the treatment combination of V₂G₃ (Table 3.23). At 6th day, the highest (5.51 mg/100 g) quantity of Fe was reported from the treatment combination of V₁G₀ and the lowest (2.12 mg/100 g) was reported from the treatment combination of V₂G₃.

X

×

Jed

末

9.4.7 Manganese content

Manganese content of mango pulp was perceived to be differed significantly in both the varieties mean at different days after storage (Appendix 3.11). The results denoted that Mn content extended gradually with the increase of storage period (Table 3.22). It also stated that Langra performed better in Mn content accumulation comparing to Khirshapat. At 12th day, the highest (1.17 mg/100 g) quantity of Mn was recorded from Langra whereas; the lowest (0.99 mg/100 g) was recorded from Khirshapat (Table 3.22). There were no available research reports regarding Mn content in the scientific literature. The results of the present research findings invented that Langra was better than Khirshapat in Mn content accumulation but, it was increased during storage period.

Different doses of GA3 solution had highly significant in respect of Mn content of mango pulp at different days after storage (Appendix 3.11). It indicated that Mn content increased successively from initial to 6^{th} days, and then it fell off sharply in control. On the other hand, it expanded from initial to 9^{th} day thereafter; it abated in the fruit treated with G₁ treatment. At the same time, very lower rising trend of Mn content was notified from the fruit treated with G₃ treatment (Fig. 3.19). At 12^{th} day, Mn content ranged between 0.89 to 1.27 mg per 100 g of fresh pulp of mango. The maximum (1.27 mg/100 g) was noted from G₂ treated fruits and the minimum (0.89 mg/100 g) was noted from control.

The combined effect of varieties and used different doses of GA3 solution in relation to Mn content of mango pulp were found to be non significant at various days after storage except 9th day (Appendix 3.11). The results denoted a developing trend of Mn content in different treatment combination with the rising of storage duration (Table 3.23). At 9th day, the highest (1.38 mg/100 g) was obtained from the treatment combination of V₁G₁, and the lowest (0.82 mg/100 g) was reported from the treatment of V₂G₃.

9.4.8 Zinc content

Variation in respect of Zn content of mango pulp due to the effect of varieties exhibited highly significant at different days after storage (Appendix 3.12). The results explained that Zn content fell off sharply with the rising of storage duration in both the varieties (Table 3.24). It also narrated that Khirshapat was perceived to

Th-

大

produce more Zn content as compared to Langra. Zn content of mango pulp ranged between 1.27 to 1.53 mg per 100 g of mango pulp at initial level and 0.77 to 0.91 mg per 100 g at 12th day. The highest (1.53 and 0.91 mg/100 g) was noticed from Khirshapat and the lowest (1.27 and 0.77 mg/100 g) was noticed in Langra from initial and 12th day. There were no available research findings in terms of Zn content in the scientific review. The results of the present studies revealed that green mango especially, Khirshapat contained more Zn than Langra, but, its quantity continuously decreased with the advancement of storage duration.

Different doses of GA3 solution were found to be significant variation in Zn content of mango pulp at different days after storage (Appendix 3.12). The results noticed a diminishing trend of Zn content of mango pulp with the hastening of storage period from the fruit treated with different doses of GA3 solution. It also explained that the falling off trend was very high in control and very low in the fruit treated with G₃ treatment (Fig. 3.20). At 12^{th} day, the maximum (1.01 mg/100 g) was recorded from the fruit treated with G₃ treatment and the lowest (0.65 mg/100 g) was recorded from control. Zn content of mango pulp came down in the storage period was probably due to dissemination of Zn from pulp to stone and peel at stored condition or Zn content of mango pulp may be depressed or suppressed as influenced by metabolic activities during storage.

The combined effect of varieties and imposed different doses of GA3 solution in respect of Zn content of mango exhibited significant variation at different days after storage except 3 and 6th day (Appendix 3.12). A falling trend of Zn content of mango pulp was recorded from various treatment combinations (Table 3.25). At 12th day, the maximum (1.06 mg/100 g) quantity of Zn was recorded from Khirshapat along G₃ treatment whereas; the lowest (0.54 mg/100 g) value was recorded from Langra without treatment, respectively.

9.5 Shelf life

There was found to be significant variation between the varieties on shelf life of mango (Appendix 3.12). The longest shelf life (13.92 days) was obtained from Khirshapat and the shortest (12.67 days) was obtained from Langra (Fig. 3.21). Variation in shelf life between varieties might be possible due to genetical. The present findings from the data revealed that Khirshapat was better than Langra in preservation.

T

K



Fig. 3.17 Effect of different doses of GA3 on copper content of mango pulp at different days after storage. Vertical bars represent LSD at 0.05 level



G0 - G1 - G2 - G3 6 I I Т ٦ 5 Iron content (mg/ 100 g) 4 3 2 1 D Initial 3 6 9 12 Days after storage



Fig. 3.19 Manganese content of mango pulp as influenced by different doses of GA3 at different days after storage. Vertical bars represent LSD at 0.05 level

Graphs represe	ent
$G_0 = Control$	
$G_{1} = 100 \text{ ppm}$	of GA 3 solution

Fig. 3.20 Effect of different doses of GA3 on zinc content of mango pulp at different days after storage. Vertical bars represent LSD at 0.05 level

 $G_2 = 200 \text{ ppm of GA 3 solution}$ $G_3 = 400 \text{ ppm of GA 3 solution}$

Days after storage Fig. 3.18 Iron content of mango pulp as influenced by different doses of GA3 at different days after storage. Vertical bars represent LSD at 0.05 level

Treatments	Ма	gnesium cont	ent (mg/100	g) at differen	Copper content (mg/100 g) at different days					
Variety (V)	Initial	3	6	9	12	Initial	3	6	9	12
V ₁	16.29 a	16.80 a	17.33 a	17.44 a	17.20 a	0.33 b	0.30 b	0.26 b	0.21b	0.18 b
V ₂	15.40 b	16.13 b	16.69 b	16.81 b	16.53 b	0.36 a	0.32 a	0.29 a	0.23 a	0.20 a
Level of significance	***	***	***	***	***	***	***	***	***	***

K

Y,

Table 3.20 Changes of magnesium and copper content of mango pulp in varieties during storage at ambient condition

F

Table 3.21 Combined effects of varieties and different doses of Gibberellic acid solution on magnesium and copper content of postharvest mango pulp during storage at ambient condition

Treatments combination	Magn	esium conte	ent (mg/100) g) at differe	ent days	Co	pper conte	nt (mg/100	g) at differe	nt days
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12
V ₁ G ₀	16.55 a	17.66 a	18. <mark>5</mark> 9 a	17.55 ab	16.45 d	0.34 b-d	0.28 d	0.22 f	0.16 f	0.12 e
V ₁ G ₁	16.35 b	16.75 b	17.24 c	17.72 a	16.72 c	0.31 e	0.29 cd	0.25 de	0.20 de	0.15 d
V ₁ G ₂	16.22 bc	16.52 c	16.92 d	17.51 b	18.28 a	0.32 de	0.30 cd	0.27 cd	0.23 bc	0.20 c
V ₁ G ₃	16.05 c	16.28 d	16.58 e	16.96 c	17.35 b	0.33 c-e	0.31 bc	0.29 bc	0.25 ab	0.23 b
V ₂ G ₀	15.45 de	16.48 c	17.46 b	16.47 d	15.48 f	0.36 ab	0.30 cd	0.24 ef	0.18 ef	0.14 de
V ₂ G ₁	15.55 d	15.95 e	16.44 e	16.92 c	15.92 e	0.35 a-c	0.31 bc	0.27 cd	0.22 cd	0.18 c
V ₂ G ₂	15.35 ef	16.65 bc	17.06 cd	17.65 ab	18.15 a	0.36 ab	0.33 ab	0.30 b	0.26 a	0.23 b
V ₂ G ₃	15.25 f	15.45 f	15.78 f	16.18 e	16.58 cd	0.37 a	0.35 a	0.33 a	0.27 a	0.26 a
Level of significance	NS	***	***	***	***	NS	NS	NS	NS	NS
CV%	0.67	0.64	0.62	0.62	0.62	3.10	3.44	3.76	4.79	5.62

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

 $G_2 = 200 \text{ ppm of GA3 solution}$

18

Y

Table represents $V_1 = Langra$ $V_2 = Khirshapat$

- $V_2 = Knirsnapat$ $G_0 = Control$
- $G_1 = 100 \text{ ppm of GA3 solution}$
- G₃ = 400 ppm of GA3 solution * indicates at 5% level ** indicate at 1 % level

*** indicate at 0.1% level NS means non significant

202

Treatments		Iron content	(mg/100 g)	at different	days	Mang	Manganese content (mg/100 g) at different days					
Variety (V)	Initial	3	6	9	12	Initial	3	6	9	12		
Vı	2.32 a	2.96 a	4.02 a	4.68 a	4.57 a	0.60 a	0.80 a	1.04 a	1.16 a	1.17 a		
V ₂	1.44 b	2.06 b	3.14 b	3.80 b	3.74 b	0.44 b	0.65 b	0.87 b	1.01 b	0.99 b		
Level of significance	***	***	***	***	***	***	***	***	***	***		

4

Vi

Table 3.22 Changes of iron and manganese content of postharvest mango pulp between varieties during storage at ambient condition

K.

Table 3.23 Combined effects of varieties and different doses of Gibberellic acid solution on iron and manganese content of postharvest mango pulp during storage at ambient condition

Treatments combination	Irc	on content	(mg/100 g)	at different	: days	Man	ganese cor	ntent (mg/1	00 g) at diff	erent days
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12
V ₁ G ₀	2.54 a	3.55 a	5.51 a	5.04 b	3.14 f	0.64 a	0.96 a	1.33 a	1.19 b	0.96 f
V ₁ G ₁	2.42 ab	3.13 b	4.14 c	5.34 a	4.74 c	0.62 b	0.84 b	1.12 c	1.38 a	1.24 b
V ₁ G ₂	2.25 b	2.76 c	3.48 d	4.47 c	5.45 a	0.59 c	0.74 d	0.92 d	1.14 c	1.36 a
V ₁ G ₃	2.05 c	2.38 d	2.97 f	3.88 e	4.95 b	0.56 d	0.66 e	0.79 e	0.94 f	1.12 d
V_2G_0	1.72 d	2.68 c	4.67 b	4.18 d	2.27 g	0.48 e	0.82 c	1.16 b	1.04 d	0.81 g
V ₂ G ₁	1.55 d	2.26 d	3.23 e	4.44 c	3.82 e	0.45 f	0.67 e	0.92 d	1.18 b	1.03 e
V ₂ G ₂	1.35 e	1.85 e	2.54 g	3.54 f	4.74 c	0.43 g	0.58 f	0.76 f	0.98 e	1.17 c
V ₂ G ₃	1.15 f	1.45 f	2.12 h	3.02 g	4.12 d	0.41 h	0.52 g	0.65 g	0.82 g	1.96 f
Level of significance	NS	NS	NS	NS	NS	NS	NS	NS	*	NS
CV %	5.65	4.23	3.19	2.50	2.55	4.06	2.93	2.22	1.96	1.96

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

K

Table represents

- $V_1 = Langra$ $V_2 = Khirshapat$ $G_0 = Control$ $G_1 = 100 \text{ ppm of GA3 solution}$
- $G_2 = 200 \text{ ppm of GA3 solution}$ $G_3 = 400 \text{ ppm of GA3 solution}$ * indicates at 5% level

*** indicate at 0.1% level NS means non significant

** indicate at 1 % level

7

Y

-

T

来

Applied different doses of GA3 solution in this investigation in terms of shelf life of mango showed highly significant (Appendix 3.12). The shelf life of mango ranged between 8.17 to 17.00 days from different doses of GA3 solution. The longest shelf life (17.00 days) was reported from the fruit treated with G_3 treatment followed by the shelf life of the fruits treated with G_2 (14.83 days), and G_1 (13.17 days) treatments whereas; the shortest shelf life (8.17 days) was reported from control, (Fig. 3.22). The longest shelf life obtained from the fruit treated with G_3 treatment might be probably due to suppression or depression of physiological and biochemical activities in responding for slower senescence of harvested fruits and consequently led to the longest shelf life. The results of the present investigation are in conformity with the findings of Ranjan *et al.* (2005), Jain and Mukherjie (2001) and Mondal *et al.* (1995).





Fig. 3.21 Effect of varieties on shelf life of mango

Fig. 3.22 Effect of different doses of GA3 on shelf life of mango. Vertical bars represent LSD at 0.05 level

Graph represents	
$V_1 = Langra$	$G_1 = 100 \text{ ppm of GA3 solution}$
$V_2 = Khirshapat$	$G_2 = 200 \text{ ppm of GA3 solution}$
G ₀ =Control	$G_3 = 400 \text{ ppm of GA3 solution}$

The combined effects of varieties and subjected to different doss of GA3 solution in terms of shelf life of mango were observed non significant (Appendix 3.12). The shelf life was ranged between 7.67 to 18.00 days from various treatment combinations. The longest shelf life (18.00 days) was noticed from the treatment combination of V_2G_3 , whereas the shortest (7.67 days) was noticed from V_1G_0 , respectively.

Treatments		Zinc conte	nt (mg/100 g) at o	different days		Shelf life
Variety (V)	Initial	3	6	9	12	Total days
V ₁	1.27 b	1.15 b	1.06 b	0.91 b	0.77 b	12.67 b
V ₂	1.53 a	1.30 a	1.19 a	1.05 a	0.91 a	13.92 a
Level of significance	***	***	***	***	***	**

Table 3.24 Changes of zinc content of mango pulp and shelf life of postharvest mango in varieties during storage at ambient condition

4

Table 3.25 Combined effects of varieties and different doses of Gibberellic acid solution on zinc content of mango pulp and shelf life of postharvest mango during storage at ambient condition

(4)

×,

A

Treatments combination		Zinc conte	nt (mg/100 g) at	different days		Shelf life
Varieties × Treatments	Initial	3	6	9	12	Total days
V ₁ G ₀	1.25 e	1.12 f	0.95 g	0.76 f	0.54 f	7.67 h
V ₁ G ₁	1.28 d	1.14 ef	1.05 f	0.88 e	0.72 e	12.67 f
V ₁ G ₂	1.26 de	1.16 e	1.11 e	0.98 d	0.86 c	14.33 d
V_1G_3	1.27 de	1.19 d	1.13 de	1.03 c	0.95 b	16.00 b
V ₂ G ₀	1.48 c	1.26 c	1.14 cd	0.96 d	0.75 d	8.67 g
V ₂ G ₁	1.55 a	1.28 c	1.16 c	1.02 c	0.87 c	13.67 e
V ₂ G ₂	1.52 b	1.31 b	1.19 b	1.08 b	0.96 b	15.33 c
V ₂ G ₃	1.56 a	1.35 a	1.25 a	1.15 a	1.06 a	18.00 a
Level of significance	NS	NS	**	**	**	NS
CV%	1.52	1.73	1.90	2.16	2.53	5.66

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

-

Table represents $G_2 = 200 \text{ ppm of GA3 solution}$ *** indicate at 0.1% level $V_1 = Langra$ $G_2 = 200 \text{ ppm of GA3 solution}$ *** indicate at 0.1% level $V_2 = K \text{ hirshapat}$ $G_3 = 400 \text{ ppm of GA3 solution}$ *** indicate at 0.1% level $G_0 = \text{Control}$ * indicates at 5% levelNS means non significant $G_1 = 100 \text{ ppm of GA3 solution}$ ** indicate at 1 % level

205

*

¥

2

,k

×

*

10.0 Experiment 4

Influence of different doses of Bavistin DF on physicochemical properties and shelf life of mango varieties during storage

10.1 Changes in skin color

Different doses of Bavistin DF (BDF) significantly affected the skin color of both the fruits namely Langra and Khirshapat (Table 4.1). Both varieties demonstrated the original green color at the initial stage of harvesting. At 3^{rd} day, Langra developed trace in yellow color at control (B₀) and green color at 250 ppm (B₁), 500 ppm (B₂) and 750 ppm (B₃), and Khirshapat was perceived light green at control and retained its green color at B₁, B₂ and B₃ treatments.

At 6th day, Langra was noted yellow, greenish yellow, yellowish green and trace in yellow color at control, B_1 , B_2 and B_3 treatments, respectively. On the other hand, Khirshapat was notified greenish yellow, trace in yellow and yellowish green color at B_0 , B_1 , and B_2 treatments but, it retained its original green color at B_3 treatment.

At 9th day, Langra was modified deep yellow color at control, greenish yellow at B₁ and yellowish green at B₂ as well as trace in yellow at B₃ treatments, which Khirshapat exhibited yellow color at control, yellowish green color at B₁ and B₂ and light green color at B₃ treatment.

At 12th day of storage, Langra was transformed into black spotted yellow color at control and yellow, greenish yellow and yellowish green color at B_1 , B_2 and B_3 treatments, respectively. On the other hand, Khirshapat was converted into light spotted yellow at B_0 , yellow, greenish yellow and trace in yellow at B_1 , B_2 and B_3 treatments, respectively.

At 15^{th} day of storage, Langra was found to be decomposed at B₀ and B₁, yellow and greenish yellow at B₂ and B₃ treatments, respectively. At the same time Khirshapat was notified as completely rotten at B₀ and B₁ treatment and changed into deep yellow and greenish yellow at B₂ and B₃ treatments, respectively. The results of the present study are in partially agreement with the findings of Dhemre and Waskar (2004).

				[Days after storage		
Varieties	Treatments	Initial	3	6	9	12	15
	Bo	Green	Trace in yellow	Yellow	Deep yellow	Black spotted yellow	_
	B ₁	Green	Green	Greenish yellow	Greenish yellow	Yellow	-
V ₁	B₂	Green	Green	Yellowish green	Yellowish green	Greenish yellow	Yellow
	B ₃	Green	Green	Trace in yellow	Trace in yellow	Yellowish green	Greenish yellow
	Bo	Green	Trace in yellow	Greenish yellow	Yellow	Light spotted yellow	_
	B1	Green	Green	Trace in yellow	Yellowish green	Yellow	_
V ₂	Bz	Green	Green	Yellowish green	Yellowish green	Greenish yellow	Deep yellow
	B ₃	Green	Green	Green	Light green	Yellowish green	Greenish yellow

4

-

Table 4.1 Changes in skin color of two mango varieties as influenced by different doses of Bavistin DF during storage at ambient condition

H)

207

-

Table indicates

42

V1 = Langra B_0 =Control B_2 =500 ppm of BDF solutionV2 = Khirshapat B_1 =250 ppm of BDF solution B_3 =750 ppm of BDF solution

X

1

Y

*

10.2 Changes in physical characters during storage environments

Different parameters oriented to the physical changes of mangoes are presented and interpreted in the following sub-headings.

10.2.1 Physiological weight loss

Varieties were found to be highly significant in terms of PWL at different days after storage (Appendix 4.1 and Table 4.2). At each day, Khirshapat (V_2) successively showed more PWL comparing to Langra with the rising of storage duration (Table 4.2). Higher (10.86%) and lower (9.85%) of PWL were obtained from Khirshapat and Langra at 12th day, respectively. The results also denoted that total PWL progressively grew up with the increase of storage duration. The data also explained that Langra was better than Khirshapat owing to its lesser PWL. Water loss through lenticel seems to be the possible reason of physiological weight loss in the fruits during storage. Lower lenticel density in Langra facilitated lesser water loss leading to minimum total weight loss (Azad, 2001). Singh *et al.* (2000) also observed more or less the similar findings.

Analysis of variance of mango pulp in relation to PWL as influenced by different doses of BDF solution was found to be highly significant at different days after storage (Appendix 4.1). At different days, the results noted that control treatment was very swift in PWL compared to B_1 , B_2 and B_3 treatments (Fig. 4.1). At 12th day, the maximum PWL (11.68%) was perceived in B_1 and the minimum (9.35%) in B_3 . These phenomena happened might be probably due to 750 ppm (B_3) of BDF solution diminished the metabolic activities of mango pulp resulting in lower-PWL. These results of the present investigation are strongly supported by the findings of Dhemre and Waskar (2004). Parmar and Chundawat (1989) also observed the more or less similar results. These results are also conformity with the findings of Ahmed and Singh *et al.* (2000).

The combined effect of varieties and different doses of BDF showed significant variation in PWL at different days after storage except 12^{th} day (Appendix 4.1). At different days, the results stated that various treatments combination showed in PWL successively with the rising of storage duration. At 9^{th} day, there found the maximum PWL (10.65%) at V₂B₀ and the minimum (7.25%) at V₁B₃. It also revealed

k

太

that Khirshapat lost the highest amount of water along with B_3 treatment followed by other treatment combinations.

10.2.2 Moisture content

The analysis of variance of imposed varieties showed highly significant variation in moisture content at different days after storage except initial day (Appendix 4.1 and Table 4.2). At different days, the results interpreted that moisture content increased with the increase of storage period. The increasing trend was more or less similar from initial to 9th days and thereafter its increasing tendency decreased due to starting decay. It also denoted that each day of storage, Khirshapat absorbed more moisture comparing to Langra. Higher (88.13%) and lower (85.05%) were derived from Khirsapat and Langra at 12th day (Table 4.2). The reports were quite different than the present results as stated by Shajahan *et al.* (1994) and they observed that moisture content increased in Langra than Khirsapat. But, these results are in agreement with the findings of Azad (2001). This variation might be possible due to genetical, location, weather effect, soil quality or maturity of the fruit etc.

Variation among the means of different doses of BDF solution in connection with moisture content was perceived to be significant at different days after storage except initial day (Appendix 4.1). At different days, moisture content increased continuously with the advancement of storage duration. The last growing up trend of moisture content was noticed from control and B₁ treatment at 6 and 9th days (Fig. 4.2) whereas; B₂ and B₃ treatments also showed their increasing trend. Untreated fruit absorbed the highest moisture content (87.45%) at 9th day and the lowest (85.05%) was noted from B₃ treatment, respectively. The increasing trend of moisture content from initial to 6th days might be possibly due to metabolic activities and osmotic pressure inside the mango fruit as well as its decreased might be possible due to suppression of metabolic activities that resulted in decaying and drying.

The combined effect of varieties and different doses of BDF solution in terms of moisture content demonstrated non significant variation at different days after storage (Appendix 4.1 and Table 4.3). At different days of storage, moisture content

to

1

*

was absorbed in mango pulp with the advancement of storage period. The treatment combination of V_2B_0 , V_2B_1 and V_2B_2 produced the maximum moisture content (88.60%, 89.00% and 89.30%) at 6, 9 and 12^{th} days, respectively. In this storage period, the lowest values (82.70%, 83.50% and 84.30%) were obtained from the treatment combination of V_1B_3 , respectively (Table 4.3). Moisture content increased in the present study is in partially agreement with the findings of El-Mahmoudi and Eisawi (1968) in banana.

10.2.3 Dry matter content

The variation in varieties means in connection with dry matter content exhibited highly significant variation at different days after storage (Appendix 4.2 and Table 4.4). At different days of storage, dry matter content diminished successively with the rising of storage duration. It denoted that Langra produced comparatively more dry matter comparing to Khirshapat. At initial day, Lngra gave higher (18.65%) dry matter while, Khirshapat had lower (15.48%) and at 12th day, Langra achieved higher (14.95%) while Khirshapat achieved lower (11.88%), respectively (Table 4.4). These results are in partially supported by the findings of Hossain (1999).

Different doses of BDF solution used in this study in terms of dry matter content of mango pulp exhibited highly significant variation at different days after storage except initial day (Appendix 4.2). At different days of storage, dry matter content diminished continuously with the increase of storage duration (Fig. 4.3). It also revealed that B₃ treatment gave the highest dry matter (15.75%) at 6th day while; the lowest dry matter (12.95%) was recorded from control.

The combined effect of varieties and different doses of BDF solution on dry matter content of mango pulp showed non significant at different days after storage (Appendix 4.2 and Table 4.5). At initial day, the highest dry matter content (18.80%) was obtained from the treatment combination of V_1B_3 which was statistically similar with the treatment combination of V_1B_0 , V_1B_1 and V_1B_2 whereas; the lowest (15.30%) was recorded from the treatment combination of V_2B_1 , V_2B_2 , and V_2B_3 . At 12th day, the highest value (15.80%) was reported from the treatment combination of

2

x

A-

Se

4

 V_1B_0 which was statistically at par with V_1B_3 and the lowest value (10.70%) was notified from V_2B_2 (Table 4.5). The successive decrease in dry matter content with the increase of storage period might be possibly due to breaking down of the complex carbohydrates into simple molecules and H_2O as well as adding water through osmotic process and metabolic activities. The abating trend in dry matter content of mango pulp is not supported by the findings of Hossain (1999).

10.2.4 Ash content

Variation in varieties means in connection with ash content of mango pulp showed significant variation at different days after storage (Appendix 4.2 and Table 4.4). At different days of storage, values of ash content fell off continuously with the advancement of storage duration. It narrated that Langra gave comparatively more ash than Khirshapat at all days of storage. Higher (1.03%) ash content was derived from Langra at initial day whereas; Khirshapat produced the lowest (0.87%) and again at 12th day higher (0.83%) was recorded from Langra and lower achievement (0.67%) was recorded from Khirshapat (Table 4.4).

Different doses of BDF solution were selected as non significant in connection with ash content of mango pulp at different days after storage (Appendix 4.2). The results indicated that ash content imposed by different doses of BDF solution diminished slightly from B_3 treatment and markedly from control (Fig. 4.4). At 9th day, the highest (0.85%) was noticed from B_3 and the lowest (0.72%) was noticed from control.

The combined effect of varieties and different doses of BDF solution were manifested to be non-significant in connection with ash content of mango pulp at various days after storage (Appendix 4.2 and Table 4.5). At various days, the results obtained from investigation that ash content gradually decreased with the rising of storage period (Table 4.5). It may suggest from the present data that ash content demonstrated good correlation with the dry matter content.

10.3 Changes in biochemical properties of mango during storage environments

Different behavior of biochemical properties in connection with mango pulp as affected by various doses of BDF solution are presented and interpreted in the following sub-headings.

B0 - B1 - B2 - D B3 14 Ι 12 Physiological weight loss (%) 10 8 6 4 2 0 6 9 3 12 Days after storage



Fig. 4.1 Effect of different doses of Bavistin DF (BDF) on physiological weight of mango at different days after storage. Vertical bars represent LSD at 0.05 level







influenced by different doses of Bavistin DF

(BDF) at different days after storage.

Fig. 4.3 Dry matter content of mango pulp as influenced by different doses of Bavistin DF (BDF) at different days after storage. Vertical bars represent LSD at 0.05 level

 $\begin{array}{ll} \mbox{Graphs represent} & & \\ \mbox{B}_0 = \mbox{Control} & & \\ \mbox{B}_1 = 250 \mbox{ ppm of BDF solution} & \\ \mbox{B}_3 = 750 \mbox{ ppm of BDF solution} & \\ \end{array}$

Vertical bars represent LSD at 0.05 level

at the

4

1

ta

F

Treatments	Physiolo	ogical weight	loss (%) at o	different days	Moisture content (%) at different days					
Variety (V)	3	6	9	12	Initial	3	6	9	12	
V1	5.49 b	6.94 b	8.33 b	9.85 b	81.35 b	82.68 b	83.73 b	84.65 b	85.05 b	
V ₂	6.33 a	8.00 a	9.26 a	10.86 a	84.53 a	85.85 a	86.85 a	87.78 a	88.13 a	
Level of significance	***	***	***	***	***	***	***	***	***	

Table 4.2 Changes of physiological weight loss of mango and moisture content of mango pulp between varieties during storage condition

tes.

+

Table 4.3 Combined effects of varieties and different doses of Bavistin DF solution on physiological weight loss and moisture content of postharvest mango at ambient condition

Treatments combination	Physiological weight loss (%) at different days				Moisture content (%) at different days				
Varieties × Treatments	3	6	9	12	Initial	3	6	9	12
V ₁ B ₀	6.90 b	8.50 b	9.85 b	11.12 b	81.50 c	83.80 d	85.50 d	85.10 f	84.20 f
V ₁ B ₁	5.70 e	7.10 e	8.55 e	10.15 cd	81.40 c	82.60 e	83.60 e	85.90 e	85.50 e
V ₁ B ₂	4.85 g	6.25 g	7.65 g	9.25 e	81.30 c	82.30 ef	83.10 f	84.10 g	86.20 d
V ₁ B ₃	4.50 h	5.90 h	7.25 h	8.90 e	81.20 c	82.00 f	82.70 g	83.50 h	84.30 f
V ₂ B ₀	7.60 a	9.85 a	10.65 a	12.25 a	84.70 a	87.00 a	88.60 a	88.20 b	87.30 c
V ₂ B ₁	6.65 c	8.15 c	9.50 c	11.10 b	84.60 ab	85.80 b	86.70 b	89.00 a	88.50 b
V ₂ B ₂	5.80 d	7.30 d	8.70 d	10.30 c	84.50 ab	85.50 b	86.30 c	87.30 c	89.30 a
V ₂ B ₃	5.25 f	6.70 f	8.20 f	9.80 d	84.30 b	85.10 c	85.80 d	86.60 d	87.40 c
Level of significance	**	* * *	**	NS	NS	NS	NS	NS	NS
CV%	0.88	0.70	0.60	2.21	0.24	0.24	0.24	0.25	0.23

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents

$V_1 = Langra$	B ₁ =250 ppm of BDF solution	* indicates at 5% level
V ₂ = Khirshapat	B ₂ =500 ppm of BDF solution	** indicate at 1 % level
B ₀ =Control	B ₃ = 750 ppm of BDF solution	*** indicate at 0.1% level

NS means non significant

4

t

A

Ar

*
Treatments	C	bry matter c	ontent (%)	at different	Ash content (%) at different days						
Variety (V)	Initial	3	6	9	12	Initial	3	6	9	12	
V ₁	18.65 a	17.33 a	16.28 a	15.35 a	14.95 a	1.03 a	0.97 a	0.91 a	0.86 a	0.83 a	
V ₂	15.48 b	14.15 b	13.15 b	12.23 b	11.88 b	0.87 b	0.81 b	0.76 b	0.70 b	0.67 b	
Level of significance	***	***	***	***	***	**	**	**	**	**	

Table 4.4 Changes of dry matter and ash content of mango pulp between varieties during storage at ambient condition

4

Table 4.5 Combined effects of varieties and different doses of Bavistin DF solution on dry matter and ash content of postharvest mango pulp at room temperature

Treatments combination	(Dry matter co	ntent (%) a	at different o	lays		Ash cont	ent (%) at o	t (%) at different days		
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12	
V ₁ B ₀	18.50 a	16.20 c	14.50 d	14.90 c	15.80 a	1.02	0.91 ab	0.82 ab	0.80 ab	0.82 ab	
V ₁ B ₁	18.60 a	17.40 b	16.40 c	14.10 d	14.50 b	1.02	0.97 a	0.92 a	0.82 ab	0.78 ab	
V1B2	18.70 a	17.70 ab	16.90 b	15.90 b	13.80 c	1.03	0.98 a	0.94 a	0.89 a	0.79 ab	
V ₁ B ₃	18.80 a	18.00 a	17.30 a	16.50 a	15.70 a	1.04	1.00 a	0.96 a	0.92 a	0.95 a	
V ₂ B ₀	15.30 b	13.00 f	11.40 g	11.80 g	12.70 d	0.86	0.75 b	0.67 b	0.64 b	0.67 b	
V ₂ B ₁	15.40 b	14.20 e	13.30 f	11.00 h	11.50 e	0.87	0.81 ab	0.76 ab	0.65 b	0.63 b	
V ₂ B ₂	15.50 b	14.50 e	13.70 e	12.70 f	10.70 f	0.87	0.82 ab	0.78 ab	0.73 ab	0.64 b	
V ₂ B ₃	15.70 b	14.90 d	14.20 d	13.40 e	12.60 d	0.88	0.84 ab	0.81 ab	0.77 ab	0.73 ab	
Level of significance	NS	NS	NS	NS	NS	NS	NS	NS	NS	NS	
CV %	1.24	1.28	1.37	1.47	1.51	11.25	11.98	12.74	13.64	15.24	

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents

$V_1 = Langra$	B ₁ =250 ppm of BDF solution	* indicates at 5% level
$V_2 = Khirshapat$	B ₂ =500 ppm of BDF solution	** indicate at 1 % level
B ₀ =Control	$B_3 = 750 \text{ ppm of BDF solution}$	*** indicate at 0.1% level

NS means non significant

41

10

N

4

X

10

1

1º

F

10.3.1 Vitamin C content

Variation between varieties means in terms of vitamin C content of mango pulp was highly statistical significant at different days after storage (Appendix 4.3 and Table 4.6). At various days, the results denoted that Langra was better in performing of vitamin C accumulation as compared to Khirshapat (Table 4.6). It also stated that quantity of vitamin C came down with the rising of storage period. At the initial stage, Langra produced higher (131.86 mg/100 g) quantity of vitamin C while; Khirshapat had lower (41.30 mg/100 g). At 12th day, Langra was noticed as higher (15.35 mg/100 g) producer of vitamin C and Khirshapat was lower (7.88 mg/100 g) producer (Table 4.6).These results revealed that vitamin C content gradually came down with the passing of storage duration in both the varieties. It might be probably due to rising of ethylene synthesis resulting in oxidation of ascorbic acid. The results of the present study are in agreement with the findings of Azad (2001), Shyamalamma (1995), Gafur *et al.* (1994) and Absar *et al.* (1993).

Variation among the means of different doses of BDF solution in connection with vitamin C content was performed highly statistical significant at different days after storage (Appendix 4.3). At initial day, green mangoes treated with B₃ treatment produced the highest (88.24 mg/100 g) amount of vitamin C while; the lowest (85.25 mg/100 g) was notified from the untreated mangoes. After initial day, the results diminished successively with the advancement of storage time (Fig. 4.5). It also denoted that the diminishing tendency was hastily in control, but it was delay in B₃ treated fruit. At 12th day, the maximum (16.00 mg/100 g) was derived from B₃ treatment and the minimum (7.35 mg/100 g) was derived from control, respectively.The reduction of vitamin C content in both treated and untreated mangoes at different storage period might be possible due to oxidation of ascorbic acid and B₃ treatment was possibly causing delay ripening resulting in lower oxidation in vitamin C. These results of the present investigation are in agreement with the findings of Ahmed and Singh (2000). Parmar and Chundawat (1989) also found more or less similar result.

The combined effect of varieties and different doses of BDF solution showed significant variation in respect of vitamin C content at different days after storage

1

1

1-

-

(Appendix 4.3 and Table 4.7). At various days, the results were found that vitamin C content came down with the extending of storage period. The quantity of vitamin C ranged between 3.90 to 19.70 mg per 100 g of fresh mango pulp at 12^{th} day. The highest (19.70 mg/100 g) was obtained from the treatment combination of V₁B₁ and the lowest (3.90 mg/100 g) was obtained from the treatment combination of V₂B₀.

10.3.2 Titratable acidity

Variation between varieties means in connection with titratable acidity was perceived to be highly significant at different days after storage (Appendix 4.4 and Table 4.8). At various days of storage, Langra was noted higher titratable acid producer as compared to Khirshapat. Titratable acidity abated with the advancement of storage period. The abating trend was hastily from initial to 3rd day and thereafter, it was slower. At initial day, higher (3.77%) was derived from Langra while lower (2.47%) was noticed from Khirshapat. At 12th day, higher (0.31%) was recorded in Langra while; Khirshapat gave lower amount (0.24%). The decreasing trend of titratable acidity during storage period was reported by Upadhyay and Tripathi (1985), Leon and Lima (1968) and Medlicott *et al.* (1986) also found the similar results. According to them, acidity was reduced during storage growth on attainment of maturity and ripening. The results of the present investigation might be possibly due to genetical dissimilarities between two varieties.

Different doses of BDF solution imposed to this investigation in terms of titratable acidity showed significant variation among the means at various days after storage (Appendix 4.4). At various days of storage, titratable acid content diminished hastily from initial to 3 days and then, it diminished steadily (Fig. 4.6). In all the storage period, higher titratable acidity (3.19, 1.20, 0.92, 0.74 and 0.42%) was derived from B_3 treatment from initial to 12^{th} days followed by 3.09, 0.68, 0.40, 0.22 and 0.11% from untreated mangoes (Fig. 4.6). These results are supported by the findings of Ahmed and Singh (2000). These phenomena happened might be possible due to B_3 treatment delayed ripening that caused in lower diminishing trend of titratable acidity while control treatment caused ripening fast resulting in high decreasing trend of titrable acid content.

k

*

1

1

The combined effect of varieties and different doses of BDF solution in relation to titratable acidity of mango pulp demonstrated significant variation at different days after storage except initial days (Appendix 4.4 and Table 4.9). At different days of storage, there perceived a decreasing trend of titratable acid content with the rising of storage period (Table 4.9). At 6th day, the highest (1.00%) quantity was recorded from the treatment combination of V₁B₃ and the lowest acid concentration (0.30%) was recorded from the treatment combination of V₂B₀ (Table 4.9). This occurrence might be probably due to the reduction of acid oxidation at V₂B₃ combination as well as genetical variation in between varieties.

10.3.3 Pulp pH

The analysis of variance between the varieties exhibited significant variation in terms of pulp pH of mango at different days after storage except at 6th day (Appendix 4.4 and Table 4.8). At various days of storage, there observed a growing up trend of pulp pH with the increase of storage period. In each storage period, pulp pH was found more in Khirshapat compared to Langra. Higher pulp pH (6.96) was noted from Khirshapat at 12th day whereas; lower (6.85) was noted from Langra. The growing up trend of pulp pH was also observed by Yuniarti (1980), Kumar *et al.* (1993) and Shahjahan *et al.* (1994). This phenomenon might be possible due to oxidation of acid during storage resulting in higher pH and also might have been genetical dissimilarities between varieties.

Different doses of BDF solution subjected to this trial showed significant variation in pulp pH at different days after storage (Appendix 4.4). The results indicated that the growing up trend of pulp pH was perceived from different treated and untreated mangoes at various days of storage (Fig. 4.7). Pulp pH was higher in control at all stages of storage followed by the fruits treated with B_1 , B_2 and B_3 treatments, respectively. pH value of mango pulp was higher (7.05) in control which was statistically at par with B_1 and B_2 treatment whereas; the fruits treated with B_3 produced lower (6.57) value at 12^{th} day (Fig. 4.7). The results of the present investigation at B_3 treatment interrupted the loss of acid oxidation resulting in lower pH value. These results are in agreement with the findings of Ahmed and Singh (2000).

The combined effect of varieties and different doses of BDF solution imposed to this study in pulp pH were noticed to be non significant at different days after storage (Appendix 4.4). There was found a rising trend of pulp pH from various treatment combinations at different days of storage (Table 4.9). At 6th day, the highest (6.90) pH value was reported from the treatment combination of V₂B₀ which was statistically at par with V₁B₀ and the lowest (6.70) was reported from the treatment of combination of V₁B₃, respectively.

10.3.4 Total soluble solid (Brix %) content

Statistically highly significant variation was observed in TSS content between two varieties at different days after storage (Appendix 4.5). The results exhibited that TSS content of mango pulp developed in a continuous stream with the expansion of storage period. The developing trend was hastily from initial to 6th day thereafter; it increased slower. From initial to 6th day, Khirshapat was better in TSS accumulation than Langra, but, after 6th day, Langra performed better than Khirshapat up to 12th day. At 9th day, higher (17.90%) TSS quantity was noted from Khirshapat and lower (17.00%) was noted from Langra (Table 3.10). The higher percentage of TSS during storage in mango was reported by Singh (1968). Absar *et al.* (1993) reported that TSS was increased with maturity of mango fruit. But, they found highest TSS in Langra. Mollah and Siddique (1973) reported that TSS value varied from cultivar to cultivar. These might be possible due to genetically differences between varieties.

Different doses of BDF solution implied to the postharvest mangoes in this study were noticed significant variation in terms of TSS content at different days after storage (Appendix 4.5). At different days of storage, the results showed that TSS accumulation increased with the increase of storage duration. It also explored that TSS content was hastily grown up from untreated mangoes from initial to 6^{th} day and then, it came down significantly (Fig. 4.8). The other treatment viz., B₁ also increasingly produced TSS from initial to 9^{th} day and thereafter, it decreased sharply. Mango fruits treated with B₂ also produced more or less similar enhancing trend from initial to 12^{th} days. But, the fruits treated with B₃ treatment gave very slower motion in TSS accumulation at various days after storage. The highest (21.25 and 21.30%)

8

7

6

5

3 2

1 0

Initial

Pulp pH 4 т

Т

3

4

X

1.

A.

*

Results and Discussion





Fig. 4.6 Titratable acidity of mango pulp

as influenced by different doses of BDF at

B3

12

Fig. 4.5 Effect of different doses of BDF on vitamin C content of mango pulp at different days after storage. Vertical bars represent LSD at 0.05 level





Fig. 4.7 Pulp pH of mango as influenced by different doses of BDF at different days after storage. Vertical bars represent LSD at 0.05 level

9

Graphs represent B₀ =Control B1=250 ppm of BDF solution

6

Days after storage

Fig. 4.8 Effect of different doses of BDF on total soluble solid content of mango pulp at different days after storage. Vertical bars represent LSD at 0.05 level

B₂=500 ppm of BDF solution $B_3 = 750 \text{ ppm of BDF solution}$

Treatments		Vitamin C con	itent (mg/100 g)	at different days	
Variety (V)	Initial	3	6	9	12
V,	131.86 a	107.05 a	54 43 a	32.90 a	15.15 a
V ₂	41.30 b	29.64 b	16.30 b	10.43 b	7.88 b
Level of significance	***	***	***	***	***

Table 4.6 Changes of vitamin C content of mango pulp in varieties during storage at ambient condition

7

r

Table 4.7 Combined effects of varieties and different doses of Bavistin DF solution on vitamin C content of postharvest mango at ambient condition

Treatments combination		Vitamin C cont	ent (mg/100 g) at	different days	
Varieties × Treatments	Initial	3	6	9	12
V ₁ B ₀	131.30 b	84.75 d	37.30 d	18.50 d	10.80 e
V ₁ B ₁	130.70 b	110.50 c	55.40 c	33.20 c	13.60 c
V ₁ B ₂	132.60 a	114.30 b	60.70 b	38.60 b	16.50 b
V ₁ B ₃	132.90 a	118.70 a	64.30 a	41.30 a	19.70 a
V ₂ B ₀	41.50 d	22.45 h	11.20 h	7.30 h	3.90 h
V ₂ B ₁	39.80 e	29.60 g	15.30 g	9.20 g	6.60 g
V ₂ B ₂	40.30 de	31.80 f	17.50 f	11.90 f	8.70 f
V ₂ B ₂	43.60 c	34 70 e	21.20 e	13.30 e	12.30 d
Level of significance	*	***	***	***	***
CV%	0.85	0.98	0.32	0.58	0.92

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents

$V_1 = Langra$	B ₁ =250 ppm of BDF solution	 indicates at 5% level
V ₂ = Khirshapat	B ₂ =500 ppm of BDF solution	** indicate at 1 % level
B₀ =Control	$B_3 = 750 \text{ ppm of BDF solution}$	*** indicate at 0.1% level

NS means non significant

*

. 8

Treatments	Titratable acidity (%) at different days					Pulp pH at different days						
Variety (V)	Initial	3	6	9	12	Initial	3	6	9	12		
V.	3 77 2	1 10 a	0.78 a	0.55 a	0 31 a	3.55 b	4.55 b	5.65	6.53 b	6.85 b		
V ₁	2.47 h	0.92 h	0.65 b	0.55 a	0.24 b	3.65 a	4.65 a	5.73	6.65 a	6.96 a		
Level of significance	***	***	***	***	***	*	*	NS	*	*		

Table 4.8 Changes of titratable acidity and pH of postharvest mango pulp in varieties during storage at room temperature

*

*

Table 4.9 Combined effects of varieties and different doses of Bavistin DF solution on titratable acidity and pulp pH of postharvest mango pulp during storage at ambient condition

Treatments combination		Titratable a	cidity (%) at	different da	ys		Pulp	pH at differe	ent days	
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12
V ₁ B ₀	3.70 b	0.88 e	0.50 f	0.25 e	0.12 e	3.70 ab	4.70 ab	6.70 a	7.00 ab	7.00 ab
V ₁ B ₁	3.75 ab	1.10 c	0.75 d	0.50 c	0.28 c	3.60 bc	4.60 bc	5.60 b	6.90 D	7.00 au
V ₁ B ₂	3.80 ab	1.15 b	0.88 b	0.65 b	0.35 b	3.50 bc	4.50 bc	5.20 cd	6.20 C	0.90 0
V ₁ B ₃	3.83 a	1. 25a	1.00 a	0.80 a	0.48 a	3.40 c	4.40 c	5.10 d	6.00 d	6.50 C
V ₂ B ₀	2.48 cd	0.47 f	0.30 g	0.18 f	0.09 e	3.80 a	4.80 a	6.80 a	7.10 a	7.10 a
V ₂ B ₁	2.40 d	0.95 d	0.68 e	0.35 d	0.20 d	3.70 ab	4.70 ab	5.60 b	7.00 ab	7.10 a
V ₂ B ₂	2.43 d	1.10 c	0.78 d	0.52 c	0.30 c	3.60 a-c	4.60 a-c	5.30 c	6.30 c	7.00 ab
V ₂ B ₃	2.55 c	1.15 b	0.84 c	0.68 b	0.35 b	3.50 bc	4.50 bc	5.20 cd	6.20 c	6.63 c
Level of significance	NS	***	***	*	**	NS	NS	NS	NS	NS
CV%	1.89	2.17	2.99	4.35	7.04	2.95	2.31	1.86	1.61	1.47

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents

+

221

1

$V_1 = Langra$	B ₁ =250 ppm of BDF solution	* indicates at 5% level
$V_2 = Khirshapat$	B ₂ =500 ppm of BDF solution	** indicate at 1 % level
B ₀ =Control	$B_3 = 750 \text{ ppm of BDF solution}$	*** indicate at 0.1% level

NS means non significant

2

Ne.

gathering of TSS content was perceived from B_0 and B_1 treatment at 6 and 9th days while, the lowest (12.90 and 14.85%) was noted from B_3 treatment (Fig. 4.8). The results of the present studies are strongly supported by the findings of Dhemre and Waskar (2004). Ahmed and Singh (2000) also found the similar results. These happened possibly due to ripening condition resulting in maximizing TSS accumulation in control and 750 ppm of BDF solution resisted in ethylene synthesis that caused delay ripening and ultimately in lower TSS accumulation. It also revealed that TSS accumulation is strongly related to ripening and it caused falling off owing to decaying.

The combined effect of varieties and applied different doses of BDF solution in connection with TSS content were perceived to be significant at different days after storage except initial and 3^{rd} day (Appendix 4.5). It was exposited an enhancing behavior of TSS content at different days after storage (Table 4.11). The highest accumulation (22.00, 21.80 and 21.60%) was obtained from the treatment combination of V₁B₀, V₁B₁ and V₁B₂ at 6, 9 and 12th days, while; the lowest value (13.20, 14.50 and 17.50%) was notified from the treatment combination of V₂B₃, respectively.

10.3.5 Total sugar content

*

Highly significant variation was manifested between both the varieties means in terms of total sugar content of mango pulp at different days after storage (Appendix 4.5 and Table 4.10). It elucidated that TSC gathered in a continuous stream with expanding of storage duration. This gathering trend was more or less hastily from initial to 9th days in both the varieties, thereafter; it expanded slightly slower. At all days of storage, Khirshapat produced more quantity of TSC than Langra. At initial day, Khirshapat had higher (6.09%) while; Langra provided lower (5.57%). At 12th day, it gave higher quantity (19.64%) and lower (19.07%) was noted in Langra. Upadhyay and Tripathi (1985) reported that total sugar content was expanded gradually, when stored for 6 days at room temperature. Sugar content increased during ripening (Srivastava, 1967). These results are in conformity with the findings of Shahjahan *et al.* (1994). Tsuda *et al.* (1999) also found the

similar results. The increase in TSC might be possible due to conversion of complex starch or carbohydrate into simple compound.

Different doses of BDF solution subjected to the investigation in connection with total sugar content of mango pulp demonstrated significant variation at different days after storage except initial (Appendix 4.5). At different days, the results found that TSC increased hastily with expanding of storage period (Fig. 4.9). The increasing trend was very swift in untreated mango followed by other treatments *viz*. B_1 , B_2 and B_3 treatment, respectively. The highest quantity of TSC (21.06 and 21.36%) was obtained from control and B_1 treated mangoes at 9 and 12th days, while; the lowest (12.40% and 15.70%) was reported from B_3 treatment. The results of the present investigation are in conformity with the reports of Dhemre and Waskar (2004). The enhancing trend of total sugar at untreated mangoes might be perhaps due to breaking down of complex carbohydrate into simple compound but, B_3 treatment made delay ripening at storage period.

The combined effect of varieties and implied different doses of BDF solution in this study in terms of total sugar content of mango pulp exhibited non significant variation at different days after storage (Appendix 4.5). The results indicated that total sugar content accumulated progressively with the advancement of storage period (Table 4.11). At 9th day, the maximum (21.31%) quantity of TSC was formed from the treatment combination of V₂B₀ while, the minimum (12.10%) was formed from the treatment combination of V₁B₃, respectively (Table 4.11).

10.3.6 Reducing sugar content

a

Analysis of variance showed significant effect on reducing sugar content of mango pulp at different days after storage except at 6th day (Appendix 4.6). The results observed an expanding trend of reducing sugar with the rising of storage period (Table 4.12). It also annotated that Khirshapat was better in gathering of reducing sugar than Langra at different days of storage. Higher (5.47%) quantity of this sugar was observed from Khirshapat while; lower (5.20%) was noted from Langra at 12th day of storage (Table 4.12). These results are in agreement with the findings of Upadhyay and Tripathi (1985). Casttrillo *et al.* (1992) stated that reducing sugar content was augmented during storage period. Khirshapat providing

more reducing sugar might be possibly due to genetical variation in both the varieties.

Different doses of BDF subjected to this study were perceived to be significant in respect of reducing sugar content of mango pulp at different storage period except initial (Appendix 4.6). The results explained that reducing sugar of mango pulp was increased progressively at different days of storage. It also stated that untreated mangoes were better in forming of reducing sugar as compared to the other treatments. Control treatment was considered as more progressive producer of reducing sugar up to 9th day and then, it came down owing to starting decay. At 12th day, the maximum (6.27%) amount of reducing sugar was obtained from B₁ treatment and the lowest (4.32%) was obtained at B₃ treatment. The results of the present study are in conformity with the findings of Dhemre and Waskar (2004). Lower increasing trend of reducing sugar content treated with B₃ treatment might be possibly due to delay ripening that resulted in lesser conversion of carbohydrates into simple's molecules.

The combined effect of varieties and different doses of BDF solution of mango pulp demonstrated non significant variation in terms of reducing sugar content of mango pulp at different days after storage (Appendix 4.6). The results exposited that reducing sugar content grew up progressively at three days interval up to 9th day thereafter, it came down from the treatment combination of V₂B₀ (Table 4.13). At 9th day, the highest (6.44%) quantity was noticed from the treatment combination of V₂B₀ and the lowest (3.90%) was noticed from V₁B₃.

10.3.7 Non reducing sugar content

*

The variation between the varieties means demonstrated highly significant in respect of non-reducing sugar content at different days after storage (Appendix 4.6). The results were noticed as an enhancing trend of non reducing sugar content at different days of storage. At all the days, it revealed that Khirshapat was much better than Langra in receiving of non reducing sugar content (Table 4.12). At 12th day, higher (14.20%) amount of non reducing sugar was recorded from Khirshapat and lower (13.88%) amount was recorded from Langra. These results are in conformity with the report of Ali and Mazhar (1960). They reported that non

reducing sugar content of ripe fruits was 11.20%. The results obtained from the investigation might be possible due to varietals dissimilarities.

Different doses of BDF solution imposed to this trial were noticed to be significant variation in connection with non reducing sugar content of mango pulp at different days after storage except at initial stage (Appendix 4.6). The results stated that non reducing sugar content of mango pulp was formed progressively at various days. It denoted that untreated fruits were noticed better in achieving of more quantity of non reducing sugar followed by the other treatments. This increasing trend was markedly up to 9th day and thereafter, it increased slowly owing to becoming hackneyed. Lower rising trend was perceived from the fruit treated with B₃ treatment. At 12th day, the highest result (15.28%) was recorded from control and lowest value (11.39%) was recorded from B₃ treatment (Table 4.12). These events might be probably due to B₃ treatment retarded ethylene synthesis of mango pulp resulting in delay ripening and little amount of non reducing sugar achieving. These results are in agreement with the statement of Ahmed and Singh (2000).

The combined effect of varieties and different doses of BDF solution exhibited non significant in terms of non reducing sugar content of mango pulp at different days after storage (Appendix 3.6). The results were exposited a growing up trend of non reducing sugar from different treatment combination at various days (Table 3.13). At 9th day, the highest (14.87%) quantity of non reducing sugar was notified from the treatment combination of V₂B₀ while; the lowest (8.42%) was notified from the treatment combination of V₁B₃, respectively.

10.3.8 Crude fibre content

*

The analysis of variance of mango varieties subjected to this investigation was perceived to be significant in respect of crude fiber content of mango pulp at different days after storage (Appendix 4.7). The results denoted that a growing up trend of crude fibre was noticed from both the varieties at different days of storage. It stated that Langra was found better in accumulation of crude fiber content. At initial day, Langra gave higher (1.22%) quantity of crude fibre as compared to Khirshapat (1.12%). The quantity of crude fibre came down in a continuous stream with the increase of storage period from both the varieties. At 12th day, higher

(0.51%) quantity of crude fiber was obtained from Langra while; lower (0.39%) was obtained from Khirshapat (Table 4.14). There were no available research findings in connection with crude fibre content of mango pulp in the scientific literature. The falling off of crude fibre content with the rising of storage period might be probable due to metabolic activities resulting in hydrolysis of cellulose and lignin into simple molecules.

Various doses of BDF solution used in this trial were observed to be significant on crude fiber content of mango pulp at different days after storage except initial day (Appendix 4.7). The results elucidated that crude fibre content decreased successively with the advancement of storage period (Fig. 4.10). At all days of storage, it was noticed that crude fibre content was comparatively more at the fruit treated with B₃ treatment followed by the other treatments. At 3rd day, the maximum (1.07%) amount of crude fiber was recorded from B₃ treatment which was statistically at par with B₂ treatment and the lowest (0.77%) was recorded from B₀ treatment (Table 4.14). At 12th day, the maximum (0.60%) was obtained from B₃ treatment while, the lowest (0.38%) was obtained from B₀ treatment which was also statistically at par to B₁ and B₂ treatment (Table 4.14). The results of the present study are partially supported by the findings of Mathooko (2000).The diminishing trend of crude fiber influenced by B₃ treatment might be possible due to retardation to ripening resulting in lower declining trend of crude fiber content during storage.

The combined effect of varieties and different doses of BDF solution were observed to be non significant variation in terms of crude fiber content of mango pulp at different days after storage (Appendix 4.7). The results annotated that crude fibre content came down continuously with the rising of storage period (Table 4.15). It denoted that the treatment combination of V₁B₃ was better in accumulation of crude fibre content at all storage period. At 6th day, the highest (1.00%) quantity of crude fibre was reported from the treatment combination of V₁B₃ and the lowest (0.39%) was reported from the treatment combination of V₂B₀, respectively.

10.3.9 Total lipid content

The analysis of variance of varieties showed non significant variation in terms of lipid content in mango at different days after storage except 9th day (Appendix

4.7). The results were noticed an increasing trend of lipid content in mango pulp with the advancement of storage period. At all days of storage, Langra was perceived higher in forming lipid content than Khirshapat. At 9th day, higher (0.60%) amount of lipid was obtained from Langra and lower (0.58%) was obtained from Khirshapat (Table 4.14). These occurrences might be possible due to genetically dissimilarities between two varieties.

Different doses of BDF solution to this study in terms of lipid content were observed to be highly significant at different days after storage (Appendix 4.7). The results were exposited an increasing trend of lipid content with the rising of storage period at various days of storage (Fig. 4.11). It illustrated that control treatment received comparatively higher amount of lipid followed by B₁, B₂ and B₃ treatments from initial to 9th days of storage; and then, it came down due to starting decay. At 9th day, the highest (0.76%) amount of lipid was derived from control whereas; the lowest (0.44%) was derived from the fruit treated with B₃ treatment. At 12th day, B₂ treated fruit produced the highest (0.71%) amount of lipid which was statistically at par with B₁ and the lowest (0.57%) was recorded from the fruit treated with B₃ treatment which was also statistically at par with control. These occurrences might be possible due to B₃ treatment caused delay ripening that resulted in lower production of lipid content and keeping the quality good. These results are partially supported by the statement of Bandyoppadhyay and Nair (1990).

The interaction effects of varieties and implied different doses of BDF solution in this trial showed significant variation in terms of lipid content of mango pulp at different days after storage except 9th day (Appendix 4.7). The results indicated that a slightly growing up trend of lipid content in mango pulp was noticed with the advancement of storage duration (Table 4.15). It revealed that the treatment combination of V₁B₀ was found as more producers (0.78%) of lipid at initial to 9th day and then, it fell off due to starting bad fruits situation. At this time, lower (0.43%) quantity was noticed from the treatment combination of V₂B₃.

10.3.10 Water soluble protein content

Y

The analysis of variance of mango varieties was perceived to be significant variation in connection with WSPC in mango at different days after storage except 3,

7

9 and 12th day (Appendix 4.8). The results elucidated an increasing trend of WSPC with the increase of storage duration. It denoted that Langra was better in WSPC accumulation as compared to Khirshapat at all the stages of storage. At 6th day, higher (0.91%) synthesis of WSPC was derived in Langra while; lower (0.79%) was recorded in Khirshapat (Table 4.16). Anon (1962) suggested that protein content might be differed from cultivar to cultivar and stages of maturity as well as ripening. These events was possible due to some of seed protein of mango might have to be disseminated to pulp portion through complex metabolic activities during ripening and also might have the genetical variation between the varieties.

Different doses of BDF solution implied to this investigation in terms of WSPC demonstrated significant variation at different days after storage except initial and 12^{th} day (Appendix 4.8). Various results of WSPC were found an augmenting trend in mango pulp with the increase of storage period (Fig. 4.12). It indicated that WSPC was gathered more in untreated fruit followed by the fruit treated with B₁, B₂ and B₃, treatments. The growing up trend of WSPC in control was hastily from initial to 6th days thereafter; it declined due to starting decay. At the same time, the increasing trend from the fruit treated with B₃ treatment was very slow due to delay ripening. At 9th day, the highest (1.23%) gathering of WSPC was notified from control which was statistically at par with B₁ treatment and the lowest (0.74%) was notified from B₃ treatment which was also statistical at par with B₂ treatment. It elucidated that extension of WSPC synthesis was strongly depended upon fruit ripening during storage. Lakshminarayana (1980) reported that mango fruits contained different quantity of protein during storage period. Peter *et al.* (2007) recommended that that protein content increased with the advances of storage duration.

The combined effect of varieties and imposed different doses of BDF solution in this experiment exhibited non significant variation in respect of WSPC at different days after storage (Appendix 4.8). The results reported from the studies annotated that WSPC increased in a continuous stream from initial to 9th days from the treatment combination of V₁B₀ then, it abated due to starting decomposition while, the lowest increasing trend was observed from the treatment combination of V₂B₃. At 9th day, the highest (1.25%) quantity of WSPC was recorded from the treatment

*

7





Fig. 4.9 Effect of different doses of BDF on total sugar content of mango pulp at different days after storage. Vertical bars represent LSD at 0.05 level

Fig. 4.10 Crude fibre content of mango pulp as influenced by different doses of BDF at different days after storage. Vertical bars represent LSD at 0.05 level





Fig. 4.11 Lipid content of mango pulp as influenced by different doses of BDF at different days after storage. Vertical bars represent LSD at 0.05 level

Graphs represent $B_0 = Control$ $B_1=250 \text{ ppm of BDF solution}$ Fig. 4.12 Effect of different doses of BDF on water soluble protein content of mango pulp at different days after storage. Vertical bars represent LSD at 0.05 level

 $B_2=500$ ppm of BDF solution $B_3=750$ ppm of BDF solution

Treatments		TSS conte	ent (%) at d	ifferent days	5		Total sugar	content (%)	at different da	ays
Variety (V)	Initial	3	6	9	12	Initial	3	6	9	12
V ₁	7.73 b	10.60 b	15.75 a	17.90 a	18.65 a	5.57 b	8.39 b	13.29 b	16.70 b	19.07 b
V ₂	8.73 a	11.58 a	15.53 b	17.00 b	17.58 b	6.09 a	8.95 a	13.85 a	17.26 a	19.64 a
evel of significance	***	***	**	***	***	***	***	***	***	***

Table 4.10 Pattern of total soluble solid and total sugar content of postharvest mango pulp between varieties during storage at ambient condition

Table 4.11 Combined effects of varieties and different doses of Bavistin DF solution on total soluble solid and total sugar content of postharvest mango pulp at ambient condition

Treatments combination		TSS conte	ent (%) at d	ifferent days	Total sugar content (%) at different days						
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12	
V ₁ B ₀	8.00 c	12.50 b	22.00 a	19.00 c	16.00 g	5.65 b	10.37 b	18.80 b	20.82 b	19.83 e	
V ₁ B ₁	7.80 cd	10.80 d	15.80 c	21.80 a	18.80 c	5.58 b	8.11 e	13.79 d	19.09 d	21.09 b	
V ₁ B ₂	7.60 d	9,60 q	12.60 g	15.60 e	21.60 a	5.53 b	7.78 f	11.08 f	14.78 f	19.98 e	
V ₁ B ₃	7.50 d	9.50 g	12.60 g	15.20 f	18.20 d	5.50 b	7.30 g	9.50 h	12.10 h	15.40 g	
V ₂ B ₀	9.00 a	13.50 a	20.50 b	17.50 d	14.50 h	6.15 a	10.87 a	19.29 a	21.31 a	20.32 d	
V ₂ B ₁	8.80 ab	11.80 c	14.80 d	20.80 b	17.80 e	6.10 a	8.63 c	14.31 c	19.61 c	21.63 a	
V ₂ B ₂	8.60 b	10.60 e	13.60 e	15.20 f	20.50 b	6.00 a	8.40 d	11.70 e	15.40 e	20.60 c	
V ₂ B ₃	8.50 b	10.40 f	13.20 f	14.50 g	17.50 f	6.10 a	7.90 f	10.10 g	12.70 g	16.00 f	
Level of significance	NS	NS	***	***	***	NS	NS	NS	NS	NS	
CV%	2.03	0.96	1.14	0.81	0.68	1.82	1.22	0.78	0.62	0.55	

In a column values having the same letter(s) do not differ significant as per DMRT at 5% level

Table represents $V_1 = Langra$ $B_1 = 250$ ppm of BDF solution* indicates at 5% level $V_2 = Khirshapat$ $B_2 = 500$ ppm of BDF solution** indicate at 1 % level $B_0 = Control$ $B_3 = 750$ ppm of BDF solution*** indicate at 0.1% level

NS means non significant

1

41

Treatments	Reducing sugar content (%) at different days						Non-reducing sugar content (%) at different days					
Variety (V)	Initial	3	6	9	12	Initial	3	6	9	12		
Vi	1.45 b	2.17 b	3.82 b	5.17 b	5.20 b	4.13 b	6.22 b	9.47 b	11.53 b	13.88 b		
V ₂	1.72 a	2.46 a	4.11 a	5.44 a	5.47 a	4.32 a	6.49 a	9.74 a	11.82 a	14.20 a		
Level of significance	**	***	***	***	***	***	***	***	***	***		
Treatments (B)												
Bo	1.63	3.10 a	6.10 a	6.30 a	4.80 c	4.28	7.53 a	12.95 a	14.77 a	15.28 a		
Bı	1.62	2.20 b	3.90 b	6.15 b	6.27 a	4.24	6.17 b	10.15 b	13.20 b	15.09 b		
B ₂	1.56	2.05 c	3.25 c	4.95 c	5.95 b	4.21	6.04 c	8.14 c	10.14 c	14.39 c		
B ₃	1.53	1.92 d	2.62 d	3.82 d	4.32 d	4.18	5.69 d	7.19 d	8.59 d	11.39 d		
Level of significance	NS	***	***	***	***	NS	***	***	***	***		

Table 4.12 Behavior of reducing and non reducing sugar content of postharvest mango pulp in varieties and influenced by different doses of Bavistin DF solution during storage condition

7

Table 4.13 Combined effects of varieties and different doses of Bavistin DF solution on reducing and non reducing sugar content of postharvest mango pulp at ambient condition

Treatments combination		Reducing sugar	r content (%) at different d	lays	Non	-reducing sug	gar content (%) at differer	nt days
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12
V ₁ B ₀	1.48 bc	2.95 b	5.95 b	6.15 bc	4.65 e	4.17 bc	7.42 b	12.85 b	14.67 b	15.18 b
V1B1	1.48 bc	2.05 de	3.75 d	6.05 c	6.17 b	4.13 c	6.06 de	10.04 d	13.04 d	14.92 c
V ₁ B ₂	1.42 c	1.90 ef	3.10 f	4.80 e	5.80 c	4.11 c	5.88 ef	7.98 f	9.98 f	14.18 e
V ₁ B ₃	1.40 c	1.78 f	2.48 h	3.68 g	4.18 g	4.10 c	5.52 g	7.02 h	8.42 h	11.22 g
V ₂ B ₀	1.77 a	3.24 a	6.24 a	6.44 a	4.94 d	4.38 a	7.63 a	13.05 a	14.87 a	15.38 a
V ₂ B ₁	1.75 a	2.35 c	4.05 c	6.25 b	6.37 a	4.35 ab	6.28 c	10.26 c	13.36 c	15.26 ab
V ₂ B ₂	1.70 a	2.20 cd	3.40 e	5.10 d	6.10 b	4.30 a-c	6.20 cd	8.30 e	10.30 e	14.60 d
V2B3	1.65 ab	2.05 de	2.75 g	3.95 f	4.45 f	4.25 a-c	5.85 f	7.35 g	8.75 g	11.55 f
Level of significance	NS	NS	NS	NS	NS	NS	NS	NS	NS	NS
CV%	6.71	4.58	2.68	2.00	1.99	2.51	1.67	1.10	0.91	0.76

In a column values having the same letter(s) do not differ significantly as per DMRT at 5 %level

*

Table represents

V ₁ = Langra	B ₁ =25 Oppm of BDF solution	* indicates at 5 %level
$V_2 = Khirshapat$	B ₂ =5 00ppm of BDF solution	** indicate at 1 % level
B₀ =Control	$B_3 = 75$ Oppm of BDF solution	*** indicate at 0.1% level

NS means non significant

4

4

ź.

Treatments		Crude fibre (%) at different days					Lipid content (%) at different days						
Variety (V)	Initial	3	6	9	12	Initial	3	6	9	12			
V1	1.22 a	1.01 a	0.77 a	0.65 a	0.51 a	0.18	0.31	0.48	0.60 a	0.64			
Vz	1.12 b	0.90 b	0.69 b	0.53 b	0.39 b	0.17	0.30	0.46	0.58 b	0.63			
Level of significance	***	***	**	***	***	NS	NS	NS	***	NS			

4

Table 4.14 Behavior of crude fibre and lipid content of postharvest mango pulp in varieties during storage environments at ambient condition

¥

Table 4.15 Combined effects of varieties and different doses of BDF solution on crude fibre and lipid content of postharvest mango pulp during storage at ambient condition

Treatments combination		Crude fi	bre (%) at di	fferent days			Lipid content (%) at different days					
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12		
V ₁ B ₀	1.20 a-d	0.84 d	0.49 d	0.46 de	0.44 cd	0.20 a	0.40 a	0.69 a	0.78 a	0.63 ab		
V ₁ B ₁	1.21 a-c	1.01 a-c	0.71 c	0.51 cd	0.46 b-d	0.18 ab	0.33 b	0.48 b	0.68 c	0.64 ab		
V ₁ B ₂	1.23 ab	1.08 a	0.88 b	0.73 b	0.49 bc	0.16 b	0.26 c	0.36 c	0.51 d	0.71 a		
V ₁ B ₃	1.25 a	1.10 a	1.00 a	0.90 a	0.65 a	0.16 b	0.25 c	0.34 c	0.44 e	0.58 b		
V ₂ B ₀	1.10 d	0.70 e	0.39 e	0.36 e	0.32 e	0.19 ab	0.39 a	0.67 a	0.73 b	0.58 b		
V ₂ B ₁	1.11 cd	0.91 cd	0.71 c	0.38 e	0.32 e	0.17 ab	0.32 b	0.47 b	0.67 c	0.70 a		
V ₂ B ₂	1.12 cd	0.97 bc	0.77 c	0.61 c	0.36 de	0.16 b	0.25 c	0.35 c	0.50 d	0.70 a		
V ₂ B ₃	1.13 b-d	1.03 ab	0.88 b	0.75 b	0.55 b	0.16 b	0.24 c	0.33 c	0.43 e	0.56 b		
Level of significance	NS	NS	NS	NS	NS	NS	NS	NS	**	NS		
CV%	4.54	5.42	7.28	10.49	11.82	6.05	6.15	3.19	1.79	6.43		

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents

$V_1 = Langra$	B ₁ =250 ppm of BDF solution	 indicates at 5% level 	NS means non significant
$V_2 = Khirshapat$	B ₂ =500 ppm of BDF solution	** indicate at 1 % level	
B ₀ =Control	B ₃ =750 ppm of BDF solution	*** indicate at 0.1% level	

232

+

+

combination of V_1B_0 which was statistically at par with V_1B_1 , V_2B_0 and V_2B_1 whereas; the lowest (0.70%) was recorded from the treatment combination of V_2B_3 , which was also statistically at par with V_1B_3 and V_2B_2 (Table 4.17).

10.4 Changes in mineral contents of mango during storage environments

The data obtained from different behavior of mineral contents during storage period of mango fruits are presented and fairly interpreted in the following subheadings.

10.4.1 Phosphorus content

.Xe

T

Analysis of variance of varieties in connection with P contents showed highly significant at different days after storage (Appendix 4.8). The results gave a signal that P content increased smoothly in both the varieties with the expansion of storage period (Table 4.16). It was observed that Langra produced more quantity of P as compared to Khirshapat. The increasing trend of P in Langra was more or less similar up to 9th day thereafter, it abated due to becoming spoilage of fruits. At 12th day, higher (23.60 mg/100 g) quantity of P was recorded in Langra and lower (21.62 mg/100 g) was recorded in Khirshapat. Anon (1962) elucidated that P content of mango pulp might be differed from cultivar to cultivar. Better performance in increasing trend of P content in Langra might be probably due to genetical dissimilarities. It also revealed that the expansion of P content in mango was intimately associated with ripening during storage. The data of the present study are in partially supported by the findings of Nadkarni (1963) when he worked with 16 cultivars and found the ranged between 10-30 mg/100 g among the cultivars.

Different doses of BDF solution subjected to this trial in relation to P content exhibited highly significant variation at different days after storage (Appendix 4.8). The results narrated that P content increased steadily at various days of storage. It denoted that P content in control augmented at slow rate from initial to 6th days thereafter; it augmented at very slow motion than its previous trend. After 9th day, it declined possibly due to decomposition of fruits (Fig. 4.13). At 9th day, the maximum (23.43 mg/100 g) quantity of P content was obtained from the fruit treated with control and the minimum (20.40 mg/100 g) was obtained from the fruit treated with

 B_3 treatment. Lower amount of P content from the fruit treated with B_3 treatment might be probably due to delay ripening that resulted in lower accumulation of P content. Peter *et al.* (2007) stated that P content was increasingly changed during storage. Watt and Merrill (1963) found 13.00 mg per 100 g of fresh green mango pulp.

The combined effect of varieties and implied various doses of BDF solution demonstrated non significant variation on P content of mango pulp at different days after storage except 12^{th} day (Appendix 4.8). The results found a growing up trend of P content with the rising of storage period (Table 4.17). It annotated that P content increased gradually from initial to 9^{th} day with the treatment combination of V₁B₀ and then, it fell off due to bad fruit situation. In this time, P content was noticed lower increasing trend from the fruit treated with the treatment combination of V₂B₃. At 12^{th} day, the highest (24.45 mg/100 g) quantity of P was obtained from the treatment combination of V₁B₁ whereas; the lowest (20.66 mg/100 g) was obtained from the treatment was regarded as the best treatment combination in keeping the quality good in preservation followed by the other treatment combinations. This treatment combination strictly impeded the ripening of mango fruits during storage.

10.4.2 Potassium content

*

x

Ve

MA

Adopted varieties in terms of K content demonstrated highly significant at different days after storage (Appendix 4.9). The results explored a rising trend of K content was observed in both the varieties with the increase of storage period at different days (Table 4.18). In all the storage period, Langra was manifested more producer of K content comparing to Khirshapat. It also annotated that the increase trend of K content was suspended at 9th day. At this period, higher (0.28%) amount of K was recorded from Langra and lower (0.26%) was recorded from Khirshapat (Table 4.18). These occurrences might be possible due to deterioration of fruits that resulted in lower metabolic activities. Langra was better accumulator of K content storage

14

>

4

7-

period, K content increased might be possible due to transmission of K from stone and peel to pulp of mango during high metabolic activity.

Different doses of BDF solution imposed to this experiment in terms of K content of mango pulp were perceived to be significant at different days after storage (Appendix 4.9). The results exposited that K content accumulated in a continuous stream with the advancement of storage period, but, K content from the untreated fruit came down after 9th day, while other treatments viz., B₁, B₂ and B₃ treatments retained their increasing behavior (Fig. 4.14). In this period, it noticed a very lower extending trend of K content in the fruit treated with B₃ treatment. At 9th day, the maximum (0.29%) of K content was recorded from the untreated fruit but it was statistically at par with B_1 and B_2 treatment while; the lowest (0.25%) was recorded from the fruit treated with B₃ treatment which was also statistically at par with B_1 and B_2 treatment. Lower quantity of K in the fruit treated with B_3 treatment might be possible due to delay ripening that caused lower transmission of K with good keeping quality. Peter et al. (2007) showed that K content was increased during storage. Watt and Merrill (1963) reported 189 mg K per 100 g of fresh green mango. The results of the present investigation interpreted that K content of mango pulp was grown up slightly in the storage period, but, B_3 treatment interrupted its growing up trend during storage period.

The combined effect of varieties and imposed different doses of BDF solution in relation to K content of mango pulp demonstrated non significant variation at different days after storage (Appendix 4.9). The results denoted that K content of mango pulp was increased gradually from different treatment combination with the expansion of storage time (Table 4.19). But, only the treatment combination of V₁B₀ produced the higher content of K up to 9th day. In this period, the highest (0.30%) quantity was reported from the treatment combination of V₁B₀ which was statistically at par with V₁B₁, V₁B₂, V₁B₃, V₂B₀, and V₂B₁, whereas, the lowest (0.24%) was reported from the treatment combination of V₂B₃ which was also statistically at par with V₂B₂.

3

1

10.4.3 Calcium content

Analysis of variance of induced varieties to this investigation in connection with Ca content demonstrated highly significant variation at different days after storage (Appendix 4.9). The results were perceived an increasing trend of Ca content with the extending of storage time from both the varieties (Table 4.18). It also narrated that Khirshapat was better in accumulation of Ca content as compared to Langra. At 12th day, higher (22.98 mg/100 g) quantity of Ca was recorded in Khirshapat and lower (21.25 mg/100 g) was recorded in Langra. These occurrences might be possible due to genetical variation between two varieties. These results are in partially supported by the findings of Nadkarni (1963). Calcium content differed from cultivar to cultivar during storage period was reported by Anon (1962).

Different doses of BDF solution on Ca content of mango pulp demonstrated significant variation at different days after storage (Appendix 4.9). At various days of storage, the results of Ca content extended in a continuous stream with the passing of storage time (Fig. 4.15). It also stated that Ca content of control extended sharply from initial to 6th day and then, it extended smoothly and thereafter, it declined due to starting of decomposition. At the same time, Ca content from the fruit treated with B₃ treatment gathered very smoothly. At 9th day, the highest (24.76 mg/100 g) quantity of Ca was recorded from untreated fruit while; the lowest (16.69 mg/100 g) was recorded from the fruit treated with B₃ treatment (Table 4.18). These phenomena caused by B₃ treatment might be possible due to delay ripening that caused in lower dissemination of Ca content from peel and stone to pulp of mango. Increased form of Ca content of mango pulp was stated by Anon (1963) and reported to be 10.00 mg of Ca per 100 g of fresh green mango pulp. But, these results are not supported by the findings of Peter *et al.* (2007).

The combined effect of varieties and different doses of BDF solution in terms of Ca content of mango pulp had significant variation at different days after storage except 6th day (Appendix 4.9). The results of Ca content were noticed an increasing trend from various treatment combinations (Table 4.19). At 9th day, the highest (25.58 mg/100 g) quantity of Ca was observed from the treatment combination of V₂B₀ and the lowest (15.94 mg/100 g) was observed from the treatment combination

De

y

+

of V_1B_3 . The results of the present investigation annotated that the treatment combination of V_1B_3 was better in mango preservation.

10.4.4 Magnesium content

Highly significant variation was observed in terms of Mg content between the varieties means at different days after storage (Appendix 4.10). The results indicated that Mg content augmented gradually with the advancement of storage period (Table 4.20). It also denoted that Mg content extended steadily of initial to 9 days thereafter; it came down slightly due to starting decay of the fruit. At 9th day, higher (18.08 mg/100 g) was derived from Langra and lower (17.11 mg/100 g) was derived from Khirshapat. These phenomena might be probably due to genetical dissimilarities between the varieties. There were no available research findings in connection with Mg content during storage in the scientific literature. But, the data of the present study revealed that Langra contained more Mg over Khirshapat.

Different doses of BDF solution used in this trial exhibited highly significant on Mg content of mango pulp at different days after storage (Appendix 4.10). The results indicated a smooth gathering trend of Mg content of mango pulp with the extension of storage period (Fig. 4.16). It was also noticed that Mg content of control extended gradually from initial to 6 days and then, it abated very sharply. The highest (18.72 and 18.00 mg/100 g) quantity of Mg was obtained from control and B₁ treatment at 6 and 9th day while, the lowest (16.83 and 17.18 mg/100 g) was obtained from the fruit treated with B₃ treatment, respectively. The increasing tendency of Mg content in mango pulp during storage period might have been related to starting of ripening. This result is very much similar to the report of Peter *et al.* (2007).

The combined effect of varieties and various doses of BDF solution were observed to be non significant variation in connection with Mg content at various days after storage (Appendix 4.10). The results were noticed an extending trend of Mg content of mango pulp with the growing up of storage time (Table 4.21). At 6th day, the highest (19.15 mg/100 g) quantity of Mg was manifested from the treatment combination of V₁B₀ while; the lowest (16.25 mg/100 g) was reported from the treatment combination of V₂B₃.

+

1

3

F





Fig. 4.13 Effect of different doses of BDF on Phosphorus content of mango pulp at different days after storage. Vertical bars represent LSD at 0.05 level



Fig. 4.14 Potassium content of mango pulp as influenced by different doses of BDF at different days after storage. Vertical bars represent LSD at 0.05 level



Fig. 4.15 Calcium content of mango pulp as influenced by different doses of BDF at different days after storage. Vertical bars represent LSD at 0.05 level

Graphs represent $B_0 = Control$ $B_1=250 \text{ ppm of BDF solution}$ Fig. 4.16 Effect of different doses of BDF on Magnesium content of mango pulp at different days after storage. Vertical bars represent LSD at 0.05 level

 $B_2=500$ ppm of BDF solution $B_3=750$ ppm of BDF solution to

239

4

1

*

4

 (\mathbf{x}_{i})

+

Table 4.16 Changes of water soluble protein and phosphorus content of postharvest mango pulp in varieties during storage environments at ambient condition

Treatments	Water	soluble pro	otein content	(%) at diff	ferent days		Phosphorus (mg/100 g) at different days					
Variety (V)	Initial	3	6	9	12	Initial	3	6	9	12		
V1	0.56 a	0.72	0.91 a	1.03	1.21	19.06 a	20.24 a	21.78 a	23.16 a	23.60 a		
V2	0.48 b	0.63	0.79 b	0.95	1.15	16.75 b	17.94 b	19.38 b	20.85 b	21.62 b		
Level of significance	*	NS	*	NS	NS	***	***	***	***	***		

Table 4.17 Combined effects of varieties and different doses of BDF solution on water soluble protein and phosphorus content of postharvest mango pulp at ambient condition

Treatments combination	Water	r soluble prot	ein content (%) at differe	nt days	P	hosphorus (mg/100 g)	0 g) at different days		
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12	
V ₁ B ₀	0.58	0.87 a	1.22 a	1.25 a	1.22	19.65 a	21.75 a	24.26 a	24.59 a	23.12 d	
V ₁ B ₁	0.59	0.70 a-c	0.85 b	1.20 a	1.25	19.25 b	20.27 b	21.78 b	24.15 b	24.45 a	
V ₁ B ₂	0.55	0.68 a-c	0.80 bc	0.90 b	1.22	18.78 c	19.69 c	20.83 c	22.34 c	23.95 b	
V ₁ B ₃	0.53	0.64 bc	0.75 bc	0.78 bc	1.15	18.55 d	19.26 d	20.23 d	21.55 e	22.86 e	
V ₂ B ₀	0.49	0.78 ab	1.15 a	1.21 a	1.18	17.25 e	19.38 d	21.87 b	22.28 c	20.78 g	
V ₂ B ₁	0.50	0.62 bc	0.72 bc	1.10 a	1.18	16.89 f	17.92 e	19.43 e	21.93 d	23.54 c	
V ₂ B ₂	0.45	0.58 bc	0.68 bc	0.78 bc	1.14	16.50 g	17.42 f	18.38 f	19.92 f	21.48 f	
V ₂ B ₃	0.48	0.54 c	0.62 c	0.70 c	1.10	16.35 g	17.05 g	17.85 g	19.25 g	20.66 g	
Level of significance	NS	NS	NS	NS	NS	NS	NS	NS	NS	***	
CV%	14.35	15.68	12.50	10.71	8.99	0.59	0.56	0.52	0.52	0.47	

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents

$ \begin{array}{ll} V_1 = Langra \\ V_2 = Khirshapat \\ B_0 = Control \end{array} \begin{array}{ll} B_1 = 250 \mbox{ ppm of BDF solution} \\ B_2 = 500 \mbox{ ppm of BDF solution} \\ B_3 = 750 \mbox{ ppm of BDF solution} \end{array} $	 indicates at 5% level indicate at 1 % level indicate at 0.1% level 	NS means non significant
---	--	--------------------------

Treatments		Potassium c	ontent (%)	at different	days	Cal	cium conte	nt (mg/100 g	g) at differer	nt days
Variety (V)	Initial	3	6	9	12	Initial	3	6	9	12
V ₁	0.23 a	0.24 a	0.26 a	0.28 a	0.28 a	10.74 b	13.29 b	16. 6 3 b	19.92 b	21.25 b
V ₂	0.19 b	0.22 b	0.24 b	0.26 b	0.26 b	12.39 a	15.01 a	18.44 a	21.64 a	22.98 a
Level of significance	***	***	***	***	***	***	***	***	***	***

X

41

4

Table 4.18 Changes of potassium and calcium content of postharvest mango pulp between varieties during storage environments at ambient condition

*

Table 4.19 Combined effects of varieties and different doses of BDF solution on potassium and calcium content of postharvest mango pulp during storage at ambient condition

Treatments combination	F	otassium cor	ntent (%) at	different da	ays	Calci	Calcium content (mg/100 g) at different days					
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12		
V ₁ B ₀	0.23 ab	0.26 a	0.28 a	0.30 a	0.28 ab	10.75 d	15.48 b	22.32 b	23.93 c	20.32 g		
V ₁ B ₁	0.24 a	0.25 ab	0.26 ab	0.28 ab	0.30 a	10.82 d	13.42 d	16.62 d	22.26 d	23.35 c		
V ₁ B ₂	0.22 a-c	0.23 a-c	0.25 ab	0.27 ab	0.28 ab	10.65 d	12.25 e	14.23 g	17.55 f	22.17 d		
V ₁ B ₃	0.21 a-d	0.22 bc	0.24 ab	0.26 ab	0.27 ab	10.72 d	12.02 f	13.34 h	15.94 g	19.16 h		
V ₂ B ₀	0.20 b-e	0.23 a-c	0.25 ab	0.27 ab	0.25 b	12.58 a	17.15 a	23.98 a	25.58 a	21.98 e		
V ₂ B ₁	0.18 de	0.22 a-c	0.24 ab	0.26 ab	0.27 ab	12.48 ab	15.48 b	18.67 c	24.32 b	25.42 a		
V ₂ B ₂	0.19 с-е	0.21 bc	0.23 b	0.25 b	0.26 ab	12.35 b	13.95 c	15.92 e	19.22 e	23.84 b		
V ₂ B ₃	0.17 e	0.20 c	0.22 b	0.24 b	0.25 b	12.16 c	13.47 d	15.17 f	17.45 f	20.68 f		
Level of significance	NS	NS	NS	NS	NS	*	**	NS	**	**		
CV%	5.17	4.66	4.31	3.98	3.93	0.92	0.75	1.15	0.51	0.48		

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

4

Y

Table represents

$V_1 = Langra$	B ₁ =250 ppm of BDF solution	 indicates at 5% level 	NS means non significant
V ₂ = Khirshapat	B ₂ =500 ppm of BDF solution	** indicate at 1 % level	
Bo =Control	B ₃ =750 ppm of BDF solution	*** indicate at 0.1% level	

240

X

×

Je-

10.4.5 Copper content

Variation between varieties mean in terms of Cu content of mango pulp demonstrated highly significant at different days after storage (Appendix 4.10). The results denoted that Cu content of mango pulp diminished in a continuous stream with the passing of storage period (Table 4.20). It also narrated that Cu content of Khirshapat was higher as compared to Langra. At initial day, higher (0.36 mg/100 g) content of Cu was noted in Khirshapat and lower (0.33 mg/100 g) was noticed in Langra, respectively. At 12th day, Khirshapat gave higher (0.19 mg/100 g), quantity of Cu and lower (0.16 mg/100 g) was noted in Langra (Table 4.20). These happened might be probably due to genetical dissimilarities between the varieties. There were no available research reports on Cu content of mango pulp at the scientific literature. The results of the present studies revealed that green mangoes contained more copper comparing to stored mangoes and Khirshapat was fairly good in copper accumulation than Langra. The present research findings also revealed that Cu content came down gradually during storage period.

Different doses of BDF solution implied to the present study in connection with Cu content of mango pulp exhibited significant variation at different days after storage (Appendix 4.10). The results narrated that Cu content fell off gradually with the advancement of storage period. The diminishing trend from control was higher than the fruit treated with B_1 , B_2 and B_3 , treatments respectively (Fig. 4.17). At 12^{th} day, higher quantity of Cu (0.24 mg/100 g) was noticed from the fruit treated with B_3 treatment but, it was statistically at par with B_2 treatment while; lower (0.11 mg/100 g) was noticed from the from the from the investigation was fairly good over the other treatments in Cu preservation. So, the present findings revealed that B_3 treatment was much better in mango preservation.

The combined effect of varieties and applied different doses of BDF solution in relation to Cu content of mango pulp were found to be non significant variation at various days after storage (Appendix 4.10). The results stated that Cu content in different treatment combination came down with the extension of storage duration. At 12^{th} day, the highest (0.24 mg/100 g) was derived from the treatment combination of V₂B₃, which was statistically at par with V₁B₃, V₂B₂ and V₁B₂ and the

lowest (0.09 mg/100 g) was derived from the treatment combination of V_1B_0 but, it was statistically at par with V_1B_1 and V_2B_2 (Table 4.21).

10.4.6 Iron content

1

*

Ja-

The analysis of variance of varieties in connection with Fe content of mango pulp was observed to be highly significant at different days after storage (Appendix 4.11). The results found an extending trend of Fe content with the advancement of storage duration. The growing up trend of Fe content was more or less similar from initial to 9 days thereafter, it abated slightly. At 12th day, higher (4.72 mg/100 g) quantity of Fe was reported from Langra and lower (3.30 mg/100 g) was reported from Khirshapat (Table 4.22). These results are in partially supported by the findings of Nadkarni (1963). Fe content differed from cultivar to cultivar during storage period was supported by the report of Anon (1963).

Highly significant variation was noticed due to the effect of different doses of BDF solution in connection with Fe content of mango pulp at various days after storage (Appendix 4.11). The results obtained from the study indicated that Fe content was achieved hastily from initial to 6^{th} day and then, it came down significantly at control and similarly B₁ treated fruits abated of Fe content after 9^{th} day. But, Fe content recorded from the other fruits treated with B₂ and B₃ treatment extended successively up to 12^{th} day (Fig. 4.18). At 12^{th} day, the maximum (4.59 mg/100 g) quantity was derived from the fruit treated with B₂ treatment but, it was statistically at par with B₃ while; the lowest (2.89 mg/100 g) was derived from the untreated fruits at 12^{th} day might be due to starting decomposition. On the other hand, B₃ treated fruits were fairly good because of its lower achievement of Fe content with the increase of storage duration.

The combined effect of varieties and different doses of BDF solution were observed to be non significant variation in relation to Fe content of mango pulp at different days after storage except 9 and 12^{th} day (Appendix 4.11). The results narrated that enhancing trend of Fe content was found from initial to 6 days thereafter, it came down very fast from the treatment combination of V₁B₀ while; the lowest trend was found from the treatment combination of V₂B₃ (Table 4.23). At 9th

X

3m-

day, the highest (5.35 mg/100 g) quantity was recorded from the treatment combination of V_1B_1 and the lowest (3.15 mg/100 g) was recorded from the treatment combination of V_2B_2 .

10.4.7 Manganese content

Manganese content of mango pulp was noticed to be differed significantly in both the varieties mean at different days after storage (Appendix 4.11). The results elucidated that Mn content extended gradually with the extension of storage period (Table 4.22). It also stated that Langra performed better in Mn accumulation comparing to Khirshapat. At 9th day, higher (1.20 mg/100 g) quantity of Mn was rnoted in Langra while; lower (1.05 mg/100 g) was noted in Khirshapat (Table 4.22). There were no available research reports in terms of Mn content in the scientific literature. The results of the present research revealed that Langra was fairly good than Khirshapat in Mn accumulation and it extended during storage period.

Different doses of BDF solution were found to be highly significant in connection with Mn content of mango pulp at different days after storage (Appendix 4.11). The results explained that Mn content extended continuously from initial to 6^{th} days, and then it came down sharply in control. On the other hand, it increased from initial to 9^{th} day thereafter; it diminished in the fruit treated with B₁ treatment. At the same time, very lesser augmenting trend of Mn content was perceived from the fruit treated with B₃ treatment (Fig. 4.19). At 12^{th} day, Mn content ranged between 0.89 to 1.33 mg per 100 g of fresh mango pulp. The maximum (1.33 mg/100 g) was recorded from B₂ treated fruits and the minimum (0.89 mg/100 g) was recorded from control. There were no available research findings in terms of Mn content in the scientific literature. But, the data of the present research revealed that B3 treatment showed profound effect in delay ripening which resulted in lesser extending trend of Mn and keeping the quality good in preservation.

The combined effect of varieties and imposed different doses of BDF solution in relation to Mn content of mango pulp were observed to be non significant at various days after storage (Appendix 3.11). The results were exposited that a growing up trend of Mn content was perceived in different treatment combination with the extension of storage duration (Table 4.23). At 9th day, the highest (1.42 mg/100 g) was listed from the treatment combination of V_1B_1 and the lowest (0.87 mg/100 g) was listed from the treatment combination of V_2B_3 .

10.4.8 Zinc content

X

to

X

Variation in respect of Zinc content of mango pulp due to the effect of varieties showed highly significant at different days after storage (Appendix 4.12). The results indicated that Zn content came down markedly with the extension of storage duration from both the varieties (Table 4.24). It was also exposited that Khirshapat was noticed better in Zn content accumulation as compared to Langra. At initial stage, higher (1.47 mg/100 g) was derived from control and lower (1.26 mg/ 100 g) was derived from Langra. Again, at 12th day, higher (0.85 mg/100 g) was noticed from Khirshapat and the lesser (0.73 mg/100 g) was recorded from Langra. There were no available research findings in terms of Zn content in the scientific review. The results of the present studies invented that green mango especially, Khirshapat received more quantity of Zn than Langra, but, it continuously reduced with the increase of storage duration.

Different doses of BDF solution were found to be significant variation in connection with Zn content of mango pulp at different days after storage (Appendix 4.12). The results observed an extending trend of Zn content of mango pulp with the advancement of storage period from the fruit treated with different doses of BDF solution. It also stated that the coming down trend was very high in control and very low in the fruit treated with B₃ treatment (Fig. 4.20). At 12^{th} day, the maximum (0.96 mg/100 g) was noticed from the fruit treated with B₃ treatment and the lowest (0.65 mg/100 g) was noticed from control. Zn content of mango pulp decreased during storage period was possibly due to transmission of Zn from pulp to stone and peel at stored condition or Zn content of mango pulp might have been depressed or suppressed as influenced by metabolic activities during storage.

The combined effect of varieties and subjected to different doses of BDF solution on Zn content of mango pulp demonstrated non significant variation at different days after storage except initial day (Appendix 4.12). An expanding trend of Zn content of mango pulp was recorded from various treatment combinations (Table 4.25). At initial day, the maximum (1.52 mg/100 g)

2

X

M.

p

×



Fig. 4.17 Effect of different doses of BDF on copper content of mango pulp at different days after storage. Vertical bars represent LSD at 0.05 level



Days after storage

Fig. 4.18 Iron content of mango pulp as influenced by different doses of BDF at different days after storage. Vertical bars represent LSD at 0.05 level



Fig. 4.19 Manganese content of mango pulp as influenced by different doses of BDF at different days after storage. Vertical bars represent LSD at 0.05 level

Graphs represent $B_0 = Control$ $B_1=250 \text{ ppm of BDF solution}$

Fig. 4.20 Effect of different doses of BDF on Zinc content of mango pulp at different days after storage. Vertical bars represent LSD at 0.05 level

 B_2 =500 ppm of BDF solution B_3 =750 ppm of BDF solution Results and Discussion

>

×

Treatments	Mag	nesium conte	nt (mg/100	g) at differ	ent days	Copper content (mg/100 g) at different days						
Variety (V)	Initial	3	6	9	12	Initial	3	6	9	12		
V ₁	17.01 a	17.53 a	18.03 a	18.08 a	17.81 a	0.33 b	0.28 b	0.24 b	0.20 b	0.16 b		
™ V₂	16.02 b	16.49 b	17.01 b	17.11 b	16.85 b	0.36 a	0.32 a	0.27 a	0.23 a	0.19 a		
evel of significance	***	***	***	***	***	***	***	***	***	***		

Table 4.20 Pattern of magnesium and copper content of postharvest mango pulp during storage environments at ambient condition

¥.

1

Table 4.21 Combined effects of varieties and different doses of BDF solution on magnesium and copper content of postharvest mango at ambient condition

Treatments combination	Magnesium content (mg/100 g) at different days					Copper content (mg/100 g) at different days				
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12
V ₁ B ₀	17.25 a	18.22 a	19.15 a	18.12 b	17.16 d	0.31 c	0.25 c	0.19 d	0.14 d	0.09 d
V ₁ B ₁	17.15 a	17.55 b	18.05 c	18.55 a	17.55 c	0.32 bc	0.27 bc	0.22 cd	0.17 cd	0.13 cd
V ₁ B ₂	16.75 b	17.23 c	17.52 d	17.92 c	18.42 a	0.34 a-c	0.30 ab	0.26 a-c	0.22 bc	0.19 ab
V ₁ B ₃	16.88 b	17.12 c	17.41 d	17.72 d	18.12 b	0.33 a-c	0.31 ab	0.28 ab	0.25 ab	0.23 a
V ₂ B ₀	16.35 c	17.30 c	18.28 b	17.28 e	16.27 e	0.36 ab	0.30 ab	0.24 bc	0.18 cd	0.13 cd
V ₂ B ₁	16.11 d	16.55 d	17.05 e	17.45 e	16.45 e	0.35 a-c	0.31 ab	0.26 a-c	0.21 bc	0.17 bc
V ₂ B ₂	15.85 e	16.15 e	16.45 f	17.05 f	17.55 c	0.36 ab	0.32 a	0.28 ab	0.24 ab	0.21 ab
V ₂ B ₃	15.75 e	15.95 f	16.25 g	16.65 g	17.12 d	0.37 a	0.34 a	0.31 a	0.28 a	0.24 a
Level of significance	NS	NS	NS	NS	NS	NS	NS	NS	NS	NS
CV%	0.64	0.67	0.61	0.60	0.62	3.10	3.54	4.16	5.02	6.10

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents			
$V_1 = Langra$	B ₁ =250 ppm of BDF solution	* indicates at 5% level	NS means non significant
V ₂ = Khirshapat	B ₂ =500 ppm of BDF solution	** indicate at 1 % level	
B ₀ =Control	$B_3 = 750$ ppm of BDF solution	*** indicate at 0.1% level	

246

¥.

Y

Treatments	1	Iron content	(mg/100 g)	at different	days	Mai	Manganese content (mg/100 g) at different days					
Variety (V)	Initial	3	6	9	12	Initial	3	6	9	12		
V ₁	2.59 a	3.26 a	4.44 a	5.02 a	4.72 a	0.63 a	0.84 a	1.07 a	1.21 a	1.20 a	-	
V ₂	1.70 b	2.32 b	3.50 b	3.80 b	3.30 b	0.47 b	0.68 b	0.91 b	1.05 b	1.05 b		
Level of significance	***	***	***	***	***	***	***	***	***	***		

Table 4.22 Behavior of iron and manganese content of postharvest mango pulp in varieties during storage environments at ambient condition

31

Table 4.23 Combined effects of varieties and different doses of BDF solution on iron and manganese content of postharvest mango pulp at ambient condition

Treatments combination	Iron content (mg/100 g) at different days					Manganese content (mg/100 g) at different days				
Varieties × Treatments	Initial	3	6	9	12	Initial	3	6	9	12
V ₁ B ₀	2.82 a	3.82 a	5.92 a	5.32 a	3.42 c	0.65 a	0.98 a	1.34 a	1.19 c	0.96 e
V ₁ B ₁	2.65 ab	3.35 b	4.45 c	5.35 a	4.64 b	0.64 a	0.87 b	1.12 c	1.42 a	1.24 b
V ₁ B ₂	2.52 bc	3.12 c	3.92 d	5.02 b	5.52 a	0.62 ab	0.78 d	0.97 d	1.21 bc	1.40 a
V ₁ B ₃	2.35 c	2.75 d	3.45 e	4.38 cd	5.31 a	0.60 b	0.72 e	0.86 e	1.02 d	1.19 c
V ₂ B ₀	1.88 d	2.78 d	4.88 b	4.25 d	2.35 d	0.49 c	0.82 c	1.18 b	1.03 d	0.82 f
V ₂ B ₁	1.73 de	2.32 e	3.42 e	4.45 c	3.75 c	0.46 cd	0.69 e	0.94 d	1.25 b	1.07 d
V ₂ B ₂	1.64 ef	2.25 e	3.05 f	3.15 f	3.65 c	0.47 cd	0.63 f	0.82 e	1.05 d	1.25 b
V ₂ B ₃	1.53 f	1.94 f	2.64 g	3.35 e	3.46 c	0.45 d	0.56 g	0.71 f	0.87 e	1.05 d
Level of significance	NS	NS	NS	***	***	NS	NS	NS	NS	NS
CV%	4.96	3.80	2.67	2.41	4.78	3.87	2.81	2.14	1.88	1.89

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents $V_1 = Langra$ $B_1 = 250 \text{ ppm of BDF solution}$ * indicates at 5% level $V_2 = Khirshapat$ $B_2 = 500 \text{ ppm of BDF solution}$ * indicate at 1 % level $B_0 = Control$ $B_3 = 750 \text{ ppm of BDF solution}$ ** indicate at 0.1% level

NS means non significant

X

1

247

¥

+

quantity of Zn was obtained from Khirshapat along with B_3 treatment while; the lowest (1.24 mg/100 g) value was obtained from Langra using no treatment.

10.5 Shelf life

4

There was perceived to be significant variation between the varieties in terms of shelf life of mango (Appendix 4.12 and table 4.24). The longest shelf life (13.58 days) was recorded in Khirshapat and the shortest (12.25 days) was recorded in Langra (Fig. 4.21). Variation in shelf life between varieties might be possible due to genetical. The results of the present investigation revealed that Khirshapat was much better over Langra in preservation.

Subjected to different doses of BDF solution in this investigation In connection with shelf life of mango exhibited highly significant (Appendix 4.12). The shelf life of mango ranged between 7.83 to 16.67 days was recorded from different doses of BDF solution. The longest shelf life (16.67 days) was obtained from the fruit treated with B₃ treatment followed by the shelf life of the fruits treated with B₂ (14.00 days), and B₁ (13.17 days) treatments while; the shortest shelf life (7.83 days) was obtained from control (Fig. 4.22). The longest shelf life obtained from the fruit treated with B₃ treatment might be probably due to suppression or depression of physiological and biochemical activities that was responsible for slower senescence of harvested fruits, and consequently led to the longest shelf life. The results of the present investigation are in conformity with the findings of Dhemre and Waskar (2004) and Ahmed and Singh (2000).

The combined effect of varieties and imposed different doss of BDF solution in respect of shelf life of mango were found to be non significant (Appendix 4.12). The shelf life was found to be differed between 7.33 to 17.33 days from various treatment combinations. The longest shelf life (17.33 days) was found from the treatment combination of V_2B_3 while, the shortest (7.33 days) was noticed from the treatment combination of V_1B_0 (Table 4.25).

×

A

Sec.

5

ÌF.

3

+





Fig. 4.21 Effect of varieties on shelf life of mango.

Fig. 4.22 Effect of different doses of BDF on shelf life of mango. Vertical bars represent LSD at 0.05 level

 $\begin{array}{ll} \mbox{Graph represents} & & \\ \mbox{V}_1 = \mbox{Langra} & & \mbox{B}_1 = 23 \\ \mbox{V}_2 = \mbox{Khirshapat} & & \mbox{B}_2 = 50 \\ \mbox{B}_0 = \mbox{Control} & & \mbox{B}_3 = 7 \end{array}$

 $B_1=250$ ppm of BDF solution $B_2=500$ ppm of BDF solution $B_3=750$ ppm of BDF solution
Treatments Variety (V)		Zinc content (mg/100 g) at							
	Initial	3	6	9	12	Total days			
V1	1.26 b	1.19 b	1.06 b	0.90 b	0.73 b	12.25 b			
V ₂	1.47 a	1.30 a	1.18 a	1.02 a	0.85 a	13.58 a			
Level of significance	***	***	***	***	***	**			

Table 4.24 Pattern of zinc content of mango pulp and shelf life of postharvest mango in varieties during storage environments at ambient condition

Y.

Table 4.25 Combined effects of varieties and different doses of BDF solution on zinc content and shelf life of postharvest mango pulp at ambient condition

Treatments combination		Shelf life				
Varieties × Treatments	Initial	3	6	9	12	Total days
V ₁ B ₀	1.26 c	1.13 f	0.97 e	0.76 e	0.53 e	7.33 e
V ₁ B ₁	1.28 c	1.17 ef	1.02 d	0.85 d	0.67 d	12.67 d
V ₁ B ₂	1.24 c	1.21 de	1.10 c	0.96 c	0.82 c	13.00 d
V ₁ B ₃	1.25 c	1.25 cd	1.14 b	1.03 b	0.91 b	16.00 ab
V ₂ B ₀	1.45 b	1.26 c	1.10 c	0.89 d	0.66 d	8.33 e
V ₂ B ₁	1.46 b	1.29 bc	1.15 b	0.98 c	0.81 c	13.67 cd
V ₂ B ₂	1.44 b	1.32 ab	1.21 a	1.07 b	0.93 b	15.00 bc
V ₂ B ₃	1.52 a	1.34 a	1.24 a	1.13 a	1.01 a	17.33 a
Level of significance	*	NS	NS	NS	NS	NS
CV%	1.56	1.70	1.90	2.21	2.68	6.29

In a column values having the same letter(s) do not differ significantly as per DMRT at 5% level

Table represents $V_1 = Langra$ $B_1 = 250 \text{ ppm of BDF solution}$ * indicates at 5% level $V_2 = Khirshapat$ $B_2 = 500 \text{ ppm of BDF solution}$ * indicate at 1 % level $B_0 = Control$ $B_3 = 750 \text{ ppm of BDF solution}$ *** indicate at 0.1% level

NS means non significant

*

¥

4.

GRAPTER FIVE

×

2

-

>

*

4

SUMMARY AND GONGLUSION

CHAPTER 5

SUMMARY AND CONCLUSION

11.1 Summary

X

L

The present investigation was conducted at the Protein and Enzyme Laboratory of the Department of Biochemistry and Molecular Biology of the University of Rajshahi during the period from January, 2005 to January, 2008. These research activities were also carried out in the regional Soll Resources Development Institute (SRDI) and Science Laboratory, BSCIR, Sympur, Rajshahi. The objectives of the investigation were to find out the technologically sound and economically viable methods for the reduction of postharvest losses and to determine the pattern of physicochemical changes, storability and shelf life of postharvest mango varieties. Four different experiments were undertaken in this investigation for achieving the goals.

The experiment 1 was involved with two factors consisting of varieties (*viz.*, Langra and Khirshapat) and different storage treatments {(viz., control, paraffin coating, perforated polyethylene cover, unperforated polyethylene cover, hot water $(55\pm1)^0$ C treatment and low temperature in refrigerator $(4\pm1)^0$ C}. The experiment 2 was included with two factors constituting of varieties as in experiment 1 and different doses of Maleic hydrazide solution (namely, control, 200, 400 and 600 ppm). The experiment 3 was assigned with two factors comprising of varieties as in experiment 1 and different doses of Gibberellic acid solution (viz., control, 100, 200 and 400 ppm). The experiment 4 was laid out with two factors including varieties as in experiment 1 and different doses of Bavistin DF solution (viz., control, 250, 500 and 750 ppm)

Mangos of both the varieties were collected from mango growers of Kansat, Shibgonj Upazila of Chapai Nowabgonj district and Charghat Upazila of Rajshahi district. The other materials subjected as postharvest treatments were bought from Rajshahi City Market, Rajshahi by early order. The experiments were laid out in Randomized Complete Block Design with three replicates. Various observations were made on different fruit characteristics, physicochemical properties and shelf life. External fruit features were evaluated with naked eyes and standard color charts were also used for the determination of skin color of mango. Physlcochemical

Summary and Conclusion

Chapter 5

Se.

x

analyses of mango pulp were performed with the determination of edible portion, pulp to peel ratio, physiological weight loss, moisture content, dry matter content, ash content, vitamin C content, titratable acidity, pulp pH, TSS, total sugar, reducing sugar and non reducing sugar content, crude fibre, lipid content, water soluble protein content, different mineral as phosphorus, potassium, calcium, magnesium, copper, iron, manganese and zinc content. Shelf life of mango was determined externally by daily monitoring the physical appearance of the treated and untreated fruits as the total days required for ripening up to their suitability for the last consumption.

Three mango fruits were collected at random from three replicates of which each fruit was taken from the same type of treatment combination at an interval of 3rd, 6, 9 and 12th days of storage for physicochemical studies. The data obtained from the biochemical methods were statistically analyzed and the mean differences were evaluated with DMRT and LSD (LSD used only as graphical illustration) as well as interpreted systematically. The results of the different experiments are summarized below.

The results of the experiment 1 revealed that variety significantly influenced all the parameters studied. Markedly variation in between the two varieties in terms of skin color of treated and untreated mangoes during storage were encountered. Green color of untreated Langra was changed hastily than Khirshapat whereas, different storage treatments made delay their changes. Khirshapat treated with low temperature in refrigerator demonstrated better performance in holding green color up to 15th day of storage while, Langra showed up to 12th day. In addition, Langra increasingly produced higher quantity of edible portion, pulp to peel ratio, crude fibre, lipid, water soluble protein as well as different minerals namely, phosphorus, potassium, calcium, magnesium, iron and manganese content whereas, it decreased by providing more quantity of dry matter, ash, vitamin C and titratable acidity with the advancement of storage duration as compared to Khirshapat. On the other hand, Khirshapat successively lost more physiological weight and conserved more moisture content with the advancement of storage period comparing to Langra. It showed the gradual increase of pulp pH, TSS, sugar (total, reducing and non reducing) content followed by Langra. It also absorbed higher amount of copper and zinc content

Jr.

h

comparing Langra but, these minerals decreased gradually with the increase of storage period. Khirshapat exhibited the longer (16.44 days) shelf life than langra (15.22 days).

٩,

Different storage treatments subjected to this experiment played the significant role on physicochemical characters and shelf life at different days after storage except ash, total sugar and non reducing sugar at initial day as well as zinc content at 3rd day. Among the storage treatments, untreated mangoes were found to provide progressively more quantity of edible portion, pulp to peel ratio, physiological weight loss, moisture content, pulp pH, TSS, sugar (total, and non reducing), lipid, water soluble protein, phosphorus, potassium, calcium, magnesium, iron as well as manganese content at a certain period of storage, and consequently, these composition declined owing to starting decomposition of fruits but, these characters perceived lower increasing tendency at low temperature in refrigerator in all days of storage period without considering initial day in both the cases except PWL where unperforated polyethylene cover markedly reduced PWL in all storage duration. In addition, the mango fruits treated with low temperature in refrigerator were noted higher producer but, it showed lesser decreasing tendency in terms of dry matter, ash, vitamin C, titratable acidity, crude fiber, copper and zinc content at different days of storage followed by other treatments due to delay ripening resulted in lower trend while, these components were also noticed as higher decreasing trend with lower accumulation from control without considering initial day of storage. The longest (36.50 days) shelf life of mangoes was obtained from the fruit treated with low temperature in refrigerator while, the shortest (7.83 days) was obtained from control.

The combined effect of varieties and applied different storage treatments were found to be significant in connection with physiochemical parameters at different days after storage, except at edible portion at all days, pulp to peel ratio at initial day, moisture content at 6, 9 and 12th day, dry matter at 6, 9 and 12th day, ash content at all days, tritrable acidity at 3 and 6th day, pulp pH at initial and 3rd day, TSS at initial, 3rd and 6th day, total sugar at all days, reducing sugar at initial, 3rd and 6th day, non reducing sugar at initial, 3rd, 6th and 12th day, phosphorus at initial day, potassium and calcium at all days, magnesium at 9 and 12th day, copper

253

The.

and iron at all days, manganese at 6, 9 and 12^{th} day as well as zinc content at initial, 3, 6 and 9^{th} day.

Among different treatment combinations, it was noted that Langra along with control treatment without considering initial day successively increased higher quantity in edible portion, pulp to peel ratio, lipid, water soluble protein, phosphorus and potassium content up to 9th day and thereafter, these components decreased; augmented in magnesium and iron content increasingly up to 6^{th} day then, these minerals diminished with the advancement of storage period while, the minimum increasing tendency was noticed from the treatment combination of Khirshapat having low temperature in refrigerator owing to retardation of ethylene synthesis resulting in delay. But, the treatment combination of Khirshapat with control progressively showed in PWL, augmented in moisture content up to 6th day then, it decreased; increased in pulp pH, augmented in TSS up to 6th day and then, it declined slightly; accumulated in sugar(total reducing and non reducing) as well as calcium content up to 9th day thereafter, these compositions came down due to starting rotten followed by other treatment combination while, the minimum extending trend was noted from the treatment combination of Langra with low temperature in refrigerator owing to delay ripening that resulted in lower synthesis without considering initial day of storage in all the cases.

Langra using low temperature in refrigerator showed higher quantity but, it diminished by degrees with the extention of storage period in terms of dry matter, ash content, vitamin C, titrable acidity as well as crude fibre content over the other treatment combination while, higher decreasing tendency was obtained from the treatment combination of Khirshapat using control treatment but, dry matter and ash content slightly increased after 6 days of storage without considering initial day of storage.

Khirshapat without treatment accumulated higher in copper and zinc content at initial day but, this variety using low temperature in refrigerator diminished very mildly at different days after storage followed by other treatment combinations whereas, highest diminishing trend was perceived from the treatment combination of Langra with control. The longest (38.00 days) shelf life of mangoes was found from the treatment combination of Khirshapat along with low temperature in refrigerator

3m

while, the shortest (7.33 days) was noticed from the treatment combination of Langra without treatment.

The results of the experiment 2 explored that mango varieties showed significant variation in all the parameters studied. Sharp variation between the two varieties in connection with skin color of treated and untreated mangoes during storage was observed. Green color of Langra altered markedly as compared to Khirshapat from untreated mangos whereas; different doses of Maleic hydrazide (MH) solution retarded their rapid changes of green color. Khirshapat treated with 600 ppm (M₃) of MH showed better performance in retaining green color up to 9th day of storage thereafter, it converted its original color into modified color while, Langra demonstrated up to 6th day of storage. The results further elucidated that Langra increasingly accumulated higherr quantity of crude fibre, lipid, water soluble protein as well as different minerals viz., phosphorus, potassium, calcium, magnesium, iron and manganese contents while, it diminished the higher quantity of dry matter, ash, vitamin C and titratable acidity with the advancement of storage duration over Khirshapat. But, Khirshapat continuously showed higher physiological weight loss and contained higher moisture content with the extension of storage period as compared to langra. It exhibited the gradual increase in pulp pH, TSS, sugar (total, reducing and non reducing) content comparing to langra. It also developed higher quantity of copper and zinc content comparing to Langra but, these two mineral contents came down gently with the increase of storage period. Khirshapat showed longer (13.17 days) shelf life over Langra (11.83 days).

Different doses of MH solution imposed to this experiment significantly affected different physicochemical parameters and shelf life at different days after storage except moisture, titratable acidity, non reducing sugar, crude fibre and copper content at initial day as well as ash content at all days and water soluble protein content at initial and 12th day. Among various doses of MH solution, untreated mangoes lost physiological weight at all the storage duration but, these mangoes increasingly absorbed more moisture; accumulated higher pulp pH, TSS, sugar (total, reducing and non reducing), lipid, water soluble protein, phosphorus, potassium, calcium, magnesium, iron as well as manganese content at a certain period of storage and thereafter, these properties fell off owing to beginning decay of

Summary and Conclusion

Chapter 5

1 -

fruits except pulp pH but, these composition manifested lower rising tendency at 600 ppm of MH solution in all days of storage period without considering initial day in all cases. In addition, the mango fruits treated with M_3 treatment were recorded the highest producer but, they demonstrated smaller diminishing tendency in respect of dry matter, ash, vitamin C, titratable acidity, crude fiber, copper and zinc content at different days of storage as compared to other treatments due to delay ripening that caused lower diminishing trend while, these characters were also noted higher abating trend with lower accumulation from control without considering initial day of storage. The longest (15.50 days) shelf life of mangoes was recorded from the fruits treated with M_3 treatment while, the shortest (8.00 days) was obtained from control (M_0).

The combined effect of varieties and imposed different doses of MH solution were perceived to be significant in relation with physiochemical composition at different days after storage, except moisture, dry matter, ash, pulp pH, total sugar, reducing sugar, non reducing sugar, crude fibre, water soluble protein, potassium, iron, manganese at all days, tritrable acidity at initial, 6, 9 and 12th day, TSS at initial and 3^{rd} day, lipid at initial, 3 and 6^{th} day, magnesium at 3 and 9^{th} day, copper at 3, 6 and 9th day and zinc content at initial, 3, 6 and 9th day as well as shelf life. Among different treatment combinations, the results were found that Langra along with M_0 treatment without considering initial day gradually increased higher quantity in TSS up to 6th day, lipid, water soluble protein, phosphorus and potassium content up to 9th day and then, these properties fell off; augmented in magnesium and iron content increasingly up to 6th day and consequently these mineral contents came down with the extension of storage period while, the minimum increasing tendency was recorded from the treatment combination of Khirshapat along with M₃ treatment because of interruption of ethylene synthesis resulting in delay ripening. The treatment combination of Khirshapat and control continuously showed PWL; accumulated in higher moisture content up to 6th day then, it decreased; enhanced in pulp pH; extended in TSS at 3rd day and then, it declined slightly; synthesized in sugar (total, reducing and non reducing) as well as calcium content up to 9th day thereafter, these properties fell off owing to deterioration of fruit situation over other treatment combinations while, the lowest extending trend was noticed from the

×.

X

treatment combination of Langra with M_3 treatment due to delay ripening that resulted in lower synthesis without considering initial day of storage in all the cases.

Langra using M_3 treatment exhibited higher quantity but, it came down gently with the advancement of storage period in respect of dry matter, ash, vitamin C, titrable acidity as well as crude fibre content followed by the other treatment combinations while, higher decreasing tendency was noticed from the treatment combination of Khirshapat using control treatment but, dry matter and ash slightly augmented after 6th day of storage without considering initial day of storage.

Khirshapat without treatment developed higher in copper and zinc content at initial day but, this variety using M_3 treatment abated very softly at different days after storage over the other treatment combinations whereas, the highest decreasing trend was obtained from the treatment combination of Langra with control. The longest (16.00 days) shelf life of mangos was counted from the treatment combination of Khirshapat along with M_3 treatment while, the shortest (7.33 days) was counted from the treatment combination of Langra using no treatment.

The findings of the experiment 3 invented that mango varieties demonstrated significant variation in all the parameters investigated except pulp pH, ash, reducing sugar and water soluble protein content at initial, 3, 6 and 12th days of storage, respectively. There was exposition a hastily variation in between the two varieties in relation to skin color of treated and untreated mangoes during storage. Green color of untreated Langra was exchanged sharply as compared to Khirshapat whereas; different doses of GA3 solution resisted their fast changes of green color. Khirshapat treated with 400 ppm of GA3 solution exhibited better performance in holding green color up to 9th day of storage thereafter, it transformed its original color into rotted colors while, Langra gave its original color up to 6th day of storage and thereafter, it was modified as well. The results further denoted that Langra largely extended higher quantity of TSS at 9 and 12th day, crude fibre, lipid, water soluble protein as well as different minerals viz., phosphorus, potassium, calcium, magnesium, iron and manganese content while, it decreased by producing higher quantity of dry matter, ash, vitamin C and titratable acidity with the extending of storage duration over Khirshapat. But, Khirshapat showed higher PWL in a continuous stream and held in higher moisture with the augmentation of storage period as compared to Langra. It

*

×

was found the gradual extension of pulp pH, TSS, sugar (total, reducing and non reducing) contents comparing to langra. It also accumulated higher quantity of copper and zinc content comparing to Langra but, these two mineral contents came down gently with the increase of storage period. Khirshapat demonstrated longer (13.92 days) shelf life over Langra (12.67 days).

Different doses of GA3 solution applied to this experiment were perceived to be significant in terms of different physicochemical properties at different days after storage and shelf life except moisture, reducing sugar and crude fibre content at initial day as well as ash content at all days and water soluble protein content at initial and 12th day. Among different doses of GA3 solution, untreated mangoes were recorded markedly losing higher quantity of PW at all storage period but, these mangoes increasingly retained more moisture content; produced higher pulp pH, TSS, sugar (total, reducing and non reducing), lipid, water soluble protein, phosphorus, potassium, calcium, magnesium, iron as well as manganese content at a certain period of storage and thereafter, these properties fell off owing to deterioration of fruit situation except pulp pH but, these properties were noted lower enlarging tendency at 400 ppm of GA3 solution in all days of storage period without considering initial day in all the cases. In addition, the mango fruits treated with G_3 treatment were noted higher producer but, they showed lower decreasing tendency in connection with dry matter, ash, vitamin C, tltratable acidity, crude fiber, copper and zinc content at different days of storage as compared with the other treatments due to delay ripening causing in lower falling off trend while, these characters were also reported higher abating trend with lower accumulation from untreated mangoes without considering initial day of storage. The longest (17.00 days) shelf life of mangoes was counted from the fruit treated with G_3 treatment while, the shortest (8.17 days) was counted from control (G₀).

The interaction effect of varieties and different doses of GA3 solution was noticed to be non significant in terms of some physiochemical properties and shelf life of mango at different days after storage, but the combined effect also showed significant variation in some properties *viz.*, physiological weight loss at all days of storage; vitamin C, calcium and magnesium content at 3, 6, 9 and 12th day; TSS, lipid, phosphorus and zinc content at 6, 9 and 12th days; non reducing sugar and

1

in.

T

1

×

manganese content at 9th day; tritrable acidity at 3 and 6th days as well as reducing sugar content at 6th day of storage. Among different treatment combinations, the results noted that Langra along with G_0 treatment without considering initial day gradually increased higher quantity in TSS up to 6 day, lipid, water soluble protein, phosphorus and potassium content up to 9th day and then, these composition came down and accumulating magnesium and iron content increasingly up to 6th day and thereafter, these mineral contents decreased with the extension of storage period while, the lowest increasing tendency was recorded from the treatment combination of Khirshapat using G₃ treatment owing to retardation of ethylene synthesis. The treatment combination of Khirshapat and control largely exhibited PWL; retained higher moisture content up to 6th day then, it diminished; expanded pulp pH; increased TSS at 3rd day and then, it decreased slightly; accumulated sugar (total, reducing and non reducing) as well as calcium content up to 9th day consequently, these compositions diminished owing to starting rotten over other treatment combinations whereas, the minimum increasing trend was recorded from the treatment combinations of Langra with G₃ treatment due to delay of ripening that resulted in lower synthesis without considering initial day of storage in all the cases.

Langra using G_3 treatment showed higher quantity but, it diminished by fits and bound with the extending of storage period in respect of dry matter, ash, vitamin C, titrable acidity as well as crude fibre content followed by other treatment combinations whereas, higher decreasing tendency was noted from the treatment combination of Khirshapat using control treatment but, dry matter and ash content slightly grew up after 6 days of storage without considering initial day of storage.

Khirshapat using G_3 treatment produced higher quantity of copper and zinc content but, it fell off by degrees at different days after storage over the other treatment combinations whereas, the highest abating trend was recorded from the treatment combination of Langra with control. The longest (18.00 days) shelf life of mangoes was counted from the treatment combination of Khirshapat along with G_3 treatment while, the shortest (7.67 days) was counted from the treatment combination of Langra using without treatment.

The findings of the experiment 4 revealed that mango varieties had significant variation in all the parameters studied except pulp pH at 6th day, lipid content at

<u>___</u>

X

×

initial, 3, 6 and 12th days and water soluble protein content at 3, 9, and 12th days of storage period. The results showed sharp variation between two varieties in terms of skin color of treated and untreated mangos during storage. Green color of untreated Langra was modified sharply over Khirshapat while; different doses of Bavistin DF solution strictly hindered their rapid changes of green color. Khirshapat treated with 750 ppm of BDF solution showed better performance in bearing green color up to 6th day of storage and then, it changed its original color into converted colors whereas; Langra developed its original color up to 6th day of storage and thereafter, it was bartered as well. The results further denoted that Langra largely increased higher quantity of TSS at 6, 9 and 12th day, crude fibre, lipid, water soluble protein as well as different minerals viz., phosphorus, potassium, calcium, magnesium, iron and manganese content while, it diminished by producing higher quantity of dry matter, ash, vitamin C and titratable acidity with the increasing of storage period over Khirshapat. But, Khirshapat showed more PWL in a continuous stream and held in higher moisture with the increasing of storage period as compared to Langra. There noticed the gradual increase of pulp pH, TSS at initial and 3 days; sugar (total, reducing and non reducing) content comparing to Langra. It also absorbed higher quantity of copper and zinc content comparing to Langra but, these two minerals decreased little by little with the extension of storage period. Khirshapat had longer (13.58 days) shelf life over Langra (12.25 days).

Different doses of BDF solution applied to this experiment were manifested to be significant in terms of different physicochemical properties at different days after storage and shelf life except moisture, dry matter, sugar (total, reducing and non reducing) and crude fibre content at initial day as well as ash content at all days and water soluble protein content at initial and 12th day. Among different doses of BDF solution, untreated mangos were noticed to reduce sharply large quantity of PW at all storage period but, these mangoes progressively contained more moisture; augmented higher pulp pH, TSS, sugar (total, reducing and non reducing), lipid, water soluble protein, phosphorus, potassium, calcium, magnesium, iron as well as manganese content at a certain period of storage and thereafter, these properties diminished due to starting of decay of fruits except pulp pH but, these properties showed lower increasing tendency at 750 ppm of BDF solution in all days of storage

☑ Summary and Conclusion

Chapter 5

X

X

A

X

period without considering initial day in all the cases. In addition, the mango fruits treated with B_3 treatment were perceived higher producer but, they showed decreasing tendency in terms of dry matter, ash, vitamin C, titratable acidity, crude fiber, copper and zinc content at different days of storage over the other treatments due to delay ripening while, these characters were also recorded higher diminishing trend with the lower accumulation from untreated mangoes without considering initial day of storage. The longest (16.67 days) shelf life of mangoes was counted from the fruit treated with B_3 treatment while, the shortest (7.83 days) was counted from control.

The interaction effect of varieties and different doses of BDF solution were recorded to be non significant in connection with some physiochemical characters and shelf life of mango at different days after storage, but, the combined effect also demonstrated significant variation in some properties vis., physiological weight loss at 3, 6 and 9th day, vitamin C at all days, tritrable acidity at 3, 6, 9 and 12th day, TSS at 6, 9 and 9th day, lipid at 9th day, phosphorus at 12th day, calcium at initial, 3, 9 and 12th day, iron at 9 and 12th day and zinc at initial day of storage. Among different treatment combinations, Langra along with Bo treatment without considering initial day mildly increased higher quantity in TSS at 6th day, lipid, water soluble protein, phosphorus and potassium content up to 9th day and then, these compositions decreased; magnesium and iron content increased up to 6th day and thereafter, these minerals decreased with the increase of storage period. On the other hand, the lowest increasing tendency was noticed from the treatment combination of Khirshapat using B_3 treatment owing to interruption of ethylene synthesis resulting in delay ripening. The treatment combination of Khirshapat and control largely showed PWL; absorbed higher moisture up to 6th day then, it decreased; extended pulp pH, increased in TSS at 3rd day and then, it decreased slightly; received sugar (total, reducing and non reducing) as well as calcium content up to 9th day consequently, these compositions fell off owing to starting rotten over other treatment combinations whereas, the minimum increasing trend was obtained from the treatment combination of Langra with B_3 treatment due to delay ripening resulting in lower synthesis without considering initial day of storage in all the cases.

the state

2h

A

£

Langra using B_3 treatment showed higher quantity of dry matter, ash, vitamin C, titrable acidity as well as crude fibre content followed by other treatment combination but, they were reduced gradually with the augmenting of storage period whereas, higher diminishing tendency was noticed from the treatment combination of Khirshapat using control treatment but, dry matter and ash content slightly increased after 6th day of storage without considering initial day of storage.

Khirshapat using B_3 treatment gave higher quantity of Copper and Zinc content but decreased softly at different days after storage followed by other treatment combinations whereas, the highest failing trend was noted from the treatment combination of Langra with control. The longest (17.33 days) shelf life of mangoes was noted from the treatment combination of Khirshapat along with B_3 treatment while, the shortest (7.33 days) was found from the treatment combination of Langra using without treatment.

11.2 Conclusion

The overall findings of the present investigation trialed with four different experiments facilitated to draw the following conclusion:

- 1. Langra was found to be better in achieving higher quantity of edible portion, pulp to peel ratio, dry matter, ash, vitamin C, titratable acidity, crude fibre, lipid, and water soluble protein, phosphorus, potassium, magnesium, iron and manganese content overall the the parameters studied, but its shelf life was shorter and rapidly changed its skin color as compared to Khirshapat. Further investigation is to be needed for comparison with other varieties.
- 2. Among different postharvest treatments, most of the physical, biochemical and mineral properties were progressively increased with times from untreated fruits in case of all the experiments, and the remaining properties were gradually decreased that resulted in rapid ripening and consequently, it deteriorated. On the other hand, the mangoes treated with low temperature in refrigerator (4±1)°C, 600, 400 and 750 ppm of MH, GA3 and BDF solution were found to be better in interrupting the changes of physical, biochemical and mineral contents which resulted in delay ripening that caused longer shelf life (*viz.*, 36.50, 15.50, 17.00 and 16.67 days) in different experiments, respectively.

*

á

à

The state

3. From the interaction effects of both the varieties in all the experiments, it was found that Langra along with low temperature in refrigerator (4±1)⁰ C, 600, 400 and 750 ppm of MH, GA3 and BDF solution respectively, showed delay changes of physical, biochemical and minerals properties that made delay ripening. On the other hand, Khirshapat using above mentioned treatments strongly retarded the changes of these properties which exhibited the longest shelf life (namely 38.00, 16.00, 18.00 and 17.3 days). It also using unperforated polyethylene cover extensively reduced PWL. So, this method may be adjusted with other systems.

11.3 Recommendation

The following recommendations are suggested based on the data obtained from the present experiments:

- In terms of nutrient content, Langra was found to be higher quality as compared to Khirshapat. So, cultivation of this variety might be helpful for improving the nutritional status of the people of our country.
- Khirshapat showed the longest shelf life. So, based on shelf life consideration, Khirshapat may be recommended for commercial production.
- 3. Physical, biochemical and mineral properties drastically changed in control treatment, but using low temperature in refrigerator was manifested to be the best storage environment for reduction of losses and extension of shelf life of mango. In addition, at 600, 400 and 750 ppm of MH, GA3 and BDF solution were also found to be suitable for reduction of losses and extension of shelf life in other experiments, respectively. So, these approaches may be brought to farmers for processing and storage of postharvest mango.
- Unperforated polyethylene cover was found better than other systems in respect of reduction of PWL. So, this system may also be used commercially by the mango growers.



REFERENCES

2

¥.

X

X

- Absar, N., M.R. Karim and M. Al-Amin. 1993. A comparative study on the changes in the physicochemical composition of ten varieties of mango in Bangladesh at different stages of maturity. Bangladesh J. Agril. Res., **18** (2): 201-208.
- Ahmed, M. S. and S. Singh. 2000. Studies on extension of storage life of Amrapali mango. Orissa J. Hort. **28** (2):73-76.
- Ali, N. and H. Mazher. 1960. The tree, flower and fruit characteristics of mango. Punjab Fruit J., **23**: 81-86.
- Amiruzzaman, N. 1990. Post harvest handling and processing of fruits and Vegetables. In: Kitchen Gardening and Homestead Productive Activities CIRDAP action Research Series No. 11, p. 22

Anonymous . 1962. Wealth of India, Raw Materials, C.S.I.R., New Delhi.

- Ashwani, K. and S. S. Dhawan. 1995. Effect of postharvest treatment on the enhancement of ripening of mango (*Mangifera indica*) fruit cv. Dashehari. Haryana J. Hort. Sci., **24** (2): 109-115.
- Azad, M. I. 2001. Reduction of postharvest losses and extension of shelf life of mango. Ph.D Thesis, Department of Horticulture, BAU, Mymensingh.
- Baez, S. R., W. A. Bringas, W. A. Mendoza, C. J. Ojeda and R. J. Mercado. 2001.
 Postharvest performance of mango 'Tommy Atkins' treated with hot water and wax. Proceeding of the Interamerican Society For-Tropical-Horticulture. 44:39-43.
- Baez-Sanudo, R., J. Siller-Cepeda, T. E. Bringas and S.M. Baez. 1993. Determination of maturity indices in the main mango cultivars produced in mexico. Proc. Interamerican Soc. Trop. Hort., **37**: 148-154{Cited from Postharvest News and Information, **6** (6): 2775, 1995}.
- Bajpai, P. N., H. S. Shukla and O. P. Chatur Vedi. 1985. History, Importance and Scope tropical and sub-tropical, ed. T. K. Bose, B. Mitra, Naya Prokash, 206
 Bidhan Sarani, Calcutta, India, pp. 1 - 20.
- Bandyopadhyay, G and Nair. 1990. Lipid Composition and Flavor Changes in Irradiated Mango (var. Alphonso). J. Food Sci., **55** (6), 1579–1580.
- Banik, A. K. 1995. Studies on the postharvest treatments on shelf life of mango with special reference to major diseases. Ph.D. thesis, BCKV, West Bengal, p. 52.

- Barua, G. 2003. Shelf life and quality of mango as influenced by different doses of postharvest treatments. MS Thesis, Department of Horticulture, BAU, Mymensingh.
- BBS (Bangladesh Bureau of Statistics). 1997. Year Book of Agricultural Statistics of Bangladesh. Bangladesh bureau of Statistics. Statistical Division, Ministry of Planning, government of the peoples, Republic of Bangladesh, Dhaka, P. 171.
- BBS (Bangladesh Bureau of Statistics). 1999. Statistical Pocket Book of Bangladesh, 1998, Bangladesh Bureau of Statistics, Ministry of Planning, Govt. of the Peoples Republic of Bangladesh, Dhaka.
- BBS (Bangladesh Bureau of Statistics). 2006. Statistical Year Book of Bangladesh. Planning Division, Ministry of Planning, Government of People's Republic of Bangladesh, Dhaka, Bangladesh, p. 157-158.
- BBS. 2000. Monthly Statistical Bulletin, April.

R

1

- Benitez, M. M., A. L. Acedo, P. Jitareerat and S. Kanlavanarat. 2006. Mango fruit softening response to postharvest treatments. Acta-Hort., 712 (2):811-816.
- Bessey O. A. and C. G. King. 1933. The distribution of Vitamin C in plant and animal tissues and its determination., J. Biol. Chem., **103**: 687
- Bligh, E. G. and W. Dyer. 1959. Total lipid extraction and purification. Can. J. Biochem. Physiol. **37**. p. 911.
- Boonraung, C., U. Farungsang and S. Sangchote. 1993. Postharvest handling of tropical fruits. ACIAR Proc., **50**:469.
- Bose, T. K. 1985. Fruits of India Tropical and Subtropical. Naya Prokash, Calcutta-6, India, p. 69.
- Bugante, R.D., M. C. C. Lizada and M. B. D. E. Ramos. 1997. Disease control in Philippines' Carabao' mango with preharvest bagging and postharvest hot water treatment. Acta Horticulturae, 455:797-804.
- Candole, A. D. 1984. Origin of Cultivated Plants. Vegal Paul Trench and Co. London, pp. 1-67.
- Carrilo Lopez, A., J. B. Valdez Torres, V. R. Rojos and E. M. Yahia. 1995. Ripening and quality of mango fruit as affected by coating with 'semperfresh'. Acta Hort., **370**: 203-216.
- Castrillo, M. and M. Bermudez. 1992. Postharvest ripening in wax-coated Bacado mango. Int. J. Food Sci. Tech., **27** (4): 457-463.
- Castrillo, M., N. J. Kruger and F.R. Whatley. 1992. Sucrose metabolism in mango during mango ripening. Plant Sci., **84** (1): 45-51.

- Chandra, J. and V. N. Pathak. 1992. Effect of plastic film wrapping on postharvest fungal rot of mango fruits. Indian Phytopath., **45** (1): 128-130.
- Chattapadhyay, P. K. 1989. Studies on the shelf life of mango following treatment with chemicals and cooling. Hort. J., **2** (1):12-15.
- Cheema, G.S., D.V. Karmaker and B.M. Joshi. 1939. Investigation on cold storage of mangoes. Indian. Coun. Agric. Res., New Delhi, Misc., Bull 21.
- Chhatpar, H. S., A. K. Matto and V. V. Modi. 1972. Some problems pertaining to storage and ripening in mango fruit. Acta Hort., **24**: 243.
- Dara, P., T. Wantapha and A. W. Pakhlni. 1982. The effect of irradiation on the storage life of quality of mangoes. In: Proc. 2nd conf. on utilization of Atomic energy in Agric. held in Bankok, Thailand., p. 15.

7

1

- Dhalla, R. and S.W. Hanson. 1988. Effect of permeable coating on the storage life of fruit. II. Prolong treatment of mangoes (*Mangifera indica* L.) cv. Julie. Int. J. Food Sci. Tech., 23: 107-112.
- Dhemre, J. K. and D. P. Waskar. 2004. Effect of postharvest treatments on shelf life and quality of Kesar mango fruits during storage. J. Maharashtra Agril. Univ. 29 (2): 159-163.
- Durigan, J. F., G. H. A. Teixeira, N. M. Castanharo and R. E. Domarco. 2004. Postharvest conservation of 'Tommy Atkins' mango fruit influenced by gamma radiation, wax, hotwater and refrigeration. Acta Hort. 645: 601-604.
- El-Mahmoudi, L.T. and M.T. Eisawi. 1968. Studies on the storage banana. Agric. Res. Rev., Cairo, **46** (3): 41-51.
- Esquerra, E. B. and M. C. C. Lizada. 1990. The postharvest behavior and quality of 'Carabao' mangoes subjected to vapour heat treatment. ASEAN Food J., **5** (1):6-11.
- FAO (Food and Agricultural Organization). 2006. Production Year Book. Rome, Italy.
- Fawaz, S. A. 2006. Effect of postharvest treatments on physicochemical characters of mango fruits (cv. Bullock's heart) during storage. Ann. Agril. Sci. Moshtohor.
 44 (2):705-716.
- Feng, S.Q., Y.X. Chen, H.Z. Wu and S. T. Zhou. 1991. The methods for delaying ripening and controlling postharvest diseases of mango. Acta Agric. Univ. Pekinensis, **17** (4):61-65{Cited from Postharvest News and Information,**5** (1) 231, 1994}.

- Feungchan, S. 1992. Gibberellin-delayed ripening of mango. Khon Kaen Agric. J., 20 (5): 245-248 [Cited from Postharvest News and Information, 4 (3): 12991, 1993].
 - Fonseca, M. J. O., C. C. L. Salamao, P. R. Cecon and R. Puschmann. 2001. Use of fungicides and wax in storage and postharvest quality of Haden mango (Mangifera indica L.) in cold storage. Revista-Brasileria-de-Armazenamento, **26** (**2**): 28-35.

Th

*

- Fonseca, M. J. O., P. R. Cecon, L. C. C. Salomao and R. Pushmann. 2004. Fungicides and wax in postharvest preservation of mango haden. Acta. Hort. 645: 557-563.
- Fonseca, M. J. O., P. R. Cecon, L. C. C. Salomao and R. Pushmann. 2004. Pulp and skin pigments in 'Mango Haden' treated with fungicide and wax. Acta, Hort., **645**: 633-638.
- Gafur, A., M. Z. Shafique, M. O. H. Helali, M. Ibrahim, M. M. Rahman, M. S. Alam and A. Gafur. 1997. Studies on extension of postharvest storage life of mango (Mangifera indica L.). Bangladesh J. Sci. Ind. Res., 32 (1):148-152.
- Gajbhiye, V.T., R.H. Singh, R. Kumar and C.P. Singh. 2000. Efficacy of postharvest treatment of Carbendazim against stem end rot / anthracnose diseases and its persistence in mango fruits. Ann. of plant protection Sci., 8 (2): 230-232.
- Gautam, B., S. K. Sarkar and Y. N. Reddy. 2003. Effect of postharvest treatments on shelf life and quality of Banganpalli mango. Indian J. Hort. 60 (2): 135-139.
- Gofur, M. A., M. Z. Shafique, O. H. Helali, M. Ibrahim, M. M. Rahman and A. Hakim. 1994. Effect of various factors on the vitamin C content of some mango varieties grown in Rajshahi region. Bangladesh J. Sci. Ind. Res., 29 (3): 163-171.
- Gomez, K. A. and A. A. Gomez. 1984. Statistical Procedures for Agricultural Research. John Weley and Sons. Inc. New York. pp. 67-215.
- Gomez-Lim M. A. 1997. Post-harvest Physiology. In: The Mango: Botany, Production, and Uses (Edited by Litz R E) CAB International pp. 425 - 446.
- Guoqiang, G., C. Yuxin and Y. Shamtao. 1994. Studles on room temperature storage of mango. Scient. Agricultura Sinica, 27 (3): 82-88 [Cited from Food Sci. Tech. Abstr., 27 (3):37, 1995].
- Haque, M.A. 1988. "Kalar Bagan" Banana Research Project, BAU., Mymensingh. Pp. 1-4.

Hassan, M. K., M. F. Mondal and M. S. Hoque. 1998. Studies on the storage behavior of mango. Bangladesh J. Agril. Sci., **26** (2):311-318.

1

3.

A

-t

1

- Hayes, W. B.1966. The Principles of fruit growing in India. Kitabistan , Publication , Allahabad, India, pp. 368 - 381.
- Hermanto, S. C. and Yuniarti. 1994. Hot water control of anthracnose on mango varieties 'Arumanis' 'Golek' and 'Manalogi' [*Colletotricum gloeosporioides*].
 Aust. Centre Intl. Agril. Res., pp. 453-454.
- Hossain, A. K. M. A. and A. Ahmed. 1994. A monograph on mango varieties of Bangladesh, pp. 1-89.
- Hossain, M. M. 1999. A study of the physicochemical changes during storage and shelf life of mango. MS Thesis, Department of Horticulture BAU. Mymensingh.

Huddar, A. G., B. C. Bharali and K. R. Thimmaraju. 1989. Note on extension of storage life of mango fruits by Tal-prolong. Acta Horticulturae, **231**:668-669.

Ilangantilake, S and V. Salokhe. 1990. Postharvest biotechnology to increase storage life of mango. Asian Inst. Tech., Div. Agril. Food Engg. Bangkok. pp. 1-39.

- Ilangantilake, S., L. Turla and C. C. Ren.1989. Pretreatment and hypoboric storage for increase of storage life of mango. St. Joseph, Michigan, USA. ASAE paper – American Soc. Agril. Engg., 6036(89): 16-20{Cited from Postharvest New Information, 2 (2): 638, 1991}.
- Illeperuma, C. K. and P. Jayasuriya. 2002. Prolonged storage of "Karuthacolomban" mango by modified atmosphere packaging at low temperature. J. Hort. Sci. Biotech, 77 (2): 153-157.
- Inyang, U.E. and A.U. Agbo. 1995. Effect of hot ash treatment of mango fruits on the physico-chemical changes during ripening. Trop. Sci., **35** (3):259-262.
- Jacobi, K. K. and D. Gowanlock. 1995. Ultra-structural studies of 'Kensington' mango (*Mangifera indica* L.) heat injuries. Hort. Sci., **30** (1):102-103.
- Jacobi, K. K. and J. E. Giles. 1997. Quality of 'Kensington' mango (*Mangifera indica*L.) following combined vapour heat disinfestation and hot water disease control treatments. Postharvest Biol. Technol., **12** (3):285-292.
- Jacobi, K. K. and K. S. Wong. 1992. Quality of 'Kensington' mango (*M. indica* L.) following hot water and vapour heat treatment. Postharvest Biol. and Technol., 1 (4):349-359.
- Jacobi, K. K., E. A. Macrae and S. E. Hetherington. 2000. Effect of hot air conditioning of 'Kensington' mango fruit on the response to hot water treatment. Postharvest Biol. Tech. 21 (1): 39-49.

- Jacobi, K. K., E. A. Macrae and S. E. Hetherington. 2001. Loss of heat tolerance in 'Kensington' mango fruit following heat treatment. Postharvest Biol. Tech. 21 3: 321-330.
- Jacobi, K. K., K. S. Wong and J. E. Giles. 1996. Conditioning with hot air reduces damage caused to 'Kensington' mango (*Mangifera indica* L.) by hot water disinfestation treatment. Aust. J. Expt. Agric., **36** (4):507-512.
- Jagirdar, S.A. P. and A.K. Maniyar. 1960. Mango culture in Sind. Punjab Fruit J., 23 (82-83): 221-223.

F.

T

1

- Jain, S. K. and S. Mukherjee. 2001. Postharvest application of GA3 to delay ripening in mango (*Mangifera indica* L.) cv. Langra. J. of Eco-physiology. 4 (1/2): 27-30.
- Jain, S. K., S. Mukherjee and N. K. Gupta. 2001. Effect of postharvest treatments and storage condition on the quality of mango during storage. Haryana J. Hort. Sci., **30** (3/4): 183-187.
- Jayaraman, J. 1981. Laboratory Manual in Biochemistry (Ist ed.) Wiley Eastern Ltd. New Delhi, India.
- Joarder, G. K. 1981. Preservation of mango at cold temperature in Bangladesh. Bangladesh. J. Agric., **5** (3):142-152.
- Joarder, G.K. 1980. Preservation of mango at cold temperature. Bangladesh Coun. Sci. Ind. Res. Lab., Dhaka, Bangladesh, pp. 11-19.
- Johnson, G. I., L. M. Coates, A. W. Corke and I. A. Wells. 1984. Post Harvest Control of mango diseases. Aust. Centre Intl. Agril. Res., pp. 70-71.
- Johnson, G. I. and S. Sangehote. 1994. Control of postharvest disease of tropical fruits: Challenges for the 21st century. In: Champ, B. R. Highley, E. and G.I. Jonson (eds). Postharvest Handling of Tropical Fruits. ACIAR proc. 50. Canberra, pp. 140-161.
- Joshi, G.D. and S.K. Roy. 1988. Influence of maturity, transport and cold storage on biochemical composition of 'Alphonso' mango fruit. J. Moharastra Agril. Univ., 13 (1): 12-15.
- Kalra, S.K. and D.K. Tandon. 1983. Ripening behaviour of Dashehari mango in relation to harvest period. Scientia Hortic., **19** (3-4): 263-269.
- Khader, A. K. 1985. Postharvest Technology of Horticultural Crops, Cooperative Extension. Univ. California Special Pub. No. **331**. pp. 35-43.

Khader, M. A. J. B. M., K. Chellappan, O. A. A. Pillai and P. K. Chattapadhaya. 1995. Banana, in: Fruit of India. Tropical and Sub-tropical, ed. T. K. Bose, B. Mitra, Naya Prakash, 206 Bidhan Sarani, Calcutta, India, P. 124.

X

F

A.

*

- Khader, S. E. S. A. 1989. Delaying ripening by postharvest treatment of GA3 in mango. Indian J. Hort., **46** (4): 444-448.
- Khader, S. E. S. A. 1991. Effect of postharvest application of GA3 on postharvest behaviour of mango fruits. Scientia Hortic., **47** (3-4): 317-321.
- Khader, S. E. S. A. 1992. Effect of gibberellic acid and vapour gard on ripening, amylase and peroxidase activities and quality of mango fruits during storage.
 J. Hort. Sci., 67 (6): 855-860.
- Khumlert, R. 1992. Effect of gibberellic acid (GA3) on some physico-chemical characteristics of 'Kheaw' and 'Sawoey' mango fruit. Collage Laguna (The Philippines), p. 95.
- Koolpluksee, M., S. Mesta and S. Subhadrabandhu. 1993. Effect of modified atmospheres on quality and chilling injury of 'Nam Dok Mai' mango fruit. Kaset. J. Nat. Sci., 27 (2): 115-124.
- Krishnamaruthy, S. 1989. Effect of Tal-prolong on shelf life and quality attributes of mango. Acta Hort., **231**:675-678
- Kruger, F. J., M. Kritzinger and M. Bezuidenhout. 1997. Trials aimed at improving the storage potential of Heidi. Yearbook-South African Mango Growers' Association, 17:100-104.
- Kumar, A. and S.S. Dhawan. 1995. Effect of postharvest treatments on the enhancement of ripening of mango (*Mangifera indica*) fruits cv. Dashehari. Haryana J. Hort. Sci., **24** (2): 109-115.
- Kumar, P. and S. Singh. 1993. Effect of GA3 and Ethrel on ripening and quality of mango cv. Amrapali. Hort. J., 6 (1): 19-23.
- Kumar, R., R.A. Kaushik and A.S. Chharia. 1992. Effect of postharvest treatments on the quality of mango during storage. Haryana J, Hort. Sci., **21** (1/2): 46-55.
- Kumar, S., D.K. Das, A.K. Singh and U.S. Prasad. 1993. Changes in non-volatile organic acid composition and pH during maturation and ripening of two mango varieties. Indian J. Plant Physiol., 36 (2): 107-111.

Lakshminarayana, S. 1975. Relation of time of harvest on respiration, chemical constituents and storage life of mangoes. Proc. Pla. State Hort. Soc., **88**: 477-481.

X

3r

3

- Lakshminarayna. S. 1980. Mango, Tropical and Sub-tropical fruit composition, properties and uses. AVI. Westport. C. T., P. 184.
- Lam, P.E. and L.S. Wong. 1988. Eating quality of ethylene ripened Harumanis mangoes after cold storage. MARDI Res. J., **16** (1): 85-90.
- Lashley, D. 1984. Advances in postharvest technology and new technology in food production. Proc. Seminar. St. Augustine (Trinidad Tobago), pp. 173-183.
- Leon, S. Y. D. and L. S. D. Lima. 1968. Postharvest changes in some physical and chemical properties of Picomangoes (*Mangifera indica* L.) Philippines J. Sci., 77: 337-347.

Lianni, Z. H. Ying, D. Yicai and L. Mingai. 1994. Studies on postharvest physiology of mango fruits. J. Trop. Sub-Trop. Bot., **2** (1): 64-69 [Cited from Hort. Abstr., **66** (2): 1834, 1996].

- Lonsdale, J. H. 1992a. In search of effective postharvest treatment for the control of postharvest diseases of mangoes. Year book-South African Mango Growers' Association, **12**:153-180.
- Lonsdale, J. H., H. Greef and W. Brooks. 1991a. The efficacy of prochloraz, choramizol-sulphate and quazatine on postharvest diseases of mango. South African Mango Growers' Association Year book, **11**:35-38.
- Lonsdale, T. Gough and R. E. Lunt. 1991b. Control of postharvest decay of mangoes using hot water in combination with radurisation or modified atmosphere packaging. Yearbook-South African Mango Growers' Association, **11**:42-44.

Loomis, W. E. and C. A. Shull. 1937. Methods in Plant Physiology. Mc Hill, New York.

- Lowry, O. H., N. J. Rosebrough, A. L., Farr and R. J., Randall. 1951. Protein measurement with the Folin- Ciocalteau's Reagent., J. Biol. Chem. 193: 265-275
- Majumder, G., V.V. Modi and V.A. Palejwala. 1981. Effect of plant growth regulator on mango ripening. Indian J. Exp. Biol., **19** (9): 885-886.
- Manzano, J.E., Y. Perez and E. Rojas. 1997. Coating waxes on Haden mango fruits (Mangifera indica L.) variety for export. Acta Hort., **455**: 738-746.

- Martinez, B.E., G.C. Guevara, M.J. Contreras and R. Rodriguez. 1997. Preservation of mango Azucar variety (*Mangifera indica* L.) at different storage stages. Acta Hort., **455**: 747-754.
- Mathooko, F. M. 2000. Manual of Third Country Group Training Programme in Applied Food Analysis: Postharvest Physiology- JKUAT KENYA.
- McCollum, T.G., S. Aguino and R.E. McDonald. 1993. Heat treatment inhibits mango chilling injury. Hort. Sci., **28** (3): 197-198.
- Medlicott, A.P., M. Bhogol and S.B. Reynolds. 1986. Changes in peel pigmentation during ripening of mango (*Mangifera indica* cv. Tommy Atkins). Ann. Appl. Biol., **109**: 651-656.
- Medlicott, A.P., S.B. Rynolds and A.K. Thompson. 1986. Effect of temperature on ripening of mango fruit (*Mangifera incia* L. var. Tommy Atkins). J. Sci. Food. Agric., **37**: 469-474.
- Mendez, M.J.E. 1994. Effect of gibberellic acid (GA3) application on fruits of several mango varieties. Proc. Inter Amer. Soc. Trop. Hort., **38**: 43-49.
- Mendez, M.J.E. 1995. Response of several mango varieties to gibberellic acid application. Amer. Soc. Agril. Eng. pp. 477-487.

-

A

- Miller, G.L. 1972. Use of Dinitro Salicylic acid reagent for determination of reducing sugar. Anal. Chem., **31**: 426-428
- Miller, W.R., D.H. Spalding and P.W. Hale. 1986. Film wrapping of mangoes at advancing stage of postharvest ripening. Trop. Sci., **26** (1): 9-17.
- Mohammed, A. and C. K. Sankat. 1995. Intermittent warming and cooling cycles in alleviating chilling injury of the Julie mango (*Mangifera indica* L.). Amer. Soc. Agril. Engg., pp.575-571.
- Mohiuddin, A. B. M., M. A. Rahim, F. Gheyas and A. M. Farooque. 1991. Anote on the prolongation of shelf life of mango (*Mangifera indica* L.) with some coating materials. Bangladesh Hort., **19** (1):45-47.
- Mollah, Z. and M.A. Siddique. 1973. Studies on some mangoes varieties of Bangladesh. Bangladesh Hort., **1** (2): 16-24.
- Mondal, M. F. 2000. Production and Storage of fruits (In Bengali). Published by Mrs. Afla Mondal, BAU campus, Mymenslngh-2202. pp. 11 - 17.

- Mondal, M. F., M. A. Rahman and M.A.J. Pramanik. 1995. Effects of different postharvest treatments on physico-chemical changes and shelf life of mango. Bangladesh Hort. 23 (1& 2): 1-5.
- Mondal, M. F., M. K. Hassan and M.S. Hoque. 1998. Physicochemical changes and shelf life of mango as influenced by postharvest treatments. Prog. Agric., 9 (1& 2):77-82.

Mukherjee, P. K. 1949. Mango and its relatives. Sci. Cult., 15: 5-9.

X.

*

1

4

- Mukherjee, P.K. 1956. The behaviour of mango varieties Dashehari and Langra under low temperature. Ann. Rep. Fruit Res. St., Saharanpur, India, p. 60.
- Murthy, S.K. and K.P.G. Rao. 1982. Regulation of ripening by chemicals in 'Alphonso' mango. Scientia Hortic., **16** (2): 1799-183.
- Mushtaq, K., M. S. Javed and A. Bari 2005. Postharvest losses in mango: the case study of Pakistani Punjab. Proceeding of the International Conference of Postharvest Technology and Quality Management in Arid-Tropics. 31 January to 2 February, 2005, pp. 89-94.
- Muy, R. D., J. C. Siller, J. P. Diaz and B. T. Valdez. 2004. Storage condition and waxing affect water status and quality of mango. Revista Fitotcnia-Mexicana. 27 (2): 201-209.
- Nadkarni, B. Y. 1963. Indian J. Med. Res., 51: 1111-16.
- Nair, S., Z. Singh and S. S. Tan. 2001. Heat treatments affect development chilling injury, respiration, ethylene production and fruit quality of mango. Acta Hort. Belgium, ISHS., 553 (2):549-550.
- Nyanjage, M. O., H. Wainwright and C. F. H. Bishop. 1998. The effects of hot water treatment with cooling and/ or storage on the physiology and disease of mango fruits (*Mangifera indica* L.). J. Hort. Sci. Biotechnol., **73** (5): 589-597.
- Nyanjage, M. O., H. Wainwright and C. F. H. Bishop. 2001. Effect of hot water treatments and storage temperatures on the ripening and the use of electrical impedance as an index for assessings postharvest changes in mango fruits. Annals. Applied Biol. **139** (1):21-29.
- O' Hare, T. J. 1995. Effect of ripening temperature on quality and compositional changes of mango(*Mangifera indica* L.) var. Kensington. Australia J. Expt. Agric., **35**(2): 259-263.
- Oosthuyse, S. A., M. E. Mobile, B. V. Straten and A. J. Winter. 1995. Effect of hydrocooling on field heat removal of mangoes and of differently water

temperatures on fruit quality after four weeks of cool storage. Year book-South African Mango Growers' Association, **15**: 54-57.

- Pal, R. K. 1998. Ripening and physiological properties of mango as influenced by ethereal and carbide. J. Food Sci. Technol. 35 (4): 358-360
- Paramanik, M.A.J. 1995. Effect of different postharvest treatments on physicochemical changes during storage and shelf life of mango. M.S. Thesis, Dept. Hort., BAU, Mymensingh.
- Parmar, P. B. and B. S Chundawat. 1989. Effect of various postharvest treatments on the physiology of Keser mango. Acta Hort., **231**: 679-684.
- Peter, M., L. Fweja, B. Chove, J. Kinabo, V. George, and K. Mtebe. 2007. Physical and chemical characteristics of off vine ripened mango (*Mangifera indica L.*) fruit (Dodo). African J. Biotech. Vol. 6 (21), pp. 2477-2483.
- Petersen, L. 2002. Analytical Methods- Soil, Water, Plant material, Fertilizer. Soil Resources Management and Analytical Services, Soil Resource Dev., Inst. Danida, Dhaka. pp. 61-70.
- Purohit, A. G. 1985. Fruit trees for social forestry. Indian Hort., 30 (3):3.

Purseglove, J. W. 1972. Mangoes of West India. Acta Hort,, 24:107-174.

-t

1

R

- Puttaraju, T. B. and T. V. Reddy. 1997. Effect of precooling on the quality of mango (cv. Mallika). J. Food Sci. Technol., **34** (1): 24-27.
- Quroshi, S.U. and M. B. Meah.1991. Postharvest Loss in Mango owing to stem rot. Int. J.Trop. Agric., **9** (2): 98: 105.
- Ranganna, S. 1979. Manual of analysis of fruits and vegetable products. Tata McGraw Hill Pub. Co. Ltd., New Delhi, p. 634
- Rangavalli, K., C. Ravisankar and P.H. Prasad. 1993. Postharvest changes in mango (Mangifera indica L.) var. Baneshan. South Indian Hort., **41** (3): 169-170.
- Ranjan, A.; R. N. Raj and K. K. Prasad. 2005. Effect of postharvest application of calcium salts and GA3 on storage life of mango (*Mangifera indica* L.) cv. Langra. J. Applied Biol. Patna India, **15** (1): 69-73.
- Rashid, A. and W.A. Farooqui. 1984. Studies on the effect of gamma irradiation and maleic hydrazide on the shelf life of mangoes. J. Agril. Res., **22** (2): 151-158.
- Reddy, N. S. and K. Haripriya. 2002. Extension of storage life of mango cvs. Bangalora and Neelum. South-Indian-Hort. India. **50** (1/3): 7-18.

- Rubbbi, S. F., M. A. Rahman and K. Q. Rahman. 1985. Studies on the processing and preservation of mango. Proc. 4th Natn. Symp. of Bangladesh Soc. Hort. Sci., pp. 138-148.
- Saaiman, W.C. and J. H. Lonsdale, 1994. The effect of prochloraz in combination with guazatine on postharvest diseases of mangoes. Yearbook-South African Mango Growers' Association, **14**: 62-64.
- Sadasivam, and Manickam. 1992. Biochemical Methods for Agricultural Sciences. Wiley Eastern Ltd. New Delhi, pp. 20 - 21.
- Saiman, W. C. 1995. Short wave infra-red radiation as an alternative to the hot water bath for the control of postharvest disease of mango. Year book-South African Mango Growers' Association, 14: 62-64.
- Salles, J. R. D. J and J. C. Tavares. 1999. Postharvest life of Mango (*Mangifera indica* L. cv. Tommy Atkins): the influence of temperature and state of maturity. Revista Brasileira de Fruticultura, **21** (2): 171 176.

t

- +-

A

T

- Salunkhe, D.K. and B.B. Desai. 1984. Postharvest Biotechnology of Fruit. Vol 1. CRC Press, Inc. Boca Raton, Florida. pp. 77-94.
- Samad, M. A., A. M. M. farruque and A. MaleK. 1975. A Study on the Biochemical Characteristic of the fruit of some mango variety of Bangladesh. Bangladesh J. Sci. Res., **12** (2): 28-32.

Samson, J. A. 1986. Tropical Fruits. 2nd edition, Jhon Wiley & Sons Inc., 605, Third Avenue, New York.

Samson, J.A. 1980. Tropical fruits. Longman Group Ltd., London, p. 121.

- Sankat, C. K., R. Moharaja and B. Lanckner. 1993. Ripening quality of 'Julie' Mangoes stored at low temperature. Acta Hort., **368**: 712-722.
- Sarker, S.C. and A.A.A. Muhshi. 1978. Study of the sugar and starch contents of mango at different stages of maturity. Bangladesh Hort., **6** (1-2): 14-19.
- Saucedo Veloz, C., G. Mena Nevarez and S. H. Chavez Franco. 1995. Effect of hydrothermal treatments for quarantine purpose on the physiology and quality of 'Manila' mango. Amer. Soc. Agril. Engg., pp. 276-281.
- Shahjahan, M. S., M. A. Sheel, M. A. Zaman and M. A. Sakur. 1994. Optimization of harvesting maturities for major mango cultivars in Bangladesh. Bangladesh J. Sci. Res., **12** (22): 209-215.

- Shahjahan, M.S., M.A. Sheel, M.A. Zaman and M.A. Sakur. 1994. Optimization of harvesting maturities for major mango varieties in Bangladesh. Bangladesh J. Sci. Res., **12** (2): 209-215.
- Shams- Ud-din.1997. Importance and role of processing and preservation of food crops in Bangladesh. Proceeding of the national seminar on the "Role of Agricultural Engineering in national Development" held at the BAU, Mymensingh, 15-16, May, 1997. Published by Agril. Engg. Division. BARC. Dhaka, pp. 22-32.
- Sheikh, J.I., F. Muhammad, T. Abbas and M. Yusuf. 1992. Effect of different chemicals on postharvest physical changes in mango fruits during storage. J. Argil. Sci., **30** (1): 111-117.
- Shivarama Reddy, L., K. R. Thimmaraju and L. S. Reddy. 1989. Effect of pre packing and postharvest treatments on the storage behaviour of mango fruits cv. Alphonso. Acta Hort., 231:670-674.

3

A

A

A

- Shyamalamma, S., T.K. Subbaiah, G.K. Mukunda and M.M. Khan. 1995. Effect of Maturity at harvest on the post harvest behaviour of Mango. Phala Sanskasana, Division Hort. Univ. Agril. Sci. GKVK, Bangalore, Karnataka, India, 1 (107): 43-47.
- Siddiqui, A. B. and F. M. Scanlan. 1995. Nutritive value of fruits. In: Fruit Production Manual. Hort. Res.Dev. Proj. (FAO/UNDP/ADB Project: BGD/87/ 025), pp. 1-286.
- Simao, S. and F.P. Gomes. 1996. Distribution of Brix in mango. Revistade Agric., **69** (1): 3-10 [Cited from Hort. Abst. **67** (10): 1152, 1997].
- Singh, A. 1990. Banana. In: Fruit physiology and production, kalyani Publishers, New Delhi, Pp. 285-300.
- Singh, J. N., A. Pinaki and B. B. Singh. 2000. Effect of GA3 and plant extracts on storage behavior of mango (*Mangifera indica* L.) cv. Langra. Haryana J. Hort. Sci., **29** (3/4):199-200.
- Singh, K.K., G.S. Nijjar and G. Singh. 1967. Cold storage studies on Dashehari mango. J. Res.. Ludhiana, **4**: 516.
- Singh, L. B. 1968. The mango: Botany, Cultivation and Utilization, Leonard Hall, London pp. 363-376.

- Singh, S. M. 1960. Studies on the mango shoot gall in the Tarai region of Uttar Pradesh. Its causes and control. II. Distribution, Nature, Extent and Intensity of damage and bionomics of the pest. Hort. Adv., **4**: 97.
- Singh, S., P. Kumar, V.S. Brahmachari and D.M. Singh. 1995. Effect of post harvest spray of GA3 and Ethrel on storage life of mango cv. Amrapali. Orissa J. Hort., 23 (1/2): 112-118.
- Sornsrivichai, J., P. Anusadorn, C. Oogaki and H. Gemma. 1989. Storage life and quality of mango (cv. 'Keaw Sawoey') fruit stored in seal packaging by plastic films and low pressure at different temperature. Nettai Noyyo- Jap. J. Trop. Agric., 33 (1): 6-17.
- Srinivas, R. N., T. V. Reddy, P. C. Ravi and B. V. C. Reddy. 1996. Postharvest loss Assessment of 'Totapuri' and 'Alphonso' mango. J. Food Sci. Tech., 34 (1):70-72.
- Srivastava, H.C. 1967. Grading, storage and marketing. In: The Mango: A handbook, ICAR, New Delhi. pp. 106-375.
- Takuji, I., K. Sasaki and Y. Yoshida. 1997. Changes in respiration rate, saccharide and organic acid contents during the development and ripening of mango fruit (*Mangifera inndica* L.) cultivated in a plastic house. J. Japa. Soc. Hort. Sci., 66 (3-4): 629-635.
- Tandon, D.K. and S.K. Kalra. 1983. Changes in sugar, starch and amylase activity during development of mango fruit cv. Dashehari. J. Hort. Sci., 58 (3): 449-453.
- Tandon, D.K., S.K. Kalra and H.C. Lohani. 1985. Changes in some carbohydrates in developing mango fruit cv. Langra and Mallika. Indian J. Hort., 42(315): 222-228.

X

A.

\$

- Tang, Y.L., Y.C. Zhou and X.P. Pan. 1997. Effect of plant growth regulator and prochloraz on postharvest diseases of Zihua mango fruits. Acta Phytophyl. Sinica, 24 (1): 70-74.
- Tefera, A., T. Seyoum and K. Worldetsadik. 2007. Effect of disinfection, packaging and storage environment on the shelf life of mango. Biosystem Engineering. Oxford, UK: Elsevier 96 (2): 201-212.

- Thangaraj, T. and I. Irulappan. 1988. Effectiveness of hot water treatment, waxing and cool chamber storage in prolonging the shelf life of mango. South Indian Hort., **36** (6): 327-328.
- Tripathi, J. S. 1988. Postharvest changes during storage and ripening of Gaurjeet mango fruit. Madras Agril. J., **75** (3-4): 155-156.
- Tripathi, J. S. 1988. Postharvest changes during storage and ripening of Gaurjeet mango fruits. Agril. Sci. digest, India, **8** (4): 191-192.
- Tripathi, V. K., H. B. Ram, S. P. Jain and S. Singh. 1981. Changes in developing banana fruit. Prog. Hort., **13** (1): 45-53.
- Tsuda, T., K. Chachin and Y. Ueda. 1999. Studies on keeping quality of imported 'Carabao' mango fruit from the Philippines. J. Jap. Soc. Hort. Sci. **68** (3): 669-674.
- Upadhyay, I. P., A. Noomhorm and S. G. Ilangantileke. 1994. Effects of gamma irradiation and hot water treatment on the shelf life and quality of Thai mango cv. Red. Aust. Centre for Intl. Agril. Res., pp. 348-351.
- Upadhyay, N. P. and B. M. Tripathi. 1985. Postharvest changes during storage and ripening of Gaurjeet mango (*Mangifera indica* L.) fruits. Prog. Hort., **17** (1): 25-27.

x

2

A

4

- Wanlapha, T., S. Thanom and A. Pakhini. 1980. Postharvest diseases and storage techniques of mango fruits. Proc. 5th conf. on statement of Results and Res. Pl. Path. Div., Bangkok, p. 9.
- Wasker, D.P. and S.D. Masalkar. 1997. Effect of hydro-cooling and Bavistin dip on the shelf life and quality of mango during storage under various environments. Acta Hort., 455: 687-695.
- Watt, B. K. and A. L. Merrill. 1963. Composition of Foods, Agriculture Hand Book No. 8. Washington D.C., USDA.
- Wavhal, K. N. and P. W. Athale. 1989. Studies to prolong shelf life of mango fruits. Acta Hort., **231**: 771-775.
- Wills, R. B. H., W. B. McGlossan, D. Graham, T. H. Lee and E. G. Hall. 1989.
 Postharvest: An introduction to the physiology and handling of fruits and vegetables. An Avi Book. Van Nostrand Reinhold, New York, P.173.

- Youlin, T., Z. Yuchan and P. Xiaoping. 1997. Effect on plant growth regulators and prochloraz on postharvest disease of Zihua mango fruits. Acta Phytophylacica Sinica, 24 (1): 70-74 [Cited from Hort. Abstr., 68 (2):1876, 1998].
- Yuniarti and Suhardi. 1992. Ripening retardation of 'Arumanis' mango. ASEAN Food J., **7** (4):207-208.
- Yuniarti. 1980. Physico-chemical changes of Arumanis mangoes during storage at ambient temperature. Bul. Penelition Hortikultura, Indonesia, 8 (11): 11-17 [Cited from Hort. Abstr., 53 (12): 9051, 1983].

X

X

*

Z

-

- Zambrano, J., S. Briceno, C. Medenz, J. Manzano and E. Castellanose. 1997.
 Changes during storage of wax coated mango fruits. Agronomia Trop., 47
 (1): 5-15 [Cited from Hort. Abstr., 68 (10): 9127, 1998].
- Zhu, D. M., M. chen, Z. Z. Li, Y. J. Deng, M. J. Yu. and P. Yin. 2002. Effect of Mango hot water treatment and artificial ripening on its quality. Transactions, Chinese Soc. Agril. Engg., 18(3)., 18 (3): 139 – 141.



			Temp	erature (⁰ C)	Pa	Total					
Year	Month	R	Room temperature			Out side temperature					
	Max.	Min.	Mean	Max.	Min.	Mean	Max.	Min.	Mean	(mm)	
	May	32.24	22.22	27.23	38.87	23.39	31.13	73.90	67.65	70.78	76(6)*
2005	June	33.55	29.77	31.66	37.78	27.65	32.72	82.87	78.63	80.75	135(7)
	July	32.25	28.75	30.50	34.92	25.58	30.25	86.35	83.13	84.74	391(17)
	May	32.62	22.45	27.54	39.56	24.18	31.87	75.35	67.06	71.21	103(6)
2006	June	33.85	29.92	31.89	38.45	27.92	33.19	70.03	62.43	66.23	105(8)
	July	32.58	28.97	30.78	35.38	26.14	30.76	87.71	85.42	86.57	484(18)
	May	32.62	22.45	27.54	39.63	23.75	31.69	85.87	76.90	81.39	126(9)
2007	June	33.85	29.92	31.89	38.72	27.62	33.17	91.43	88.77	90.10	450(13)
	July	32.58	28.97	30.78	35.78	26.45	31.12	85.00	81.61	83.31	344(19)

Appendix 1.1 Monthly records of maximum and minimum temperature, relative humidity and rainfall during the study period at the Department of Biochemistry and Molecular Biology, University of Rajshahi, Bangladesh

1

Y

*

+

280

* Figures in parentheses are the number of rainy days

-

*

Appendix 1.2 Analysis of variance of data on edible portion and pulp to peel ratio of mango as influenced by varieties and different storage treatments

Sources of variation		Mean sum of square										
	of	Edible portion (%) at different days					Pulp to peel ratio at different days					
	freedom	Initial	3	6	9	12	Initial	3	6	9	12	
Replication	2	0.081	0.080	0.192	0.083	0.085	0.001	0.001	0.001	0.001	0.001	
Variety -V	1	8.644**	8.930**	10.704**	9.641**	10.017**	1.613***	0.706***	1.210***	0.348***	0.372***	
Storage treatment -T	5	5.295**	16.617***	46.816***	60.252***	42.608***	0.051**	4.657***	15.127***	19.575***	27.698***	
V × T	5	0.555 ^{NS}	0.559 ^{NS}	0.746 NS	0.499 NS	0.466 NS	0.025 NS	0.046**	0.059**	0.151***	0.156***	
Error	22	1.078	1.062	0.992	1.063	1.105	0.011	0.011	0.011	0.011	0.011	

Appendix 1.3 Analysis of variance of data on physiological weight loss of mango and moisture content of mango pulp as influenced by varieties and different storage treatments 281

Sources of	Degree				Ν	lean sum of s	square			
variation	of	Physiologi	ical weight lo	oss (%) at dif	ferent days		Moisture co	ntent (%) at o	different days	
	freedom	3 D	6	9	12	Initial	3	6	9	12
Replication	2	0.000	0.000	0.001	0.001	0.001	0.001	0.000	0.001	0.001
Variety -V	1	4.000***	2.806***	4.410***	6.891***	37.027***	42.968***	39.837***	39.942***	40.006***
Storage treatment -T	5	26.028***	67.976***	207.427***	312.376***	0.093***	1.516***	6.950***	7.073***	8.061***
V × T	5	0.086***	0.140***	0.219***	0.109***	0.134 ***	0.054**	0.020 NS	0.019 NS	0.018 ^{NS}
Error	22	0.003	0.011	0.011	0.011	0.011	0.011	0.011	0.011	0.011

>

Appendix represents * indicates 5% level of probability ** indicates 1% level of probability

7

7

*** indicates 0.1% level of probability NS- Non significant

4

N.

Appendix 1.4 Analysis of variance of data on dry matter content and ash content of mango pulp as influenced by varieties and different storage treatments

2

*

М.

1

Sources of variation	Degree of freedom	Mean sum of square									
			Ash content (%) at different days								
		Initial	3	6	9	12	Initial	3	6	9	12
Replication	2	0.001	0.001	0.001	0.001	0.001	0.000	0.000	0.000	0.000	0.000
Variety -V	1	36.663***	42.968***	39.942***	39.942***	40.217***	0.126***	0.137***	0.133***	0.126***	0.130***
Storage treatment -T	5	0.106***	1.516***	6.978***	7.073***	8.053***	0.001 NS	0.005***	0.022***	0.025***	0.025***
V × T	5	0.133***	0.054**	0.018 NS	0.019 NS	0.017 ^{NS}	0.0008 ^{NS}	0.0007 ^{NS}	0.0006 ^{NS}	0.0007 ^{NS}	0.0009 NS
Error	22	0.011	0.011	0.011	0.011	0.012	0.0003	0.0003	0.0003	0.0003	0.0003

Appendix 1.5 Analysis of variance of data on vitamin C content of mango pulp as influenced by varieties and different storage treatments 282

Sources of variation	Degree of	Mean sum of square										
Sources of variation	freedom		Vitamin C content (mg/100 g) at different days									
		Initial	3	6	9	12						
Replication	2	0.000	0.000	0.000	0.000	0.028						
Variety -V	1	81607.350***	58129.21***	16218.022***	1464.976***	379.405***						
Storage treatment -T	5	24.382***	128.269***	887.248***	435.034***	244.965***						
V × T	5	13.473***	4.170*	163.464***	76.972***	4.316***						
Error	22	1.091	1.091	0.551	0.011	0.048						

7

Appendix represents * indicates 5% level of probability ** indicates 1% level of probability

1

*** indicates 0.1% level of probability NS-Non significant
Appendix 1.6 Analysis of variance of data on titratable acidity and pH of mango pulp as influenced by varieties and different storage treatments

*

Sources of De	Dograa of					Mean su	m of square	9			
variation	freedom	Т	itratable acid	ity (%) at d	ifferent da	ys	1	Pulp	pH at differ	ent days	
		Initial	3	6	9	12	Initial	3	6	9	12
Replication	2	0.000	0.016	0.001	0.000	0.000	0.001	0.001	0.001	0.001	0.001
Variety -V	1	5.153***	2.953***	0.449***	0.360***	0.053***	0.123**	0.161***	0.303***	0.490***	0.250***
Storage treatment -T	5	0.020***	1.426***	0.615***	0.561***	0.484***	0.039*	1.138***	2.859***	5.788***	5.082***
V × T	5	0.010*	0.023 ^{NS}	0.007 ^{NS}	0.017***	0.015***	0.015 ^{NS}	0.010 ^{NS}	0.038*	0.052**	0.082***
Error	22	0.003	0.017	0.011	0.001	0.0012	0.011	0.011	0.011	0.011	0.011

N

4

N 83 *

1

14

Appendix 1.7 Analysis of variance of data on total soluble solid and total sugar content of mango pulp as influenced by varieties and different storage treatments

						Mean sum	n of square				
Sources of variation	Degree		TSS conter	nt (%) at diffe	erent days		Т	otal sugar co	ntent (%) at	different day	'S
	freedom	Initial	3	6	9	12	Initial	3	6	9	12
Replication	2	0.001	0.001	0.001	0.001	0.001	0.001	0.000	0.015	0.001	0.020
Variety -V	1	7.618***	7.775***	8.094***	7.398***	8.266***	1.488***	1.361***	1.228***	1.416***	1.881***
Storage treatment -T	5	0.047**	28.081***	113.093***	87.549***	67.398***	0.010 ^{NS}	15.319***	72.049***	100.311***	61.814***
V × T	5	0.004 ^{NS}	0.008 ^{NS}	0.015 NS	0.032 *	0.090***	0.004 ^{NS}	0.005 ^{NS}	0.004 ^{NS}	0.028 ^{NS}	0.054 ^{NS}
Error	22	0.011	0.012	0.011	0.011	0.011	0.011	0.011	0.031	0.011	0.030

Appendix represents * indicates 5% level of probability ** indicates 1% level of probability

*** indicates 0.1% level of probability NS- Non significant

Appendix 1.8 Analysis of variance of data on reducing and non-reducing sugar content of mango pulp as influenced by varieties and different storage treatments

*

						Mean su	m of square	2			
Sources of variation	of	Red	ucing sugar	content (%)	at different	days	Non-	reducing su	gar content	(%) at differ	ent days
	freedom	Initial	3	6	9	12	Initial	3	6	9	12
Replication	2	0.000	0.000	0.000	0.000	0.001	0.000	0.001	0.001	0.001	0.001
Variety -V	1	0.270***	0.281***	0.286***	0.202***	0.292***	0.481***	0.348***	0.533***	0.548***	0.442***
Storage treatment -T	5	0.013**	2.145***	13.188***	12.031***	4.016***	0.013 ^{NS}	6.892***	27.117***	46.493***	36.739***
V×T	5	0.005 ^{NS}	0.004 ^{NS}	0.004 ^{NS}	0.007 *	0.011**	0.014 ^{NS}	0.005 ^{NS}	0.024 ^{NS}	0.037*	0.020 ^{NS}
Error	22	0.003	0.003	0.003	0.003	0.003	0.011	0.011	0.011	0.011	0.011

Appendix 1.9 Analysis of variance of data on crude fibre and lipid content of mango pulp as influenced by varieties and different storage treatments 284

	Deserve					Mean s	um of squar	е			
Sources of Variation	freedom		Crude fib	re (%) at d	ifferent days	5		Lipid cont	ent (%) at d	ifferent days	5
		Initial	3	6	9	12	Initial	3	6	9	12
Replication	2	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
Variety -V	1	0.148***	0.128***	0.111***	0.098***	0.096***	0.022***	0.022***	0.026***	0.031***	0.040***
Storage treatment -T	5	0.002**	0.020***	0.075***	0.131***	0.222***	0.001***	0.051***	0.161***	0.142***	0.072***
V × T	5	0.003**	0.002*	0.001*	0.001***	0.001***	0.002***	0.001***	0.001***	0.001***	0.001**
Error	22	0.001	0.0012	0.0013	0.0013	0.0003	0.0003	0.0012	0.0013	0.0011	0.0011

7

Y

Appendix represents * indicates 5% level of probability ** indicates 1% level of probability

*** indicates 0.1% level of probability

NS-Non significant

18

×

4

×

7

Appendix 1.10 Analysis of variance of data on water soluble protein and phosphorus content of mango pulp as influenced by varieties and different storage treatments

*

X

11

1

Sources of Degr	Deeree					Mean su	um of squa	ire			
variation	of	Water so	luble prote	ein conten	t (%) at di	ifferent days	Pho	sphorus cor	ntent (mg/	100 g) differe	ent days
	freedom	Initial	3	6	9	12	Initial	3	6	9	12
Replication	2	0.000	0.000	0.000	0.000	0.000	0.001	0.001	0.001	0.001	0.001
Variety -V	1	0.092***	0.123***	0.097***	0.116***	0.119***	80.102** *	71.487***	71.487***	71.487***	69.890***
Storage treatment -T	5	0.005***	0.061***	0.297***	0.317***	0.184***	0.244***	4.667***	16.274***	18.563***	10.586***
V × T	5	0.003***	0.001**	0.001*	0.002***	0.001*	0.013 ^{NS}	0.196***	0.197***	0.199***	0.186***
Error	22	0.0002	0.0003	0.0010	0.0012	0.0013	0.011	0.011	0.011	0.011	0.011

285

笋.

Appendix 1.11 Analysis of variance of data on potassium and calcium content of mango pulp as influenced by varieties and different storage treatments

Sources of variation	Degree					Mean	sum of squa	are			
Sources of Variation	of	Pot	assium con	tent (%) at	t different d	ays	C	alcium conte	nt (mg/100 g) at different	days
	freedom	Initial	3	6	9	12	Initial	3	6	9	12
Replication	2	0.000	0.000	0.000	0.000	0.000	0.001	0.002	0.001	0.001	0.001
Variety -V	1	0.009***	0.011***	0.011***	0.010***	0.011***	27.826***	27.878***	27.405***	27.563***	27.248***
Storage treatment -T	5	0.002***	0.003***	0.004***	0.004***	0.002***	0.049**	14.937***	73.161***	69.326***	40.586***
V × T	5	0.0008 NS	0.0007 ^{NS}	0.0006 ^{NS}	0.0004 ^{NS}	0.0005 ^{NS}	0.006 ^{NS}	0.003 ^{NS}	0.002 ^{NS}	0.003 ^{NS}	0.012 ^{NS}
Error	22	0.00012	0.00013	0.00014	0.0002	0.00015	0.011	0.011	0.011	0.011	0.011

Appendix represents

indicates 5% level of probability
 indicates 1% level of probability

*** indicates 0.1% level of probability NS- Non significant

Appendix 1.12 Analysis of variance of data on magnesium and copper content of mango pulp as influenced by varieties and different storage treatments

*

*

						Mean su	m of square				
Sources of variation	Degree of	Magne	sium conten	t (mg/100	g) at differe	nt days	Co	pper content	(mg/100 g) a	at different d	ays
	freedom	Initial	3	6	9	12	Initial	3	6	9	12
Replication	2	0.001	0.001	0.001	0.001	0.002	0.000	0.000	0.000	0.000	0.000
Variety -V	1	14.138***	12.285***	12.076***	12.781***	12.555***	0.008***	0.004***	0.004***	0.003***	0.003***
Storage treatment -T	5	0.115***	1.361***	3.412***	1.914***	2.531***	0.001***	0.002***	0.005***	0.010***	0.014***
V × T	5	0.067***	0.038*	0.039*	0.026 ^{NS}	0.028 ^{NS}	0.0005 ^{NS}	-0.0006 ^{NS}	-0.0006 ^{NS}	0.0007 ^{NS}	0.0007 ^{NS}
Error	22	0.011	0.011	0.011	0.011	0.012	0.00012	0.00013	0.00014	0.00015	0.00016

286

۶.

4

Appendix 1.13 Analysis of variance of data on iron and manganese content of mango pulp as influenced by varieties and different storage treatments

						Mean s	um of square				
Sources of variation	Degree of	Iro	n content (m	ng/100 g) a	at different	days	Mang	anese conter	nt (mg/100 g) at different	days
	freedom	Initial	3	6	9	12	Initial	3	6	9	12
Replication	2	0.001	0.001	0.003	0.001	0.001	0.000	0.000	0.001	0.000	0.000
Variety -V	1	9.923***	10.240***	9.517***	9.456***	10.176***	0.325***	0.221***	0.253***	0.207***	0.207***
Storage treatment -T	5	0.189***	1.505***	5.132***	3.071***	5.811***	0.004***	0.061***	0.202***	0.117***	0.147***
V × T	5	0.007 ^{NS}	0.004 ^{NS}	0.021 ^{NS}	0.021 ^{NS}	0.015 ^{NS}	0.001***	0.001***	0.002 ^{NS}	0.002 ^{NS}	0.002 ^{NS}
Error	22	0.011	0.011	0.011	0.011	0.011	0.00012	0.00013	0.003	0.003	0.003

Appendix represents * indicates 5% level of probability ** indicates 1% level of probability

*** indicates 0.1% level of probability NS- Non significant

À.

*

Appendix 1.14 Analysis of variance of the data on zinc content and shelf life of mango as influenced by varieties and postharvest treatments

*

.

4

2

	5			Mean s	sum of square		
Sources of Variacion	freedom		Zinc con	tent (mg/100 g) a	at different days		Shelf life
	needom	Initial	3	6	9	12	Total days
Replication	2	0.000	0.000	0.000	0.000	0.000	0.083
Variety -V	1	0.292***	0.265***	0.308***	0.303***	0.286***	13.444***
Storage treatment -T	5	0.009*	0.004 ^{NS}	0.035***	0.088***	0.162***	644.867***
V × T	5	0.001 NS	0.001 ^{NS}	0.003 ^{NS}	0.002 ^{NS}	0.003**	1.311*
Error	22	0.003	0.003	0.003	0.003	0.00014	0.477

287

A

Appendix represents

indicates 5% level of probability
 indicates 1% level of probability

 \mathbb{R}^{2}

*** indicates 0.1% level of probability NS- Non significant

+

Appendix 2.1 Analysis of variance of data on physiological weight loss of mango and moisture content of mango pulp as influenced by varieties and different doses of Maleic hydrazide solution

4

					Me	ean sum of squ	are			
Sources of	Degree	Physiolo	gical weight lo	oss (%) at diff	ferent days		Moisture cor	ntent (%) at d	ifferent days	
	freedom	3	6	9	12	Initial	3	6	9	12
Replication	2	0.001	0.007	0.001	0.001	0.010	0.003	0.012	0.006	0.012
Variety -V	1	4.914***	4.646***	4.779***	4.779***	45.375***	46.204***	46.204***	47.040***	46.121***
Maleic hadrazide- M	3	7.260***	7.397***	7.194***	7.194***	0.069 ^{NS}	4.386***	8.776***	5.464***	3.885***
V × M	3	0.042*	0.026*	0.033***	0.033***	0.003 ^{NS}	0.006 ^{NS}	0.006 ^{NS}	0.013 ^{NS}	0.007 ^{NS}
Error	14	0.008	0.005	0.003	0.003	0.103	0.046	0.042	0.051	0.042

.

4

Appendix 2.2 Analysis of variance of data on dry matter and ash content of mango pulp as influenced by varieties and different doses of Maleic hydrazide solution

						Mean sum o	f square				
Sources of variation	Degree of	l	Dry matter co	ontent (%) at	different day	S		Ash conter	nt (%) at dif	ferent days	
	freedom	Initial	3	6	9	12	Initial	3	6	9	12
Replication Variety -V	2	0.004 46.204***	0.009 46.204***	0.012 46.204***	0.004 47.040***	0.013 45.844***	0.001 0.113**	0.001 0.120**	0.001 0.115**	0.001 0.116**	0.002 0.111**
Maleic hadrazide- M	3	0.056*	4.386***	8.776***	5.464***	3.956***	0.0009 ^{NS}	0.011 NS	0.021 NS	0.022 NS	0.010 ^{NS}
Error	3 14	0.006	0.023	0.006	0.013	0.006	0.0008	0.0008	0.0009	0.0007	0.0009

Appendix represents

* indicates 1% level of probability

*** indicates 0.1% level of probability NS- Non significant

** indicates 1% level of probability N

-

14

Appendix 2.3 Analysis of variance of data on vitamin C content of mango pulp as influenced by varieties and different doses of Maleic hydrazide solution

				Mean sum of squar	е		
Sources of variation	Degree of		Vitamin	C content (mg/100 g) at	t different days		
	freedom	Initial	3	6	9	12	
Replication	2	0.125	0.005	0.005	0.000	0.001	
Variety -V	1	47680.988***	36539.889***	9594.001***	3153.334 ***	335.254***	
Maleic hadrazide M	3	14.690***	622.621***	433.706***	242.224***	80.984***	
V × M	3	9.042**	191.345***	143.046***	90.824***	0.094*	
Error	14	1.131	0.658	0.045	0.037	0.018	

Appendix 2.4 Analysis of variance of data on titratable acidity and pH of mango pulp as influenced by varieties and different doses of Maleic hydrazide solution

N							Mean sum	of square				
68	Sources of variation	Degree of		Titratable ac	idity (%) at d	lifferent days			Pulp	pH at differ	ent days	
		freedom	Initial	3	6	9	12	Initial	3	6	9	12
	Replication	2	0.002	0.002	0.000	0.000	0.000	0.007	0.000	0.001	0.001	0.001
	Variety -V	1	9.526***	0.196***	0.132***	0.083***	0.025***	0.042*	0.060*	0.060**	0.060*	0.060*
	Maleic hadrazide- M	3	0.013 ^{NS}	0.381***	0.324***	0.297***	0.121***	0.065**	4.00***	1.655***	3.295***	1.960***
	V × M	3	0.002 ^{NS}	0.032*	0.003 ^{NS}	0.001 ^{NS}	0.001 ^{NS}	0.005 ^{NS}	-0.0002 ^{NS}	0.0009 ^{NS}	0.0007 ^{NS}	-0.0002 ^{NS}
0.	Error	14	0.010	0.008	0.007	0.003	0.001	0.008	0.010	0.004	0.008	0.011

Appendix represents

* indicates 1% level of probability

*** indicates 0.1% level of probability NS- Non significant

** indicates 1% level of probability NS

Appendix 2.5 Analysis of variance of data on total soluble solid and total sugar content of mango pulp as influenced by varieties and different doses of Małeic hydrazide solution

A'

-

a a						Mean su	m of square	2			
Sources of variation	of		TSS cont	ent (%) at dif	ferent days			Total sugar	content (%) a	at different da	ys
	freedom	Initial	3	6	9	12	Initial	3	6	9	12
Replication	2	0.001	0.004	0.001	0.009	0.004	0.001	0.001	0.001	0.004	0.001
Variety -V	1	7.260***	7.260***	0.094*	2.734***	0.240***	1.354***	1.500***	1.500***	1.354***	1.354***
Maleic hadrazide- M	3	0.100**	38.575***	157.034***	49.594***	39.375***	0.064**	12.423***	105.555***	104.043***	44.149***
V × M	3	0.0007 ^{NS}	0.0009 ^{NS}	0.424***	0.144***	5.700***	0.004 ^{NS}	0.003 ^{NS}	0.003 NS	0.001 ^{NS}	0.001 ^{NS}
Error	14	0.011	0.011	0.011	0.010	0.011	0.011	0.011	0.011	0.011	0.011

Appendix 2.6 Analysis of variance of data on reducing and non reducing sugar content of mango pulp as influenced by varieties and different doses of Maleic hydrazide solution

290

14.

Courses of	Design					Mean sur	n of square				
variation	of	Red	ucing sugar	content (%)	at different o	days	Non	reducing su	gar content (%) at different	ent days
	freedom	Initial	3	6	9	12	Initial	3	6	9	12
Replication	2	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.001
Variety -V	1	0.437***	0.540***	0.540***	0.540***	0.540***	0.277***	0.240***	0.240***	0.184**	0.184**
Maleic hadrazide- M	3	0.042*	2.250***	16.105***	10.630***	6.105***	0.008 ^{NS}	4.170***	40.000***	48.662***	21.921***
V × M	3	0.001 ^{NS}	0.0006 ^{NS}	-0.004 ^{NS}	0.0008 ^{NS}	0.0009 ^{NS}	0.005 NS	0.003 ^{NS}	0.003 ^{NS}	0.001 ^{NS}	0.001 ^{NS}
Error	14	0.011	0.011	0.011	0.011	0.011	0.011	0.011	0.011	0.011	0.011

Appendix represents

* indicates 1% level of probability

** indicates 1% level of probability

*** indicates 0.1% level of probability

lity NS- Non significant

*

4

Appendix 2.7 Analysis of variance of data on crude fibre and lipid content of mango pulp as influenced by varieties and different doses of Maleic hydrazide solution

+

1

4

						Mean sum	of square				
Sources of variation	Degree of		Crude fibre	(%) at diffe	erent days			Lipid cont	ent (%) at c	lifferent days	5
	freedom	Initial	3	6	9	12	Initial	3	6	9	12
Replication	2	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
Variety -V	1	0.086***	0.086***	0.076***	0.076***	0.104***	0.001*	0.001*	0.001*	0.002***	0.007***
Maleic hadrazide- M	3	0.005 ^{NS}	0.150***	0.335***	0.245***	0.090***	0.002***	0.023***	0.134***	0.143***	0.036***
V × M	3	0.0009 ^{NS}	0.0008 ^{NS}	0.0006 ^{NS}	0.0007 ^{NS}	0.001 ^{NS}	0.0007 ^{NS}	0.0008 ^{NS}	0.0008 ^{NS}	0.001**	0.001***
Error	14	0.003	0.003	0.003	0.003	0.003	0.0004	0.0003	0.00014	0.00015	0.0016

Appendix 2.8 Analysis of variance of data on water soluble protein and phosphorus content of mango pulp as influenced by varieties and different doses of Maleic hydrazide solution

291

d,

4

×

						Mean su	m of square				
Sources of variation	Degree of	Water so	luble protei	n content (%) at differe	ent days	Phos	sphorus cont	ent (mg/100	g) different	days
	freedom	Initial	3	6	9	12	Initial	3	6	9	12
Replication	2	0.001	0.001	0.004	0.001	0.001	0.001	0.001	0.001	0.001	0.001
Variety -V	1	0.034 ^{NS}	0.043 NS	0.090*	0.057*	0.020 ^{NS}	64.452***	40.482***	40.249***	36.977***	20.981***
Maleic hadrazide- M	3	0.001 ^{NS}	0.060*	0.316***	0.335***	0.023 ^{NS}	0.778***	14.867***	32.232***	24.572***	4.576***
V × M	3	0.002 NS	0.0008 ^{NS}	0.001 ^{NS}	0.002 NS	0.001 ^{NS}	0.062*	1.909***	1.905***	1.829***	0.739***
Error	14	0.009	0.011	0.011	0.011	0.012	0.011	0.011	0.011	0.011	0.011

Appendix represents

* indicates 1% level of probability

** indicates 1% level of probability

*** indicates 0.1% level of probability

oility NS- Non significant

Appendix 2.9 Analysis of variance of data on potassium and calcium content of mango pulp as influenced by varieties and different doses of Maleic hydrazide solution

						Mean s	um of square	2			
Sources of variation	Degree	Pc	tassium cor	ntent (%) a	t different d	ays	Cal	cium conten	t (mg/100 g)	at different o	lays
	freedom	Initial	3	6	9	12	Initial	3	6	9	12
Replication	2	0.000	0.000	0.000	0.000	0.000	0.001	0.001	0.001	0.001	0.001
Variety -V	1	0.011***	0.018***	0.011***	0.005***	0.005***	19.820***	22.349***	22.465***	22.640***	22.640***
Maleic hadrazide- M	3	0.001**	0.002***	0.001***	0.002***	0.0004*	0.213***	17.977***	104.242***	93.451***	23.167***
V × M	3	0.0008 ^{NS}	0.0006 ^{NS}	0.0003 ^{NS}	0.0004 ^{NS}	0.0006 ^{NS}	0.173***	0.115***	0.119***	0.104**	0.101**
Error	14	0.00011	0.00012	0.00013	0.00014	0.00015	0.011	0.011	0.011	0.011	0.011

Appendix 2.10 Analysis of variance of data on in magnesium and copper content of mango pulp as influenced by varieties and different doses of maleic hydrazide

292

4

2

x

						Mean sun	n of square				
variation	of	Magne	sium conter	nt (mg/100	g) at differe	ent days	Cop	oper conten	t (mg/100 g) at different	: days
	freedom	Initial	3	6	9	12	Initial	3	6	9	12
Replication	2	0.002	0.001	0.001	0.002	0.001	0.000	0.000	0.000	0.000	0.000
Variety -V	1	5.723***	3.488***	8.284***	2.350***	1.084***	0.005***	0.007***	0.005***	0.005***	0.003***
Maleic hadrazide- M	3	0.779***	1.255***	2.695***	0.953***	14.328***	0.0007 ^{NS}	0.001***	0.005***	0.011***	0.017***
V × M	3	0.110**	0.012 ^{NS}	0.206***	0.041 ^{NS}	0.059*	0.0008*	0.0005 ^{NS}	0.0009 ^{NS}	0.0003 ^{NS}	0.0003*
Error	14	0.013	0.011	0.011	0.013	0.011	0.00017	0.00021	0.00018	0.00019	0.00020

Appendix represents

* indicates 1% level of probability

*** indicates 0.1% level of probability NS- Non significant

** indicates 1% level of probability

2

1

Appendix 2.11 Analysis of variance of data on iron and manganese content of mango pulp as influenced by varieties and different doses of Maleic hydrazide solution

4

X

Courses of	Desma					Mean su	m of square	9			
variation	of	Irc	on content ((mg/100 g) a	t different o	days	Man	ganese cont	ent (mg/100	g) at differ	ent days
	freedom	Initial	3	6	9	12	Initial	3	6	9	12
Replication	2	0.001	0.001	0.001	0.001	0.001	0.000	0.000	0.000	0.000	0.000
Variety -V	1	3.768***	3.650***	3.912***	3.888***	4.010***	0.194***	0.222***	0.217***	0.234***	0.240***
Maleic hadrazide- M	3	0.362***	1.763***	10.657***	4.248***	3.985***	0.005***	0.092***	0.318***	0.163***	0.115***
V × M	3	0.006 ^{NS}	0.006 ^{NS}	0.004 ^{NS}	0.003 ^{NS}	0.002 ^{NS}	0.002 ^{NS}	0.0006 ^{NS}	0.001 ^{NS}	0.001 ^{NS}	0.001 ^{NS}
Error	14	0.011	0.011	0.011	0.011	0.011	0.00011	0.00012	0.00013	0.00014	0.00015

4

4

A

Appendix 2.12 Analysis of variance of data on zinc content of mango pulp and shelf life of mango as influenced by varieties and different doses of Maleic hydrazide solution

	1. A			Mean	sum of square		
Sources of	Degree of		Zinc conte	nt (mg/100 g) a	t different days		Shelf life
Variation	rreedom	Initial	3	6	9	12	Total days
Replication	2	0.000	0.000	0.000	0.000	0.000	0.375
Variety -V	1	0.317***	0.094***	0.098***	0.090***	0.105***	10.667***
Maleic hadrazide- M	3	0.004**	0.009***	0.032***	0.077***	0.144***	61.444***
V × M	3	0.0006 ^{NS}	0.0007 ^{NS}	0.0005 ^{NS}	0.001 ^{NS}	0.004**	0.333 NS
Error	14	0.00016	0.00017	0.00018	0.00019	0.00020	0.518

Appendix represents * indicates 1% level of probability

4

293

¥

*** indicates 0.1% level of probability

** indicates 1% level of probability

NS- Non significant

Appendix 3.1 Analysis of variance of data on physiological weight loss of mango and moisture content of mango pulp as influenced by varieties and different doses of Gibberellic acid solution

4

4

it

A

					M	ean sum of squ	are			
Sources of variation	Degree of	Physiol	ogical weight	loss (%) at d	ifferent days		Moisture cor	ntent (%) at d	ifferent days	
	freedom	3	6	9	12	Initial	3	6	9	12
Replication	2	0.001	0.049	0.003	0.000	0.054	0.020	0.005	0.002	0.004
Variety -V	1	4.860***	7.673***	4.815***	4.779***	22.330***	26.355***	27.735***	25.010***	25.215***
Gibberellic acid- G	3	7.342***	6.178***	7.131***	7.651***	0.128 ^{NS}	5.919***	9.030***	5.897***	4.015***
V × G	3	0.052***	0.176*	0.027*	0.049**	0.068 ^{NS}	0.077 ^{NS}	0.085 ^{NS}	0.010 ^{NS}	0.015 ^{NS}
Error	14	0.005	0.044	0.005	0.005	0.117	0.058	0.045	0.039	0.041

Appendix 3.2 Analysis of variance of data on dry matter and ash content of mango pulp as influenced by varieties and different doses of Gibberellic acid solution

						Mean sum o	f square				
Sources of variation	Degree of		Dry matter o	content (%) a	t different day	/S		Ash conter	nt (%) at di	fferent days	
	freedom	Initial	3	6	9	12	Initial	3	6	9	12
Replication	2	0.002	0.002	0.004	0.002	0.004	0.001	0.008	0.001	0.003	0.001
Variety -V	1	24.301***	24.301***	26.460***	25.010***	25.215***	0.069*	0.056 ^{NS}	0.069*	0.077*	0.063*
Gibberellic acid- G	3	0.141*	4.998***	9.225***	5.814***	4.015***	0.001 ^{NS}	0.004 ^{NS}	0.024 ^{NS}	0.019 ^{NS}	0.011 ^{NS}
V × G	3	0.006 ^{NS}	0.006 ^{NS}	0.030 ^{NS}	0.010 ^{NS}	0.015 ^{NS}	0.0006 ^{NS}	0.0008 ^{NS}	0.0003 ^{NS}	0.001 ^{NS}	0.0007 ^{NS}
Error	14	0.039	0.039	0.041	0.039	0.041	0.010	0.016	0.010	0.012	0.011

Appendix represents

* indicates 5 % level of probability

*** indicates 0.1% level of probability

** indicates 1% level of probability

NS means non significant

4

Y

h

Appendix 3.3 Analysis of variance of the data on vitamin C content of mango pulp as influenced by varieties and different doses of Gibberellic acid solution

Sources of uprintion	Dearce of			Mean sum of so	luare	
Sources of variation	freedom		Vitamin	C content (mg/100	g) at different days	
		Initial	3	6	9	12
Replication	2	0.153	0.114	0.127	0.061	0.001
Variety -V	1	44032.666***	35669.543***	12099.203***	3264.334***	376.834***
Gibberellic acid- G	3	4.286*	756.069***	663.647***	272.194***	92.904***
V × G	3	0.818 ^{NS}	154.827***	225.909***	88.514***	0.934***
Error	14	1.169	1.102	1.199	0.811	0.028

N

X

Appendix 3.4 Analysis of variance of data on titratable acidity and pH of mango pulp as influenced by varieties and different doses of Gibberellic acid solution

295

4

Sources of Degree	Mean sum of square												
variation	of		Titratable aci	dity (%) at o	different day	/S	Pulp pH at different days						
	freedom	Initial	3	6	9	12	Initial	3	6	9	12		
Replication	2	0.000	0.000	0.000	0.000	0.000	0.004	0.001	0.001	0.004	0.001		
Variety -V	1	9.754***	0.173***	0.135***	0.086***	0.020***	0.034 ^{NS}	0.375***	0.375***	0.375***	0.060 NS		
Gibberellic acid- G	3	0.022**	0.368***	0.344***	0.313***	0.131***	0.044 *	0.175***	0.175***	2.320***	0.400***		
V × G	3	0.018**	0.035***	0.005*	0.002 ^{NS}	0.001 ^{NS}	0.004 ^{NS}	0.015 ^{NS}	0.015 ^{NS}	0.015 ^{NS}	0.0003 ^{NS}		
Error	14	0.003	0.003	0.001	0.001	0.00011	0.011	0.011	0.011	0.011	0.016		

Appendix represents

* indicates 5 % level of probability

** indicates 1% level of probability

*** indicates 0.1% level of probability

NS means non significant

2

Y

Appendix 3.5 Analysis of variance of data on total soluble solid and total sugar content of mango pulp as influenced by varieties and different doses of Gibberellic acid solution

A

1

4

A

Sources of Dears	0					Mean sum	n of square					
variation	variation of Replication 2		TSS conte	ent (%) at dif	ferent days		Total sugar content (%) at different days					
		Initial	3	6	9	12	Initial	3	6	9	12	
Replication	2	0.001	0.002	0.001	0.004	0.012	0.001	0.001	0.005	0.001	0.001	
Variety -V	1	6.615***	6.720***	1.500***	0.375***	0.540***	1.229***	1.500***	1.354***	1.500***	1.215***	
Gibberellic acid- G	3	0.105**	32.300***	87.145***	39.090***	36.710***	0.118**	12.945***	113.474***	106.745***	44.415***	
V × G	3	0.005 ^{NS}	0.007 ^{NS}	0.970***	0.045**	0.050**	0.006 ^{NS}	0.010 ^{NS}	0.014 ^{NS}	0.010 ^{NS}	0.015 ^{NS}	
Error	14	0.011	0.013	0.011	0.006	0.006	0.013	0.011	0.011	0.016	0.011	

Appendix 3.6 Analysis of variance of data on reducing and non reducing sugar content of mango pulp as influenced by varieties and different doses of Gibberellic acid solution

Sources of Degree					De la composition	Mean sum	of square					
variation	of	Re	ducing sugar	content (%) a	t different d	ays	Non-reducing sugar content (%) at different days					
	freedom In 2 0.0	Initial	3	6	9	12	Initial	3	6	9	12	
Replication	2	0.001	0.005	0.143	0.001	0.001	0.002	0.001	0.001	0.001	0.001	
Variety -V	1	0.833***	0.901***	0.0005 ^{NS}	0.586***	0.586***	0.064*	0.076*	0.113**	0.271***	0.158**	
Gibberellic acid- G	3	0.018 ^{NS}	2.223***	17.027***	11.751***	6.166***	0.053*	4.466***	43.398***	48.086***	21.888***	
V × G	3	0.010 ^{NS}	0.008 ^{NS}	0.661*	0.023 ^{NS}	0.023 ^{NS}	0.009 ^{NS}	0.028 ^{NS}	0.006 ^{NS}	0.043*	0.011 ^{NS}	
Error	14	0.011	0.011	0.181	0.011	0.011	0.012	0.011	0.011	0.011	0.011	

Appendix represents

* indicates 5 % level of probability *** indicates 0.1% level of probability

1

Y

** indicates 1% level of probability

NS means non significant

296

*

-

Appendix 3.7 Analysis of variance of data on crude fibre and lipid content of mango pulp as influenced by varieties and different doses of Gibberellic acid

14

Sources of variation D	Degree					Mean su	m of square	2			
	of		Crude fibr	e (%) at di	fferent days	6		Lipid cont	ent (%) at	different day	/S
	freedom	Initial	3	6	9	12	Initial	3	6	9	12
Replication	2	0.001	0.001	0.001	0.001	0.003	0.000	0.000	0.000	0.000	0.000
Variety -V	1	0.079*	0.211***	0.213***	0.211***	0.0130***	0.002***	0.002***	0.006***	0.015***	0.014***
Gibberellic acid- G	3	0.001 ^{NS}	0.089**	0.323***	0.269***	0.050***	0.002***	0.027***	0.132***	0.100***	0.026***
V × G	3	0.0006 ^{NS}	0.0007 ^{NS}	0.0003 ^{NS}	0.0005 NS	0.009 NS	0.0005 ^{NS}	-0.0004 ^{NS}	0.001**	0.002***	0.002***
Error	14	0.011	0.011	0.011	0.009	0.005	0.00011	0.00012	0.00013	0.00014	0.00015

Appendix 3.8 Analysis of variance of data on water soluble protein and phosphorus content of mango pulp as influenced by varieties and different doses of Gibberellic acid solution

Sources of variation Degree					Mean s	um of squar	e				
Sources of Variation	of	Water se	oluble prote	in content	(%) at diffe	rent days	Phos	sphorus cont	tent (mg/10	0 g) differen	t days
	freedom	Initial	3	6	9	12	Initial	3	6	9	12
Replication	2	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.016	0.001	0.001
Variety -V	1	0.057**	0.069*	0.089*	0.090*	0.029 ^{NS}	42.294***	41.659***	43.902***	42.693***	42.135***
Gibberellic acid- G	3	0.005 ^{NS}	0.080**	0.334***	0.356***	0.032 ^{NS}	1.210***	6.924***	21.448***	16.098***	4.378***
V × G	3	0.002 ^{NS}	0.0003 NS	0.0005 ^{NS}	0.003 ^{NS}	0.0007 NS	0.007 ^{NS}	0.013 NS	0.081*	0.055*	0.075**
Error	14	0.006	0.009	0.011	0.011	0.011	0.011	0.011	0.018	0.011	0.011

Appendix represents

* indicates 5 % level of probability

** indicates 1% level of probability

*** indicates 0.1% level of probability

NS means non significant

1

*

V

Appendix 3.9 Analysis of variance of data on potassium and calcium content of mango pulp as influenced by varieties and different doses of Gibberellic acid solution

H

A

2

A

Sources of variation De		Mean sum of square										
Sources of variation	Degree of	Pot	tassium con	tent (%) at	t different d	lays	Calcium content (mg/100 g) at different days					
	freedom	Initial	3	6	9	12	Initial	3	6	9	12	
Replication	2	0.000	0.000	0.000	0.000	0.000	0.001	0.001	0.001	0.001	0.001	
Variety -V	1	0.002***	0.002***	0.001**	0.003***	0.003***	21.489***	21.263***	21.113***	24.786***	25.896***	
Gibberellic acid- G	3	0.001***	0.002***	0.002***	0.002***	0.001**	0.239***	18.437***	109.634***	95.970***	25.369***	
V × G	3	0.0005 ^{NS}	0.0006 ^{NS}	0.0007 ^{NS}	0.0007 ^{NS}	0.0005 ^{NS}	0.007 ^{NS}	0.067**	0.055*	0.220***	0.196***	
Error	14	0.00011	0.00012	0.00013	0.00014	0.00015	0.011	0.011	0.011	0.011	0.011	

Appendix 3.10 Analysis of variance of data in magnesium and copper content of mango pulp as influenced by varieties and different doses of Gibberellic acid solution

298

*

Sources of variation Degree		Mean sum of square										
Sources of Variation	of	Magne	sium conter	nt (mg/100	g) at differe	nt days	Co	pper conten	t (mg/100 g) at differer	it days	
	freedom	Initial	3	6	9	12	Initial	3	6	9	12	
Replication	2	0.001	0.001	0.001	0.001	0.004	0.000	0.000	0.000	0.000	0.000	
Variety -V	1	4.779***	2.693***	2.516***	2.381***	2.673***	0.007***	0.005***	0.005***	0.003***	0.005***	
Gibberellic acid- G	3	0.153***	1.507***	3.497***	1.132***	5.879***	0.001*	0.002***	0.007***	0.010***	0.016***	
V × G	3	0.030 ^{NS}	0.471***	0.450***	0.423***	0.204***	0.0002 ^{NS}	0.0003 ^{NS}	0.0005 ^{NS}	0.0005 ^{NS}	0.0007 NS	
Error	14	0.011	0.011	0.011	0.011	0.011	0.00021	0.00022	0.00023	0.00024	0.00024	

Appendix represents

* indicates 5 % level of probability

** indicates 1% level of probability

*** indicates 0.1% level of probability

ability NS means non significant

5

V

Appendix 3.11 Analysis of variance of data on iron and manganese content of mango pulp as influenced by varieties and different doses of Gibberellic acid solution

Sources of variation De	Degree					Mean su	m of squar	e						
	of	Iron content (mg/100 g) at different days						Manganese content (mg/100 g) at different days						
freedor	freedom	Initial	3	6	9	12	Initial	3	6	9	12			
Replication	2	0.001	0.001	0.002	0.001	0.001	0.000	0.000	0.000	0.000	0.000			
Variety -V	1	4.568***	4.806***	4.691***	4.726***	4.158***	0.154***	0.140***	0.168***	0.149***	0.189***			
Gibberellic acid- G	3	0.317***	1.593***	7.364***	2.477***	6.292***	0.006***	0 101***	0 314***	0 163***	0.154***			
V × G	3	0.002 ^{NS}	0.001 ^{NS}	0.004 ^{NS}	0.002 ^{NS}	0.012 ^{NS}	0.0003 ^{NS}	0.0004 ^{NS}	0.001 ^{NS}	0.002*	0.001 ^{NS}			
Error	14	0.011	0.011	0.013	0.011	0.011	0.00011	0.00012	0.00013	0.00014	0.00015			

299

*

X

Appendix 3.12 Analysis of variance of data on zinc content of mango pulp and shelf life of mango as influenced by varieties and different doses of Gibberellic acid solution

Sources of variation Degree				Mean	sum of square			
	of		Zinc co	ntent (mg/100 g) a	t different days		Shelf life	
Paplication		Initial	3	6	9	12	Total days	
Replication	2	0.000	0.000	0.000	0.000	0.000	0.042	
Variety -V	1	0.413***	0.131***	0.094***	0.118***	0.122***	9 375**	
Gibberellic acid- G	3	0.003**	0.007***	0.023***	0.060***	0.122	84 819***	
V × G	3	0.001 ^{NS}	0.0005 ^{NS}	0.003**	0.003**	0.004**	0.375 ^{NS}	
Error 14 0.		0.00021	0.00022	0.00023	0.00024	0.00025	0.565	

Appendix represents

* indicates 5 % level of probability *** indicates 0.1% level of probability

×

** indicates 1% level of probability

NS means non significant

4

L

Appendix 4.1 Analysis of variance of data on physiological weight loss of mango and moisture content of mango pulp as influenced by varieties and different doses of Bavistin DF solution

1

1

Sources of Degree		Mean sum of square											
variation	variation of	Physio	logical weight	loss (%) at di	ifferent days		Moisture content (%) at different days						
freedom	3	6	9	12	Initial	3	6	9	12				
Replication	2	0.001	0.001	0.000	0.036	0.004	0.004	0.004	0.005	0.004			
Variety -V	1	4.208***	6.773***	5.273***	6.100***	60.484***	60,484***	58.594***	58.594***	56.734***			
Bavistin DF - B	3	6.558***	9.566***	7.398***	6.367***	0.134 ^{NS}	3.874***	9.094***	6.674***	5.534***			
V × B	3	0.026**	0.076***	0.016**	0.016 ^{NS}	0.004 ^{NS}	0.004 ^{NS}	0.004 ^{NS}	0.004 ^{NS}	0.004 ^{NS}			
Error	14	0.003	0.003	0.003	0.053	0.041	0.041	0.041	0.045	0.041			

Sources of Degree					Mean sum o	of square							
variation	of		Dry matter	content (%) a	at different da	t different days			Ash content (%) at different days				
	freedom	Initial	3	6	9	12	Initial	3	6	9	12		
Replication Variety -V	2 1	0.005 60.484** *	0.004 60.484** *	0.004 58.594***	0.004 58.594***	0.004 56.734** *	0.001 0.147**	0.001 0.154**	0.001 0.144**	0.001 0.154**	0.002 0.167**		
Bavistin DF - B V × B Error	3 3 14	0.134 ^{NS} 0.004 ^{NS} 0.045	3.874*** 0.004 ^{NS} 0.041	9.094*** 0.004 ^{ns} 0.041	6.674*** 0.004 ^{NS} 0.041	5.534*** 0.004 ^{NS} 0.041	0.0005 ^{NS} 0.0007 ^{NS} 0.011	0.009 ^{NS} 0.0005 ^{NS} 0.011	0.022 ^{NS} 0.0003 ^{NS} 0.011	0.021 ^{NS} 0.0007 ^{NS} 0.011	0.022 ^{NS} 0.002 ^{NS} 0.013		

Appendix represents

*** indicates 0.1% level of probability

indicates 5 % level of probability
indicates 1% level of probability

NS means non significant

1

Y

×

Mean sum of square Sources of Degree of Vitamin C content (mg/100 g) at different days variation freedom Initial 3 9 12 6 Replication 2 0.010 0.005 0.000 0.001 0.001 Variety -V 1 49204.870*** 35956.171*** 8721.094*** 3030.754*** 317.554*** Bavistin DF - B 3 9.178*** 623.696*** 81.284*** 384.064*** 243.294*** V × B 3 2.715* 154.586*** 99.504*** 88.914*** 0.254*** Error 14 0.547 0.448 0.013 0.016 0.011

Appendix 4.3 Analysis of variance of data on vitamin C content of mango pulp as influenced by varieties and different doses of Bavistin DF solution

1

1

Appendix 4.4 Analysis of variance of data on titratable acidity and pH of mango pulp as influenced by varieties and different doses of Bavistin DF solution

Sources of Degree of		Mean sum of square											
variation	Variation freedom	Т	itratable ac	idity (%) at	different da	ys		Pulp	pH at differ	ent days		-	
		Initial	3	6	9	12	Initial	3	6	9	12		
Replication	2	0.000	0.000	0.000	0.000	0.000	0.001	0.001	0.001	0.001	0.002		
Variety -V	1	10.218***	0.189***	0.105***	0.083***	0.032***	0.060*	0.060*	0.034 ^{NS}	0.094*	0.070*		
Bavistin DF - B	3	0.016**	0.323***	0.309***	0.303***	0.104***	0.100**	0.100**	3.234***	1.394***	0.317***		
V × B	3	0.007 ^{NS}	0.039***	0.005***	0.002*	0.003**	0.0003 ^{NS}	0.0005 ^{NS}	0.004 ^{NS}	0.004 ^{NS}	0.0003 ^{NS}		
Error	14	0.003	0.0003	0.0004	0.0005	0.0006	0.011	0.011	0.011	0.011	0.010		

Appendix represents

>

301

* indicates 5 % level of probability

** indicates 1% level of probability

*** indicates 0.1% level of probability NS means non significant

×

Y

Appendix 4.5 Analysis of variance of data on total soluble solid and total sugar content of mango pulp as influenced by varieties and different doses of Bavistin DF solution

Courses of	Deserve	Mean sum of square											
variation	of		TSS conte	nt (%) at di	ferent days		Total sugar content (%) at different days						
	freedom	Initial	3	6	9	12	Initial	3	6	9	12		
Replication	2	0.005	0.001	0.009	0.001	0.004	0.001	0.001	0.001	0.001	0.001		
Variety -V	1	6.000***	5.704***	0.304**	4.860***	6.934***	1.638***	1.882***	1.865***	1.865***	1.904***		
Bavistin DF - B	3	0.295***	11.944***	91.094***	52.850***	33.854***	0.020 ^{NS}	10.748***	98.343***	93.705***	37.530***		
V × B	3	0.0003 ^{NS}	0.004 ^{NS}	2.204***	0.330***	0.164***	0.005 ^{NS}	0.005 ^{NS}	0.006 ^{NS}	0.006 ^{NS}	0.005 ^{NS}		
Error	14	0.028	0.011	0.032	0.020	0.015	0.011	0.011	0.011	0.011	0.011		

Appendix 4.6 Analysis of variance of data on reducing and non reducing sugar content of mango pulp as influenced by varieties and different doses of Bavistin DF solution

302

Sources of Der variation fre	Deeree of	1.00	Mean sum of square											
	freedom	Reducing sugar content (%) at different days					Non-reducing sugar content (%) at different days							
		Initial	3	6	9	12	Initial	3	6	9	12			
Replication	2	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.001			
Variety -V	1	0.446**	0.505***	0.505***	0.421***	0.421***	0.222***	0.437***	0.429***	0.513***	0.624* **			
Bavistin DF - B	3	0.013 ^{NS}	1.704***	13.750***	8.080***	5.169***	0.011 ^{NS}	3.903***	38.977***	47.618***	19.623***			
V × B	3	0.0003 ^{NS}	0.0005 ^{NS}	0.0006 ^{NS}	0.003 ^{NS}	0.003 ^{NS}	0.001 ^{NS}	0.006 ^{NS}	0.007"5	0.006 ^{NS}	0.012 ^{NS}			
Error	14	0.011	0.011	0.011	0.011	0.011	0.011	0.011	0.011	0.011	0.011			

Appendix represents

* indicates 5 % level of probability

** indicates 1% level of probability

*** indicates 0.1% level of probability NS means non significant Appendix 4.7 Analysis of variance of data on crude fibre and lipid content of mango pulp as influenced by varieties and different doses of Bavistin DF solution

Courses of	Degree of freedom	Mean sum of square											
variation		Crude fibre (%) at different days					Lipid content (%) at different days						
		Initial	3	6	9	12	Initial	3	6	9	12		
Replication	2	0.000	0.001	0.000	0.001	0.000	0.000	0.000	0.000	0.000	0.001		
Variety -V	1	0.069***	0.066***	0.041**	0.094***	0.090***	0.0009 ^{NS}	0.001 ^{NS}	0.001 ^{NS}	0.002***	0.0007 ^{NS}		
Bavistin DF - B	3	0.002 ^{NS}	0.103***	0.275***	0.230***	0.063***	0.002***	0.029***	0.150***	0 131***	0.023***		
V × B	3	0.0005 ^{NS}	0.001 ^{NS}	0.005 ^{NS}	0.001 ^{NS}	0.0008 ^{NS}	0.0007 ^{NS}	-0.0003 ^{NS}	0.0003 ^{NS}	0.001**	0.003 ^{NS}		
Error	14	0.003	0.003	0.003	0.004	0.003	0.00031	0.00032	0.00033	0.00034	0.002		

Appendix 4.8 Analysis of variance of data on water soluble protein and phosphorus content of mango pulp as influenced by varieties and different doses of Bavistin DF solution

303

Courses of	Degree	Mean sum of square											
Sources of variation Replication Variety -V Bavistin DF - B V × B		Water so	luble protei	n content (%) at differ	ent days	Phosphorus content (mg/100 g) different days						
	freedom	Initial	3	6	9	12	Initial	3	6	9	12		
Replication	2	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.002	0.001		
Variety -V	1	0.041*	0.051 ^{NS}	0.076*	0.043 ^{NS}	0.022 ^{NS}	32.017***	31.740***	34.344***	32.063***	23.522***		
Bavistin DF - B	3	0.003 ^{NS}	0.064**	0.312***	0.336***	0.009 ^{NS}	1.203***	6.672***	18.996***	12.906***	6.175***		
V × B	3	0.001 ^{NS}	0.0003 ^{NS}	0.001 ^{NS}	0.002 ^{NS}	0.001 ^{NS}	0.012 ^{NS}	0.008 ^{NS}	0.003 ^{NS}	0.010 ^{NS}	0.782***		
Error	14	0.006	0.011	0.011	0.011	0.011	0.011	0.011	0.011	0.013	0.011		

Appendix represents

* indicates 5 % level of probability

** indicates 1% level of probability

*** indicates 0.1% level of probability

NS means non significant

Appendix 4.9 Analysis of variance of data on potassium and calcium content of mango pulp as influenced by varieties and different doses of Bavistin DF solution

Sources of	Degree	Mean sum of square											
variation	of	Po	tassium con	tent (%) at	different d	ays	Calc	cium content	t (mg/100 g) at different	days		
	freedom	Initial	3	6	9	12	Initial	3	6	9	12		
Replication	2	0.000	0.000	0.000	0.000	0.000	0.001	0.001	0.030	0.001	0.001		
Variety -V	1	0.010***	0.004***	0.003***	0.000	0.000	16 494***	17 750***	10 620***	17 802***	17.957***		
Bavistin DF - B	3	0.001**	0.001***	0.001***	0.005	0.004	0.074**	15 707***	19.020	89 048***	23.389***		
V × B	3	0.0003 ^{NS}	0.0003 ^{NS}	0.0007NS	0.001	0.001	0.074**	15.707***	90.079	0 084**	0.084**		
Error	14	0.00041	0.00043	0.00052	0.00053	0.00063	0.011	0.096**	0.048	0.011	0.011		

Appendix 4.10 Analysis of variance of data on magnesium and copper content of mango pulp as influenced by varieties and different doses of Bavistin DF solution

Sources of Degr variation of freed	Degree	Mean sum of square											
	of	Magnesium content (mg/100 g) at different days						Copper content (mg/100 g) at different days					
	freedom	Initial	3	6	9	12	Initial	3	6	9	12		
Replication Variety -V Bavistin DF - B	2 1 3	0.001 5.910*** 0.361***	0.002 6.531*** 1.779***	0.001 6.304*** 4.382***	0.001 5.645*** 0.710***	0.001 5.578*** 2.005***	0.000 0.007*** 0.0003*	0.000 0.007*** 0.003***	0.000 0.00 7 *** 0.007***	0.000 0.006*** 0.013***	0.000 0.005*** 0.018***		
Error	3 14	0.019	0.017 ^{NS} 0.013	0.022 ^{№5} 0.011	0.027 ^{NS} 0.011	0.017 ^{NS} 0.011	0.0004 ^{№s} 0.00062	0.0001 ^{NS} 0.00063	0.0002 ^{NS} 0.00074	0.0003 ^{NS} 0.00092	0.0005		

Appendix represents

* indicates 5 % level of probability

** indicates 1% level of probability

*** indicates 0.1% level of probability NS means non significant

.

Appendix 4.11 Analysis of variance of data on iron and manganese content of mango pulp as influenced by varieties and different doses of Bavistin DF solution

Sources of variation		Mean sum of square												
	Degree	Iroi	t different o	lays	Manganese content (mg/100 g) at different days									
	freedom	Initial	3	6	9	12	Initial	3	6	9	12			
Replication	2	0.001	0.001	0.001	0.001	0.055	0.000	0.000	0.000	0.000	0.000			
Variety -V	1	4.753***	5.273***	5.273***	8.894***	12.113***	0.154***	0.158***	0.154***	0.154***	0.135***			
Bavistin DF - B	3	0.180***	0.942***	6.274***	1.567***	3.545***	0.002*	0.075***	0.251***	0.153***	0.192***			
V x B	3	0.004 ^{NS}	0.020 ^{NS}	0.020 ^{NS}	0.292***	0.397***	0.0004 ^{NS}	0.0005 ^{NS}	0.0007 ^{NS}	0.0008 ^{NS}	0.0007 ^{NS}			
Error	14	0.011	0.011	0.011	0.011	0.037	0.00032	0.00043	0.00054	0.00063	0.00033			

Appendix 4.12 Analysis of variance of the data on zinc content of mango pulp and shelf life of mango as influenced by varieties and different doses of Bavistin DF solution

305

Adama ntatio

Jaiwarsing Jaiwarsing

6018401

		Mean sum of square									
Sources of variation	Degree of freedom		Shelf life								
		Initial	3	6	9	12	Total days				
Replication	2	0.000	0.000	0.000	0.000	0.000	0.042				
Variety -V	1	0.265***	0.076***	0.083***	0.083***	0.086***	10.667**				
Bavistin DF - B	3	0.002*	0.011***	0.029***	0.075***	0.153***	82.278***				
V × B	3	0.002*	0.0003 ^{NS}	0.0005 ^{NS}	0.0007 ^{NS}	0.001 ^{NS}	0.333 ^{NS}				
Error	14	0.00072	0.00074	0.00035	0.00065	0.00082	0.661				

Appendix represents

* indicates 5 % level of probability

** indicates 1% level of probability NS mea

*** indicates 0.1% level of probability NS means non significant